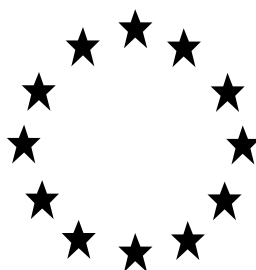


Directive 98/8/EC concerning the placing biocidal products on the market

Inclusion of active substances in Annex I or IA to Directive 98/8/EC

Assessment Report



Boric oxide Product-type 8 (Wood preservative)

20 February 2009

Annex I – the Netherlands

Boric oxide (PT 8)**Assessment report**

Finalised in the Standing Committee on Biocidal Products at its meeting on 20 February 2009 in view of its inclusion in Annex I to Directive 98/8/EC

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1. STATEMENT OF SUBJECT MATTER AND PURPOSE

1.1. Procedure followed

This assessment report has been established as a result of the evaluation of Boric oxide as product-type 8 (wood preservative), carried out in the context of the work programme for the review of existing active substances provided for in Article 16(2) of Directive 98/8/EC concerning the placing of biocidal products on the market¹, with a view to the possible inclusion of this substance into Annex I or IA to the Directive.

Boric oxide (CAS no. 1303-86-2) was notified as an existing active substance, by Rio Tinto Minerals, hereafter referred to as the applicant, in product-type 8.

Commission Regulation (EC) No 1451/2007 of 4 December 2007² lays down the detailed rules for the evaluation of dossiers and for the decision-making process in order to include or not an existing active substance into Annex I or IA to the Directive.

In accordance with the provisions of Article 7(1) of that Regulation, the Netherlands was designated as Rapporteur Member State to carry out the assessment on the basis of the dossier submitted by the applicant. The deadline for submission of a complete dossier for Boric oxide as an active substance in Product Type 8 was 28 March 2004, in accordance with Annex V of Regulation (EC) No 2032/2003.

On 27 March 2004, the Netherlands competent authorities received a dossier from the applicant. The Rapporteur Member State accepted the dossier as complete for the purpose of the evaluation on 24 September 2004.

On 7 July 2006, the Rapporteur Member State submitted, in accordance with the provisions of Article 14(4) and (6) of Regulation (EC) No 1451/2007, to the Commission and the applicant a copy of the evaluation report, hereafter referred to as the competent authority report. The Commission made the report available to all Member States by electronic means on 25 July 2006. The competent authority report included a recommendation for the inclusion of Boric oxide in Annex I to the Directive for product-type 8.

In accordance with Article 16 of Regulation (EC) No 1451/2007, the Commission made the competent authority report publicly available by electronic means on 23 October 2006. This report did not include such information that was to be treated as confidential in accordance with Article 19 of Directive 98/8/EC.

In order to review the competent authority report and the comments received on it,

1 Directive 98/8/EC of the European Parliament and of the Council of 16 February 1998 concerning the placing of biocidal products on the market. OJ L 123, 24.4.98, p.1

2 Commission Regulation (EC) No 1451/2007 of 4 December 2007 on the second phase of the 10-year work programme referred to in Article 16(2) of Directive 98/8/EC of the European Parliament and of the Council concerning the placing of biocidal products on the market. OJ L 325, 11.12.2007, p. 3

consultations of technical experts from all Member States (peer review) were organised by the Commission. Revisions agreed upon were presented at technical and competent authority meetings and the competent authority report was amended accordingly.

On the basis of the final competent authority report, the Commission proposed the inclusion of Boric oxide in Annex I to Directive 98/8/EC and consulted the Standing Committee on Biocidal Product on 20 February 2009.

In accordance with Article 15(4) of Regulation (EC) No 1451/2007, the present assessment report contains the conclusions of the Standing Committee on Biocidal Products, as finalised during its meeting held on 20 February 2009.

1.2. Purpose of the assessment report

This assessment report has been developed and finalised in support of the decision to include Boric oxide in Annex I to Directive 98/8/EC for product-type 8. The aim of the assessment report is to facilitate the authorisation in Member States of individual biocidal products in product-type 8 that contain Boric oxide. In their evaluation, Member States shall apply the provisions of Directive 98/8/EC, in particular the provisions of Article 5 as well as the common principles laid down in Annex VI.

For the implementation of the common principles of Annex VI, the content and conclusions of this assessment report, which is available at the Commission website³, shall be taken into account.

However, where conclusions of this assessment report are based on data protected under the provisions of Directive 98/8/EC, such conclusions may not be used to the benefit of another applicant, unless access to these data has been granted.

1.3. Overall conclusion in the context of Directive 98/8/EC

The overall conclusion from the evaluation is that it may be expected that there are products containing Boric oxide for the product-type 8, which will fulfil the requirements laid down in Article 5 of Directive 98/8/EC. This conclusion is however subject to:

- i. compliance with the particular requirements in the following sections of this assessment report,
- ii. the implementation of the provisions of Article 5(1) of Directive 98/8/EC, and
- iii. the common principles laid down in Annex VI to Directive 98/8/EC.

Furthermore, these conclusions were reached within the framework of the uses that were proposed and supported by the applicant (see [Appendix II](#)). Extension of the use pattern beyond those described will require an evaluation at product authorisation level in order to establish whether the proposed extensions of use will satisfy the requirements of Article 5(1) and of the common principles laid down in Annex VI to Directive 98/8/EC.

³ <http://ec.europa.eu/comm/environment/biocides/index.htm>

2. OVERALL SUMMARY AND CONCLUSIONS

2.1. Presentation of the Active Substance

2.1.1. Identity, Physico-Chemical Properties & Methods of Analysis

Identity of the active substance

The CAS no. of boric oxide used in this CA-report is 1303-86-2 (pure boric oxide)

The EC no. of boric oxide used in this CA-report is 215-125-8.

The minimum purity specification is 97.5% (w/w) expressed as B₂O₃.

There are no additives, no impurities > 1 g/kg and no (eco)toxicologically relevant impurities.

Physical and chemical properties of the active substance

Boric oxide is a white, odourless solid. No melting point is found in the range 25-360°C. A boiling point is not applicable.

The water solubility, dissociation constant, partition coefficient, and surface tension for boric oxide as such cannot be determined because boric oxide is converted into boric acid upon dissolution in water: B₂O₃ + 3H₂O = 2B(OH)₃. The water solubility, dissociation constant, partition coefficient, and surface tension found will be the ones for an equivalent amount of boric acid.

Vapour pressure and Henry's law constant are not applicable, because the melting point lies above 300°C and at ambient temperature vapour pressure is expected to be less than 10⁻⁵ Pa. The water solubility for boric oxide as such cannot be determined because boric oxide is converted into boric acid upon dissolution in water. The water solubility for boric oxide was calculated from water solubility experiments with boric acid. Calculated water solubility of boric oxide: 27.2 g/L in water at 20 °C (pH ~3.9).

The octanol/water coefficient of boric oxide as such cannot be determined because boric oxide is converted into boric acid upon dissolution in water. The octanol/water coefficient of boric acid is log P_{ow} = -1.09 (phosphate buffer, pH=7.5 at 22°C).

The active substance is a non-flammable solid and a flash point is not applicable. The active substance has no explosive and no oxidising properties.

Analysis of the active substance as manufactured

There are no methods to determine the active substance as manufactured or in the formulation. Instead, analytical methods are developed to determine the B₂O₃ and Na₂O content in boron based compounds.

Soil

Total boron can be extracted from soil by hot acid extraction with HNO₃ and HCl under reflux or by microwave digestion with HNO₃ and HF, with addition of HCl and H₂O₂ when needed. Several extraction methods are used to estimate the bioavailable boron fraction, of which the hot water soluble extraction is the most widely used.

Water

Boron can be detected in aqueous solutions by spectrophotometry after reaction with azomethine-H, or by ICP-AES.

Air

No data submitted. No data necessary.

Summary information on the identity and physico-chemical properties and analytical methods of boric oxide can be found in Appendix I to this document (List of Endpoints).

2.1.2. *Intended Uses and Efficacy*

Product type and field of use envisaged

Main group 2: preservatives

Product type 8: wood preservative

Boric oxide is intended to be used as a preventive and curative wood preservative for wood and construction timbers in Use Classe 1, 2, 3 and 4a according to CEN 335-1 standard.

User

Industrial, Professional, Non-professional.

Function

Fungicide and/or insecticide

Application

Boric oxide in the form of solid rods are used either curatively (as a fungicide and insecticide) to treat timber including joinery in UC1, 2, 3 and 4a applications or are inserted at the time of manufacture of the wooden components to act as a fungicide or an insecticide on a preventive basis. The assessment of the biocidal activity of the active substance demonstrates that it has a sufficient level of efficacy against the target organism(s) and the evaluation of the summary data provided in support of the efficacy of the accompanying product, establishes that the product may be expected to be efficacious.

In addition, in order to facilitate the work of Member States in granting or reviewing authorisations, and to apply adequately the provisions of Article 5(1) of Directive 98/8/EC and the common principles laid down in Annex VI of that Directive, the intended uses of the substance, as identified during the evaluation process, are listed in [Appendix II](#).

2.1.3. *Classification and Labelling*

It is proposed to classify borates with reproduction category 2 and assign risk phrases R60-61. This is in line with the proposal of the EU commission working group of specialised experts in the field of reproduction toxicology (2004).

Proposed classification for the active substance boric oxide (identical to 67/548/EEC)

Class of danger	T
R phrases	R60-61

2.2. Summary of the Risk Assessment

2.2.1. Human Health Risk Assessment

The toxicokinetics and toxicological effects of boric acid, disodium tetraborate, boric oxide (B_2O_3) and disodium octaborate tetrahydrate are likely to be similar on a boron equivalents basis. Therefore, the data obtained from studies with different borates can be read across in the human health assessment for each individual substance.

2.2.1.1. Hazard assessment

Toxicokinetics

Absorption of borates via the oral route is nearly 100%. For the respiratory route also 100% absorption is assumed. Dermal absorption through intact skin is very low. For risk assessment of borates a dermal absorption of 0.5% is used as a reasonable worst case approach. In the blood boric acid is the main species present. Boric acid is not further metabolised. Boric acid is distributed rapidly and evenly through the body, with concentrations in bone 2-3 higher than in other tissues. Boron is excreted relatively rapidly with elimination half-lives of 1h in the mouse, 3h in the rat and 21h in humans, and has low potential for accumulation. Boric acid is mainly excreted in the urine.

Acute toxicity

Boric acid and other borates are of low acute toxicity: LD₅₀ oral rat > 2000 mg/kg; LD₅₀ dermal rat > 2000 mg/kg; LC₅₀ inhalation rat > 2 mg/l (highest attainable concentration). Acute oral toxicity for boric oxide: LD₅₀ oral rat > 2600 mg/kg. No dermal or inhalation toxicity studies with boric oxide were available.

Irritation and Corrosivity

Boric oxide is not a skin irritant or an eye irritant.

Sensitization

No sensitization studies with boric oxide were available. No signs of skin sensitization were observed in Buehler method skin sensitization tests with boric acid or other borates and no evidence of skin sensitization in humans exposed occupationally to borates has been reported. Boric oxide does not need to be classified as a skin sensitizer.

Repeated dose toxicity

In the repeated dose studies with mouse and rat (and dog although studies are not acceptable), consistently effects on the testes and on blood parameters were found. In the 90 days study in the mouse and the 2 year study in the rat the animals appeared to be more sensitive to the effects on the haematopoietic system than on the testes. Similar results were obtained from studies with disodium tetraborates. The 2 years feeding study with boric acid in rats yielded an overall NOAEL (100 mg/kg bw, equal to 17.5 mg B/kg bw/day), based on reduction bodyweight, clinical sign of toxicity, in males testicular atrophy and reductions in red cell volume and Hb at a dose of 334 mg/kg bw/day (58.5 mgB/kg bw/day). This finding is supported by the study with disodium tetraborate decahydrate.

Genotoxicity

Boric oxide has no genotoxic potential.

Carcinogenicity

In carcinogenicity studies in mice and rats no evidence of a carcinogenic effects of boric acid or sodium tetraborate decahydrate was observed. It is concluded that boric acid is unlikely to pose a carcinogenic risk.

Reproductive toxicity

Teratogenicity

The most sensitive species for developmental effects appears to be the rat. The NOAEL for embryotoxic/teratogenic effects of boric acid was 55 mg/kg bw/day (9.6 mg B/kg bw/day), based on a reduction in mean foetal body weight/litter and an increased incidence in short rib X111, which is considered a variation, at 76 mg/kg bw/day (13.3 mg B/kg bw/day). The relevance of the observed developmental effects for humans is not clear.

Fertility

In a multigeneration reproduction toxicity study in the rat with boric acid severely impaired reproductive potency was observed at 336 mg/kg bw/day. At this dose also marked reductions (70%) in relative testes weights were observed. At lower doses no reproductive effects or effects on testes weight were observed. These findings suggest that a reduction in testes weight will result in an impaired fertility. Since this study was seriously flawed, no definitive conclusions on the effects of boron on fertility in the rat can be drawn. Other repeated dose studies in several animal species have consistently demonstrated that the testis is a primary target organ for boron. Based on the data from the 2 years feeding study with boric acid in rats, the overall NOAEL for fertility is therefore 100 mg/kg bw/day, equal to 17.5 mg B/kg bw/day. This conclusion is supported by the study with disodium tetraborate decahydrate.

Neurotoxicity

No neurotoxic studies were available for the present evaluation. CNS depression was observed in poisoning cases in humans. Apart from the CNS effects that occur at these very high doses there are no indications that boric acid or other borates have neurotoxic properties. No neurotoxicity studies are required.

Human data

It has been suggested that the potential lethal oral dose of boric acid is 3-6g in children and 15-20 g for adults. However, lethal doses are not well documented. Acute effects may include nausea, vomiting, gastric discomfort, skin flushing, excitation, convulsions, depression and vascular collapse. Multiple exposure (high levels > 1g) results in various symptoms which may appear singly or together and include dermatitis, alopecia, loss of appetite, nausea, vomiting, diarrhoea, and focal or generalised central nervous system irritation or convulsions. Infants aged from 6 to 16 weeks who ingested sodium tetraborate at doses of about 0.98 g boric acid/day for periods of 4 to 12 weeks showed convulsions, generalised seizures and focal seizures.

Workers exposed occupationally to borax dust at concentrations of about 4 mg/m³ reported eye irritation, dry mouth, nose or throat, sore throat, acute and chronic respiratory irritation and productive cough.

Effects on human reproduction

In human epidemiological cohort study in Turkey no effects on reproduction were observed in populations exposed to high levels of boron through drinking water. However, this conclusion

was based on the number of children born over a period of 15 years. Other endpoints such as time to pregnancy and number of miscarriages were not included. In a study in the USA, the fertility of male workers of a borax mine was studied. The study revealed that the workers exposed to low ($<0.82 \text{ mg/m}^3$) or high ($>5.05 \text{ mg/m}^3$) levels of boron in dust fathered more live births than was estimated on the basis of the data of the US general population. The extent to which the workers are comparable to the US general population however was not clear. Also in this study other endpoints such as time to pregnancy were not established.

2.2.1.2. Critical endpoints and assessment factors

Overall NOAEL

The toxicological data base reveals that the major targets for toxicity of borates are the testes and the blood. In the repeated dose studies with mouse, rat and dog, consistently effects on the testes and on blood parameters were found. In a 90 days study in the mouse the animals appeared to be more sensitive to the effects on the haematopoietic system than on the testes. In the rat effects on both the testes and on the blood were observed at dose levels of 334 mg boric acid/kg bw/day. The NOAEL in this study was 100 mg/kg bw/day (17.5 mg B/kg bw/day). Similar results were obtained from studies with disodium tetraborate decahydrate at equimolar doses of boron. Based on the NOAEL for embryotoxic/teratogenic effects of boric acid of 55 mg/kg bw/day (9.6 mg B/kg bw/day) the overall NOAEL is 9.6 mg B/kg bw/day.

Total assessment factor

The available information on kinetics and dynamics does not allow refinement of the standard assessment factors (10 for interspecies variation and 10 for intraspecies variation).

There are indications that the sensitivity for the effects on the testes does not differ markedly between subchronic and chronic exposure. Therefore the application of an additional assessment factor for extrapolation from subchronic to chronic exposure is not required.

ADI (acceptable daily intake) and ARfD (acute reference dose).

The borates will be used as wood preservatives. The borates are not used on wood products that are used for food preparation or feeding stuff. Furthermore finished wood products containing borates and manufactured for structural and building material are not appropriate to be used and would not be used to make products that would come in to contact with food or feeding stuff. Therefore the derivation of an ADI and ARfD is considered not necessary.

Drinking water limit

Based on the overall NOAEL (9.6 mg boron/kg bw/day), which was found in the developmental study in the rat and the overall assessment factor of 100, a tolerable daily intake (TDI) of $9.6/100 = 0.096 \text{ mg/kg bw/day}$ can be established. If 10% of the TDI is allocated to boron in drinking water, and assuming that an adult weighs 60 kg and consumes 2 litre drinking water per day, a drinking water guideline value 0.29 mg/L can be derived.

The WHO established in 1998 a drinking water guideline value of 0.5 mg/L, based on an overall NOAEL of 9.6 mg boron/kg bw per day from a developmental toxicity study in the rat, and a total assessment factor of 60. Drinking water on average contains 0.1-0.3 mg boron per litre, while in some regions much higher concentrations have been measured (EHC 204, 1998). Boron in the soil and aquatic environment is predominantly originating from natural weathering of rock and sea water evaporation (EHC 204, 1998). Accordingly, human exposure to boron will probably not have changed significantly over time.

It is clear that a drinking water limit of 0.29 mg/L, as estimated on the basis of the present toxicological data set, will generally not be exceeded.

AOEL (acceptable operator exposure level).

Using the standard assessment factor of 100 (10 for interspecies- and 10 for intraspecies variation) an oral AOEL of 0.096 mg B/kg bw/day can be derived based on the NOAEL for embryotoxic/teratogenic effects of boric acid of 9.6 mg B/kg bw/day. Based on the NOAEL of 17,5 mg B/kg bw/day in a 2-year study in the rat and an assessment factor of 300 due to serious effects and based on the NOAEL of 21.8 mg B/kg bw/day in a teratogenicity study in the rabbit and an assessment factor of 300 due to serious effects AOEL values can be derived of 0.06 and 0.07 mg B/kg bw/day, respectively. For the risk assessment, a rounded (systemic) AOEL value of 0.1 mg B/kg bw/day was used for boric oxide, even though values of 0.06 and 0.07 mg B/kg bw/day could be derived from the critical effects.

MOE (margin of exposure)

Professional users.

For the risk assessment of borates the systemic NOAEL of 9.6 mg/kg bw/day will be used. A MOE of 100 would be considered acceptable, on the basis of the standard assessment factors of 100 (10 x 10) for interspecies and intraspecies variability.

Non-professional users

The developmental studies are considered to be the most relevant for the risk characterization of non-professional users of borates. A MOE of 100 with respect to the NOAEL of 9.6 mg B/kg bw/day from the developmental toxicity study in the rat is considered acceptable.

2.2.1.3. Risk characterisation

Boric oxide is used in fused rods. The rods are inserted in predrilled holes. Boric oxide is not volatile. Considering the process of application, the inhalatory and dermal exposure will be minimal.

Since the inhalatory and dermal exposure to boric oxide is considered minimal, it is considered that the risk to the user is negligible.

2.2.2. Environmental Risk Assessment

At environmentally relevant conditions boric oxide will dissolve to boric acid. Therefore, data on the ecotoxicity of boric acid can be used for boric oxide as well. Furthermore, it is assumed that of all the possible forms of borate in the environment, the boron ion is the potentially toxic compound. Therefore, effects of borates, including boric oxide, are usually expressed on the basis of boron.

Boron is a naturally occurring element that is essential to a variety of organisms. Because of boron being present in natural environments, the background concentrations should be taken into account when deriving PNEC's. Boron concentrations in freshwater systems rarely exceed 1.0 mg B/L and are usually < 0.1 mg B/L. A recent review of European monitoring data report ranges of < 0.001 to 7.5 mg B/L, the mean 95-percentile concentration over 15 European countries ranges from 0.017 mg B/L in the United Kingdom to 0.632 mg B/L in

Germany. Oceans have a natural level of boron of about 5 mg/L. Boron concentrations in soil are highly variable, ranges are 4 to 200 mg/kg, mostly as the *unavailable* earth mineral tourmaline and usual levels are 45-124 mg/kg dwt, based on observations in the United States. Boron levels in European groundwaters ranged for Italy and Spain from 0.5 to 1.5 mg/litre (mean boron concentrations); up to approximately 0.6 mg boron/litre in the Netherlands and United Kingdom, and levels in approximately 90% of samples in Denmark, France, and Germany were found to be below 0.3, 0.3, and 0.1 mg boron/litre, respectively. In the following sections, all data refer to added boron, unless stated otherwise.

2.2.2.1. PNEC derivation

The notifier proposed to base the Risk Assessment on total PNEC (i.e. including background concentrations). He argues that the background boron concentration influences a number of biological/ecological processes (e.g. optimal concentration ranges, acclimation/adaptation, and field community responses), and is crucial for the derivation of ecological relevant PNEC values.

The RMS partly agrees with this point of view but considers that other arguments have to be taken into account:

- In the added risk approach both the "Predicted Environmental Concentration" (PEC) and the "Predicted No Effect Concentration" (PNEC) are expressed as B added by man, resulting in an "added Predicted Environmental Concentration" (PEC_{add}) and "added Predicted No Effect Concentration" (PNEC_{add}), respectively. The use of the added risk approach (a method that in principle can be used for all naturally occurring substances) implies that only the anthropogenic amount of a substance, i.e. the amount added to the natural background concentration, is considered to be relevant for the effects assessment of that substance.
- There is inadequate knowledge of the natural Boron concentrations in water, sediment and soil.
- There is insufficient knowledge to incorporate bioavailability.
- No studies are available that indicate acclimatization and adaptation effects to the sensitivity of species to Boron - with the exception of deficiency occurring in some fish and plant studies.
- There is no information that Boron is essential to all biota.

The RMS therefore prefers to conduct the risk assessment on *added concentrations*.

Aquatic compartment

Based on the lowest NOEC of 1.8 mg B/L and an assessment factor of 10 the PNEC_{add,aquatic} for freshwater systems is set to **0.18 mg B/L**. Using equilibrium partitioning the PNEC_{add,sed} is set to **0.24 mg B/kg wwt**. As both PEC_{sed} and PNEC_{sed} values are based on equilibrium partitioning, no further assessment of the risk was performed.

Sewage treatment plants (STPs)

Based on a submitted microbial inhibition test with micro-organisms in activated sludge, a PNEC_{add,STP} of **1.8 mg B/L** was calculated, based on a NOEC of 17.5 mg B/L and applying an assessment factor of 10.

Air

No toxicity data for air were available. Therefore, a PNEC_{air} was not calculated.

Terrestrial compartment

The $PNEC_{add,terrestrial}$ should be regarded as the concentration that may be *added* to the background without inducing adverse effects on the terrestrial ecosystem. It is implicitly assumed that the background concentration itself does not induce a toxic effect and that the background concentration is sufficient to meet the natural demands of the organisms and that there is no risk for deficiency.

The $PNEC_{add,terrestrial}$ is set to **0.4 mg B/kg dwt soil**, equivalent to **0.35 mg B/kg wwt soil**. The applicability of a low $PNEC_{add,terrestrial}$ should be judged in the light of sensitive plant species instead of being focussed solely on tolerant species.

Non compartment specific effects relevant to the food chain (secondary poisoning)

Boron is not bioconcentrated or bioaccumulated along the food chain in both aquatic and terrestrial ecosystems.

2.2.2.2. Exposure and risk characterisation

The environmental exposure assessments have been taken from the OECD Emission Scenario Document Number 1, 2 and 3, Emission Scenario Documents on Wood Preservatives (OECD, 2003) using boric acid as the representative boron compound. Boron is released into the environment during end use (in-service period). Releases can enter the aquatic compartment (incl. sediment), soil (incl. groundwater) and to a lesser degree air, depending on the scenario. Applications involve the preventive and curative application.

The following in-service uses are recognized:

- Wood in service: Noise barrier; Fence; House cladding; Transmission pole; Fence post; Bridge

No emissions to the environment are expected during service life for Use Class 1 (timber under cover including indoor joinery (preventive)) and Use Class 2 (timber under cover including indoor roofing timbers – risk of wetting (preventive)). Therefore Use Class 1 and 2 is considered not to lead to unacceptable risk to the environment. The risk assessment for the wood in service (Use class 3 and 4a) are considered below.

As to the waste stage the ESD states the following: “Waste treatment: at this stage the unused wood preservative products or the out-of-service treated wood is disposed of with waste. Waste treatment may consist of incineration or landfill dumping. Releases during these processes are considered through leaching models and release of non-degraded substances during incineration, especially heavy metal oxides. Disposal of wastes from treatment plants or disposal of treated wood after service do not fall under the scope of this ESD document. Waste wood, waste wood dust, protection foil, cleaning solvents, used cans and unused product should be disposed of according to national waste disposal regulations”.

Aquatic Environment (fresh water)

PEC/PNECs are > 1 (in the range of 7.9 to-79) for the in-situ treatments on and in service period of treated bridges over ponds, and a risk is expected. The other uses (processing, storage and wood in service, such as noise barriers) are considered as safe with respect to the risks for surface water.

Aquatic Environment (sediment)

As the $PNEC_{add, sed}$ and the $PEC_{add, sed}$ both are based on equilibrium partitioning, risks found in the aquatic risk assessment related to the bridge over pond scenario also will result in risks for the sediment.

Sewage treatment plants (STPs)

Industrial application and use of treated wood as a noise barrier may result in emission of boric acid to a STP. A risk for micro-organisms in the STP is not expected ($PEC/PNEC < 1$).

Assessment of drinking water criterion

According to Directive 98/83/EC Annex I, part B, the limit value for boron in water is 1 mg/L. This value is exceeded in the scenario for in service of a bridge over pond (UC3) (TIME2 PEC values ranging between 1.43 and 3.34 mg B/l) and the $PEC/PNEC$ is also > 1 . European monitoring data (Wyness et al. 2003) report ranges of < 0.001 to 7.5 mg B/L, the mean 95-percentile concentration over 15 European countries ranges from 0.017 mg B/L in the United Kingdom to 0.632 mg B/L in Germany. This means that boric oxide (in service) in treated wood of bridges over ponds does **not** comply with the drinking water criteria.

Air

Because of the low vapour pressure, release of borates to air will be minimal. No $PEC_{add, air}$ and $PNEC_{add, air}$ were calculated.

Terrestrial environment

The total $PEC_{add, terrestrial}$ was calculated to be ranging between 1.8 and 99 mg B/kg wwt soil, and compared to the $PNEC_{add, terrestrial}$ of 0.35 mg B/kg wwt soil. The resulting $PEC_{add, terrestrial}/PNEC_{add, terrestrial}$ ratio ranges between 5.1 and 283, indicating a risk to the terrestrial ecosystem.

Groundwater

The $PEC_{add, grw}$ is estimated from the $PEC_{add, terrestrial}$ using equilibrium partitioning. The calculated $PEC_{add, grw}$ is evaluated according to the criteria for water that will be used for production of drinking water. According to Directive 98/83/EC Annex I, part B, the limit value for boron is 1 mg B/L. The calculated TIME2 $PEC_{add, grw}$ ranges between 4.7 and 40.5 mg B/L, which is above this value. The UC3 and 4A applications of boric oxide do not comply with the drinking water criterion.

Assessment of persistence in soil

Being an inorganic compound, boron does not comply with the persistence criteria of $DT_{90, field} < 1$ year and DT_{50} at $20^{\circ}C < 6$ months that are laid down in paragraph 85 of Annex VI to the Biocides Directive and in the TNsG on Annex I inclusion. According to the latter, the degradation triggers do not necessarily apply if the active substance is included in Annex I with regard to areas of use where a long lasting service-life of the treated material is essential and it is scientifically demonstrated that under field conditions there is no unacceptable accumulation in soil (e.g. that the $PEC/PNEC < 1$ in soil during the service-life of the treated article). The application as a wood preservative can be considered as such. It should therefore be evaluated whether the $PEC/PNEC$ in soil during service life is < 1 . It was shown above in Section 2.9.1 that for all use classes the $PEC/PNEC$ ratio is far above 1. Boric oxide does not meet the criteria for persistence in soil, and is therefore considered **persistent**.

2.2.2.3. PBT assessment

Being an inorganic compound, boron does not biodegrade in marine or freshwater and sediments, and should therefore be considered as Very Persistent (VP).

Boron is not bioconcentrated, based on the available data the BCF is < 2000 L/kg wwt.

The chronic NOEC of boron for marine or freshwater organisms is > 0.01 mg B/L and boron is not considered to have endocrine disrupting effects. However, based on toxicological data, boron is classified with Toxic for Reproduction category 2 and assigned risk phrases R60 and R61.

Therefore, boron should be considered as fulfilling the criteria for Persistence and Toxicity, but not for Bioaccumulation.

Boron does not fulfil the PBT-criteria. Therefore inclusion in Annex I is not restricted by these criteria.

2.2.3. List of endpoints

In order to facilitate the work of Member States in granting or reviewing authorisations, and to apply adequately the provisions of Article 5(1) of Directive 98/8/EC and the common principles laid down in Annex VI of that Directive, the most important endpoints, as identified during the evaluation process, are listed in [Appendix I](#).

3. DECISION

3.1. Background to the Decision

Boric oxide is requested for inclusion in Annex I as a preservative for wood to be used in the form of boron rods in Use Class 1, 2, 3 and 4a preventive and curative treated by professional and non-professional users. Other Use Classes are not included in the present assessment and no inclusion can be granted for these uses.

The unconditional inclusion of boric oxide in Annex I for this Use Class can only be granted in case efficacy can be demonstrated for all proposed uses and the proposed uses do not lead to an unacceptable risk for humans and the environment.

A conditional inclusion of boric oxide in Annex I can be granted in case efficacy can be demonstrated and at least one of the proposed uses do not lead to an unacceptable risk for humans or the environment.

The risk assessment for boric oxide is based on a fictitious model product containing 100% boric oxide. Actual products based on boric oxide contain almost 100% boric oxide (fused boron rods). Therefore the assessment of efficacy and human health effects based on the model product are representative for existing products (boron rods) that contain boric oxide.

Inclusion of boric oxide in Annex I is feasible for the aspects efficacy, human health and environmental aspect because efficacy is demonstrated for preventive application of boron rods and a safe use is identified for human exposure and for environmental exposure when the treated wood is used for Use Class 1 (wood under cover and fully protected from the weather)

and Use Class 2 (wood under cover and fully protected from the weather but where high environmental humidity can lead to occasional but not persistent wetting for indoor use and/or incidentally getting wet).

For Use Class 3 and 4a the environmental risk assessment is performed according to the TGD and TNsG's, with PEC:PNEC > 1 for the aquatic environment (bridges over ponds), terrestrial environment and the drinking water criterion as a result. At product authorisation level extra attention should be paid to potential risks for the environment for Use Classes 3 and 4a.

Methods of analysis:

All necessary methods of analysis for boric oxide are available.

Efficacy:

Pure boric oxide has indicated to be effective against fungi (wet rot, dry rot), egg larvae of wood-boring insects. However, the available data do not permit that final conclusions can be drawn concerning efficacy against all the target organisms to be controlled. Furthermore, leaching of most of the impregnated boric oxide from treated wood can occur within days/months. No information is available about the influence of leaching on the efficacy of the boric oxide product.

Labelling and classification:

Boric oxide has a proposed classification with: 'T' (Toxic), R60 (may impair fertility) and R61 (may cause harm to the unborn child). Products containing more than the specific concentration limit of 3.1% of boric acid (30th ATP to Directive 67/548/EEC) will be classified with R60-61 and shall not be authorised for marketing to, or use by the general public.

Human health:

Adverse health effects due to dermal and respiratory exposure to boric oxide is not expected using boric oxide as boron rods.

Environment:

Service life of wood:

For Use Class 1 and 2, no emissions to the environment are expected during service life of the treated wood. Use Classes 3 and 4a treated wood with boron oxide is expected to lead to risks, resulting from emissions to soil (in-service treated fences and posts, noise barriers, transmission poles and wooden houses) and water (Bridge over pond scenario) during the service life (see table 3.1-1).

In-situ treatment:

Emissions to soil and water during *in-situ* treatment with boric oxide are not expected to lead to risks.

Table 3.1-1 Summary of environmental risk of boric oxide.

	Fresh water environment	Sediment	Sewage treatment plants	Drinking water criterion	Persistence in sediment	Air	Terrestrial environment	Groundwater	Persistence in soil	PBT
<i>In-situ</i> treatment	n.a.	n.a.	n.a.	n.a.	-	-	n.a.	n.a.	+	-
Wood in service	+ ^a	+ ^a	-	+	+	-	+	+	+	-

- no risk expected

+ risk expected

n.a. not applicable

^a risk reduction possible, for instance storage on hard standing and collection of rainwater

3.2. Decision regarding Inclusion in Annex I

The active substance boric oxide shall be included in Annex I to Directive 98/8/EC as an active substance for use in product-type 8 (wood preservative), subject to the following specific provisions:

The active substance boric oxide shall have a minimum purity of 97.5% expressed as B₂O₃.

Member States shall ensure that authorisations are subject to the following conditions:

- (1) Products authorised for industrial and professional use must be used with appropriate personal protective equipment, unless it can be demonstrated in the application for product authorisation that risks to industrial and/or professional users can be reduced to an acceptable level by other means.
- (2) In view of the risks identified for the soil and aquatic compartments appropriate risk mitigation measures must be taken to protect those compartments. In particular, labels and/or safety-data sheets of products authorised for industrial use shall indicate that freshly treated timber must be stored after treatment under shelter or on impermeable hard standing to prevent direct losses to soil or water and that any losses must be collected for reuse or disposal.

3.3. Elements to be taken into account by Member States when authorising products

The risk assessment based on the dummy product showed that the use of boric oxide in Use Class 1 and 2 is safe. For the use of that product in Use Class 3 and 4a unacceptable risks were identified in the soil and water compartment due to in-service leaching of boric oxide from treated wood. Boric oxide shall not be authorized in wood preservative products for Use Class 3 and 4a if it has been concluded in the means of comparative assessment that there is a sufficient number of active substances for same use purposes for Use Class 3 and 4a with significantly lower risk and without unacceptable effects.

Boric oxide was intended to be used by industrial, professional and non-professional users. However, boric oxide was found to be 'Toxic for reproduction category 2 (R60-61)'. Products

containing more than the specific concentration limit of 3,1% boric oxide will be classified with R60-61 and according to article 5 (2) of the Directive, products with this classification shall not be authorised for marketing to, or use by the general public.

Handling and Storage

No special handling precautions are required, but dry indoor storage is recommended. Good housekeeping procedures should be followed to minimise dust generation and accumulation. No specific fire fighting measures are required since boric oxide is not flammable, combustible or explosive. The product is itself a flame retardant.

Suitable container material: paper, cardboard, plastic (polypropylene, high density polyethylene). The use of polypropylene may best be avoided for storage at lower temperatures as the material becomes brittle.

Unsuitable container material: base metals.

Emergency measures in case of accident

Inhalation: If symptoms such as nose or throat irritation are observed, remove to fresh air.

In case of contact with eyes: Rinse immediately with plenty of clean water or sterile saline solution for at least 15 minutes. If appropriate, remove contact lenses after 5 minutes rinsing. If symptoms persist, seek medical attention.

Skin contact: No treatment necessary because non-irritating.

Ingestion: Swallowing small quantities (one teaspoon) will cause no harm to healthy adults. If larger amounts are swallowed, give two glasses of water to drink and seek medical attention.

Avoid creation of dust. Use vacuum cleaners wherever possible.

Decontamination

Air: Borates are non-volatile. As a dust borates rapidly settle from the atmosphere.

Water: Borates are naturally occurring minerals and are present in surface and underground waters. Borates are rapidly dissolved in water and will disperse with dilution. Removal at low concentrations is unnecessary. Where water containing high levels of borates can be captured precipitation with lime can be used to reduce boron levels to the 100 ppm range. Treatments with boron specific ion exchange resins and activated carbon are also possible.

Soil: Borates are naturally found in rocks and soil and are an essential micronutrient for all plant growth. Contaminated soil can be leached with water or acid to reduce boron levels.

Waste Management

Prevention of waste and unintended environmental releases is practiced by most, if not all operations. Storage of treated products is covered so as to prevent leaching and run-off. Small quantities can usually be disposed to landfill sites. No special disposal treatment is required, but local authorities should be consulted about any specific local requirements. Tonnage quantities of products are not considered appropriate for landfills. Such products should, if possible, be used for an appropriate application.

Unintended side effects

Borates are essential micronutrients for all plant life but at high levels they are phytotoxic.

Combustion Products

Fused borate, glass and water.

3.4. Requirement for further information

It is considered that the evaluation has shown that sufficient data have been provided to verify the outcome and conclusions, and permit the proposal for the inclusion of boric oxide in Annex 1 to Directive 98/8/EC.

3.5. Updating this Assessment Report

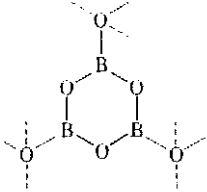
This assessment report may need to be updated periodically in order to take account of scientific developments and results from the examination of any of the information referred to in Articles 7, 10.4 and 14 of Directive 98/8/EC. Such adaptations will be examined and finalised in connection with any amendment of the conditions for the inclusion of boric oxide in Annex I to the Directive.

Appendix I: List of endpoints

Chapter 1: Identity, Physical and Chemical Properties, Classification and Labelling

Active substance (ISO Common Name)	Boric oxide
Function (<i>e.g.</i> fungicide)	Fungicide, insecticide
Rapporteur Member State	The Netherlands

Identity (Annex IIA, point II.)

Chemical name (IUPAC)	Diboron trioxide
Chemical name (CA)	Diboron trioxide
CAS No.	Two CAS numbers exist for the same compound. <ul style="list-style-type: none"> Boric oxide, CAS no 1303-86-2 Boric oxide, unstated purity, CAS no 54566-73-3. For this CA-report, CAS no 1303-86-2 is used (pure boric oxide). CAS no 54566-73-3 has an unstated purity and is not assessed in this CA-report.
EC No.	Two EC numbers exist for the same compound. <ul style="list-style-type: none"> Boric oxide, EC no 215-125-8. Boric oxide, unstated purity, EC no 259-231-2. For this CA-report, EC no is 215-125-8 is used (pure boric oxide). EC no 259-231-2 has an unstated purity and is not assessed in this CA-report.
Other substance No.	None
Minimum purity of the active substance as manufactured (g/kg or g/l)	minimum purity 97.5% (w/w), expressed as B ₂ O ₃
Identity of relevant impurities and additives (substances of concern) in the active substance as manufactured (g/kg)	no additives no impurities > 1 g/kg no (eco)toxicologically relevant impurities
Molecular formula	B ₂ O ₃
Molecular mass	69.62
Structural formula	The solid state boric oxide consists of boroxyl rings and probably is an infinite polymeric structure. 

Physical and chemical properties

Melting point (state purity)	No melting point found in range 25-360°C at atmospheric pressure (purity not stated)
Boiling point (state purity)	Not applicable
Temperature of decomposition	Endotherms observed at 100°C, 140°C and 160°C are interpreted as loss of free and crystallisation water.
Appearance (state purity)	Solid white odourless powder (purity not stated)
Relative density (state purity)	1.85 at 21.5°C (purity not stated)
Surface tension	Not applicable
Vapour pressure (in Pa, state temperature)	Not applicable. because the melting point lies above 300°C and at ambient temperature vapour pressure is expected to be less than 10 ⁻⁵ Pa.
Henry's law constant (Pa m ³ mol ⁻¹)	Not applicable. because the melting point lies above 300°C and at ambient temperature vapour pressure is expected to be less than 10 ⁻⁵ Pa.
Solubility in water (g/l or mg/l, state temperature)	<p>PH 4.25-4.41: 20.4 g/L in water at 10°C, calculated from boric acid solubility.</p> <p>pH 3.76-4.08: 27.2 g/L in water at 20°C, calculated from boric acid solubility.</p> <p>pH 3.96-4.07: 35.2 g/L in water at 30°C, calculated from boric acid solubility.</p> <p>pH 5.60-5.69: 30.5 g/L at 20°C in Na/K-phosphate solution, calculated from boric acid solubility.</p> <p>pH 5.64-5.69: 35.4 g/L at 20°C in tris-HCl solution, calculated from boric acid solubility.</p> <p>pH 5.96 – 6.01: 37.4 g/L at 20 °C in ammonium carbonate buffer, calculated from boric acid solubility.</p> <p>-----</p> <p>The water solubility for boric oxide as such cannot be determined because boric oxide is converted into boric acid upon dissolution in water. The water solubility for boric oxide was calculated from water solubility experiments with boric acid.</p> <p>Water solubility studies at pH=5, 7, 9 are not possible, because of the strong buffering capacity of boric acid solutions and ion-pair formation in the presence of alkali-metal ions like Na, K.</p>
Solubility in organic solvents (in g/l or mg/l, state temperature) (Annex IIIA, point III.1)	Not required. The a.s. is not used in organic solvents and data will not relevant information for the risk assessment.
Stability in organic solvents used in biocidal products including relevant breakdown products (IIIA, point III.2)	Not relevant.
Partition coefficient (log P _{ow}) (state temperature)	pH_5: not investigated

	pH_7: not investigated
	pH_9: not investigated
	pH_7.5: -1.09 at 22°C in potassium/sodium phosphate buffer at a concentration of 0.0097 M boron (purity 99.0% w/w expressed as H ₃ BO ₃).
	pH_unknown: -0.757 at 25 °C in water at concentration levels between 0.16 - 0.89 M boron (purity not indicated)
	pH_unknown: -0.74 in 2 M KCl at 25 °C
	pH_unknown: -0.56 in 3 M NaClO ₄ at 25 °C
	pH_unknown: -0.55 in 3 M NaClO ₄ at 35 °C
	The log P _{ow} for boric oxide as such cannot be determined because boric oxide is converted into boric acid upon dissolution in water. The log P _{ow} given is the log P _{ow} for boric acid. The difference between log P _{ow} values obtained at different temperatures, different salinity, different concentration and different analysis, is only 0.5 log P _{ow} unit. No further tests are required.
Hydrolytic stability (DT ₅₀) (state pH and temperature) (point VII.7.6.2.1)	No data available
Dissociation constant (not stated in Annex IIA or IIIA; additional data requirement from TNsG)	The dissociation constant for boric oxide as such cannot be determined because boric oxide is converted into boric acid upon dissolution in water. The dissociation constant given is the dissociation constant for boric acid. Boric acid is a Lewis acid (hydroxide ion acceptor) rather than a Brønsted acid (proton donator). For this purpose the formula for boric acid is best written as B(OH) ₃ . pKa = 9.0 at 25°C for boric acid in dilute solutions only (B ≤ 0.025 M). At higher boron concentrations, polynuclear complexes are formed and several dissociation/formation constants apply.
UV/VIS absorption (max.) (if absorption > 290 nm state ε at wavelength)	No absorption maximum or minimum found in neutral, basic or acidic medium in the range 200-750 nm.
Photostability (DT ₅₀) (aqueous, sunlight, state pH) (point VII.7.6.2.2)	No data available.
Quantum yield of direct phototransformation in water at λ > 290 nm (point VII.7.6.2.2)	No data available.
Flammability	Not highly flammable, no pyrophoric behaviour, not self-igniting.
Explosive properties	Not explosive.
Oxidative properties	Not oxidising.

Classification and proposed labelling

with regard to physical/chemical data
 with regard to toxicological data
 with regard to fate and behaviour data
 with regard to ecotoxicological data

None
Toxic, R60-61
None
None

Chapter 2: Methods of Analysis**Analytical methods for the active substance**

Technical active substance (principle of method)
 (Annex IIA, point 4.1)

Method BS 5688 : Part 8 :1979 ISO 2216-1972 [ISO title: Crude sodium borates for industrial use- Determination of sodium oxide and boric oxide contents- Volumetric method].

Impurities in technical active substance (principle of method) (Annex IIA, point 4.1)

Not required.
 No impurities > 1 g/kg and no (eco)toxicologically relevant impurities.

Analytical methods for residues

Soil (principle of method and LOQ) (Annex IIA, point 4.2)

Total boron can be extracted from soil by hot acid extraction with HNO₃ and HCl under reflux or by microwave digestion with HNO₃ and HF, with addition of HCl and H₂O₂ when needed. Detection by ICP-AES (LOD ca. 4 µg B/L) or by photospectrometry after reaction with azomethine-H (LOD 0.01 – 1 mg B/L). LOQ depends on the extraction method

Air (principle of method and LOQ) (Annex IIA, point 4.2)

Not required.

Water (principle of method and LOQ) (Annex IIA, point 4.2)

Same methods as for soil

Body fluids and tissues (principle of method and LOQ) (Annex IIA, point 4.2)

Although classified as Toxic, no method is required. The same method as for soil can be used.

Food/feed of plant origin (principle of method and LOQ for methods for monitoring purposes) (Annex IIIA, point IV.1)

Not applicable

Food/feed of animal origin (principle of method and LOQ for methods for monitoring purposes) (Annex IIIA, point IV.1)

Not applicable

Chapter 3: Impact on Human Health**Absorption, distribution, metabolism and excretion in mammals** (Annex IIA, point 6.2)

Rate and extent of oral absorption:

100%

Rate and extent of dermal absorption:

0.5%

Distribution:

Absorbed boron rapidly distributes throughout the body

Potential for accumulation:	water in humans and animals.
Rate and extent of excretion:	Low
	Excreted almost exclusively in the urine.
	Half-life < 24 hours.
Toxicologically significant metabolite	Does not metabolise Most of the simple inorganic borates such as boric acid; boric oxide; disodium tetraborate and disodium octaborate tetrahydrate exist predominantly as un-dissociated boric acid in dilute aqueous solution at physiological pH, leading to the conclusion that the main species in the plasma of mammals is un-dissociated boric acid. Since other borates dissociate to form boric acid in aqueous solutions, they too can be considered to exist as un-dissociated boric acid under physiological conditions. For example disodium octaborate tetrahydrate a solid solution of boric acid and disodium tetraborate decahydrate

Acute toxicity (Annex IIA, point 6.1)

Rat LD ₅₀ oral	> 2600 mg/kg bw
Rat LD ₅₀ dermal	> 2000 mg/kg bw (read across boric acid)
Rat LC ₅₀ inhalation	> 2120 mg/m ³ (read across boric acid)
Skin irritation	Non Irritant
Eye irritation	Non Irritant
Skin sensitization (test method used and result)	Non Sensitiser – Buehler and Human

Repeated dose toxicity (Annex IIA, point 6.3)

Species/ target / critical effect	Testis, blood
Lowest relevant oral NOAEL / LOAEL	17.5 mg boron /kg bw/day
Lowest relevant dermal NOAEL / LOAEL	No data available
Lowest relevant inhalation NOAEL / LOAEL	No data available

Genotoxicity (Annex IIA, point 6.6)

Negative

Carcinogenicity (Annex IIA, point 6.4)

Species/type of tumour	Negative
lowest dose with tumours	Not relevant

Reproductive toxicity (Annex IIA, point 6.8)

Species/ Reproduction target / critical effect	Rat; mouse and dog testis; Testicular atrophy
Lowest relevant reproductive NOAEL / LOAEL	17.5 mg boron/kg bw/day
Species/Developmental target / critical effect	Rat: reduction in mean fetal body weight/litter, increased

	incidence in short rib X111) Rabbit: increased incidence in major heart and/or great vessel malformations
Lowest relevant developmental NOAEL / LOAEL	9.6 mg boron/kg bw/day

Neurotoxicity / Delayed neurotoxicity (Annex IIIA, point VI.1)

Species/ target/critical effect	Not relevant
Lowest relevant developmental NOAEL / LOAEL.	Not relevant

Other toxicological studies (Annex IIIA, VI/XI)

Not relevant

Medical data (Annex IIA, point 6.9)

<p>The human oral lethal dose is regularly quoted as 2-3 g boric acid for infants, 5-6 g boric acid for children and 15-30 g boric acid for adults. Acute effects of poisoning may include nausea, vomiting, gastric discomfort, skin flushing, excitation, convulsions, depression and vascular collapse. Multiple exposure to high levels of boron may result in dermatitis, alopecia, loss of appetite, nausea, vomiting, diarrhoea, and focal or generalised central nervous system irritation or convulsions.</p> <p>Workers exposed to borax dust at about 4 mg/m³ reported eye irritation, dry mouth, nose or throat, sore throat, acute and chronic respiratory irritation and productive cough.</p> <p>In human populations exposed to high boron levels in drinking water or male workers of a borax mine no effects on reproduction were observed.</p>
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Summary (Annex IIA, point 6.10)

ADI (if residues in food or feed)

AOEL (Operator/Worker Exposure)
acute/semi-chronic and chronic

Drinking water limit

ARfD (acute reference dose)

Value	Study	Safety factor
Not relevant		
Rounded 0,1* mg B/kg bw/day.	developmental study rat	100
	EU Council Directive 98/83/EC. Quality of Water Intended for Human Consumption	
not relevant		

* A rounded (systemic) AOEL value of 0.1 mg B/kg bw/day was used for boric oxide based on the NOAEL for embryotoxic/teratogenic effects of 9.6 mg B/kg bw/day and a standard assessment factor of 100 even though an AOEL value of 0.06 mg B/kg bw/day (based on a NOAEL of 17.5 mg B/kg bw/day in a 2-year study in the rat and an assessment factor of 300 due to serious effects) and an AOEL value of 0.07 mg B/kg bw/day (based on a NOAEL of 21.8 mg B/kg bw/day in a teratogenicity study in the rabbit and an assessment factor of 300 due to serious effects) could be derived from the critical effects.

Acceptable exposure scenarios (including method of calculation)

Professional users	Exposure is considered negligible. Therefore, adverse health effects due to the use of boric oxide fused rods are not expected.
Non-professional users	Exposure is considered negligible. Therefore, adverse health effects due to the use of boric oxide fused rods are not expected. However, biocidal products containing more than 3.1% boric oxide and classified with R60-61 shall not be authorised for marketing to, or use by the general public.
Indirect exposure as a result of use	Exposure is considered negligible. Therefore, adverse health effects due to the use of boric oxide fused rods are not expected.

Chapter 4: Fate and Behaviour in the Environment

Route and rate of degradation in water (Annex IIA, point 7.6, IIIA, point XII.2.1, 2.2)

Hydrolysis of active substance and relevant metabolites (DT ₅₀) (state pH and temperature)	hydrolytically stable
Photolytic / photo-oxidative degradation of active substance and resulting relevant metabolites	photolytically stable
Quantum yield of direct phototransformation in water at $\lambda > 290$ nm (point VII.7.6.2.2)	not applicable
Readily biodegradable (yes/no)	No
Biodegradation in seawater	not applicable
Non-extractable residues	Not applicable
Distribution in water / sediment systems (active substance)	no information supplied
Distribution in water / sediment systems (metabolites)	not applicable

Route and rate of degradation in soil (Annex IIIA, point VII.4, XII.1.1, XII.1.4; Annex VI, para. 85)

Mineralization (aerobic)	not applicable
Laboratory studies (range or median, with number of measurements, with regression coefficient)	DT _{50lab} (20°C, aerobic): not applicable
	DT _{90lab} (20°C, aerobic): not applicable
	DT _{50lab} (10°C, aerobic): not applicable
	DT _{50lab} (20°C, anaerobic): not applicable
	degradation in the saturated zone: not applicable
Field studies (state location, range or median with number of measurements)	DT _{50f} : not applicable
	DT _{90f} : not applicable
Anaerobic degradation	not applicable
Soil photolysis	not applicable
Non-extractable residues	not applicable
Relevant metabolites - name and/or code, % of applied a.i. (range and maximum)	not applicable
Soil accumulation and plateau concentration	not available

Adsorption/desorption (Annex IIA, point XII.7.7; Annex IIIA, point XII.1.2)

K _a , K _d	Average K _F 2.6 L/kg (range 0.4 – 8.41 L/kg; n = 9); average 1/n 0.83 (range 0.726 – 0.955)
K _{a_{oc}} , K _{d_{oc}}	no clear relationship with OC
pH dependence (yes / no) (if yes type of dependence)	no

Fate and behaviour in air (Annex IIIA, point VII.3, VII.5)

Direct photolysis in air	does not photolyse
Quantum yield of direct photolysis	not applicable
Photo-oxidative degradation in air	not applicable
Volatilization	Negligible – vapour pressure 9.9 x 10 ⁻⁶ Pa at 25°C

Monitoring data, if available (Annex VI, para. 44)

Soil (indicate location and type of study)	Range 45-124 mg/kg worldwide
Surface water (indicate location and type of study)	Range < 0.001 – 0.632 mg B/L in surface water (review of monitoring values for 15 EU countries)
Ground water (indicate location and type of study)	not available
Air (indicate location and type of study)	Estimated as 20 ng/m ³ (monitoring values near coastlines)

Chapter 5: Effects on Non-target Species

Toxicity data for aquatic species (most sensitive species of each group)

(Annex IIA, point 8.2, Annex IIIA, point 10.2)

Species	Time-scale	Endpoint	Toxicity
Fish			
<i>Catostomus latipinnis</i>	acute (96 h)	LC50	125 mg B/L
<i>Brachydanio rerio</i>	chronic (34 d)	NOEC	1.8 mg B/L
Invertebrates			
<i>Daphnia magna</i>	acute (48 h)	L(E)C50	141 mg B/L
<i>Daphnia magna</i>	chronic (21 d)	NOEC	10 mg B/L
Algae			
<i>Selenastrum capricornutum</i>	chronic (74.5 h)	NOEC (growth inhibition)	17.5 mg B/L
Microorganisms			
Activated sludge	chronic (3 h)	NOEC (respiration inhibition)	17.5 mg B/L
An assessment factor of 10 was applied to the lowest chronic NOEC of 1.8 mg B/l. Resulting in a PNEC _{add, aquatic} of 0.18 mg B/l.			

Effects on earthworms or other soil non-target organisms (most sensitive species for each group)

Acute toxicity for earthworms
(Annex IIIA, point XIII.3.2)

14-days LC₅₀
Lumbricus terrestris: 473 mg B/kg dwt (geomean of 501 and 447)

Reproductive toxicity for earthworms
(Annex IIIA, point XIII.3.2)

63-days NOEC
Eisenia andrei: 54 mg B/kg dwt soil (geomean of 39, 79, 52, 52)

Reproductive toxicity for springtails

28-days NOEC Juvenile reproduction
Folsomia candida: 14 mg B/kg dwt soil (geomean of 14 and 14)

Toxicity to plants

120-days NOEC root/shoot ratio
Hordeum vulgare: 2 mg B/kg dwt soil

An assessment factor of 5 was applied to the lowest chronic NOEC of 2 mg B/kg dwt. Resulting in a PNEC_{add, soil} of 0.4 mg B/kg dwt or 0.35 mg B/kg wwt.

Effects on soil micro-organisms (Annex IIA, point 7.4)

Nitrogen mineralization

20-days EC₁₀: 54 mg B/kg dwt soil

Carbon mineralization

no data available

Effects on terrestrial vertebratesAcute toxicity to mammals
(Annex IIIA, point XIII.3.3)Oral LD₅₀
Rat: > 2600 mg/kg bwAcute toxicity to birds
(Annex IIIA, point XIII.1.1)LD₅₀
Colinus virginianus: > 527 mg B/kg bwDietary toxicity to birds
(Annex IIIA, point XIII.1.2)5-days LC50
Colinus virginianus: > 983 mg B/kg fd; > 571 mg B/kg
bw.d
Anas platyrhynchos: > 2100 mg B/kg fdReproductive toxicity to birds
(Annex IIIA, point XIII.1.3)

no data available

Effects on honeybees (Annex IIIA, point XIII.3.1)

Acute oral toxicity

no data available

Acute contact toxicity

no data available

Effects on other beneficial arthropods (Annex IIIA, point XIII.3.1)

Acute oral toxicity

no data available

Acute contact toxicity

no data available

Acute toxicity to

no data available

Bioconcentration (Annex IIA, point 7.5)

Bioconcentration factor (BCF)

BAF < 10 L/kg (fish)
BAF < 30 L/kg (plankton and invertebrates)Depuration time (DT₅₀)
(DT₉₀)

virtually all boron was eliminated within 24 days

Level of metabolites (%) in organisms accounting
for > 10% of residues

no data available

Appendix II: List of Intended Uses

Object and/or situation	Member State or Country	Product name	Organisms controlled	Formulation		Application			Applied amount, assuming a wood density of 500 kg/m ³ (pine wood)			
				Type	Conc of as	Method kind	No	Interval between applications	kg/m ³ , BAE in wood	% (w/w), BAE in wood	% (w/v) BAE in treatment solution	kg/m ² , BAE on wood
Wood preservation (PT08) Preventive, UC 1,2 and 3	All Member States	model product containing boric oxide	decay fungi; wood boring insects	boron rods; stand alone product	> 97.5%	professional; in-situ; rod insertion in predrilled holes that are sealed later	1	na	2.0	0.4	na	na
Wood preservation (PT08) Preventive UC 4a and curative UC 1, 2, 3 and 4a	All Member States	model product containing boric oxide	decay fungi; wood boring insects	boron rods; stand alone product	> 97.5%	professional; in-situ; rod insertion in predrilled holes that are sealed later	1	na	4.0	0.8	na	1.8 x 10 ⁻²

na not applicable

BAE boric acid equivalents. For boric oxide the conversion factor from boric oxide weight units (e.g. gram) to boric acid equivalent weight units (e.g. gram) is 1.78. The conversion factor from BAE to boron equivalent is 0.175.

UC1 Use Class 1 timber under cover and fully protected from weather and not exposed to wetting (preventive and curative)

UC2 Use Class 2 timber under cover and fully protected from the weather but where high environmental humidity can lead to occasional but not persistent wetting (preventive and curative)

UC3 Use Class 3 timber not covered and not in contact with ground. It is either continually exposed to weather or is protected from weather but subject to frequent wetting (preventive and curative)

UC4a Use Class 4a application to timbers in service in ground contact (preventive and curative).

Appendix III: List of studies

Data protection is claimed by the applicant in accordance with Article 12.1(c) (i) and (ii) of Council Directive 98/8/EC for all study reports marked “Y” in the “Data Protection Claimed” column of the table below. For studies marked Yes(i) data protection is claimed under Article 12.1(c) (i), for studies marked Yes(ii) data protection is claimed under Article 12.1(c) (ii). These claims are based on information from the applicant. It is assumed that the relevant studies are not already protected in any other Member State of the European Union under existing national rules relating to biocidal products. It was however not possible to confirm the accuracy of this information.

BEU = Borax Europe Limited

Reference list of studies on the active substance

Section No / Reference No	Author(s)	Year	Title. Source (where different from company) Company, Report No. GLP (where relevant) / (Un)Published Electronic File Name if Appropriate	Data Protection Claimed (Yes/No)	Owner
A2.1/01	CRC	1999	CRC Handbook of Chemistry and Physics, electronic version 1999. Non-GLP Published	No	
A2.1/02	Merck Index	1996	Merck Index on CD-rom, version 12.1, 1996	No	
A2.5/01	CRC	1999	CRC Handbook of Chemistry and Physics, electronic version 1999. Non-GLP Published	No	
A2.10/01	Smith, R.A., Ascherl, F.M	1999	Issues concerning the measurement of borate in occupational environments. Am. Ind. Hyg. Assoc. J., 60, No. 5, p.651-658. (September - October 1999) GLP Not relevant Published	No	
A2.10/02	McCormack and Dickinson,	1993	Collaborative Project to Assessment the Effectiveness of Boron Rods as a Secondary Preservative Treatment in Wood Poles. Imperial College of Science Technology and Medicine. Final Report. GLP Not relevant Unpublished	No	
A2.10/03	IPCS,	1998	Environmental Health Criteria 204: Boron. International Programme on Chemical Safety involving US EPA, UNEP, WHO and International Labour Organisation, CI, USA GLP Not relevant Published	No	
A2.10/04	Wyness et al.,	2003	A summary of boron surface water quality data throughout the European Union. Science of the Total Environment 2003: (314-316), 255-269 GLP Not relevant Published	No	

Section No / Reference No	Author(s)	Year	Title. Source (where different from company) Company, Report No. GLP (where relevant) / (Un)Published Electronic File Name if Appropriate	Data Protection Claimed (Yes/No)	Owner
A3.1.1/01	Mellor's Comprehensive	1980	Supplement to Mellor's comprehensive treatise on inorganic and theoretical chemistry, Volume V, Boron, Part A: Boron-Oxygen Compounds, Longman London and New York, ISBN 0-582-46277-0, pp 170-213, pp 224-320, pp 343-344, pp 356-387, and pp 409-500. Non-GLP Published	No	
A3.1.1/02	Cuthbert JE, Mullee DM	2003	Boric Oxide: Determination of General Physico-Chemical Properties. Shardlow, Derbyshire, UK: SafePharm Laboratories SPL Project Number: 1231/004. Borax Europe Ltd GLP Unpublished	Yes	BEU
A3.1.2/01	Kirk-Othmer Encyclopedia	1992	Kirk-Othmer Encyclopedia of Chemical Technology, John Wiley & Sons Inc., 4th Edition, Volume 4, pp 368-386. Non-GLP Published	No	
A3.1.3/01	Mellor's Comprehensive	1980	Supplement to Mellor's comprehensive treatise on inorganic and theoretical chemistry, Volume V, Boron, Part A: Boron-Oxygen Compounds, Longman London and New York, ISBN 0-582-46277-0, pp 170-213, pp 224-320, pp 343-344, pp 356-387, and pp 409-500. Non-GLP Published	No	
A3.1.3/02	Cuthbert JE, Mullee DM	2003	Boric Oxide: Determination of General Physico-Chemical Properties. Shardlow, Derbyshire, UK: SafePharm Laboratories SPL Project Number: 1231/004. Borax Europe Ltd GLP Unpublished	Yes	BEU
A3.2/01	Mellor's Comprehensive	1980	Supplement to Mellor's comprehensive treatise on inorganic and theoretical chemistry, Volume V, Boron, Part A: Boron-Oxygen Compounds, Longman London and New York, ISBN 0-582-46277-0, pp 170-213, pp 224-320, pp 343-344, pp 356-387, and pp 409-500. Non-GLP Published	No	
A3.2/02	Tremain SP	1998	Boric acid: determination of vapour pressure. Derby, UK: Safepharm Laboratories Ltd, SPL project number 1231/002. Borax Europe Ltd. GLP Unpublished	Yes	BEU

Section No / Reference No	Author(s)	Year	Title. Source (where different from company) Company, Report No. GLP (where relevant) / (Un)Published Electronic File Name if Appropriate	Data Protection Claimed (Yes/No)	Owner
A3.4/01	Spruit WET, Van Baar BLM and Wils ERJ	2005	Some spectral data of boric oxide. Rijswijk, The Netherlands: TNO Defence, Security and Safety, report TNO-DV2 2005-PU002 Borax Europe Ltd. GLP. Unpublished	Yes	BEU
A3.5/01	Hoare AM	2003	Boric Oxide, B ₂ O ₃ , Justification for waiving the Water Solubility study, SafePharm Laboratories Ltd, 2003. Non-GLP Unpublished	Yes	BEU
A3.5/02	Mellor's Comprehensive	1980	Supplement to Mellor's comprehensive treatise on inorganic and theoretical chemistry, Volume V, Boron, Part A: Boron-Oxygen Compounds, Longman London and New York. ISBN 0-582-46277-0, pp 170-213, pp 224-320, pp 343-344, pp 356-387, and pp 409-500. Non-GLP Published	No	
A3.5/03	Borax Holdings	no date	Boron products and their applications, London, UK, Borax Holdings Ltd. Non-GLP Published	Yes	BEU
A3.5/04	Dawber JG and Matusin DH	1982	Potentiometric and polarimetric studies on the reaction of boric acid and tetrahydroxyborate ion with polyhydroxy compounds J. Chem. Soc. Faraday Trans. 1, 78, 2521-2528. Non-GLP Published	No	
A3.5/05	Cordia JA, Bal EA, Mak WA, Wils ERJ.	2003	Determination of some physico-chemical properties of Optibor EP. Rijswijk, The Netherlands: TNO Prins Maurits Laboratory. report PML 2002-C42rr, August 2003 Borax Europe Ltd. GLP, provisional: unauthorised version Unpublished	Yes	BEU
A3.5/06	Spruit WET	2005	Determination of the water solubility of boric acid. Rijswijk, The Netherlands: TNO Defence, Security and Safety, Draft report TNO-DV2 2005-PU11, March 8, 2005. GLP, provisional: unauthorised draft version Unpublished	Yes	BEU
A3.6/01	Bell RP, Edwards JO and Jones RB	1967	The structure and acidity of boric acid and their relation to reaction mechanisms. In: The chemistry of boron and its compounds. pp 209-221 Non GLP Published	No	
A3.6/02	Kankaanpera A and Salomaa P.	1969	The structure of boric acid in aqueous solution. Acta Chem. Scand. 23, No. 2, 712-713 Non-GLP Published	No	

Section No / Reference No	Author(s)	Year	Title. Source (where different from company) Company, Report No. GLP (where relevant) / (Un)Published Electronic File Name if Appropriate	Data Protection Claimed (Yes/No)	Owner
A3.6/03	Jenkins HO	1945	Transactions of the Faraday Society 41, 138-140 Non-GLP Published	No	
A3.6/04	WHO.	1998	Guidelines for drinking water quality. 2nd Edition Addendum to Vol 2 Boron, page 15, World Health Organisation, Geneva. Non-GLP Published	No	
A3.6/05	Hahn FL and Klockmann R	1930	Die höheren Dissoziationsstufen der Phosphorsäure und der Borsäure Z physik. Chem., 151, 80-86. Non-GLP Published	No	
A3.6/06	Ingri N	1963	Equilibrium studies of polyanions containing B ^{III} , Si ^{IV} , Ge ^{IV} , and V ^V . Sven. Kem. Tidskr. 75(4):199-230. Non-GLP Published	No	
A3.6/07	Maeda M	1979	Raman spectra of polyborate ions in aqueous solution. J Inorg. Nucl. Chem. vol 41: 1217-1220 Non-GLP Published	No	
A3.6/08	Farmer J	1982	Structural chemistry in the borate industry, Chem and Ind Non-GLP Published	No	
A3.6/09	Kirk-Othmer Encyclopedia	1992	Kirk-Othmer Encyclopedia of Chemical Technology, John Wiley & Sons Inc., 4th Edition, Volume 4, pp 368-386. Non-GLP Published	No	
A3.6/10	Holleman AF and Wiberg E and Wiberg N	1995	Lehrbuch der anorganischen Chemie. 101 st ed. De Gruyter, Berlin	No	
A3.6/11	De Vette HQM, Otto C, Schoonmade MA	2001	Hydrolysis as a function of pH and identification of breakdown products. A study on the identification and comparison of the dissociation products of Polybor tech, Borax Manufacturing grade and Boric acid manufacturing grade in aqueous solution using Raman spectroscopy. TNO report no v99-1153 (TX-2001-001)	Yes	BEU
A3.7/01	CRC	1999	CRC Handbook of Chemistry and Physics, electronic version 1999. Non-GLP Published	No	
A3.9/01	Hoare AM	2003	Boric Oxide, B ₂ O ₃ . Justification for waiving the octanol/water partition coefficient study. Derby, UK: SafePharm Laboratories Ltd. Non-GLP Unpublished	Yes	BEU

Section No / Reference No	Author(s)	Year	Title. Source (where different from company) Company, Report No. GLP (where relevant) / (Un)Published Electronic File Name if Appropriate	Data Protection Claimed (Yes/No)	Owner
A3.9/02	Barres M	1967	Contribution to the study of the isopolycondensation of alkaline borates by electrometry and partition Rev. Chim. Miner., 4, 803-838; Non-GLP Published	No	
A3.9/03	Cordia JA, Bal EA, Mak WA, Wils ERJ.	2003	Determination of some physico-chemical properties of Optibor EP. Rijswijk, The Netherlands: TNO Prins Maurits Laboratory. report PML 2002-C42rr, August 2003 Borax Europe Ltd. GLP, provisional: unauthorised version Unpublished	Yes	BEU
A3.10/01	Kirk-Othmer Encyclopedia	1992	Kirk-Othmer Encyclopedia of Chemical Technology, John Wiley & Sons Inc., 4th Edition, Volume 4, pp 368-386. Non-GLP Published	No	
A3.13/01	Kirk-Othmer Encyclopedia	1992	Kirk-Othmer Encyclopedia of Chemical Technology, John Wiley & Sons Inc., 4th Edition, Volume 4, pp 368-386. Non-GLP Published	No	
A3.13/02	Wurster, E.D.	1963	Surface tension of aqueous 5-mol borax solutions. Ansheim, California, USA: US Borax Research Corporation, Technical Service Report No. 29A-1. Non-GLP Unpublished	Yes	BEU
A3.14/01	Kirk-Othmer Encyclopedia	1992	Kirk-Othmer Encyclopedia of Chemical Technology, John Wiley & Sons Inc., 4th Edition, Volume 4, pp 368-386. Non-GLP Published	No	
A3.15/01	Kirk-Othmer Encyclopedia	1992	Kirk-Othmer Encyclopedia of Chemical Technology, John Wiley & Sons Inc., 4th Edition, Volume 4, pp 368-386. Non-GLP Published	No	
A3.15/02	Mak WA	2004	Statement 014.16017/01.35. Rijswijk, The Netherlands: TNO Prins Maurits Laboratory Non-GLP Unpublished	Yes	BEU
A3.16/01	Mak WA	2004	Statement 014.16017/01.35. Rijswijk, The Netherlands: TNO Prins Maurits Laboratory Non-GLP Unpublished	Yes	BEU

Section No / Reference No	Author(s)	Year	Title. Source (where different from company) Company, Report No. GLP (where relevant) / (Un)Published Electronic File Name if Appropriate	Data Protection Claimed (Yes/No)	Owner
A4.1/01	BSI	1979	BS 5688: Part 8: 1979, ISO 2216-1972 British standard methods of test for boric acid, boric oxide, disodium tetraborates, sodium perborates and crude sodium borates for industrial use. Part 8. Determination of sodium oxide and boric oxide contents of crude sodium borates (ISO title: Crude Sodium Borates For Industrial Use- Determination Of Sodium Oxide And Boric Oxide Contents- Volumetric Method) Non-GLP Published	No	BSI/ISO
A5/01	Ingri N	1963	Equilibrium studies of polyanions containing B ^{III} , Si ^{IV} , Ge ^{IV} , and V ^V . Sven. Kem. Tidskr. 75(4):199-230. Non-GLP Published	No	
A5/02	Kirk-Othmer Encyclopedia	1992	Kirk-Othmer Encyclopedia of Chemical Technology, John Wiley & Sons Inc., 4th Edition, Volume 4, pp 368-386. Non-GLP Published	No	
A5/03	De Vette HQM, Otto C, Schoonmade MA	2001	Hydrolysis as a function of pH and identification of breakdown products. A study on the identification and comparison of the dissociation products of Polybor tech, Borax Manufacturing grade and Boric acid manufacturing grade in aqueous solution using Raman spectroscopy. TNO report no v99-1153 (TX-2001-001)	Yes	BEU
A5/04	Maeda M	1979	Raman spectra of polyborate ions in aqueous solution. J Inorg. Nucl. Chem. vol 41: 1217-1220 Non-GLP Published	No	
A5/05	Lloyd JD	1993	The mechanisms of action of boron containing preservatives. Thesis submitted for degree of Doctor of Philosophy of the University of London and the Diploma of Membership of Imperial College (page 45 and 47 Tables 4 and 5) Non-GLP Published	No	
A5.3/01	Harrow KM	1950	Toxicity of water-soluble wood-preservatives to wood-destroying fungi. NZ Journal Sci. & Tech. Vol B 31, (5):14-19. Non-GLP Published	No	
A5.3/02	Baechler RH and Roth HG	1956	Laboratory leaching and decay tests on pine and oak blocks treated with several preservative salts. Amer. Wood Preserver's Assoc. Proc. Annu. Meet., 52:24-34 Non-GLP Published	No	

Section No / Reference No	Author(s)	Year	Title. Source (where different from company) Company, Report No. GLP (where relevant) / (Un)Published Electronic File Name if Appropriate	Data Protection Claimed (Yes/No)	Owner
A5.3/03	Carr DR	1957	Toxicities of some water-borne wood preservatives to wood destroying fungi. New Zealand Forest Service, FRI report, VF 841.12:844.2 (931) CAR, 32 pages. Non-GLP Unpublished	No	
A5.3/04	Becker G, Hof T, Jacquot C, Lohwag K, Rennerfelt E and Walchli O	1966	Vergleichsversuche zur Laboratoriumsprüfung der pilzwidrigen Wirksamkeit von Holzschutzmitteln. Holz Roh-Werkst. 24 (2):53-58. Non-GLP Published	No	
A5.3/05	Gallagher LU	1968	Test of combined toxicity of Timbor and decay resistance of redwood heartwood. UK: Ministry of Technology, Forest Products Research Laboratory, Experiment 504 Non-GLP Unpublished	No	
A5.3/06	Smith RS	1969	Wood preservative toxicity evaluation using wood weight loss and fungal respiration methods. Wood Science 2(1):44-53 Non-GLP Published	No	
A5.3/07	Bechgaard C, Borup L, Henningson B and Jermer J	1979	Remedial treatment of creosoted railway sleepers of redwood by selective application of boric acid. Stockholm, Sweden: Swedish Wood Preservation Institute, report no 135E, ISSN 0346-7090. Non-GLP Published	No	
A5.3/08	Doi S, Mori M and Mineki Y	1994	Preservative efficacy of boric acid triethanol amine solution against wood decay fungi. Stockholm, Sweden: The International Research Group on Wood Preservation, document IRG/WP 94-30050 Paper prepared for the 25th Annual Meeting, Bali, Indonesia, May-June, 1994 Non-GLP Published	No	
A5.3/09	Cookson LJ and Pham K	1996	Relative tolerance of twenty basidiomycetes to boric acid. Mat und Org 29 (3) 187-196. Non-GLP Published	No	CSIRO
A5.3/10	Cummins JE and Wilson HB	1936	The preservation of timber against the attacks of powder post borer (<i>Lyctus brunneus</i> Stephens) by impregnation with various chemicals. J. Counc. Sci. Ind. Res., 9(1):37-56. Non-GLP Published	No	

Section No / Reference No	Author(s)	Year	Title. Source (where different from company) Company, Report No. GLP (where relevant) / (Un)Published Electronic File Name if Appropriate	Data Protection Claimed (Yes/No)	Owner
A5.3/11	Cummins JE	1939	The preservation of timber against the attacks of powder post borer (<i>Lyctus brunneus</i> Stephens) by impregnation with boric acid. J. Coun. Sci. Ind. Res., 12(1):30-49. Non-GLP Published	No	
A5.3/12	Spiller D	1948	Toxicity of boric acid to the common house borer, <i>Anobium punctatum</i> . Wellington, New Zealand: Department of Scientific and Industrial Research, Extract from the NZ Journal Sci. & Tech. Vol B 30 (20):22-30. Non-GLP Published	No	
A5.3/13	Taylor JM	1967	Toxicity of boron compounds to the common furniture beetle and house longhorn beetles. International Pest Control 9(1):14-17. Non-GLP Published	No	
A5.3/14	Mauldin JK and Kard BM	1996	Disodium octaborate tetrahydrate treatments to slash pine for protection against Formosan subterranean termite and eastern subterranean termite. J. of Economic Entomology 89:682-688. Non-GLP Published	No	
A5.3/15	Findlay WPK	1959	Boron compounds for the preservation of timber against fungi and insects. 6 th Wood Protection Congress of the German Society for Wood Research July, 1959 Reprinted in Pest Technology 2(6):124-127 Non-GLP Published	No	
A5.3/16	Becker G	1959	Beitrag zur Kenntnis der Wirksamkeit von Borverbindungen als Holzschutzmittel gegen Insekten und Pilze. Holz als Roh und Werkstoff 17(12):484-489. Non-GLP Published	No	
A5.3/17	Bravery AF and Carey JK	1983	A review of the data on the toxicity of boric acid to fungi and insects attacking wood. Building Research Advisory Service, Job no. P820648 Non-GLP Published	No	
A5.3/18	Borax Europe Ltd	2005	Explanatory note on efficacy assessment, application rates, environmental exposure assessment, boric acid. Borax Europe Ltd Non-GLP Unpublished	Yes	BEU

Section No / Reference No	Author(s)	Year	Title. Source (where different from company) Company, Report No. GLP (where relevant) / (Un)Published Electronic File Name if Appropriate	Data Protection Claimed (Yes/No)	Owner
A5.4.1/01	Lloyd JD	1998	Borates and their biological applications. Stockholm, Sweden: The International Research Group on Wood Preservation. Doc no. IRG/WP98-30178. Paper prepared for the 29th annual meeting , Maastricht, The Netherlands, June 1998 Non-GLP Published.	No	
A5.4.1/02	Lloyd JD and Dickinson DJ	1991	Comparison of the inhibitory effects of borate, germanate, tellurate, arsenite and arsenate on 6-phosphogluconate dehydrogenase. Stockholm, Sweden: The International Research Group on Wood Preservation. Doc no. IRG/WP1508 Paper prepared for the 22 nd annual meeting , Kyoto, Japan, May 1991. No GLP Published	No	
A5.4.1/03	Lloyd JD, Dickinson DJ and Murphy RJ	1990	The probable mechanisms of action of boric acid and borates as wood preservatives. Stockholm, Sweden: The International Research Group on Wood Preservation. Doc no. IRG/WP1450. Paper prepared for the 21 st annual meeting , Rotorua, New Zealand, May 1990 Non GLP Published	No	
A5.4.1/04	Lloyd JD, Dickinson DJ and Murphy RJ	1991	The effect of sorbitol on the decay of boric acid treated Scots pine. Stockholm, Sweden: The International Research Group on Wood Preservation. Doc no. IRG/WP1509. Paper prepared for the 22 nd annual meeting , Kyoto, Japan, May 1991 Non-GLP Published	No	
A6.1.1/01	Denton, S.	1996	Acute Toxicity. Final Report: Anhydrous Boric Acid. Acute Oral Toxicity Study in the Rat. Report number 1341/7-1032, Coming Hazleton (Europe), Harrogate, N Yorkshire, HG3 1PY, UK. (TX-096-003) GLP, Unpublished	Yes	BEU
A6.1.1/02	Denton, S.	1995	Acute Toxicity. Final Report: Anhydrous Boric Acid. Acute Oral Toxicity Study in the Rat. Report number 1341/3-1032, Coming Hazleton (Europe), Harrogate, N Yorkshire, HG3, 1PY, U.K. GLP, Unpublished (TX095-003)	Yes	BEU
A6.1.4/01	Cuperus, L., Mazur, P, Stubblefield, S, Opsahl, W., Robles, H.	1990	Acute Eye Irritation. Evaluation of the Ocular Irritation of Anhydrous Boric Acid in Rabbits. (PO1875). Biological Test Center, Irvine, CA 92714. (TX-90-001) GLP, Unpublished	Yes	BEU

Section No / Reference No	Author(s)	Year	Title. Source (where different from company) Company, Report No. GLP (where relevant) / (Un)Published Electronic File Name if Appropriate	Data Protection Claimed (Yes/No)	Owner
A6.1.4/02	Young, J.A., Briggs, B.G. and Doyle, R.I.	1973	Acute Dermal Irritation. Corrosivity study on a series of seven materials IBR-US Inc, International Bio-Research, Miami, Ohio, 45147, USA. (TX-73-29) GLP, Unpublished	Yes	BEU
A6.2-10/01	Farmer J	1982	Structural chemistry in the borate industry, Chem and Ind Non-GLP Published	No	
A6.2-10/02	Ingri, N	1963	Equilibrium studies of polyanions containing BIII, SiIV, GeIV, and VV. Sven. Kem. Tidskr. 75(4):199-230. Non-GLP Published	No	
A6.2-10/03	Kirk-Othmer Encyclopedia	1992	Kirk-Othmer Encyclopedia of Chemical Technology, John Wiley & Sons Inc., 4th Edition, Volume 4, pp 368-386. Non-GLP Published	No	
A6.2-10/04	Holleman AF and Wiberg E and Wiberg N	1995	Lehrbuch der anorganischen Chemie. 101st ed. De Gruyter, Berlin	No	
A6.2-10/05	De Vette HQM, Otto C, Schoonmade MA	2001	Hydrolysis as a function of pH and identification of breakdown products. A study on the identification and comparison of the dissociation products of Polybor tech, Borax Manufacturing grade and Boric acid manufacturing grade in aqueous solution using Raman spectroscopy. TNO report no v99-1153 (TX-2001-001)	Yes	BEU
A6.2-10/06	Maeda, M	1979	Raman spectra of polyborate ions in aqueous solution. J Inorg. Nucl. Chem. vol 41: 1217-1220 Non-GLP Published	No	
A6.2-10/07	Rickards, H and Arian, Z	2004	Rate of Dissolution Study of Borax's Products, Borax Europe Limited Internal Technical Report. TR-2004-01 No GLP Unpublished	Yes	BEU
A7.1.1.1.1/01	De Vette HQM, Otto C, Schoonmade MA	2001	Hydrolysis as a function of pH and identification of breakdown products. A study on the identification and comparison of the dissociation products of Polybor tech, Borax Manufacturing grade and Boric acid manufacturing grade in aqueous solution using Raman spectroscopy. TNO report no v99-1153 (TX-2001-001)	Yes	BEU
A7.1.1.1.1/02	Farmer J	1982	Structural chemistry in the borate industry, Chem and Ind Non-GLP Published	No	
A7.1.1.1.1/03	Holleman AF and Wiberg E and Wiberg N	1995	Lehrbuch der anorganischen Chemie. 101st ed. De Gruyter, Berlin	No	
A7.1.1.1.1/04	Kemp, P.H.	1956	"The Chemistry of Borates Part 1", copyright	No	

Section No / Reference No	Author(s)	Year	Title. Source (where different from company) Company, Report No. GLP (where relevant) / (Un)Published Electronic File Name if Appropriate	Data Protection Claimed (Yes/No)	Owner
A7.1.1.1.1/05	Maeda, M	1979	Raman spectra of polyborate ions in aqueous solution. J Inorg. Nucl. Chem. vol 41: 1217-1220 Non-GLP Published	No	
A7.1.1.1.1/06	Hanstveit A O	2004	Boric Acid (CAS No. 10043-35-3): Statement on Hydrolysis as a function of pH and identification of breakdown products TNO Report No V6004/05. Report to Borax Europe Limited Unpublished	Yes	BEU

Reference list of studies on the product

Section No / Reference No	Author(s)	Year	Title. Source (where different from company) Company, Report No. GLP (where relevant) / (Un)Published Electronic File Name if Appropriate	Data Protection Claimed (Yes/No)	Owner
B5.10/01	Cookson LJ and Pham K	1996	Relative tolerance of twenty basidiomycetes to boric acid. Mat und Org 29 (3) 187-196. Non-GLP Published	No	CSIRO
B7/01	De Vette HQM, Otto C, Schoonmade MA	2001	Hydrolysis as a function of pH and identification of breakdown products. A study on the identification and comparison of the dissociation products of Polybor tech, Borax Manufacturing grade and Boric acid manufacturing grade in aqueous solution using Raman spectroscopy. TNO report no v99-1153 (TX-2001-001)	Yes	BEU
B7/02	Kirk-Othmer Encyclopedia	1992	Kirk-Othmer Encyclopedia of Chemical Technology, John Wiley & Sons Inc., 4th Edition, Volume 4, pp 368-386. Non-GLP Published	No	

Reference list of literature and studies that were added to the dossier after submission of the summary dossier by the applicant

Section No / Reference No	Author(s)	Year	Title. Source (where different from company) Company, Report No. GLP (where relevant) / (Un)Published	Data Protection Claimed (Yes/No)	Owner
Doc IIIA	Abou-Shakra, F.R., Havercroft, J.K. and Ward	1989	Lithium and Boron in Biological Tissues and Fluids. Trace Elements in Medicine 6, 142-146.	No	
Doc IIA	Adelhardt, M. and Fogh, A.	1983	Enquiries to a centre for information on poisoning during a period of 12 years (English translation). Boraks – er de farligt (12 ars forespørgsler til giftinformationscentralen) Ugeskr. Laeger. 145 3808 – 3810.	No	
	Adriano, D.C., Kaplan, D.I., Burkman, W.G. and Mills, G.L.	1988	Long-term phytoavailability of soil applied organo-borates. J. Environ. Qual. 17, 485-492.	No	
	AE	1998	Development of plant toxicity tests for assessment of contaminated soils. Orton, Canada: Aquaterra Environmental.	No	
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Appendix IV: List of standard terms and abbreviations

Stand. term / Abbreviation	Explanation
A	ampere
ACh	acetylcholine
AChE	acetylcholinesterase
ADI	acceptable daily intake
ADME	administration distribution metabolism and excretion
ADP	adenosine diphosphate
AE	acid equivalent
AF	assessment factor
AFID	alkali flame-ionisation detector or detection
A/G	albumin/globulin ratio
ai	active ingredient
ALD ₅₀	approximate median lethal dose, 50%
ALT	alanine aminotransferase (SGPT)
<i>Ann.</i>	Annex
AOEL	acceptable operator exposure level
AMD	automatic multiple development
ANOVA	analysis of variance
AP	alkaline phosphatase
approx	approximate
ARC	anticipated residue contribution
ARfD	acute reference dose
as	active substance
AST	aspartate aminotransferase (SGOT)
ASV	air saturation value
ATP	adenosine triphosphate
BAF	bioaccumulation factor
BCF	bioconcentration factor
bfa	body fluid assay
BOD	biological oxygen demand
bp	boiling point
BPD	Biocidal Products Directive
BSAF	biota-sediment accumulation factor
BSE	bovine spongiform encephalopathy
BSP	bromosulphophthalein

Stand. term / Abbreviation	Explanation
Bt	<i>Bacillus thuringiensis</i>
Bti	<i>Bacillus thuringiensis israelensis</i>
Btk	<i>Bacillus thuringiensis kurstaki</i>
Btt	<i>Bacillus thuringiensis tenebrionis</i>
BUN	blood urea nitrogen
bw	body weight
c	centi- ($\times 10^{-2}$)
°C	degrees Celsius (centigrade)
CA	controlled atmosphere
CAD	computer aided design
CADDY	computer aided dossier and data supply (an electronic dossier interchange and archiving format)
cd	candela
CDA	controlled drop(let) application
cDNA	complementary DANN
CEC	cation exchange capacity
<i>cf</i>	confer, compare to
CFU	colony forming units
ChE	cholinesterase
CI	confidence interval
CL	confidence limits
cm	centimetre
CNS	central nervous system
COD	chemical oxygen demand
CPK	creatinine phosphatase
cv	coefficient of variation
Cv	ceiling value
d	day(s)
DES	diethylstilboestrol
DIS	draft international standard (<i>ISO</i>)
DMSO	dimethylsulfoxide
DNA	deoxyribonucleic acid
dna	designated national authority
DO	dissolved oxygen
DOC	dissolved organic carbon
dpi	days post inoculation

Stand. term / Abbreviation	Explanation
DRP	detailed review paper (<i>OECD</i>)
DT _{50(lab)}	period required for 50 percent dissipation (under laboratory conditions) (define method of estimation)
DT _{90(field)}	period required for 90 percent dissipation (under field conditions) (define method of estimation)
dw	dry weight
DWQG	drinking water quality guidelines
ε	decadic molar extinction coefficient
EC ₅₀	median effective concentration
ECD	electron capture detector
ED ₅₀	median effective dose
EDI	estimated daily intake
EINECS	European inventory of existing commercial substances
ELINCS	European list of notified chemical substances
ELISA	enzyme linked immunosorbent assay
e-mail	electronic mail
EMDI	estimated maximum daily intake
EN	European norm
EPMA	electron probe micro-analysis
ERL	extraneous residue limit
ESPE46/51	evaluation system for pesticides
EUSES	European Union system for the evaluation of substances
F	field
F ₀	parental generation
F ₁	filial generation, first
F ₂	filial generation, second
FBS	full base set
FELS	fish early-life stage
FIA	fluorescence immuno-assay
FID	flame ionisation detector
F _{mol}	fractional equivalent of the metabolite's molecular weight compared to the active substance

Stand. term / Abbreviation	Explanation
FOB	functional observation battery
f _{oc}	organic carbon factor (compartment dependent)
fp	freezing point
FPD	flame photometric detector
FPLC	fast protein liquid chromatography
g	gram(s)
GAP	good agricultural practice
GC	gas chromatography
GC-EC	gas chromatography with electron capture detector
GC-FID	gas chromatography with flame ionisation detector
GC-MS	gas chromatography-mass spectrometry
GC-MSD	gas chromatography with mass-selective detection
GEP	good experimental practice
GFP	good field practice
GGT	gamma glutamyl transferase
GI	gastro-intestinal
GIT	gastro-intestinal tract
GL	guideline level
GLC	gas liquid chromatography
GLP	good laboratory practice
GM	geometric mean
GMO	genetically modified organism
GMM	genetically modified micro-organism
GPC	gel-permeation chromatography
GPS	global positioning system
GSH	glutathione
GV	granulosevirus
h	hour(s)
H	Henry's Law constant (calculated as a unitless value)
ha	hectare(s)
Hb	haemoglobin
HC5	concentration which will be harmless to at least 95 % of the

Stand. term / Abbreviation	Explanation
	species present with a given level of confidence (usually 95 %)
HCG	human chorionic gonadotropin
Hct	haematocrit
HDT	highest dose tested
hL	hectolitre
HEED	high energy electron diffraction
HID	helium ionisation detector
HPAEC	high performance anion exchange chromatography
HPLC	high pressure liquid chromatography or high performance liquid chromatography
HPLC-MS	high pressure liquid chromatography - mass spectrometry
HPPLC	high pressure planar liquid chromatography
HPTLC	high performance thin layer chromatography
HRGC	high resolution gas chromatography
H _s	Shannon-Weaver index
Ht	haematocrit
HUSS	human and use safety standard
I	indoor
I ₅₀	inhibitory dose, 50%
IC ₅₀	median immobilisation concentration or median inhibitory concentration 1
ICM	integrated crop management
ID	ionisation detector
IEDI	international estimated daily intake
IGR	insect growth regulator
im	intramuscular
inh	inhalation
INT	2-p-iodophenyl-3-p-nitrophenyl-5-phenyltetrazoliumchloride testing method
ip	intraperitoneal
IPM	integrated pest management
IR	infrared

Stand. term / Abbreviation	Explanation
ISBN	international standard book number
ISSN	international standard serial number
IUCLID	International Uniform Chemical Information Database
iv	intravenous
IVF	<i>in vitro</i> fertilisation
k (<i>in combination</i>)	kilo
k	rate constant for biodegradation
K	Kelvin
K _a	acid dissociation constant
K _b	base dissociation constant
K _{ads}	adsorption constant
K _{des}	apparent desorption coefficient
K _F	Freundlich distribution constant, indicating the relationship between de amount sorbed and equilibrium concentration in in water.
kg	kilogram
K _H	Henry's Law constant (in atmosphere per cubic metre per mole)
K _{oc}	organic carbon adsorption coefficient
K _{om}	organic matter adsorption coefficient
K _{ow}	octanol-water partition coefficient
K _p	solid-water partition coefficient
kPa	kilopascal(s)
l, L	litre
LAN	local area network
LASER	light amplification by stimulated emission of radiation
LBC	loosely bound capacity
LC	liquid chromatography
LC-MS	liquid chromatography- mass spectrometry
LC ₅₀	lethal concentration, median
LCA	life cycle analysis
LC-MS-MS	liquid chromatography with tandem mass spectrometry

Stand. term / Abbreviation	Explanation
LD ₅₀	lethal dose, median; dosis letalis media
LDH	lactate dehydrogenase
ln	natural logarithm
LOAEC	lowest observable adverse effect concentration
LOAEL	lowest observable adverse effect level
LOD	limit of detection
LOEC	lowest observable effect concentration
LOEL	lowest observable effect level
log	logarithm to the base 10
LOQ	limit of quantification (determination)
LPLC	low pressure liquid chromatography
LSC	liquid scintillation counting or counter
LSD	least squared denominator multiple range test
LSS	liquid scintillation spectrometry
LT	lethal threshold
m	metre
M	molar
µm	micrometre (micron)
MAC	maximum allowable concentration
MAK	maximum allowable concentration
MC	moisture content
MCH	mean corpuscular haemoglobin
MCHC	mean corpuscular haemoglobin concentration
MCV	mean corpuscular volume
MDL	method detection limit
MFO	mixed function oxidase
µg	microgram
mg	milligram
MHC	moisture holding capacity
MIC	minimum inhibitory concentration
min	minute(s)

Stand. term / Abbreviation	Explanation
MKC	minimum killing concentration
mL	millilitre
MLT	median lethal time
MLD	minimum lethal dose
mm	millimetre
MMAD	mass median aerodynamic diameter
mo	month(s)
MOE	margin of exposure
mol	mole(s)
MOS	margin of safety
mp	melting point
MRE	maximum residue expected
MRL	maximum residue level or limit
mRNA	messenger ribonucleic acid
MS	mass spectrometry
MSDS	material safety data sheet
MTD	maximum tolerated dose
MT	material test
MW	molecular weight
n.a.	not applicable
n-	normal (defining isomeric configuration)
n	number of observations
NAEL	no adverse effect level
nd	not detected
NEDI	national estimated daily intake
NEL	no effect level
NERL	no effect residue level
ng	nanogram
nm	nanometre
NMR	nuclear magnetic resonance
no, n°	number
NOAEC	no observed adverse effect concentration
NOAEL	no observed adverse effect level
NOEC	no observed effect concentration
NOED	no observed effect dose

Stand. term / Abbreviation	Explanation
NOEL	no observed effect level
NOIS	notice of intent to suspend
NPD	nitrogen-phosphorus detector or detection
NPV	nuclear polyhedrosis virus
NR	not reported
NTE	neurotoxic target esterase
OC	organic carbon content
OCR	optical character recognition
ODP	ozone-depleting potential
ODS	ozone-depleting substances
OEL	occupational exposure limit
OH	hydroxide
OJ	Official Journal
OM	organic matter content
Pa	pascal
PAD	pulsed amperometric detection
2-PAM	2-pralidoxime
pc	paper chromatography
PC	personal computer
PCV	haematocrit (packed corpuscular volume)
PEC	predicted environmental concentration
PEC _A	predicted environmental concentration in air
PEC _S	predicted environmental concentration in soil
PEC _{SW}	predicted environmental concentration in surface water
PEC _{GW}	predicted environmental concentration in ground water
PED	plasma-emissions-detector
pH	pH-value
PHED	pesticide handler's exposure data
PIC	prior informed consent
pic	phage inhibitory capacity
PIXE	proton induced X-ray emission
pKa	negative logarithm (to the base 10)

Stand. term / Abbreviation	Explanation
	of the acid dissociation constant
pKb	negative logarithm (to the base 10) of the base dissociation constant
PNEC	predicted no effect concentration (compartment to be added as subscript)
po	by mouth
POP	persistent organic pollutants
ppb	parts per billion (10 ⁻⁹)
PPE	personal protective equipment
ppm	parts per million (10 ⁻⁶)
PPP	plant protection product
ppq	parts per quadrillion (10 ⁻²⁴)
ppt	parts per trillion (10 ⁻¹²)
PSP	phenolsulfophthalein
PrT	prothrombin time
PRL	practical residue limit
PT	product type
PT(CEN)	project team CEN
PTDI	provisional tolerable daily intake
PTT	partial thromboplastin time
QA	quality assurance
QAU	quality assurance unit
(Q)SAR	quantitative structure-activity relationship
r	correlation coefficient
r ²	coefficient of determination
RA	risk assessment
RBC	red blood cell
REI	restricted entry interval
RENI	Registry Nomenclature Information System
Rf	retardation factor
RfD	reference dose
RH	relative humidity
RL ₅₀	median residual lifetime
RNA	ribonucleic acid
RP	reversed phase

Stand. term / Abbreviation	Explanation
rpm	revolutions per minute
rRNA	ribosomal ribonucleic acid
RRT	relative retention time
RSD	relative standard deviation
s	second
S	solubility
SAC	strong adsorption capacity
SAP	serum alkaline phosphatase
SAR	structure/activity relationship
SBLC	shallow bed liquid chromatography
sc	subcutaneous
sce	sister chromatid exchange
SCAS	semi-continuous activated sludge
SCTER	smallest chronic toxicity exposure ratio (TER)
SD	standard deviation
se	standard error
SEM	standard error of the mean
SEP	standard evaluation procedure
SF	safety factor
SFC	supercritical fluid chromatography
SFE	supercritical fluid extraction
SIMS	secondary ion mass spectroscopy
S/L	short term to long term ratio
SMEs	small and medium sized enterprises
SOP	standard operating procedures
sp	species (only after a generic name)
SPE	solid phase extraction
SPF	specific pathogen free
spp	subspecies
SSD	sulphur specific detector
SSMS	spark source mass spectrometry
STEL	short term exposure limit
STER	smallest toxicity exposure ratio (TER)
STMR	supervised trials median residue
STP	sewage treatment plant

Stand. term / Abbreviation	Explanation
t	tonne(s) (metric ton)
t _½	half-life (define method of estimation)
T ₃	tri-iodothyroxine
T ₄	thyroxine
T ₂₅	tumorigenic dose that causes tumours in 25 % of the test animals
TADI	temporary acceptable daily intake
TBC	tightly bound capacity
TCD	thermal conductivity detector
TG	technical guideline, technical group
TGD	Technical guidance document
TID	thermionic detector, alkali flame detector
TDR	time domain reflectometry
TER	toxicity exposure ratio
TER _I	toxicity exposure ratio for initial exposure
TER _{ST}	toxicity exposure ratio following repeated exposure
TER _{LT}	toxicity exposure ratio following chronic exposure
tert	tertiary (in a chemical name)
TEP	typical end-use product
TGGE	temperature gradient gel electrophoresis
TIFF	tag image file format
TLC	thin layer chromatography
Tlm	median tolerance limit
TLV	threshold limit value
TMDI	theoretical maximum daily intake
TMRC	theoretical maximum residue contribution
TMRL	temporary maximum residue limit
TNsG	technical notes for guidance
TOC	total organic carbon
Tremcard	transport emergency card
tRNA	transfer ribonucleic acid
TSH	thyroid stimulating hormone (thyrotropin)

Stand. term / Abbreviation	Explanation
TTC	2,3,5-triphenylterazoliumchloride testing method
TWA	time weighted average
UDS	unscheduled DNA synthesis
UF	uncertainty factor (safety factor)
ULV	ultra low volume
UR	unit risk
UV	ultraviolet
UVC	unknown or variable composition, complex reaction products
UVCB	undefined or variable composition, complex reaction products in biological material
v/v	volume ratio (volume per volume)

Stand. term / Abbreviation	Explanation
vis	visible
WBC	white blood cell
wk	week
wt	weight
w/v	weight per volume
ww	wet weight
w/w	weight per weight
XRFA	X-ray fluorescence analysis
yr	year
<	less than
≤	less than or equal to
>	greater than
≥	greater than or equal to

Abbreviations of Organisations and Publications

Abbreviation	Explanation
ASTM	American Society for Testing and Materials
BA	Biological Abstracts (Philadelphia)
BART	Beneficial Arthropod Registration Testing Group
BBA	German Federal Agency of Agriculture and Forestry
CA(S)	Chemical Abstracts (System)
CAB	Centre for Agriculture and Biosciences International
CAC	Codex Alimentarius Commission
CAS	Chemical Abstracts Service
CCFAC	Codex Committee on Food Additives and Contaminants
CCGP	Codex Committee on General Principles
CCPR	Codex Committee on Pesticide Residues
CCRVDF	Codex Committee on Residues of Veterinary Drugs in Food
CE	Council of Europe

Abbreviation	Explanation
CEC	Commission of the European Communities
CEFIC	European Chemical Industry Council
CEN	European Committee for Normalisation
CEPE	European Committee for Paints and Inks
CIPAC	Collaborative International Pesticides Analytical Council Ltd
CMA	Chemicals Manufacturers Association
COREPER	Comite des Representants Permanents
COST	European Co-operation in the field of Scientific and Technical Research
DG	Directorate General
DIN	German Institute for Standardisation
EC	European Commission
ECB	European Chemicals Bureau
ECCO	European Commission Co-ordination

Abbreviation	Explanation
ECDIN	Environmental Chemicals Data and Information Network of the European Communities
ECDIS	European Environmental Chemicals Data and Information System
ECE	Economic Commission for Europe
ECETOC	European Chemical Industry Ecology and Toxicology Centre
EDEXIM	European Database on Export and Import of Dangerous Chemicals
EEC	European Economic Community
EHC	Environmental Health Criteria
EINECS	European Inventory of Existing Commercial Chemical Substances
ELINCS	European List of New Chemical Substances
EMIC	Environmental Mutagens Information Centre
EPA	Environmental Protection Agency
EPAS	European Producers of Antimicrobial Substances
EPFP	European Producers of Formulated Preservatives
EPO	European Patent Office
EPPO	European and Mediterranean Plant Protection Organization
ESCORT	European Standard Characteristics of Beneficials Regulatory Testing
EU	European Union
EUPHIDS	European Pesticide Hazard Information and Decision Support System
EUROPOEM	European Predictive Operator Exposure Model
EWMP	European Wood Preservation Manufacturers
FAO	Food and Agriculture Organization of the UN
FOCUS	Forum for the Co-ordination of Pesticide Fate Models and their Use
FRAC	Fungicide Resistance Action Committee
GATT	General Agreement on Tariffs and Trade
GAW	Global Atmosphere Watch
GIFAP	Groupement International des

Abbreviation	Explanation
	Associations Nationales de Fabricants de Produits Agrochimiques (now known as GCPF)
GCOS	Global Climate Observing System
GCPF	Global Crop Protection Federation (formerly known as GIFAP)
GEDD	Global Environmental Data Directory
GEMS	Global Environmental Monitoring System
GRIN	Germplasm Resources Information Network
IARC	International Agency for Research on Cancer
IATS	International Academy of Toxicological Science
ICBP	International Council for Bird Preservation
ICCA	International Council of Chemical Associations
ICES	International Council for the Exploration of the Seas
ILO	International Labour Organization
IMO	International Maritime Organisation
IOBC	International Organization for Biological Control of Noxious Animals and Plants
IPCS	International Programme on Chemical Safety
IRAC	Insecticide Resistance Action Committee
ISCO	International Soil Conservation Organization
ISO	International Organization for Standardisation
IUPAC	International Union of Pure and Applied Chemistry
JECFA FAO/WHO	Joint Expert Committee on Food Additives
JFCMP	Joint FAO/WHO Food and Animal Feed Contamination Monitoring Programme
JMP	Joint Meeting on Pesticides (WHO/FAO)
JMPR	Joint Meeting of the FAO Panel of Experts on Pesticide Residues in Food and the Environment and the

Abbreviation	Explanation
	WHO Expert Group on Pesticide Residues (Joint Meeting on Pesticide Residues)
MITI	Ministry of International Trade and Industry, Japan
NATO	North Atlantic Treaty Organization
NAFTA	North American Free Trade Agreement
NCI	National Cancer Institute (USA)
NCTR	National Center for Toxicological Research (USA)
NGO	non-governmental organisation
NTP	National Toxicology Program (USA)
OECD	Organization for Economic Co-operation and Development
OLIS	On-line Information Service of OECD
OPPTS	Office of Prevention, Pesticides and Toxic Substances (US EPA)
OSPAR	Oslo Paris Convention (Convention for the Protection of the Marine Environment of the North-East Atlantic)
PAN	Pesticide Action Network
RIVM	Netherlands National Institute of Public Health and Environmental Protection
RNN	Re-registration Notification Network
RTECS	Registry of Toxic Effects of Chemical Substances (USA)
SETAC	Society of Environmental Toxicology and Chemistry
SI	Système International d'Unités
SITC	Standard International Trade Classification
TOXLINE	Toxicology Information On-line
UBA	German Environmental Protection Agency
UN	United Nations
UNEP	United Nations Environment Programme
WFP	World Food Programme
WHO	World Health Organization
WPRS	West Palearctic Regional Section

Abbreviation	Explanation
WTO	World Trade Organization
WWF	World Wildlife Fund