

Guidance on Information Requirements and Chemical Safety Assessment

Chapter R.7b: Endpoint specific guidance

Draft Version 4.0

March 2017



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19 **Guidance on Information Requirements and Chemical Safety Assessment** 20 **Chapter R.7b: Endpoint specific guidance**

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1 Preface

2 This document describes the information requirements under the REACH Regulation with
3 regard to substance properties, exposure, uses and risk management measures, and the
4 chemical safety assessment. It is part of a series of guidance documents that aims to
5 help all stakeholders with their preparation for fulfilling their obligations under the
6 REACH Regulation. These documents cover detailed guidance for a range of essential
7 REACH processes as well as for some specific scientific and/or technical methods that
8 industry or authorities need to make use of under the REACH Regulation.

9 The original versions of the guidance documents were drafted and discussed within the
10 REACH Implementation Projects (RIPs) led by the European Commission services,
11 involving stakeholders from Member States, industry and non-governmental
12 organisations. After acceptance by the Member States competent authorities the
13 guidance documents had been handed over to ECHA for publication and further
14 maintenance. Any updates of the guidance are drafted by ECHA and are then subject to
15 a consultation procedure, involving stakeholders from Member States, industry and non-
16 governmental organisations. For details of the consultation procedure, please see the
17 "Second revision to the Consultation Procedure for Guidance" at:

18 [http://echa.europa.eu/documents/10162/13608/mb_63_2013_revision_consultation_pr](http://echa.europa.eu/documents/10162/13608/mb_63_2013_revision_consultation_procedure_guidance_en.pdf)
19 [ocedure_guidance_en.pdf](http://echa.europa.eu/documents/10162/13608/mb_63_2013_revision_consultation_procedure_guidance_en.pdf)

20

21 The guidance documents can be obtained via the website of the European Chemicals
22 Agency at:

23 <http://echa.europa.eu/web/guest/guidance-documents/guidance-on-reach>

24

25 This document relates to the REACH Regulation (EC) No 1907/2006 of the European
26 Parliament and of the Council of 18 December 2006¹.

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¹ Regulation (EC) No 1907/2006 of the European Parliament and of the Council of 18 December 2006 concerning the Registration, Evaluation, Authorisation and Restriction of Chemicals (REACH), establishing a European Chemicals Agency, amending Directive 1999/45/EC and repealing Council Regulation (EEC) No 793/93 and Commission Regulation (EC) No 1488/94 as well as Council Directive 76/769/EEC and Commission Directives 91/155/EEC, 93/67/EEC, 93/105/EC and 2000/21/EC (OJ L 396, 30.12.2006, p.1; corrected by OJ L 136, 29.5.2007, p.3).

1 Document History

Version	Changes	Date
Version 1	First edition	May 2008
Version 1.1	Re-introduction of lost pieces of Appendix 7.8-5 "Assessment of available information on endocrine and other related effects"	August 2008
Version 1.2	Corrigendum: (i) replacing references to DSD/DPD by references to CLP; (ii) further minor editorial changes/corrections.	November 2012
Version 2.0	Second edition. Partial revision of the document to take into account the revised version (2.0) of Chapter R.11 of the Guidance on IR&CSA following amendment of Annex XIII to REACH (according to Commission Regulation (EU) No 253/2011 of 15 March 2011, OJ L 69 7 16.3.2011). Main changes in the guidance document included the following: <ul style="list-style-type: none"> • References to the updated Chapter R.11 were added and the corresponding text updated; • The repeated Figure R.7.8-1 was deleted; • Errors in the numbering of Figures, Tables, and Appendices were corrected. In particular: former Figure R.7.8-8 was relabelled Table R.7.8-4; former Figures R.7.8-9 and R.7.8-10 were changed to Figures R.7.8-8 and R.7.8-9, respectively; former Tables R.7.8-4 and R.7.8-5 were changed to Figures R.7.8-5 and R.7.8-6, respectively; former Appendices R.7.8-4 and R.7.8-5 were changed to Appendices R.7.8-3 and R.7.8-4, respectively; corresponding cross-references were updated; • Some erroneous cross-references were corrected; • The document was re-formatted to the updated ECHA corporate identity. 	November 2014
Version 3.0	Update of the guidance covering only text concerning the sediment compartment. In particular the main changes include: <ul style="list-style-type: none"> • Addition of indication of possible 	February 2016

triggers for sediment assessment not Kow/Koc driven.

- Enhanced relevance of long term studies over short term studies.
- Addition in chapter 7.8.9.1 of reference and description of the most relevant OECD, ASTM, US EPA and ISO standards following latest developments; More details on reporting needs for non-standard methods; More information on species selection and exposure pathways; Further clarification on the equilibrium partitioning method.
- Further clarifications in chapter 7.8.10.1 about species and organisms selection in the evaluation of information; reorganisation of text on composition of test sediment and further clarifications about pros and cons of artificial Vs natural; addition of considerations on effect of aging in tests; addition of reference to use of Passive Sampling Devices in test design section.
- Further elaboration in chapter 7.8.10.2 of the role of monitoring and field exposure data and their usability.
- Further elaboration in chapter 7.8.10.3 of consideration on bioavailability for organic substances.
- Addition of a new chapter 7.8.11 on species sensitivity distribution and its role in assessment of sediment toxicity; addition of reference to EFSA Opinion 2015.
- Further development of chapter 7.8.12 on uncertainties.
- Update of Figure 7.8-8 by merging boxes when $RCR > 1$.
- Clarification in chapter 7.8.14.2 that an additional factor of 10 may need to be applied to RCR for substances with adsorption/binding behaviour not triggered by lipophilicity.
- Update of table 7.8-6 on the most common benthic test species to cover OECD, ISO, US EPA, ASTM and OSPAR standard; addition of column with

	<p>relevant tests for each species.</p> <p>Additionally some further erroneous cross-references have been corrected throughout the document.</p>	
Version 4.0	<p>Partial revision of the document with respect to PBT/vPvB aspects to take into account the updated version of Chapter R.11 (v 3.0). Main changes in the guidance document include the following:</p> <ul style="list-style-type: none">• Update to Section R.7.9.3.1: XXX• Update to Section R.7.9.4.1: XXX• Update to Section R.7.9.5.2: XXX	XXX 201X

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1 Convention for citing the REACH and the CLP Regulations

2 Where the REACH and the CLP Regulations are cited literally, this is indicated by text in
3 italics between quotes.

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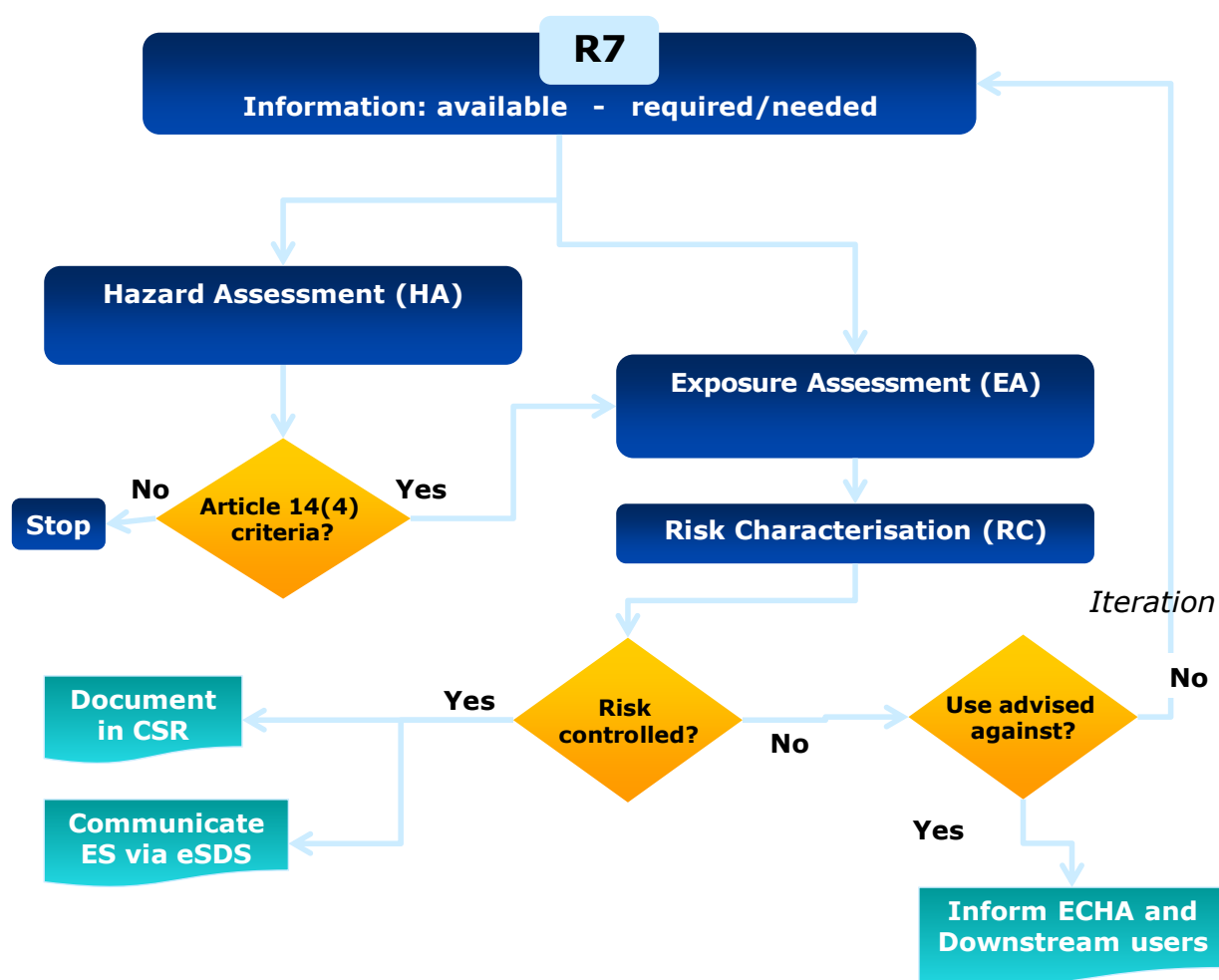
5 Table of Terms and Abbreviations

6 See Chapter R.20.

7

8 Pathfinder

9 The figure below indicates the location of part R.7(b) within the Guidance Document:



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2 **R.7.8 Aquatic toxicity; long-term toxicity to sediment organisms**

3 **R.7.8.1 Introduction to Aquatic pelagic toxicity**

4 Information on aquatic toxicity is used to assess hazard and risk to freshwater and
5 marine organisms living in the water column. In addition, the data obtained from testing
6 on freshwater species may also serve as basis for assessment of effects in marine
7 environment as well as for extrapolation of the measured effects to other compartments
8 within the aquatic ecosystem (e.g. sediment) and soil.

9 Related endpoints are (i) mammalian long-term/reproductive toxicity, where information
10 on endocrine activity obtained in toxicological studies may also be relevant for fish and
11 (ii) degradation, where information on possible (fast) primary degradation would lead to
12 inclusion of metabolites in hazard assessment of the parent compound.

13 **R.7.8.1.1 Definition of aquatic pelagic toxicity**

14 Aquatic toxicity refers to intrinsic property of a substance to be detrimental to an
15 organism in short-term and/or long-term exposure to that substance.

16 In general, it is assumed that the aquatic toxicity is mainly related to the waterborne
17 exposure of a substance and expressed as external concentration of that substance in
18 test water. There may be cases where food uptake is the predominant route of exposure
19 (i.e. for lipophilic substances). These effects are measured by employment of dietary
20 studies.

21 Some attempts have been made to relate toxic effects to internal concentration of
22 substances in the exposed organisms, e.g. by using body burden approach. This
23 approach has to be further developed and verified/validated before its application for
24 regulatory purposes (for details see [Appendix R.7.8–3](#)).

25

26 **Acute toxicity** related to waterborne exposure is generally expressed in terms of a
27 concentration which is lethal to 50% of the test organisms (lethal concentration, LC₅₀),
28 causes a measurable adverse effect to 50% of the test organisms (e.g. immobilization of
29 daphnids), or leads to a 50% reduction in test (treated) organism responses from control
30 (untreated) organism responses (e.g. growth rate in algae) following an exposure in the
31 range of hours to days, expressed as effective concentration, EC₅₀.

32 **Chronic toxicity** related to waterborne exposure refers to the potential or actual
33 properties of a substance to cause adverse effects to aquatic organisms during exposures
34 which are determined in relation to the life-cycle of the organism. Such chronic effects
35 usually include a range of sublethal endpoints and are generally expressed in terms of
36 NOEC (No Observed Effect Concentration), LOEC (Lowest Observed Effect
37 Concentration), EC_x or MATC (Maximal Acceptable Toxicant Concentration). Further
38 guidance on these terms is given in Chapter R.10.

1 Observable endpoints in chronic studies typically include survival, growth and/or
2 reproduction. Chronic toxicity exposure durations can vary widely depending on test
3 endpoint measured and test species used.

4 Although data from standard toxicity tests (internationally harmonised test guidelines)
5 are preferred, adverse effects in the water environment may also be predicted from
6 other information sources.

7 **R.7.8.1.2 Objective of the guidance on aquatic pelagic toxicity**

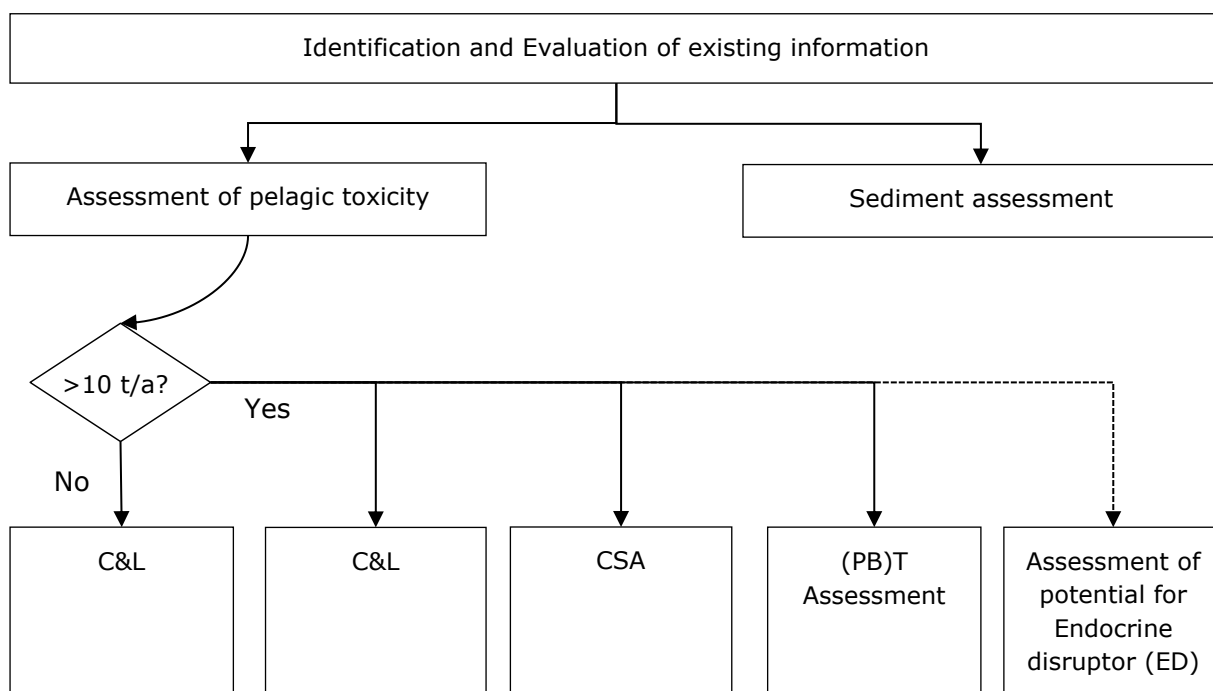
8 The main objective is to provide guidance to registrants on aquatic pelagic toxicity
9 testing and to develop an Integrated Testing Strategy (ITS) for aquatic toxicity aiming at
10 gathering data and information on substances to enable the environmental hazard
11 assessment, i.e. for use in classification and labelling and derivation of the PNEC_{water}
12 (Predicted No Effect Concentration for water) and for determination of the toxicity (T)
13 criterion in the PBT assessment. The PNEC_{water} is compared with the Predicted
14 Environmental Concentration in water (PEC_{water}) to decide whether there is a risk or not
15 to pelagic organisms from the exposure to the substance.

16 Depending on the intrinsic properties of the substance and available exposure
17 information, examination of additional possible adverse effects relevant for the aquatic
18 ecosystem could be necessary:

- 19 • Substances that are potentially capable of depositing on or sorbing to
20 sediments to a significant extent have to be assessed for *toxicity to sediment-*
21 *dwelling organisms*. In addition, marine sediment effects assessment is
22 necessary for substances that are known to be persistent in marine waters
23 and may accumulate in sediments over time. Guidance for the assessment of
24 toxic effects on sediment organisms is provided in Section [R.7.8.7](#).
- 25 • In addition, if, in the course of evaluation of available information, it is
26 confirmed or indicated that a substance displays an *endocrine mode of action*
27 in aquatic organisms, this may constitute a concern that requires further
28 investigation regarding potential adverse effects on development or
29 reproduction. If a clear link between serious adverse effects and an endocrine
30 mode of action can be established, the substance may fall under the
31 provisions of Article 57(f), which specifies that *substances - such as those*
32 *having endocrine disrupting properties (...) - for which there is scientific*
33 *evidence of probable serious effects to human health or the environment*
34 *which give rise to an equivalent level of concern* to those of CMR, PBT or vPvB
35 substances may be included in Annex XIV of substances subject to the
36 authorisation procedure. The inclusion will be decided on a case-by-case basis
37 following the preparation of an Annex XV dossier by the Competent
38 Authorities. As this kind of information is not part of the standard information
39 requirements set out in REACH Annexes VII-X (see below), this part of the
40 guidance is based on the evaluation of available information. Guidance for the
41 evaluation of available information on endocrine activity is provided in
42 [Appendix R.7.8—4](#).

1 [Figure R.7.8—1](#) summarises the general regulatory steps that are relevant for aquatic
 2 toxicity. It starts with the evaluation of existing information and, based on this
 3 information a conclusion whether evaluation of waterborne exposure is sufficient or
 4 evaluation of toxicity to sediment dwelling organisms should be included. As a second
 5 step in the hazard assessment has to be performed the classification and labelling (C&L)
 6 (for substances manufactured/imported at less than 10 tonnes per year and more than
 7 10 tonnes per year) and the determination of the $PNEC_{water}$ in the frame of the Chemical
 8 Safety Assessment (CSA) (for substances manufactures/imported at ≥ 10 t/y) as well as
 9 for PBT assessment. Guidance for gathering of and evaluation of information for these
 10 steps is provided in this document. The guidance for the evaluation of sediment toxicity
 11 is provided in a separate document. If, based on available information, a substance is
 12 suspected to exhibit endocrine activity, it might be necessary to assess the endocrine
 13 disruption potential of the substance. Guidance for this step is provided in Section
 14 [R.7.8.13](#) of this document.

15 **Figure R.7.8—1 Regulatory steps relevant for aquatic toxicity**



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19 **R.7.8.2 Information requirements for aquatic pelagic toxicity**

20 As described in **Annex VI to REACH** all available existing information should be
 21 collected and considered in the hazard assessment, regardless whether testing for a
 22 given endpoint is required or not at a specific tonnage level. Minimum information
 23 requirements are set out in Annex VII- X. If information required in Annex VII- X is not
 24 available, testing is required unless modification according to general rules described in
 25 Annex XI is possible. If the test needed (regarding ecotoxicological information)
 26 concerns Annex IX or X a testing proposal has to be prepared and submitted to the

1 Agency. Further information on general rules described in Annex XI is provided in
2 Chapter R.5 and Section R.7.8.4.1. The following paragraphs summarise requirements
3 according to Annex VII–X.

4 For substances covered by **Annex VII to REACH** short-term toxicity testing on
5 invertebrates (preferably *Daphnia*) and growth inhibition study on aquatic plants
6 (preferably algae) are required. However, these short-term studies do not need to be
7 conducted if there are mitigating factors indicating that aquatic toxicity is unlikely to
8 occur (e.g. the substance is highly insoluble in water or the substance is unlikely to cross
9 biological membranes).

10 In addition, the short-term testing on invertebrates does not need to be conducted if a
11 long-term aquatic toxicity study on invertebrates is available or if adequate information
12 on environmental classification and labelling is available.

13 If the substance is poorly water soluble the long-term toxicity testing (according to
14 Annex IX to REACH) must be considered (For more detailed description of potentially
15 mitigating factors see [Appendix R.7.8–1](#), for interpretation Section [R.7.8.5](#)).

16 For substances covered by **Annex VIII to REACH** short-term toxicity testing on fish is
17 additionally required. In analogy to the tests required on Annex VII to REACH, this test
18 does not need to be conducted if there are mitigating factors indicating that aquatic
19 toxicity is unlikely to occur (e.g. the substance is highly insoluble in water or the
20 substance is unlikely to cross biological membranes).

21 However, if the chemical safety assessment according to Annex I indicates the need to
22 investigate further effects on aquatic organisms, long-term testing as described in Annex
23 IX to REACH must be considered. Long-term testing should also be considered if the
24 substance is poorly water soluble. For explanation and interpretation see Section
25 [R.7.8.4.3](#) on exposure considerations.

26 For substances covered by **Annex IX to REACH** long-term toxicity testing on
27 invertebrates (preferably *Daphnia*) and fish is required, if the chemical safety
28 assessment according to Annex I to REACH indicates the need to investigate further the
29 effects on aquatic organisms. Examples of cases triggering further testing are presented
30 in Section [R.7.8.4.3](#) on exposure considerations.

31 In case of the long-term toxicity testing on fish, information on one of the following
32 studies must be provided: (for explanation see Section [R.7.8.5](#) on suitability of data on
33 CSA).

- 34 • Fish Early Life Stage (FELS) toxicity test (OECD TG 210): the revised OECD
35 TG 210 should be regarded as the most suitable test guideline for addressing
36 the information requirements related to fish long-term testing under REACH.
- 37 • Fish, juvenile growth test (OECD TG 215): this test can be
38 accepted/recommended, on a case-by-case basis, if there are well founded
39 justifications indicating that growth inhibition is the most relevant effect in fish
40 for the assessed substance.

41 It should be noted that the OECD TG 210 does not cover reproductive endpoints and
42 therefore, other OECD TGs should be considered for endocrine disrupting chemicals or

1 when other effects not covered by early fish development are expected to be of
2 particular relevance.

3 For substances covered by **Annex X to REACH** there are no additional information
4 requirements for pelagic aquatic toxicity.

5 As stated above the data are generated for environmental hazard assessment of
6 substances (i.e. classification, derivation of PNEC) and (PB)T assessment (see Section
7 [R.7.8.5](#) on conclusion on the endpoint).

8 It should be noted that if the registrant cannot derive a definitive conclusion (i) ("The
9 substance does not fulfil the PBT and vPvB criteria") or (ii) ("The substance fulfils the
10 PBT or vPvB criteria") in the PBT/vPvB assessment using the relevant available
11 information, he must, based on section 2.1 of Annex XIII to REACH, generate the
12 necessary information for deriving one of these conclusions, regardless of his tonnage
13 band (for further details, see Chapter R.11 of the *Guidance on Information Requirement
14 and Chemical Safety Assessment (IR&CSA)*). In such a case, the only possibility to
15 refrain from testing or generating other necessary information is to treat the substance
16 "as if it is a PBT or vPvB" (see Chapter R.11 for details).

17 **R.7.8.3 Information on aquatic pelagic toxicity and its sources**

18 Below different types of information relevant for assessing aquatic toxicity are
19 presented. This includes available testing (*in vitro* and *in vivo*) and non-testing methods
20 ((Q)SAR, read-across and categories) that generate information on aquatic toxicity
21 relevant for regulatory purposes.

22 **R.7.8.3.1 Data on aquatic pelagic toxicity**

23 Testing data on aquatic pelagic toxicity

24 ***In Vitro* Data**

25 At present, there are no EU / OECD guidelines for *in vitro* tests of relevance to aquatic
26 toxicity.

27 There are ongoing efforts to develop and validate *in vitro* methods, which in future might
28 be useful in a testing strategy for acute aquatic toxicity (e.g. ECVAM study on
29 optimisation of cytotoxicity tests and CEFIC LRi study ECO 8 aiming to replacing the
30 acute fish toxicity test using fish cell lines and fish embryos).

31 The use of fish cells in environmental toxicology was reviewed at the ECVAM workshop
32 (Castano *et al.*, 2003, ECVAM workshop report 47) and ECETOC (2005).

33 **Primary cells:** Primary cells are freshly isolated cells from various tissues: liver, gill
34 epithelia, gonads, kidney macrophages, skin epithelia, endocrine tissues, muscle cells
35 and white blood cells. Primary cells require the use of living animals. They express many
36 of the differentiated cellular structures and functions of their source tissue and are
37 particularly suitable for mechanistically oriented studies on cell-specific toxicant fate and
38 action.

1 **Fish cell lines:** More than 150 permanent fish cell lines are available, most of them are
2 fibroblast or epithelia-like and derive from tissue of salmonids and cyprinids. Most of the
3 tests with permanent cell lines (monolayers or suspension cultures) measure the basal
4 cytotoxic effects of chemical substances.

5 Results from *in vitro* studies based on mammalian systems may be of interest for the
6 assessment of endocrine activity (see Section [R.7.8.13](#)).

7 ***In vivo* data (single species)**

8 Information on aquatic toxicity may be acquired from studies performed according to
9 existing national and international guidelines as well as from scientific literature, where
10 different aspects of aquatic toxicity are examined. The available guidelines are focused
11 on measuring of adverse effects of substances due to waterborne exposure. Since there
12 are no internationally harmonised guidelines for feeding studies in pelagic species, tests
13 employed in assessment of oral exposure are designed on case-by-case basis.

14 In general, the majority of the test guidelines for pelagic system are exclusively
15 developed for testing of either freshwater or saltwater species. There are, however,
16 guidelines providing procedures that are suitable for testing of species from both water
17 systems (see Tables in [Appendix R.7.8–2](#)).

18 EU/OECD Test guidelines

19 The EU/OECD test guidelines comprise internationally agreed testing methods for
20 environmental effects. Tests undertaken using these guidelines are useful for both risk
21 assessment and classification purposes. Data obtained from a test carried out in
22 accordance with an OECD test guideline are covered by the principle of mutual
23 acceptance of data (MAD), thereby reducing the number of tests that needs to be
24 conducted saving both animals and money.

25 There are a number of the tests guidelines available. They provide information on short-
26 term and long-term toxicity to aquatic species (both freshwater and marine) due to
27 waterborne exposure. Several new test methods, including potential alternative methods
28 to vertebrate animal testing, are currently under development and validation. Both the
29 available tests guidelines and these under development are presented in Section
30 [Appendix R.7.8–2](#).

31 The information requirements of REACH are, in principle, met by studies carried out
32 according to the currently adopted OECD test guidelines. However, if required by further
33 evaluation, additional (more adequate) tests (e.g. on organisms not included in OECD
34 test guidelines) may be selected from the lists of guidelines developed by other
35 regulatory bodies (see Section [Appendix R.7.8–2](#)²).

36 Other test guidelines

² Following development in the field of eco-toxicology new test guidelines are developed and available test methods undergo changes. Their procedures may be revised or some of the guidelines may even be exchanged by other, better tests. Therefore every table that aims at compiling all available test guidelines will soon become obsolete. The table in [Appendix R.7.8–2](#) gives the status from 1998 (OECD 1998). Therefore, the user is advised to consult the organisation that has issued the selected guidelines for its current status (addresses to the organisations are also presented in [Appendix R.7.8–2](#)).

1 Acceptable alternatives to the OECD test guidelines are published by the OPPTS, US-EPA,
2 various EU countries (national standard methods) and organisations such as ASTM, ISO
3 (for detailed list of available guidelines see [Appendix R.7.9–1](#)).

4 Non-guideline studies

5 In addition to results from guideline studies, also results from non-guideline non-GLP
6 studies may be available. The studies may vary in duration, endpoints measured;
7 species exposed etc. compared to the standard test guidelines. Despite the variability in
8 the test performance the results may be useful for hazard assessment (e.g. direct in
9 calculation of PNEC or indirect in application of *Weight of Evidence*). However, these data
10 should be particularly assessed for their adequacy (reliability and relevance) and
11 completeness (for details see Section [R.7.8.4.1](#) on criteria for the evaluation of *in vivo*
12 testing data).

13 Information sources

14 Data from different tests measuring toxicity to aquatic species (results from tests
15 performed according to the test guidelines and to non-standard procedures) may be
16 gathered in different databases. Not all databases routinely make a quality check of the
17 data before their inclusion in the database. Unless the data quality is known user is
18 recommended to consult original scientific paper where these data were derived. Aquatic
19 toxicity data may also be reviewed in scientific reports. References to these databases
20 and documents are presented in [Appendix R.7.8–2](#).

21 ***In vivo* – multiple species (field data)**

22 Experimental ecosystem studies are aiming at understanding both fate and effects at
23 higher tiers of ecological integration. The design of any study is dependent on the
24 objectives and includes:

- 25 • to gain more knowledge about ecosystem structure and function (and thus
26 help to develop better ecosystem models);
- 27 • to develop and validate predictive models for chemical effect; with enough
28 information about the chemical fate in the particular experimental ecosystem
29 to be able to define NOECs, ECx or effect levels at different loading rates;
- 30 • to evaluate environmental quality standards derived from laboratory toxicity
31 data through extrapolation (improvement and refinement of extrapolation
32 models);
- 33 • to study the resilience of ecosystems in terms of time required for restoration
34 after chemical disturbance; and,
- 35 • to obtain data required for regulatory purposes of assessing fate and/or
36 effects in natural ecosystems (Crossland *et al.*, 1992).

37 Because different objectives exist for conducting model ecosystem tests, not all test
38 results may be equally useful, especially with respect to regulatory purposes.

39 Numerous expert meetings concerning the development and design of experimental
40 ecosystem studies involving all stakeholders have been held over the past 20 years. An

1 OECD guidance for the conduct of simulated freshwater lentic (standing water) tests in
2 the form of outdoor microcosms and mesocosms is available (OECD 2006a).

3 The choice of endpoints to measure during an experimental ecosystem study should not
4 be exhaustive and preferably targeted based on knowledge developed from lower tiers of
5 fate and effects assessment.

6 However, because experimental ecosystems offer the advantage of addressing ecological
7 properties that cannot be considered in lower tiers (and inherently addressed in
8 subsequent PNEC extrapolation), such as species diversity, trophic structure, species
9 interactions and so on, these may be useful to consider when designing, conducting and
10 interpreting a study (OECD 2006a).

11 Non-testing data on aquatic pelagic toxicity

12 A general guidance on the use of (Q)SAR results and chemical grouping approaches is
13 given in Sections R.6.1 and R.6.2 in Chapter R.6 of the [Guidance on IR&CSA](#). The
14 following section provides an overview of different information sources for (Q)SAR
15 predictions and grouping approaches specific for the assessment of aquatic toxicity.
16 Additional, more generic sources of information are summarised in Chapter R.4 of the
17 [Guidance on IR&CSA](#). Guidance for the evaluation of the results of these approaches is
18 provided in Section [R.7.8.4.1](#).

19 (Q)SAR

20 General guidance on QSAR is given in Section R.6.1 in Chapter R.6 of the [Guidance on](#)
21 [IR&CSA](#) and a more specific guidance on QSAR for estimating toxicity to the environment
22 is given in [Chapter R.10](#).

23 Available (Q)SAR methods can be summarised using the following categories:

- 24 • Schemes for the prediction of the mode of action/structural class of a
25 compound (baseline toxicity, excess toxicity)
- 26 • Qualitative information from structural alerts
- 27 • QSARs predictions from individual models (e.g. narcosis, other modes of
28 action, QICARs and QCARs for metals and inorganic metal compounds)
- 29 • QSARs predictions from expert systems
- 30 • Databases of (Q)SAR predictions
- 31 • Activity-activity relationships (QAARs) predictions

32 Grouping approaches

33 General guidance on grouping approaches is given in Section R.6.2 and a more specific
34 guidance on QSAR for estimating toxicity to the environment is given in [Chapter R.10](#).

35

1 **R.7.8.4 Evaluation of available information on aquatic pelagic toxicity**

2 Below criteria for evaluation of the gathered information are presented. Integration of
3 the gathered information should lead to an understanding of the toxic profile of the
4 substance, its potential exposure routes, its mechanism of action and its potential for
5 distribution in the environment.

6 Toxic effects of substances in the aquatic environment are among others related to (i)
7 intrinsic physical and chemical properties of substances and (ii) physical and chemical
8 properties of the aquatic (tests) systems. These two information have to be taken into
9 account when evaluating the available information on aquatic pelagic toxicity.

10 **Properties of substances and of test systems**

11 For most organic chemicals uptake from water is believed to be the predominant route of
12 uptake (for very hydrophobic or very sorptive substances does uptake from food become
13 important). It is believed that substances dissolved in water and taken up by organisms
14 may accumulate to a certain internal concentration, which may then cause adverse
15 effects. Therefore factors that influence bioconcentration influence also toxicity to
16 aquatic species. Molecular weight, water solubility and log K_{ow} of substances are such
17 factors. They are described in detail in [Appendix R.7.8–1](#). In addition other substance
18 related factors like degradation are described in this chapter.

19 In the context of toxicity, properties of aquatic (test) systems may or may not create
20 optimal conditions for recording possible adverse effects. Therefore they are important
21 quality parameters to be taken into account while evaluating toxicity studies. The water
22 quality parameters that influence toxicity testing are also described in [Appendix R.7.8–](#)
23 [1](#).

24 For metals and inorganic metal compounds exposure through the water is also the
25 predominant route. For many metals bioavailability and detoxification mechanisms is
26 known to modulate both accumulation and toxicity (McGeer *et al.*, 2002).

27 The criteria for evaluation of information on the physico-chemical properties of
28 substances are provided in Section R.7.1 in Chapter R.7a of the [Guidance on IR&CSA](#).
29 Furthermore consideration should be given to whether the substance being assessed can
30 be degraded, biotically or abiotically, to give stable and/or toxic degradation products.
31 Where such degradation can occur, the assessment should give due consideration to the
32 properties (including toxic effects) of the products that might arise.

33 **Other considerations**

34 Information on exposure must also be taken into account when deciding on the aquatic
35 pelagic tests to perform. Before their use the exposure data should be validated in
36 respect to their representativeness, completeness, relevance and reliability.

37 For existing data evaluation it is common that the full study information will not be
38 available to fully assess in detail all of the considerations above. The study may be of
39 good quality, however, and the study result can still be considered for use as part of a
40 *Weight of Evidence*. Under these circumstances, key information should be available to
41 give some confidence that the underlying data are of good quality. Where such
42 circumstances exist it is critical to know that the test has been carried out to

1 standardised test guidelines. The study method should be reported. In addition key
2 study information should also be provided in the technical dossier (further guidance is
3 given in the Section 8 of the [Guidance on registration](#)). These are 1) test substance
4 identification, 2) sample purity, 3) test species and 4) test duration. Without this
5 information and in the absence of other key study information or other studies for the
6 same endpoint it is extremely difficult to justify use of that particular study result on its
7 own. The study may be used in combination with other data as part of a *Weight of*
8 *Evidence* approach (see Section R.4.4 in Chapter R.4 of the [Guidance on IR&CSA](#))

9 **Other programmes/ secondary sources of data**

10 There are also circumstances where reported values have already been through a
11 screening process such as the SIDS program or through an EU existing substances risk
12 assessment (<http://esis.jrc.ec.europa.eu/>). In such circumstance the data may be
13 considered sufficiently reviewed as to not require further evaluation assuming that the
14 problems have been highlighted with the study(ies) of interest. Data reported as part of
15 other equivalent peer reviewed risk assessment programs (e.g. HERA
16 (<http://www.heraproject.com/>)US-EPA HPVC Challenge Programme) may also be
17 considered in this way although a level of expert judgement is required to evaluate the
18 quality of these programmes and further justification in the use of such a programme
19 data may be required.

20 **R.7.8.4.1 Data on aquatic pelagic toxicity**

21 **Testing data on aquatic pelagic toxicity**

22 *In vitro* data

23 Although the extrapolation of *in vitro* data to *in vivo* data is discussed in literature
24 further research in this area is needed (ECETOC, 2005) and there is currently not
25 enough information available to give guidance for the extrapolation from *in vitro* data to
26 *in vivo* data. Various publications show that, for the correlation with *in vivo* results the *in*
27 *vitro* bioavailability of the substances tested should be considered (Guelden and Seibert,
28 2005; Bernard and Dyer, 2005; Schirmer, 2006).

29 Currently, there are no validated fish cell systems available. Nevertheless, information
30 from *in vitro* studies might be considered in a *Weight of Evidence* approach provided that
31 they fulfil certain data quality aspects and comply with the Annex XI criteria.

32 Annex XI states that *suitable in vitro* methods should be well developed and fulfil certain
33 criteria, e.g. the ECVAM criteria to enter a pre-validation study (Curren *et al.*, 1995).
34 Based on these, the following information on the study/method would be useful:

- 35 • the source of data should be named (e.g. publication, study report, in-house
36 data, interlaboratory study)
- 37 • fish cell system:
 - 38 - primary cells (tissue used for isolation)
 - 39 - fish cell line and if available passage number
 - 40 - for both, culture conditions (e.g. medium, serum, serum-free)

- 1 • protocol used (e.g. incubation temperature, exposure time, replicants,
2 endpoint measured, positive and negative controls, data analysis and
3 interpretation, limitations, etc)
- 4 • status of standardisation of protocol
 - 5 - in house validated (evidence of repeatability)
 - 6 - used in other labs (evidence of reproducibility)
 - 7 - nominal or measured concentration
 - 8 - comparison to other *in vitro* / *in vivo* tests
 - 9 - data on other substances tested with the method

10 Primary cells are more suitable to evaluate specific toxic effect, e.g. isolated hepatocytes
11 for liver toxicity, metabolism or isolated gill epithelia for effects on the gill barrier
12 function, toxicant uptake and metabolism. However they require the use of living
13 animals. Cytotoxicity tests using fish cell lines are more likely to indicate acute toxic
14 effects although it is necessary to consider that they might lack of realistic toxicokinetics
15 including metabolism

16 The ongoing standardisation and validation efforts might provide validated methods
17 which will then be included into testing strategies.

18 **In vivo data (single species)**

19 **INITIAL RELIABILITY SCREENING**

20 An initial review of the reliability of data should be made in order to filter out the most
21 reliable values for consideration. For many existing substances the test data available
22 will have been generated prior to the establishment of standard protocols and Good
23 Laboratory Practices (GLP). To address the potential variability in data quality in older
24 data collections, there are various possible approaches. These include methods such as
25 those employed by the OECD (2000a), U.S. EPA (2002), Hobbs *et al.* (2005) or the
26 recommendations of Klimisch *et al.* (1997) which are introduced and described in
27 Chapter R.4 of this guidance document. Further data on structurally similar substances
28 may be available and these may add to the toxicity or ecotoxicity profile of the
29 substance under investigation.

30 Klimisch *et al.* (1997) describe the parameters that need to be considered to evaluate
31 the quality of a non-standard test. However, the authors do not describe the expert
32 judgement process by which the strengths and weaknesses in the reporting of these
33 different parameters are integrated to determine an overall quality assessment. To
34 address this limitation, the following set of quality criteria, which are a development of
35 Klimisch *et al.* (1997), should be considered (see below for further details):

- 36 • Description of the test substance.
- 37 • Description of the test procedure including exposure period.
- 38 • Data on the test species and the number of individuals tested.
- 39 • Description of measured parameters, observations, endpoints.

- 1 • Control data available and acceptable according to guidelines. For some
2 species used in environmental toxicity tests, guidelines are not available and
3 in this instance, the guideline for the taxonomically closest equivalent species
4 should be used.
- 5 • A concentration-response has been established, except in the case of limit
6 tests determining a NOEC/ECx.
- 7 • Achieved exposure concentrations were measured in the test medium or
8 vehicle. For aquatic toxicity tests, measurements should be made at least at
9 t0 and tend and exposure should be calculated in terms of geometric mean
10 measured concentrations unless measured concentrations were within 20% of
11 the nominal concentration, in which case the nominal concentrations may be
12 used.

13 If available data do not conform to the quality standards, the data should be
14 reconsidered, to determine whether any of them are acceptable under current
15 circumstances, and in particular, that they will not underestimate toxicity. For example,
16 in an environmental toxicity test the data could have been rejected due to an absence of
17 measured concentrations in the test media, but for a test substance whose
18 physical/chemical properties suggest a low potential for biodegradation / volatilisation /
19 sorption, the data may be acceptable.

20 Irrespective of whether or not data meet the full set of quality criteria, consideration
21 should be given as to whether the data:

- 22 • are outliers in a large data-set for a particular substance;
- 23 • fit with what is known of the toxicity of other related substances.

24 **Checklist**

25 After an initial screen, a number of studies will be screened out on which to focus and a
26 second stage of screening is likely to be necessary. In an ideal world this considers what
27 is essentially a minimum set of criteria which should be met. The following
28 considerations relate to the aquatic toxicity testing at this second screening:

29 **Test substance/ test substance identification**

30 It is important to be able to accurately identify the substance tested. This should include
31 an adequate description of the test substance. Ideally this should include an
32 internationally recognised identifier such as the CAS number. However, the CAS number
33 is not always unique to a substance and so a chemical description may be sufficient as
34 long as the description is sufficiently detailed to allow clear identification. For example,
35 positioning of particular moieties around a ring structure can be important from an
36 (eco)toxicity point of view so a description of dichloro- should be more clearly identified
37 as 1,3-dichlor etc. A further example can be where the term alkyl is used when an exact
38 chain length should be described.

39 It is critical to ensure that the test material which has been tested is actually consistent
40 with the substance being registered. It may be for example that the material tested is a
41 mixture of homologous chain lengths which are a different distribution to the CAS

1 number being registered. This may be acceptable. However, this information should be
2 clearly described and justified why such data can be used.

3 Chemical purity should be described and where possible identification of the impurity
4 should be made. The impurity can be important can be responsible for the majority of
5 observed toxicity of a sample even if it is present at low levels. There are cases where
6 studies have been carried out on test materials which have included with them a
7 constituent/impurity which is present intentionally (such as preservatives). In some
8 cases these studies may have been carried out intentionally on this mix in order to
9 replicate more closely the actual material used/ sold. This factor should be considered
10 when assessing the data.

11 Water solubility should be reported ideally. Results which occur above the limit of water
12 solubility should be considered in further detail – see [Appendix R.7.8–1](#).

13 **Test Organisms**

14 Details of the taxonomic identity of the organisms used in the study should be described
15 to include the genus and the species. In some cases the genus alone can be sufficient
16 information where it is known that all members of that genus are of similar sensitivity.

17 Where studies are conducted to standard methodologies such as the OECD guidelines
18 described earlier, often these have listed standard organisms for which the test method
19 is relevant. Non-standard species can also be accepted. However, these should be
20 properly identified and characterised in order to ensure that the test method is suitable.

21 **Test setup**

22 The test system should be adequately described and wherever possible the test should
23 be in accordance with an internationally accepted guideline. Non-standard methods can
24 be accepted but clear description of the methods should be made. If a non-standard
25 method is described or a standard method is followed and a judgement on whether the
26 method has been adhered to, then the following are to be considered:

27 Test procedures and conditions should be reported to include standard/recognized
28 procedures, appropriate acclimation procedures followed, certain conditions noted (test
29 temperature, dissolved oxygen levels, pH, lighting), and placement of test units to avoid
30 position effects) etc.

31

32 Test duration. This is critical information in deciding reliability of a study and must be
33 reported. These do vary by endpoint/ study. Key values have been described previously
34 under Guideline Studies. Deviations from these will make comparison with results from
35 other studies difficult even when these studies are of good quality (e.g. *Daphnia sp* EC50
36 results are commonly reported at 24 hours compared to the standard 48 hours).
37

38 Deviations from standard guidelines. Where deviations are made from the standard
39 guidelines these should be clearly described. Such studies will by default not be scored
40 as reliability 1 under Klimisch. However, with clear documentation the studies may be
41 classified as reliability 2. Without such descriptions the study may be scored as reliability
42 3 or 4, both of which would indicate less than favourable study results.

1
2 Route/Type of exposure. Delivery of the test substance is a critical factor to consider to
3 ensure suitable exposure to the test organisms. For algae, static tests are common. For
4 *Daphnia* studies static or semi-static tests are common and for fish static, semi static
5 and flow-through studies are common. The potential effect of any relevant phys-chem
6 properties of the substance such as solubility, high adsorption, precipitation etc on
7 delivery should also be documented.

8 In some studies food is added during the exposure period (e.g. green algae are added as
9 food in a *Daphnia* reproduction test). In such cases exposure may also occur via food for
10 substances that adsorb to the algae.

11
12 A description of the test medium and dilution water should be included to ensure that it
13 is for example correctly made, of specified hardness and salinity range etc. Other
14 relevant quality criteria should be included also as appropriate such as total organic
15 carbon, un-ionized ammonia. Besides ensuring that all abiotic factors fall within the
16 tolerance limits of the test organisms a proper description of other abiotic parameters,
17 e.g. dissolved organic carbon concentration (DOC), cations and anions etc., that govern
18 the speciation (i.e. availability) and subsequently may influence the uptake of certain
19 chemicals. In particular influence of abiotic factors on the bioavailability of some metals
20 and inorganic metal compounds have been studied and for certain of these chemicals
21 correction for bioavailability is possible and relevant. The term bioavailability³ is in the
22 context of environmental risk assessment of metals used to describe both the availability
23 of metals due to speciation phenomena (a part which is independent of the organism and
24 where chemical speciation models could be used as a first tier to reduce variability) and
25 the real bioaccessibility part influenced by biological/physiological factors (e.g.
26 competition effects as captured in Biotic Ligand Models).

27 Furthermore, in the case of testing essential metals and metal constituents a proper
28 description of the culture conditions, specifically related to the level of essential metals
29 and inorganic metal compounds added or already present in the culture media could give
30 valuable insight on issues such as acclimation. The way how bioavailability can be taken
31 account of in aquatic effects assessment for metals and inorganic metal compounds is
32 further elaborated in the guidance on metals.

33 Test concentrations/dose levels and number of concentrations should be known and
34 where possible evidence provided that concentrations have been maintained throughout
35 the duration of the test. Therefore, measured concentrations are preferred over nominal
36 (non-measured) concentrations. If measured concentration are <80% of nominal

³ Bioavailability of metals: A metal is considered bioavailable when it is free for uptake by an organism and when it result in a toxicity response (Newman and Jagoe, 1994; Campbell *et al.*, 1988). The main idea behind the concept of "bioavailability", is that the toxic effect of a metal does not only depend on the total (or dissolved) concentration of that metal in the surrounding environment, but also on the complex interaction between physico-chemical factors, the free metal ion considered and the biological ligand on which the metal binds and result in a toxic response of the exposed organism. In other words, the same total metal concentration does not result in the same degree of toxic effect on an organism under all environmental conditions.

1 concentrations, effect values should be related to mean measured concentrations. For
2 flow-trough studies the arithmetic mean of measured concentrations should be
3 calculated, for static or semi-static tests the geometric mean of measured concentrations
4 (see [Appendix R.7.8–1](#)). In some cases where only nominal concentrations are
5 provided, expert judgement may be required to decide whether test concentrations are
6 likely to have been maintained. Such circumstances may occur if:

- 7 • It is known that the material is abiotically and biotically stable (from e.g.
8 stability in water/ biodegradation studies etc such as OECD 111, OECD 113,
9 OECD 301A-F, OECD 310, OECD 302A-C) to conclude that the concentrations
10 are likely to have been maintained during the study.
- 11 • The test substance is soluble, well below its limit of solubility,
- 12 • Is non volatile
- 13 • Has low adsorbance to either delivery apparatus or the exposure vessels

14 For metals and inorganic metal compounds there is a strong preference for using
15 measured data because potential issues related to natural background, to analytical
16 errors and to the limited solubility of some metals and inorganic metal compounds. If it
17 is not mentioned whether the reported toxicity values are based on measured
18 concentrations, they should be considered as nominal concentrations. In cases where no
19 measured data are available the use of nominal concentrations could be considered. In
20 artificial media, where the metal background concentration is often very low compared
21 to the effects levels, nominal concentrations could usually be used as long as the tests
22 are based on soluble metal salts. When natural waters are used instead of artificial test
23 media there could be a concern with the use of nominal values when the derived
24 NOEC/EC₁₀ values are close to the reported background values of the natural water used
25 as these concentrations could potentially contribute to the observed toxicity in a
26 significant way and as result the use of a nominal values would overestimate toxicity.

27 However, it must be emphasized that most often information on metal background
28 values in natural waters is not readily available. Furthermore natural background
29 concentrations for metals can vary substantially and cannot easily be distinguished from
30 anthropogenic metal concentrations. For sparingly soluble metals measured data on the
31 dissolved fraction⁴ are always required for getting reliable toxicity test data. If the
32 solubility is exceeded the test result has to be considered as unreliable. Results from
33 tests where a visual precipitation is observed should be discarded. The absence of a
34 visual precipitation does not exclude that colloids may be present that could affect the
35 test results. For more specific guidance see section on difficult substances in [Appendix](#)
36 [R.7.8–1](#).

37 In some cases studies will have been carried out with the use of solubilisers. In these
38 circumstances it is important to consider the change in bioavailability of the test

⁴ Different definitions for the dissolved fraction exist. Most often the dissolved fraction in ecotoxicity tests refers to the fraction that passes through a filter of 0.45 µm. It should be noted, however, that this definition may not necessarily refer to the metals in solution. In the range of 0.01-0.45 µm colloid inert particles that remain suspended may exist.

1 substance and also the potential impact of the solubiliser. Studies performed without
2 solvents/solubilizers are preferred over studies with solvents. Solvent concentrations
3 should be the same in all treatments and controls. Further guidance on the interpretation
4 of studies performed with the use of solubilisers is given in OECD (2000c).
5 Where a reasonable estimation of the exposure concentration cannot be determined then
6 the test result should be considered with caution unless as part of a *Weight of Evidence*
7 approach.

8 Controls: All studies must have controls. If a solvent is used, also solvent controls are
9 necessary.

10 Test endpoints and reported data. Confidence in the reliability of a study can be
11 increased if dose-response or concentration-response is evident and some measure of
12 data quality such as GLP is reported to have been followed. Where a test result is
13 reported as a *less than* (<) value this cannot be used. Results reported as *greater than*
14 (>) can be used as additional information and may in some cases be considered directly
15 instead of a fully defined result. However, this result should be justified with
16 considerations of the test set up and phys-chem properties etc which may influence the
17 result.

18 Statistical analyses. Statistical methods for derivation of LC₅₀, EC₅₀, IC₅₀, NOEC values
19 etc should be reported. Where possible these should be presented with relevant
20 reliability criteria. However, in the absence of these a description of the method could be
21 considered acceptable.

22 Test design: Studies should be designed to enable sufficient statistical differences to be
23 established between controls and test ingredient solutions. Further guidance on number
24 of replicates, number of test organisms per replicate, number of concentrations
25 necessary for a reliable EC_x and/or NOEC/LOEC determination can be found in the
26 different OECD test guidelines.

27 Hormesis effect: Hormesis has been observed for metal as well as organic substances
28 and has been related to enhanced performance at low levels of induced stress (=at lower
29 test concentrations). In such cases it is indeed important to use the neutral control data
30 as a reference or to use specific models designed to model hormesis phenomenon
31 (Brain and Cousens, 1989, Van Ewijk and Hoekstra, 1993; Schabenberger *et al.*, 1999;
32 Cedergreen *et al.*, 2005). The need to take the activating part into account when
33 deriving an EC_x should be considered when appropriate.

34 For metals and especially, essential metals, the observation of hormesis may however
35 also indicate a metal deficiency of the control medium and this needs to be avoided (see
36 - description of the test medium). The possibility of a hormesis effects, observed for
37 essential nutrients, needs to be considered when evaluating the calculation of EC₁₀
38 values beyond the lowest tested concentration.

39

40 **Guidance of specific test types for freshwater species**

41 In the following practical guidance is given for the evaluation of data from non-standard
42 ecotoxicity tests.

1

2

3 Evaluation of data from growth inhibition testing on algae, aquatic plants (OECD 201
4 (2006c), 221 (2006d) and other standard and non-standard tests):

5 Commonly used and favoured tested species are *Pseudokirchneriella subcapitata*
6 (previously named *Selenastrum capricornutum*) *Scenedesmus subspicatus* and *Chlorella*
7 *vulgaris*. All can be considered as equally accepted preferred species.

8 The algal test is a short-term test although it provides both acute and chronic endpoints.
9 The preferred observational endpoint in this study is algal growth rate inhibition because
10 it is not dependent on the test design, whereas biomass depends both on growth rate of
11 the test species as well as test duration and other elements of test design.

12 Often both acute growth rate EC₅₀ (ErC₅₀) and biomass (EbC₅₀) endpoints are reported
13 however the latter should not be used. The reason is that direct use of the biomass
14 concentration without logarithmic transformation cannot be applied to an analysis of
15 results from a system in exponential growth. Where only the EbC₅₀ is reported, but
16 primary data are available, a re-analysis of the data should therefore be carried out to
17 determine the ErC₅₀. Where other supporting data exist as part of a *Weight of Evidence*
18 approach it may be possible to consider an EbC₅₀ value if only this value is reported.
19 However, if only an EbC₅₀ is reported and no primary data are available, it should be
20 considered to perform a new algae study to obtain a valid ErC₅₀ and NOEC or ErC₁₀
21 especially if algae are the most relevant species for the effects assessment.

22 The typical test duration for this study is 72 hours. However, 96 hours is also commonly
23 reported. This should be used as an equally acceptable value. For existing substances
24 often algae tests with a duration of >96 h are available. As it cannot be assumed that
25 the algae are in the exponential growth phase during the whole exposure period, the
26 result from such tests cannot be used, unless the available raw data show monotone
27 exponential growth of the controls. This also applies to reported chronic NOEC values.
28 Common examples of this are 7-day and 14-day reported values.

29 It is sometimes seen also when test was done according to standard test guidelines, that
30 the exponential growth ceased in the control before the end of the test period. Likewise
31 it may be seen that the validity criteria of the test were not fulfilled (pH increase etc.) or
32 growth of the algae in the exposed concentrations was increased (due to e.g. loss of test
33 substance from the test system) at the end of the test. In such cases only data from the
34 part of the test where exponential growth occurs and the validity criteria for the controls
35 are fulfilled, should be used. In many such cases this may be achieved by excluding data
36 from the last test day from the calculation of ErC₅₀ and NOEC or ErC₁₀.

37 Common problems associated with algal study measurements result from coloured test
38 materials and those with particular particle size (see [Appendix R.7.8–1](#)).

39 The most commonly used vascular plants for aquatic toxicity tests are duckweeds (*Lemna*
40 *gibba* and *Lemna minor*). The Lemna test is a short-term test although it provides both
41 acute and sub-chronic endpoints. The tests last for up to 14 days and are performed in
42 nutrient enriched media similar to that used for algae, but may be increased in strength.
43 Test design can be static, semi-static or flow-through. Frond number is the primary

1 measurement variable. Other additional measurement parameters are total frond area, dry
2 weight/fresh weight. The ECx/NOEC should be related to growth rate.

3 Evaluation of data from short-term toxicity testing on invertebrates (OECD 202 (2004b)
4 and other standard and non-standard tests):

5 In addition to *Daphnia magna*, *Daphnia pulex*, *Ceriodaphnia affinis* and *C. dubia* are
6 commonly tested species. Overall, there is no significant difference in sensitivity of *D.*
7 *magna* and *D pulex*. Good correlation has been reported between acute toxicities of all
8 three species (ECETOC 2003c). All these can be considered as equally accepted
9 preferred species.

10 Acute tests with crustacea generally begin with first instar <24 hours old juveniles. If the
11 test organisms used are >24 h old, their sensitivity might be lower and the test can be
12 accepted only in conjunction with other available data.

13 For daphnids, a test duration of 48 hours is standard. However, 24 hour LC₅₀ or EC₅₀
14 values are often reported for this study. 24 hour values can have considerable variability
15 in the repeatability of results and should not be compared to 48 hour values. The
16 standard 48 hour reported values are favoured over 24 hour values for these reasons.
17 24 hour values should be considered only in the absence of good quality 48 hour values
18 and in conjunction with other available data (non-testing, read-across, information on
19 time-dependence of effects etc). For other crustacea, such as mysids or others, a
20 duration of 96 hours is typical.

21 The observational endpoint for short-term invertebrate tests is immobilization (EC₅₀) as a
22 surrogate to mortality as it is quite difficult to make a clear judgement on mortality.
23 Immobilisation is defined as unresponsive to gentle prodding.

24 Studies are often conducted under semi-static conditions where test solutions are
25 renewed at periods (usually after 24 hours) during the study. This helps to maintain test
26 concentration during the duration of the study. These studies are preferable over those
27 studies conducted under static conditions, when the test material is known to degrade
28 rapidly (either biotically or abiotically) or where known test material properties could
29 lead to reduced test solution concentration due to adsorption processes for example.
30 Results from flow-through studies can also be used as long as test duration is as already
31 described.

32 Often a NOEC is reported for this acute study. This value cannot be used as surrogate
33 value for a chronic NOEC as reported from OECD guideline 211.

34 Evaluation of data from long-term toxicity testing on invertebrates (OECD 211 (1998b)
35 and other standard and non-standard tests):

36 Chronic tests with crustacea also generally begin with first instar juveniles and continue
37 through maturation and reproduction. At least 3 broods should be produced during the
38 exposure period. For daphnids, 21 days is sufficient for maturation and the production of
39 3 broods. For mysids, 28 days is necessary while *Ceriodaphnia dubia* produces 3 broods
40 within 7 d. Observational endpoints include time to first brood, number of offspring
41 produced per female (reproduction), growth, and survival (lethality). Reproduction and
42 lethality are the most sensitive endpoints. Where uncertainty arises from which endpoint
43 to consider, the lowest reported value should be used. Due to the test duration there is

1 higher potential for loss of test material concentration over the test period. Studies with
2 analytical support are thus preferable where available. Where such data are not
3 available, consideration of other properties which may lead to doubt over test material
4 concentration should be made, where these data are available. In addition to solubility
5 these would include biotic and abiotic degradation and adsorption potential of the test
6 material (resulting in loss to test glassware/ feed etc).

7 Typically the 21 day study may report ECx/NOEC values for survival or reproductive
8 endpoints. The lowest value should be used for establishing ECx/NOEC for reproduction
9 although in practice the two endpoints results tend to be close to each other.

10 Evaluation of data from short-term toxicity testing on fish (OECD 203 (1992a) and other
11 standard and non-standard tests):

12 A number of species are recommended for use across several OECD Test Guidelines.
13 [Appendix R.7.8–2](#) indicates commonly used recommended species from OECD Test
14 guidelines 203: Fish, Acute Toxicity Test; 204 Fish, Prolonged Toxicity Test: 14-Day
15 Study; 210: Fish, Early-life Stage Toxicity Test; 212: Fish, Short-term Toxicity Test on
16 Embryo and Sac-fry Stages and 305: bioconcentration: Flow-through Fish Test. These
17 can be considered as equally accepted preferred species.
18

19 The differences in fish species sensitivity sometimes can be substantial. This can often
20 be due to differences in toxicity of the test material rather than inherent differences in
21 species sensitivity. Often substances with the highest toxicity also have the largest
22 variation in toxicity to different species. Acute tests are generally performed with young
23 juveniles 0.1-5 g in size for a period of 96 hours. Fish larger than this range are
24 generally less sensitive.

25 Where values are reported with shorter test duration, these should be treated with
26 caution and should be used only in conjunction with other data (non-testing), read-
27 across etc. as exposure phases shorter than 96 h generally lead to higher effect values.

28 Care should be taken also when considering studies carried out where the test material
29 is readily biodegradable and where the nominal test concentration is low (<10mg/l). In
30 these cases there is high likelihood that test concentrations will be lower than nominal.

31 The observational endpoint in these tests is mortality (LC₅₀).

32 Studies are often conducted under semi-static or flow-through conditions where test
33 solutions are renewed at periods (usually after 24 hours) or continuously during the
34 study. This helps to maintain test concentration during the duration of the study. These
35 studies are preferable over those studies conducted under static conditions, when the
36 test material is known to degrade rapidly (either biotically or abiotically) or where known
37 test material properties could lead to reduced test solution concentration due to
38 adsorption processes for example.

39 Evaluation of data from long-term toxicity testing on fish (OECD 210, 212, 215 and other
40 standard and non-standard tests):

41 Only such studies can be regarded as long-term fish test, in which sensitive life-stages
42 (juveniles, eggs, larvae) are exposed. Thus, tests performed according to OECD 204
43 (Fish, Prolonged Toxicity Test: 14-Day Study (OECD 1984)) or similar guidelines cannot

1 be considered suitable long-term tests. They are, in effect, prolonged acute studies with
2 fish mortality as the major endpoint examined. The most relevant long-term fish tests
3 are described below.

4 OECD Test Guideline 210 (1992b) Fish, Early-Life Stage (FELS) Toxicity Test:

5 For the test the following freshwater species are recommended *Brachydanio rerio*,
6 *Pimephales promelas*, *Oryzias latipes*, and *Oncorhynchus mykiss* as well as saltwater
7 *Cypridon variegatus*. Among the currently available standardised test methods, the FELS
8 toxicity test is considered as the most sensitive of the fish tests. It covers several life
9 stages of the fish from the newly fertilised egg, through hatch to early stages of growth
10 and is also the only suitable test currently available for examining the potential toxic
11 effects of bioaccumulation. The required test duration is species-dependent: 60 days
12 post-hatch for rainbow trout or approximately 30 days for warm water fish.
13 Observational endpoints include hatching success, survival and growth.

14 OECD Test Guideline 212 (1998a) Fish, Short-term Toxicity Test on Embryo and Sac-Fry
15 Stages:

16 For the test the following freshwater species are recommended *Danio rerio*, *Pimephales*
17 *promelas*, *Cyprinus carpio*, *Oryzias latipes*, and *Oncorhynchus mykiss*. This test
18 measures the sensitive early life stages from the newly fertilised egg to the end of the
19 sac-fry stage. It is considerably shorter, and hence less expensive, than the FELS toxicity
20 test but it is also considered less sensitive. The method offers an alternative to the FELS
21 toxicity test for substances with log K_{ow} less than 4.

22 OECD Test Guideline 215 (2000b) Fish, Juvenile Growth test:

23 *Oncorhynchus mykiss* is recommended freshwater specie for the test, however also *Danio*
24 *rerio* and *Oryzias latipes* may be used. This test measures the growth of juvenile fish
25 over a fixed period, and it is considered a sensitive indicator of toxicity. Although it is
26 considered to be of insufficient duration to examine all the sensitive points in the fish
27 life-cycle, it provides a shorter and less expensive option to the FELS test for substances
28 of log $K_{ow} < 5$.

29 Non-standard tests using similar methods can be accepted if the studies are well
30 documented and comply with the guidelines in critical points (exposure duration,
31 endpoints studied). Studies should be performed preferably under flow-through
32 conditions or under appropriate semi-static conditions.

33 **Marine species**

34 There are few standardised marine species protocols available (see [Appendix R.7.8–2](#)).

35 In general the same criteria as described for freshwater tests should be applied for the
36 evaluation of the tests for marine species. Additional attention should be paid to the fact
37 that the solubility of the substance might be influenced by the salinity (see [Appendix](#)
38 [R.7.8–1](#) for further detail).

39

40

1 **Difficult substances**

2 A significant number of chemicals are described as 'difficult substances', which the OECD
3 (2000c) class as difficult to test for the purpose of determining their aquatic toxicity.

4 Typical characteristics of difficult substances include:

- 5 • Difficulty in maintaining substance concentration during the test, for example
6 degradation in the test medium or loss of substance from media (e.g.
7 absorption or evaporation)
- 8 • Difficulty in dissolving the substance, either due to poor solubility in test
9 medium or a multi-component substance of varying solubility
- 10 • Difficulty in being able to measure substance concentration, due to problems
11 in developing an analytical method or again multi-component substances

12 Such properties and the problems these cause for carrying out valid tests and their
13 interpretation are described in [Appendix R.7.8–2](#), and more fully in publications issued
14 by the OECD and ECETOC (ECETOC 2003a). These also describe practical ways to deal
15 with such issues. The possibility of a substance being difficult to test can often be
16 determined from its physico-chemical properties such as water solubility, volatility,
17 biodegradability, hydrolysis and photodegradability. This re-emphasises how important it
18 is to know these parameters prior to new test being carried out, or before reviewing a
19 test report.

20 ***In vivo – multiple species (field data)***

21 Model ecosystems represent the highest experimental tier in the hazard and fate
22 assessment processes. When tests are well-designed, the exposure of chemicals to
23 environmental organisms can be directly related to the route applied in model ecosystem
24 tests. The diversity of organisms and their interactions cannot be adequately modelled in
25 simpler laboratory single species tests, therefore valuable information on fate and effect
26 responses of biota can be gained. Test systems should contain sufficiently complex
27 assemblages to address the objectives. In order to be useful for environmental
28 protection, results should be statistically reliable and capable of identifying response
29 patterns.

30 **Concepts of Data Integration and Statistics**

31 Conclusions developed from model ecosystem tests are based on expert judgment using
32 a combination of univariate and multivariate statistical analyses of measured endpoints.

33 Explicit evaluation of model ecosystem data should be systematic. Combinations of both
34 univariate and multivariate analyses are preferred if the measurements collected during
35 the test are amenable to both. Effects observed through time, whether or not the
36 effects are permanent or transitory, and the nature of the exposure-response
37 relationship for important endpoints should be explored. OECD (2006a) provides
38 reporting needs for standing water studies, but similar considerations exist for flowing
39 water studies. These include information on the test substance, thorough description of
40 the test system, experimental design and measured data, and how data were evaluated.
41 As described in [Appendix R.7.9–2](#), the actual reporting of a study will largely depend on
42 the objectives of the work.

1 Evaluation of data

2 Mesocosms are not commonly employed for general chemicals partly because the dosing
3 methods employed may not be representative of the way that these chemicals reach the
4 environment (unlike pesticides which may reach ponds, ditches or rivers via drift or run-
5 off). Another reason is without doubt that only for few industrial chemicals resources
6 were available to conduct such higher tier expensive tests. In certain exceptional cases
7 (notably down the drain chemicals) lotic mesocosm data may be most useful. However,
8 if water concentrations can be maintained adequately and the mesocosm can be
9 maintained long enough that sediments reach equilibrium concentrations, the results
10 may be highly relevant in addition to laboratory tests on individual species.

11 Within the Existing Substance Regulation only for few substances results from mesocosm
12 studies were available (e.g. metals such as zinc and cadmium, acrylamide, nonylphenol).

13 In summary, the main conclusions seem to have been that mesocosm data suffer from
14 some of the following drawbacks:

- 15 • Observation intervals may be too long
- 16 • There can be overlap with other pollutants (e.g. metals) which makes
17 interpretation difficult.
- 18 • Analytical inconsistencies may occur.
- 19 • There may be difficulties in maintaining exposure concentrations over
20 prolonged periods and in confirming concentration (e.g. in relation to river
21 flow rates).
- 22 • Some potentially sensitive life stages (e.g. larval stages), endpoints or species
23 might not be included.
- 24 • Given the natural variation inherent in such test systems, very large changes
25 in population abundance may have to occur for them to be statistically
26 significant when compared to the variation in control populations.
- 27 • The number of endpoints measured may be insufficient to draw reliable
28 conclusions, or a clear concentration-effect relationship may be lacking.

29 **Non-testing data on aquatic pelagic toxicity**

30 General guidance for the evaluation of non-testing data is provided in Chapter R.6 of the
31 [Guidance on IR&CSA](#) (cross-cutting guidance QSAR). The following section includes
32 information specific for the evaluation of the reliability of non-testing data in aquatic
33 toxicity.

34 **Evaluation of QSAR results**

35 As outlined in Section R.6.1 in Chapter R.6 of the [Guidance on IR&CSA](#), the evaluation of
36 the reliability of a non-testing result includes two steps:

37 **1. Evaluation of the validity of the model or expert system**

1 The validity of a model should be assessed according to the OECD validation principles
2 for QSARs (OECD 2004a). They can be used for the evaluation of expert systems
3 respectively. An in depth interpretation of the OECD principles can be found in Worth *et*
4 *al.* (2005) and in Chapter R.6 of the [Guidance on IR&CSA](#) (cross-cutting guidance
5 QSAR). [Table R.7.8—1](#) summarizes specific aspects for the assessment of aquatic
6 toxicity endpoints.

7 **Table R.7.8—1 Specific aquatic toxicity aspects of the OECD validity criteria**

OECD Principle	Specific considerations for aquatic toxicity assessment
Principle 1: a defined endpoint	A defined endpoint is assumed if the QSAR model is based on experimental data with a) a single measured biological endpoint (eg. mortality of a specific fish species) b) comparable exposure conditions (e.g. exposure duration, same age of test organisms) and c) a single statistically derived endpoint (e.g. LC ₅₀)
Principle 2: an unambiguous algorithm	No specific considerations. Models based on linear regressions using logK _{ow} as sole descriptor are considered to have an unambiguous algorithm. General considerations for the scientific validation of (Q)SAR models are described in Section R.6.1.3.
Principle 3: a defined domain of applicability	A defined domain of applicability can be based on a) definition of the descriptor domain of the model (i.e. range of log K _{ow} of the training set) b) definition of the structural domain of the model (e.g. description of fragments and functional groups covered by the model) c) definition of the mechanistic domain of the model
Principle 4: appropriate measures of goodness-of-fit, robustness and predictivity	No specific considerations for aquatic toxicity assessment. General considerations for the scientific validation of (Q)SAR models are described in Section R.6.1.3.
Principle 5: a mechanistic interpretation (if possible)	A mechanistic interpretation is possible if the QSAR model is based on chemicals assumed to have the same mode of action (e.g. models for polar or non-polar narcosis) or on chemical classes with a known mode of action (e.g. carbamates).

8

9 The outcome of the analysis might not be a simple yes/no answer and it might be
10 impossible to conclude on the validity of the model without considering the regulatory
11 context of the decision. However results of the analysis should be reported in a

1 transparent way. Templates, so called QSAR model reporting formats (QMRFs) are
2 provided in Section R.6.1.9 in Chapter R.6 of the [Guidance on IR&CSA](#).

3

4 **2. Evaluation of the reliability of the outcome of a prediction**

5 General guidance for the evaluation of model predictions is provided in Section R.6.1.3 in
6 Chapter R.6 of the [Guidance on IR&CSA](#). The outcome of the assessment should be
7 reported in detail. Templates, so called QSAR prediction reporting formats (QPRFs) are
8 provided in Section R.6.1.10 in Chapter R.6 of the [Guidance on IR&CSA](#).

9 Evaluation of the outcome of schemes for the identification of modes of actions

10 Assessing the result of a prediction of a mode of action is mainly connected with an
11 analysis of the possible short comes of the prediction with respect to the background
12 (mechanistic domain) of the scheme. Some of the schemes include rules that focus on
13 the identification of possible structural alerts/structural classes, while other focus on the
14 active identification of chemicals acting via narcosis (e.g. Verhaar *et al.*, 1992). Some
15 information about the background of the different schemes is provided in Chapter R.10
16 of the [Guidance on IR&CSA](#) (Appendix 1).

17 In general the following issues should be considered:

- 18 • Is the characterisation based on the identification of specific structural
19 properties? E.g. was a substance identified as being narcotic because of its
20 chemical structure or just because it does not fit to any of the classes
21 described by the scheme?
- 22 • Is the chemical within the applicability domain of the characterisation
23 scheme? E.g. does the chemical include substructures that are unknown by
24 the schemes? This becomes increasingly important if the scheme is based on
25 the identification of substructures that might be responsible for excess
26 toxicity. If a substructure of the chemical is not known by the scheme, the
27 scheme might not be able to assess if this substructure will create excess
28 toxicity.

29 Evaluation of the outcome of a research for structural alerts

30 Structural alerts as described in Section [R.7.8.3](#) and Section R.10.2.2.2 in Chapter R.7c
31 of the [Guidance on IR&CSA](#), indicate the presence of substructures that might increase
32 the aquatic toxicity of the substance. Thus, if a structural alert was identified for a given
33 substance, it can be assumed that the substance exhibits excess toxicity. On the other
34 hand, the absence of a structural alert does not necessarily indicate the absence of
35 excess toxicity since lists of structural alerts are not exhaustive. Thus results from a
36 structural alert research can be used as a confirmation or evidence of excess toxicity
37 only. It cannot rule out other information if no alerts are identified. In order to assess
38 the reliability of the structural alert research the same criteria as described above should
39 be applied.

40 Evaluation of the outcome of a QSAR/QAAR prediction

1 Assessing the reliability of a QSAR/QAAR prediction for aquatic toxicity endpoints is
2 mainly connected with the question whether the substance is within the predictive space
3 of the model or not. Guidance for the assessment is provided in Section R.6.1 in Chapter
4 R.6 of the [Guidance on IR&CSA](#). Additional information about the reliability can be
5 achieved by comparing the mechanistic domain of the model with the assumed mode of
6 action of the substance.

7 Evaluation of information derived by the grouping approach

8 The reliability of results obtained by grouping approaches highly depends on the
9 selection of appropriate analogues and chemical classes. General guidance for the
10 assessment of the reliability and applicability of grouping approaches is provided in
11 Section R.6.2 in Chapter R.6 of the [Guidance on IR&CSA](#). With respect to aquatic toxicity
12 the following additional aspect should be considered:

13 Are substances used for the grouping approach that are comparable with respect to
14 substructures (e.g. do they all contain/ not contain structural alerts)?

15 Can a similar mode of action/structural class be assumed for all substances?

16 Are the substances comparable with respect to physico-chemical properties that
17 influence aquatic toxicity (e.g. comparable lipophilicity)

18 Is the metabolic pathway of the substances comparable? E.g. specific attention should
19 be paid to substances with methyl groups as the metabolic activation might differ from
20 similar compounds that do not include methyl groups.

21 The selection of chemicals for read-across and chemical categories should be combined
22 with a reliable documentation. Reporting formats are provided in Section R.6.2.6 in
23 Chapter R.6 of the [Guidance on IR&CSA](#).

24 **R.7.8.4.2 Remaining uncertainty for aquatic pelagic toxicity**

25 For the pelagic compartment generally there are more tests available than for other
26 environmental compartments. However, even for effect assessment on pelagic
27 organisms there will nevertheless normally often remain substantial uncertainty in
28 relation to estimating a concentration which will not affect structure and function of the
29 pelagic ecosystem (PNEC).

30 Often a few monospecies laboratory tests on pelagic organisms are extrapolated to a
31 PNEC value for the pelagic compartment which introduces uncertainty as it does not take
32 more complex interactions in the ecosystem into account. When only acute tests have
33 been performed, extrapolation of acute effect concentrations to chronic no effect
34 concentrations also implies uncertainty because short term data have only limited
35 predictive value for long term no effect concentrations (Ahlers *et al.*, 2006).

36 The more chronic studies are available the more likely sensitive species are represented
37 and hence the remaining is less. When the PEC/PNEC ratio is close to 1, it is preferable
38 to have a robust database with as many as possible chronic data on pelagic species
39 available, ideally including life cycle exposure.

1 The remaining uncertainty may in many cases be reduced when in an integrated
2 assessment is being made taking all available information into account (e.g. including
3 toxicity information on pelagic organisms from standard and non-standard tests, and
4 taking into account results from alternative test methods and non-testing information).

5 **R.7.8.4.3 Exposure considerations for aquatic pelagic toxicity** 6 **requirements.**

7 The information requirements for a substance as proposed by REACH may be modified
8 based on information on exposure (i.e. triggering or waiving of further testing). This
9 section considers triggering of further data requirements only (according to rules for
10 adaptation of the standard information requirements, Column 2). For waiving the specific
11 guidance on exposure based waiving should be consulted (Section R.5.1). In general,
12 further testing is proposed if the CSA indicates the need to investigate further the effects
13 on aquatic organisms, which implies long-term testing on fish and *Daphnia* for
14 substances covered by Annex VIII and Annex IX to REACH. The need to conduct further
15 testing may be triggered by the following cases, e.g.:

- 16 i. Results from a quantitative assessment, where $PEC/PNEC > 1$;
- 17 ii. Results from a qualitative assessment, where a possible risk should be
18 confirmed/rejected, e.g. when due to low water solubility of a substance,
19 short term toxicity tests do not reveal any toxicity, long-term tests are
20 performed;
- 21 iii. Information on a specific mode of action and unexpected sensitivity of a
22 group of organisms to the substance under investigation;
- 23 iv. Monitoring data showing occurrence of a substance in the aquatic
24 compartment.

25 If further tests are required, considerations provided in [Appendix R.7.8–2](#) regarding the
26 alternatives for vertebrate tests should be taken into account.

27 In the context of the PBT/vPvB assessment, a conclusion on the P and B properties
28 should be drawn before further T-testing is considered. If the substance is found to be
29 both P and B then a chronic toxicity study is required (except if the substance meets the
30 criteria for classification for carcinogenicity, mutagenicity, reprotoxicity or for chronic
31 toxicity according to Regulation 1272/2008 (CLP regulation); see section 1.1.3 points (b)
32 and (c) of Annex XIII to REACH). Normally, the testing sequence for a conclusion on T
33 based on chronic data is *Daphnia* and then fish. If the T-criterion is fulfilled by the
34 chronic algae or *Daphnia* data, a chronic fish test is not necessary and should therefore
35 not be carried out to avoid unnecessary testing on vertebrate animals.

36 **R.7.8.5 Conclusions for aquatic pelagic toxicity and integrated testing** 37 **strategy (ITS)**

38 Section [R.7.8.3](#) (information sources) presents an overview about the possibilities to
39 collect available or generate new information of different kinds (*in vivo* testing, *in vitro*
40 testing, non-testing). Section [R.7.8.4](#) gives guidance how the adequacy, i.e. reliability
41 and relevance, of every single piece of information from these different sources can be

1 judged and ranked. Section [R.7.8.5](#) is supposed to guide through the assessment of the
2 toxicity of the substance in cases where the total amount of available information is
3 suitable for regulatory decisions and in cases, where there are data gaps which have to
4 be filled.

5 The overall purpose of REACH is to provide a high level of protection for man and the
6 environment. To achieve this, the potential hazards associated with chemical substances
7 must be evaluated and to this end, information about the intrinsic properties of each
8 chemical is needed. At the same time, also according to the REACH regulation,
9 vertebrate animal testing must be restricted to the necessary minimum. Column 1 of
10 REACH Annexes VII–X specifies what is regarded as minimum information requirements.
11 Column 2 of Annexes VII–X as well as Annex XI specify possibilities to modify these
12 requirements. The prerequisite is the availability of other information that is a)
13 equivalent to the results that would be obtained by standard testing and b) adequate for
14 the three regulatory endpoints: Classification and Labelling, PBT assessment and
15 Chemical Safety Assessment. The equivalence and adequacy will have to be
16 substantiated by a *Weight of Evidence* approach, making best use of all existing
17 information.

18 *Weight of Evidence* is closely linked to *Integrated Testing Strategies* (ITS,), in that the
19 available evidence can help to determine the subsequent testing steps. Results from
20 these subsequent tests affect the *Weight of Evidence*, which leads to a new decision on
21 whether there is any need of further testing, and so on. ITS are particularly
22 characterised by flexibility and case specificity. No general ITS can be developed but a
23 case-by-case decision will always be necessary. Guidance on how to develop an
24 individual ITS has to focus on decision making criteria and underlying considerations
25 rather than on ready-to-use procedures.

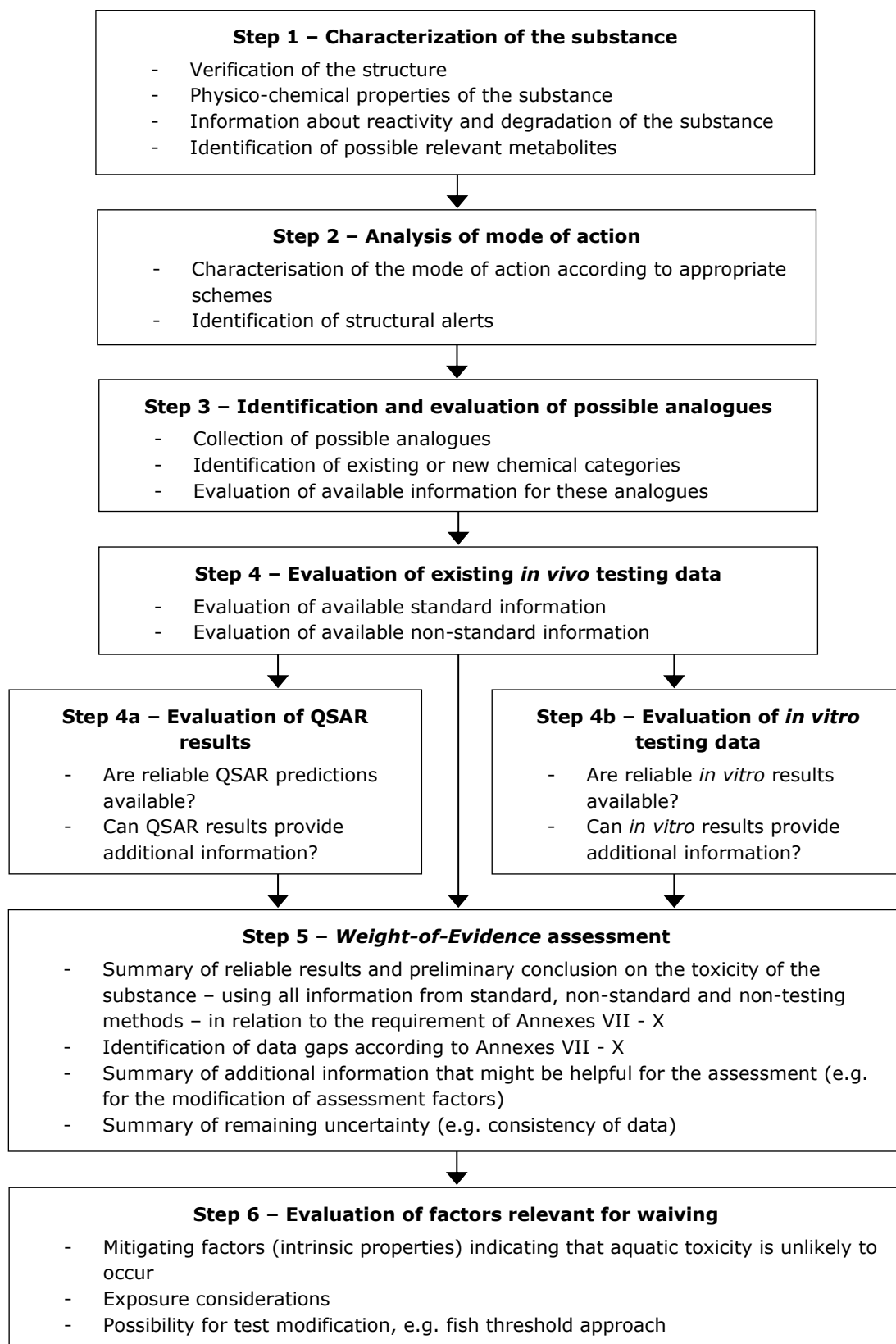
26 [Figure R.7.8—2](#) outlines a systematic approach how to use all available data on a *Weight*
27 *of Evidence* decision. It provides a step-wise procedure for the assessment of different
28 types of information, which might be helpful to come to an overall conclusion. The
29 scheme proposes a flexible sequence of steps, the order of which depends on the quality
30 and quantity of data and might be changed, e.g. for a substance with available *in vivo*
31 data of adequate quality, performance of steps 2, 3 and 4a and 4b might not be
32 necessary. On the other hand, steps 2 and 3 might be particularly helpful in cases of
33 varying data quality, and steps 4a and 4b in cases where not enough data are available.
34 Step 1, which is a collection of information on physico-chemical properties rather than an
35 assessment of available information, is a prerequisite for the further assessment of other
36 information. All steps are associated with three distinct activities:

- 37 i. the gathering of information (see detailed guidance in Section [R.7.8.3](#)),
38 ii. the evaluation of the quality of a distinct piece of information, e.g. a test report
39 or a QSAR result (see detailed guidance in Section [R.7.8.4](#)), and finally
40 iii. the overall assessment of all available information, which will be the focus of
41 this chapter. Additional guidance on generic aspects of a *Weight of Evidence*
42 approach is provided in Chapter R.4.

43 *Weight of Evidence* is a decision making activity aiming at concluding on toxicity of a
44 substance based on integration of information from different sources and various aspects

- 1 of uncertainty. It will often require expert judgement. To make this expert judgement
- 2 transparent and comprehensible it is essential that all information used, all steps carried
- 3 out in the evaluation process and all conclusions drawn are fully documented and
- 4 justified.

1 **Figure R.7.8—2 Suggestion for a *Weight of Evidence* approach**



2

3 * The scheme proposes a flexible sequence of steps, the order of which depends on the
4 quality and quantity of data and might be changed.

Step 1:

This step includes consideration of the following issues:

- Selection of the representative structure for the assessment (see Section R.6.1.7.3)

This step is essential for the assessment of the mode of action of a substance and for the potential use of non-testing techniques, e.g. QSAR models. In the case of multi-constituent substances, it may be necessary to regard two or more structures, if a single representative structure is not considered sufficient.

- Preliminary analysis of uptake and fate

A preliminary assessment of expected uptake, toxicity, and fate is performed on the basis of the information collected so far, i.e. analysis of the chemical structure, chemical and physical properties, degradation pattern, abiotic and biotic reactions involving the parent compound and other information as available.

It is important to evaluate at this stage the molecular structure and stability of the substance as well as identify the relevant metabolites. This is essential for the overall hazard assessment of a substance and especially for the evaluation of available *in vivo* tests (e.g. for the assessment if the test concentration was maintained during the test duration in cases where no analytical data are available) as well as for the use of QSAR results (in order to decide if the QSAR models should be used for a metabolite rather than the parent compound).

Further guidance is provided in Section R.6.1.7.4.

Step 2:

As described in Section [R.7.8.3](#) several schemes and programmes are available to derive information about the possible acute mode of action of a substance and to identify structural alerts. In Section [R.7.8.4](#) some help for the evaluation of the outcome of these methods is provided. For the overall assessment of the mode of action, results are available in terms of QSAR prediction reporting formats (QPRFs). In addition, information about the existence of structural alerts will be available (for more guidance see Section [R.7.8.4](#)).

The overall assessment of the acute mode of action should take the following questions into account:

- Does the chemical contain structural alerts?
- Is the characterisation of different tools consistent with respect to the mode of action?
- If the results of different classification schemes differ, is there a reasonable explanation?
- Can additional information be derived from the results?

In many cases it will be difficult to detect a specific mode of action such as inhibition of photosynthesis. Therefore the evaluation should focus on the question whether the

1 substance is likely to show baseline toxicity or if it is likely that it will exceed baseline
2 toxicity. The answer to this question will be helpful for the evaluation of QSAR
3 predictions as well as for the assessment of the reliability of experimental data and for
4 the assessment of the relative species sensitivity. For the assessment the following
5 considerations might be helpful:

6 Structural alerts

7 The presence of a structural alert gives a strong indication, that the toxicity of the
8 substance under investigation exceeds baseline toxicity with respect to the acute
9 endpoint under investigation (e.g. acute fish toxicity). On the other hand the absence of
10 a structural alert does not mean that the substance can be classified as baseline toxic.

11 Consistence of different schemes for the characterisation of the mode of action

12 As outlined in Section [R.7.8.3](#) and [R.7.8.4](#), the algorithm of different characterisation
13 schemes and the outcome (identification of specific mode of actions or identification of
14 excess toxicity) differs. Some advantages and disadvantages of the different schemes
15 are outlined in Section [R.7.8.4](#). With respect to the question if the substance shows
16 baseline toxicity, different tools should be combined.

17 It can be assumed that the characterisation of a substance as being baseline toxic is
18 reliable if different tools, based on different algorithms characterise the substance as
19 baseline toxic and if no structural alerts could be identified. For a high reliability it is
20 important that characterisation tools were included that are able to actively identify
21 baseline toxicity (e.g according to Verhaar, 1992). However it should be carefully
22 assessed if the overall assessment considers all parts of the molecule or if substructures
23 are present that were not evaluated.

24 Explanation of differences

25 If the reliability of the outcome of the assessment is low because the outcome of the
26 different schemes differs, the following considerations might be helpful:

- 27 • Can the difference be explained by different algorithms of the tools?
28 E.g. if the characterisation as baseline toxic is based on tools that do not
29 actively identify baseline toxicity a higher uncertainty can be assumed
30 because of the possibility that the substance simply can not be characterised
31 by the scheme (e.g. ECOSAR).
- 32 • Can the difference be explained because different parts of the molecule were
33 considered for the assessment?
34 In this case, the characterisation should generally be based on the most
35 conservative result (e.g. excess toxicity rather than baseline toxicity).

36 Additional information

37 Results of step 2 may help for the decision on choosing the appropriate test conditions
38 for a new test. E.g. If the substance is classified as reactive, it might be reasonable to
39 perform a semi-static or flow-through test rather than a static test.

40 Attention should be paid to the fact, that, at the current state of the art not enough
41 information is available for a characterisation of chemicals according to their chronic

1 mode of action. If tools become available and will be used for the assessment, it should
2 be clearly identified if the characterisation is valid for acute or chronic mode of actions.

3 The report of the outcome of the assessment should ideally include the following
4 information

- 5 • Description of the mode of action if possible, or description if the substance
6 can be characterised as baseline toxic or excess toxic.
- 7 • Reliability of the result
- 8 • Possible outliers and reasons for the outliers.

9 **Step 3:**

10 This step includes the following issues:

11 Identification of analogues for the verification of experimental and non-testing data

12 As the identification of possible analogues is a helpful tool for the assessment of the
13 reliability of existing data, the identification of analogues and categories might be
14 particularly helpful in cases of varying data quality.

15 In Section R.6.2.3 and in Section R.10.2.2.2 tools that might be helpful for identification
16 of analogues are described. Guidance how to conclude on possible analogues and
17 categories is provided in Section [R.7.8.4](#).

18 Analysis of substitutes for new tests

19 In certain cases, when information on a group is available it may be possible to
20 extrapolate results for studies that would otherwise be technically very difficult to
21 perform. I.e. for a substance where the hydrophobicity is just too high or solubility just
22 too low to maintain or measure a test concentration, studies on more soluble members
23 of the group could be used to predict the likely endpoint value.

24 **Step 4 – evaluation of *in vivo* data:**

25 Guidance on how to evaluate the quality of information from individual *in vivo* tests is
26 given in Section [R.7.8.4](#). The following paragraphs describe approaches for the overall
27 assessment of all available information from *in vivo* testing. This may include
28 consideration of the following issues:

29 How to deal with conflicting data?

30 When there is more than one set of data on the same species, (strain if known),
31 endpoint, duration, life stage and testing condition the greatest weight is attached to the
32 most reliable and relevant one. When there is more than one set of data with the same
33 reliability rating, it might be necessary to look into more detail at the study reports to
34 see whether a specific reason could explain the difference. If no explanation can be
35 found and the results are not more than one order of magnitude apart, they can be
36 harmonised by a geometric mean. If they are more than one order of magnitude apart,
37 this may be questionable. If the endpoint is critical for the outcome of the regulatory
38 decision, a repetition of the study may sometimes be the easiest and most efficient
39 solution, especially for non-vertebrate tests. A decision might also be possible on the

1 basis of additional available data, e.g. from studies of a lower reliability rating or from
2 non-testing methods, if these show a distinct tendency in support of a certain result.

3 Only secondary data sources available

4 Normally, data from a secondary source will lack several of the criteria required for a
5 sufficient reliability rating and can therefore not be considered for use in regulatory
6 conclusions. An exception to this can be made when these data have previously been
7 considered under widely accepted/ justified programmes which themselves contain
8 adequate review processes for data reliability.

9 Can available data, which are not adequate in themselves, provide sufficient information 10 when used in combination?

11 Some generic guidance on this issue is provided in Chapter R.4. This also mentions the
12 technique of *meta-analysis*, a statistical tool used for analysing the combined data from
13 multiple studies. Such pooling of data may increase the statistical power of certain
14 findings. It requires, however, that the studies from which data are pooled are
15 sufficiently similar with regard to critical parameters of test conditions, set-up,
16 endpoints, reporting etc.

17 There may be several studies available for the same test substance for the same
18 endpoint, which are deemed to not be fully reliable. However, when used collectively the
19 study results may indicate an effect at approximately the same concentration and time.
20 In these cases there could be justification for using all the studies collectively to
21 conclude on a specific endpoint.

22 Examples:

- 23 • Valid fish toxicity data are only available for a short exposure regime (e.g.
24 24h). Tests over 96h might be available, which cannot be judged as reliable
25 (e.g. because of poor documentation), but which provide information that the
26 main effect occurs within the first 24h. In this case the 24h value might be
27 used.
- 28 • Toxicity data are available for several time points from a 72h test. In this
29 case, the time-effect curve may allow extrapolation of the 96h value.

30 Do available data allow the derivation of a semi-quantitative result?

31 This consideration applies in relation to given effect values, for example:

- 32 • an LC₅₀ value cannot be calculated from an available acute fish tests because
33 no mortality was observed but the tested concentrations are above the EC₅₀
34 value determined for algae or *Daphnia* (retrospective threshold approach).
- 35 • an EC/LC₅₀ value cannot be derived, because test concentrations were either
36 too high or too low, but it can be stated that the LC₅₀ is either above or below
37 a specific regulatory relevant trigger value, such as C&L criteria or the T
38 criterion in PBT assessment.

39 The summary of the gathered information from the available *in vivo* studies should
40 contain the following:

- 1 • Results of standard tests available for all trophic levels?
- 2 • Reliable results of non-standard tests available for all trophic levels?
- 3 • Reliable results from aggregation of different studies available?
- 4 • Reliable half-quantitative results available?
- 5 • Description of additional information available, of the reliability of this
- 6 information and of its intended use?

7 **Step 4a:**

8 The overall assessment of QSAR results highly depends on the availability of additional
9 data such as information about the mode of action and experimental results for
10 analogues. Therefore if this step is used, information generated by step 2 and 3 should
11 ideally be available.

12 As described in Section [R.7.8.3](#), several QSAR models and programs including models
13 and expert systems are available in order to derive non-testing data. For the overall
14 assessment of the results, the outcome of the analysis of different QSAR models
15 (provided as QSAR prediction formats (QPRFs)) should be considered.

16 Step 4a aims at answering the following questions:

- 17 • Are reliable QSAR results available that can be used instead of experimental
- 18 data if data gaps are present?
- 19 • Can additional information provide a rationale for the waiving of tests?
- 20 • Can additional information provide a rationale for the performance of specific
- 21 additional tests?

22 Reliable QSAR results

23 In general, due to development of regulatory experience in use of non-testing data,
24 guidance at this point is rather tentative. The conclusion on the use of non-testing data
25 alone or in combination with experimental data on decision making will benefit from a
26 case-by-case discussion. It is foreseen to develop a manual of experience which could
27 continuously be updated, revised and improved by a suitable mechanism. This manual
28 will turn practical experience in the validity and acceptance of using (Q)SARs under
29 REACH into a continuously growing REACH QSAR guidance.

30 However the following considerations might be helpful for the conclusion:

- 31 • At the present (2006) higher confidence is based on QSAR models for acute
- 32 effects compared to QSAR models for chronic effects. Thus QSAR predictions
- 33 should focus on acute effects, while QSAR results for chronic effects will be in
- 34 most cases highly unreliable.
- 35 • In general higher confidence is provided by QSAR predictions based on
- 36 baseline toxicity compared to QSAR predictions based on specific modes of
- 37 action or chemical classes that show more than baseline toxicity. Thus if for a
- 38 substance a highly reliable classification as baseline toxic according to step 2

1 and a valid QSAR model where the substance fits into the applicability domain
2 is available the confidence in the prediction might be high.

- 3 • Reliability of the result may increase if a close analogue is available and
4 experimental results for this analogues fit to the QSAR prediction.

5 Waiving of tests

6 In general for most substances with a log K_{ow} between 1 and 6 a reliable QSAR model for
7 acute baseline toxicity will be available. Thus in most cases it will be possible to calculate
8 the baseline toxicity of the substance. If the acute effect concentration calculated for
9 baseline toxicity already triggers a regulatory decision (e.g. baseline toxicity <1 mg/L for
10 classification and labelling) this result might be used. But attention should be paid to the
11 fact that the real toxicity of the substance might be much higher due to a more specific
12 mode of action.

13 In addition, there could be cases where a substance was classified as having a specific
14 mode of action and a valid model for this specific mode of action is available. Although
15 the result of the prediction may not be reliable enough for a definitive risk assessment, it
16 might be possible to base the decision on the results as a worst case decision (see step
17 5).

18 The summary of the gathered information from the available QSAR models should
19 contain the following:

- 20 • Reliable results of QSAR predictions available?
- 21 • Other half-quantitative information available?
- 22 • Description of additional information available?
- 23 • Description of the reliability of the information and of its intended use?

24 **Step 4b:**

25 Available *in vitro* tests and their use for regulatory decision are described in Chapters R.3
26 and R.4. At the present (2006) no *in vitro* tests are available that can substitute *in vivo*
27 data. However *in vitro* data might be helpful to get further insight into the mode of
28 action of a substance:

29 Some permanent cell lines might express specific characteristics/functions of their source
30 tissue/organ. Their use for more specific modes of action has to be evaluated. Specific
31 modes of action are more likely to be detected with primary cell cultures. For example,
32 primary hepatocytes are used for studying metabolism, hepatotoxicity, genotoxicity and
33 vitogellin induction and isolated gill cells for studying the effect on the branchial
34 epithelium. Transfected permanent fish cell lines were used to detect estrogenic effects
35 of substances.

36 **Step 5:**

37 In step 5 all available data from the different steps should be integrated in the
38 assessment of the toxicity of the substance in order to understand the toxicity pattern of
39 the substance:

1 Experimental data (especially of standard tests) have the highest priority when
2 conclusions on the various endpoints (C&L, PBT assessment, PNEC derivation) have to be
3 drawn. Non-standard or *in-vitro* as well as non-testing data are important in cases where
4 standard experimental data are missing, are not reliable or inconsistent in order to verify
5 experimental data and avoid an assessment on the basis of invalid data (e.g. if two
6 acute fish toxicity tests give two different LC₅₀ values (e.g. 10 and 100 mg/L) and the
7 chemical under concern shows non-polar narcosis with an appropriate QSAR result of
8 LC₅₀ = 120 mg/L, more confidence might be given to the 100mg/L LC₅₀ value). Non-
9 testing data can be considered also as additional information to experimental data in a
10 *Weight of Evidence* approach even if experimental data exist. Moreover, they can be
11 used for elaboration of a test-design for higher-tier-tests or for a decision to perform
12 chronic tests instead of acute ones.

13 Ideally, at the end all available information (test data and non-testing information)
14 should be used for a comprehensive conclusion on the endpoint (multi task assessment).
15 This conclusion has to be substantiated and described in the text. The amount of
16 information necessary to draw such conclusions will definitely be different dependent on
17 the regulatory endpoint. For C&L, in certain cases limit tests may be sufficient as only a
18 decision has to be drawn whether the toxicity is below a certain trigger value, whereas
19 for derivation of the PNEC a quantitative figure has to be given. In the latter case it is of
20 particular importance to use all available information, as PNEC derivation means to
21 extrapolate from a few monospecies laboratory tests to maintenance of structure and
22 function of ecosystems. Especially the extrapolation from acute to chronic toxicity is
23 hardly possible. Analysis of a large number of validated data on new and existing
24 chemicals revealed that acute data have only limited predictive value for long-term
25 effects in aquatic ecosystems. The acute/chronic ratio correlates neither with acute
26 toxicity nor with baseline toxicity as modelled through log K_{ow} and no acute/chronic ratio
27 correlation is found across trophic levels, meaning that it is generally not possible to
28 conclude e.g. from *Daphnia* or algal ACR on fish ACR (Ahlers *et al.*, 2006).

29 In contrast to C+L and PBT assessment, which solely base on intrinsic properties, for
30 PNEC derivation also exposure-based decisions (PEC/PNEC ratio) have to be considered.
31 E.g. EC₅₀ values for alga and *Daphnia* are available. In addition QSAR calculations for fish
32 have been performed. From these data a high or low PEC/PNEC ratio has been derived.
33 In the first case a chronic fish test has to be considered. In the second case no additional
34 data are necessary.

35 **Step 6:**

36 Intrinsic physico-chemical properties

37 Column 2 of REACH Annexes VII and VIII contains the provision that acute studies do
38 not need to be conducted if there are mitigating factors indicating that aquatic toxicity is
39 unlikely to occur for instance if the substance is highly insoluble in water or the substance
40 is unlikely to cross biological membranes. On the other hand, REACH asks registrant to
41 consider long-term study when substance is poorly water soluble.

42 There is no scientific basis to define a cut off limit value for solubility below which no
43 toxicity could occur. There may be technical difficulties to perform the test, e.g.
44 sensitivity of the analytical method used for the determination of test concentration.
45 Such difficulties and proposed solutions should be clearly documented. Results from

1 tests above the limit of solubility should not be interpreted as pelagic toxicity, but as
2 confounded by physical effects. For further details see testing of difficult substances in
3 [Appendix R.7.8–1](#).

4 Equally, there is no scientific basis to define molecular characteristics that would render
5 a substance unlikely to cross biological membranes.

6 Thus no scientifically based cut off criteria for these mitigation factors can be provided at
7 the moment. Nonetheless, it might be possible to decide on a case-by-case basis, that
8 aquatic toxicity is unlikely to occur due to very low water solubility and unlikelihood to
9 cross biological membranes. Issues which may be considered in this regard are the
10 indicators used for low likelihood of a high bioaccumulation potential (Chapter R.11).
11 When such indicators are used in the context of triggering derogation from toxicity
12 testing on aquatic organisms however a more cautious approach should be used. The
13 reason is that indications of lack of a high bioaccumulation potential does not necessarily
14 imply lack of toxicity to aquatic organisms.

15 In any case any proposal to deviate from the standard testing requirements in reference
16 to this clause should be carefully justified. For poorly water soluble substances (e.g.
17 water solubility below 1 mg/L or below the detection limit of the analytical method of the
18 test substance) it should instead of an acute test be considered to perform a long term
19 test (REACH Annex VII and VIII, 9.1) bearing in mind any possibilities for waiving
20 (REACH Annex XI).

21 Threshold approach for toxicity testing in fish

22 This approach offers a possibility to significantly reduce the number of fish to be used in
23 acute aquatic toxicity testing when a test on fish is required. It takes into consideration
24 that only the lowest value of the acute toxicity in species of three trophic levels is
25 considered for regulatory purposes.

26 The approach was originally described as threshold/step-down approach by Hutchinson
27 *et al.* (2003) for pharmaceuticals. Several authors retrospectively evaluated acute
28 aquatic toxicity data of chemical substances (Jeram *et al.*, 2005; Hoekzema *et al.*, 2006)
29 by applying this approach. ECVAM and the ECB further developed the threshold approach
30 taking into account existing guidelines and reflecting the requirements for the limit test
31 (OECD TG 203, Annex V C.1). The ECVAM Scientific Advisory Committee (ESAC) has
32 endorsed the scientific validity of the threshold approach following the advice of the
33 ESAC peer review panel.

34 The approach is currently part of the rolling workplan for the OECD test guidelines
35 program 2006/2008 (Project 2.23: New Guidance Document on Application of the Step
36 Down Approach (or Upper Threshold Concentration) as a Limit Test for Acute Fish
37 Toxicity Testing).

38 With the lowest of the two EC₅₀ concentrations obtained for algae and *Daphnia*, (the
39 Upper Threshold Concentration, UTC), a limit test according to OECD TG 203, using 7-10
40 test and 7-10 control fish, is carried out. In case that no mortality is observed, no
41 further tests are carried out and the acute fish toxicity result (LC₅₀) is reported as
42 *greater than* (>) the UTC value. In case that mortality is observed, a full LC₅₀ test should
43 be performed.

1 The same principle could also be applied when instead of fish, fish embryos or early life
2 stages are used for toxicity testing.

3 From Integrated Testing to Integrated Assessment

4 When the *Weight of Evidence* approach has been finalised as described above, the
5 amount of validated information may in some cases largely exceed the minimum
6 information requirements of the Annexes of REACH and thus reduce the uncertainties
7 when extrapolating from monospecies laboratory tests to the structure and function of
8 ecosystems. As for PNEC derivation these uncertainties are to be covered by the
9 assessment factors it may be considered to use these factors in a more flexible way
10 according to the altered degree of uncertainty; (*it has to be mentioned that such*
11 *flexibilities on assessment factors are already foreseen, when the assessment is based*
12 *on Species Sensitivity Distribution (SSD) and on mesocosm as well as field studies and*
13 *also use of QSAR for narcotic mode of action, to be confirmed).*

14 Such a *multi-criteria assessment* should cover - beside the information mentioned above
15 - also:

- 16 • The number and representativity of species tested
- 17 • The quality of non-standard tests
- 18 • the time-dependence of the toxicity
- 19 • the steepness of concentration/effect curves
- 20 • Information from mammalian toxicity normally not used in standard
21 assessments.

22 Specific guidance on this approach with regard to potential reproductive or
23 developmental toxicity via endocrine modes of action is provided in [Appendix R.7.8—4](#).

24 At the end the derivation from the degree of uncertainty defined in the standard
25 situations and represented by certain assessment factors given by the Section R.10.3
26 has to be substantiated fully.

27 The proposal presented here is an optimal possibility to use *all available information* in
28 order to protect human health and the environment from hazardous chemicals.

29 **R.7.8.5.1 Concluding on suitability for Classification and Labelling⁵**

30 Environmental classification and labelling of a substance is generally based on data from
31 short-term tests for fish, invertebrates and algae. Information from other tests may be
32 used under the *safety net* provisions, i.e. in cases where substances do not fall under the
33 *core set of criteria* but on the basis of the available evidence concerning their toxicity
34 may nevertheless present a danger to the structure and/or functioning of aquatic

⁵ For more up-to-date information please see the *Guidance on the Application of the CLP Criteria*, section 4 and Annexes I and IV which have been updated in April 2012 in order to take into account the second Adaptation to Technical Progress (ATP) to the CLP Regulation.

1 ecosystems. There are no defined criteria for this classification; its possible application to
2 substances that cause adverse effects on development or reproduction is discussed in
3 [Appendix R.7.8–4](#).

4 Classification and labelling has to be performed for all substances registered in REACH.
5 The following strategy gives guidance how to classify a substance for the environment
6 based on its toxicity, if different levels of information are available (see also [Figure](#)
7 [R.7.8–3](#)).

8 As a first step all available information on substance has to be collected and evaluated as
9 described in Section [R.7.8.5](#) and Chapter R.3.

- 10
- 11 • If acute effect values for all three trophic levels are available, classify based
12 on the lowest effect value available and derive specific concentration limits
(M-factor) if relevant, i.e. toxicity <0.1 mg/l.
 - 13 • For substances with tonnages between 1 and 10 t/y, Annex VII requires acute
14 toxicity tests with invertebrates and algae/aquatic plants:
 - 15 a. If EC₅₀ for invertebrates and algae/aquatic plants are available
16 according to Annex VII, classify the substance based on the lowest
17 effect value; if, according to step 4a of Section [R.7.8.5](#), a reliable
18 QSAR result for fish is available or if additional information e.g. using
19 read-across can be provided, consider this value for the classification.
20 Specific concentration limits (SCLs) (M-factor) should be derived, if
21 relevant (GHS and the *Guidance on the Application of the CLP*
22 *Criteria*).
 - 23 b. If no acute data are available for invertebrates and/or algae/aquatic
24 plants, it should first be checked, if mitigating factors (water solubility,
25 molecular size) are justifiable:
 - 26 - if this is the case, no acute tests have to be performed for the
27 substance. *Safety net* classification based on fate data
28 (degradation and bioaccumulation) should nevertheless be
29 considered.
 - 30 - if the mitigation factors are not applicable, it is necessary to
31 perform an acute *Daphnia* and an acute algae test to fulfill the
32 requirements of Annex VII. If a reliable QSAR prediction for fish
33 can be made, consider this value for classification. SCLs (M-
34 factor) should be derived, if relevant.
 - 35 • For substances with tonnages >10 t/y, Annex VIII requires in addition an
36 acute fish test. However derogations from the standard information
37 requirements may be made if the provisions of REACH for this are fulfilled. In
38 the following, guidance is given to use available aquatic toxicity data on
39 classification and labelling:
 - 40 a. Acute toxicity data on invertebrates and algae/aquatic plants are
41 available and the EC₅₀ for at least on of these species is <1 mg/l. In
42 this case, no acute fish study is necessary for substances that are not

1 used in mixtures, as the available effect value(s) already trigger the
2 classification as Aquatic. Acute 1, H400. However, for substances used
3 in mixtures, an acute fish test might nevertheless be a prerequisite for
4 setting specific concentration limits (SCL, M-factor) for the
5 classification of mixtures containing the substance.

6 b. Acute toxicity data on invertebrates and algae/aquatic plants are
7 available and EC₅₀ for both species is >1 mg/l. In this case,
8 information on acute toxicity to fish is necessary for the judging
9 whether the aquatic toxicity to fish may warrant classification. Thus it
10 should be checked whether the calculation of an LC₅₀ for fish with a
11 reliable QSAR is possible or whether estimation is possible that fish
12 may be less sensitive than invertebrates and/or algae/aquatic plants
13 (see to Section [R.7.8.5](#)). Derive SCLs (M-factor) if necessary.

14 - if this is possible, this information can be used together with the
15 available effect data on invertebrates and algae/aquatic plants
16 for the purpose of classification.

17 - if this is not possible, an acute toxicity test with fish would
18 provide data which may be used for classification purposes.
19 However if alternative and adequate test methods are available
20 for acute fish toxicity they may be considered to be used
21 instead for classification (see [Figure R.7.8–3](#). E.g. a proposal to
22 use the fish embryo test (FET) as an alternative to the acute
23 fish toxicity test has been made and is currently under
24 evaluation in the OECD Guideline program (see [Appendix](#)
25 [R.7.8–4](#)). For further information, please see Guidance on the
26 Application of the CLP Criteria

27 - if data from suitable alternative test methods are not available,
28 a fish limit test following OECD TG 203 using as exposure
29 concentration the lowest EC₅₀ from acute tests on invertebrates
30 and algae/aquatic plants may be performed. If no mortality is
31 observed, this is an indication of fish not being more acutely
32 sensitive than invertebrates and algae/aquatic plants. Hence
33 classification can then be based on the lowest available EC₅₀-
34 value (for invertebrates and algae/aquatic plants). If mortality
35 occurs in the fish limit test, data from an acute fish toxicity test
36 according OECD TG 203 should be made available according to
37 the needs of the chemicals safety assessment and the LC₅₀
38 (fish) can then be used together with the EC₅₀-values for
39 invertebrates and algae/ higher plants as basis for classification
40 (GHS & Guidance on the Application of the CLP Criteria).

41 In the following, guidance is given for the specific cases, that instead of acute
42 invertebrate/fish tests long-term invertebrate/fish tests are available (Column 2 of
43 Annex VII and VIII). It is very likely that such cases do not commonly occur, and
44 therefore guidance is only given in the text and, not in the flow chart:

- 1 1. Substances with tonnages between 1 and 10 t/y (Annex VII): EC₅₀
2 algae/aquatic plants and *long-term* invertebrate instead of acute invertebrate
3 test are available.
- 4 2. Substances with tonnages ≥10 t/y (Annex VIII): EC₅₀ invertebrates and
5 algae/aquatic plant and *long-term* fish instead of acute fish are available.

6 For both points above:

- 7 a. At least one available EC₅₀ is <1 mg/l: In this case no further acute data are
8 necessary for the classification for substances that are not used in mixtures,
9 as this value triggers already the classification as Aquatic. Acute 1, H400.
10 However, for substances used in mixtures, further information on acute
11 toxicity might nevertheless be useful for classification purposes of substances.
12 The reason is that particular high acute toxicity may imply that a specific
13 concentration limit (SCL, M-factor) should then be set for the substance.
- 14 b. Available EC₅₀ >1 mg/l: In this case it should be checked whether the
15 derivation of an acute EC₅₀ from the long-term studies is possible (e.g. if raw
16 data of the study are available and at the tested concentration range included
17 immobilisation of parent invertebrates (OECD TG 202, part 2) resp. mortality
18 of fish (OECD 215) of >50 % the test parental animals). This effect value can
19 then directly be used for classification purposes together with available EC₅₀
20 values.
21 If this is not possible, it should be checked whether reliable predictions of EC₅₀
22 for invertebrates resp. fish with valid QSAR models are possible that can be
23 used for the classification of the substance. An additional option is to check
24 whether classification can be considered based on a grouping approach
25 relating to the species for which data are missing regarding acute toxicity. If
26 no estimation is possible of the acute toxicity for the aquatic organism with no
27 acute toxicity test data , then classification have to be considered based on
28 the available data on other aquatic organisms.

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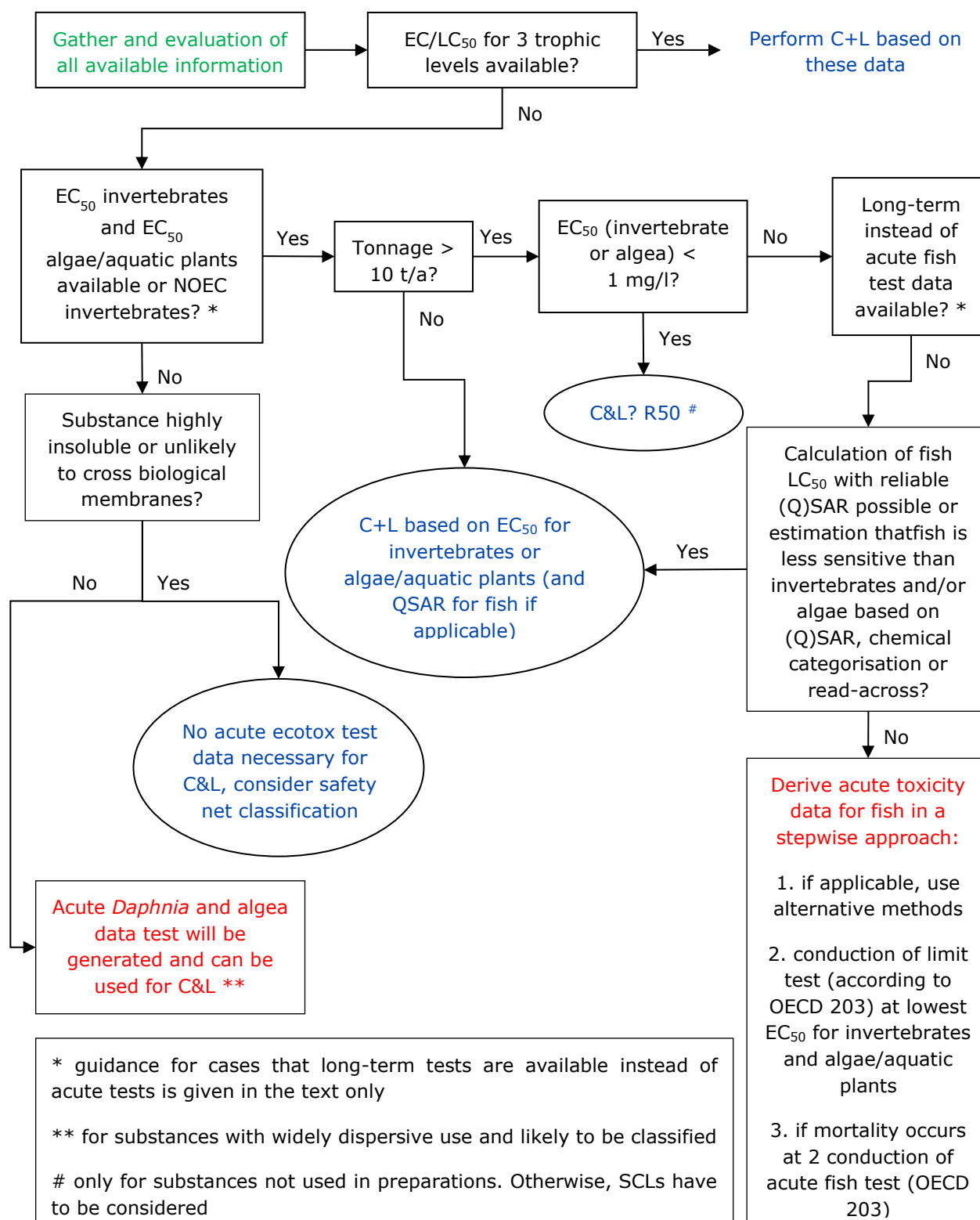
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1 **Figure R.7.8—3 Decision Scheme for Classification and Labelling⁶**

⁶ For more up-to-date information please see the *Guidance on the Application of the CLP Criteria*, section 4 and Annexes I and IV which have been updated in April 2012 in order to take into account the second Adaptation to Technical Progress (ATP) to the CLP Regulation.

1 **R.7.8.5.2 Concluding on suitability for PBT/vPvB assessment**

2 Guidance on the suitability for PBT/vPvB assessment is given in Chapter R.11 of the
3 [Guidance on IR&CSA](#).

4 **R.7.8.5.3 Conclusions on Chemical Safety Assessment (PNEC** 5 **Derivation)**

6 The Chemical Safety Assessment (CSA) is based on all available toxicity information. The
7 information should at least cover species of three trophic levels: algae/aquatic plants,
8 invertebrates (*Daphnia* preferred), and fish. The following strategy gives guidance how
9 to assess the pelagic toxicity of a substance for chemical safety assessment, if different
10 levels of information are available (see also [Figure R.7.8–4](#)).

11 A first sequence of considerations is primarily based on the availability of short-term
12 toxicity data as specified in REACH Annexes VII and VIII (combined). If results from the
13 hazard assessment or the risk characterisation indicate the need for further
14 investigations, long-term toxicity data will be considered in subsequent considerations.

15 **Short-term toxicity data**

16 1. Check available data from standard testing:

17 **Algae:** If a 72 hour ErC₅₀ value from a growth inhibition study according to OECD 201 or
18 a 96 hour ErC₅₀ value from a growth inhibition study is available this can be used directly
19 for PNEC assessment. If possible, it is recommended to calculate the 72 h growth rate
20 based on data from the test report of 96h tests.

21 **Invertebrates:** If a 48 hour EC₅₀ value from short-term toxicity testing on *Daphnia sp.*
22 according to OECD 202 or a NOEC/EC_x value from long-term toxicity testing on *Daphnia*
23 *sp.* according to OECD 211 or results from tests using equivalent test guidelines are
24 available, these can be used directly for PNEC assessment.

25 **Fish:** If an LC₅₀ value from short-term toxicity testing on fish according to OECD 203 or
26 a NOEC value from long-term toxicity testing on fish e.g according to OECD 215 (fish
27 juvenile growth test) or 210 (fish early life stage test) or OECD 212 (egg and sac-fry
28 test) or results from tests using equivalent test guidelines are available these can be
29 used directly for PNEC assessment.

30 2. Check other available data - standard testing data might be substituted by
31 one of the following:

32 **Algae:** The ErC₅₀ is the preferable and more meaningful value from a standard growth
33 inhibition (OECD 201) study. Where this is not available/ reported but an EbC₅₀ is
34 available/reported it should be considered to perform a new algae study, especially if
35 algae are the most relevant species for the effects assessment. If possible it is
36 recommended to calculate, the 72 h value based on data from the test report of 96 h
37 tests.

38 **Invertebrates:** A 24 hour EC₅₀ value from short-term toxicity testing on *Daphnia sp.*
39 according to OECD 202 but this should only be used in conjunction with other data (e.g.
40 on time-dependence of toxicity) as part of a *Weight of Evidence* approach.

1 **Other reliable experimental data** on algae/aquatic plants, invertebrates or fish (e.g.
2 data from non-standard studies or for non-standard organisms).

3 Reliable QSAR results (see Section [R.7.8.4.1](#) for evaluation of QSAR results).

4 Reliable read-across from available experimental data on a structurally related
5 substance.

6 An adequate value for growth inhibition of algae/aquatic plants or for short-term toxicity
7 in invertebrates or fish from any of the sources listed above may be used directly for
8 PNEC assessment.

9 3. Check possibilities for the prediction of relative species sensitivities:

10 The sensitivity of fish relative to invertebrates and algae might be predicted from one of
11 the following:

- 12 • Experimental data from standard studies
- 13 • Other experimental data (e.g. data from non-standard studies or for
14 non-standard organisms)
- 15 • Data generated with QSAR models
- 16 • Read-across from available experimental data on a structurally related
17 substance.

18 If there is compelling evidence, using these methods, to suggest that the fish value is
19 likely to be at least a factor of about 10 less sensitive than invertebrates or algae there
20 are no further requirements for acute fish testing. There may be other considerations for
21 testing, e.g. if a test result would help to build or improve a data base for a chemical
22 category. Consideration should also be given to needs for chronic testing e.g. whether
23 range finding data is needed to determine test concentrations etc.

24 4. Check possibilities for adaptation of the standard information requirements:

25 If there are mitigating factors, such as those specified in Section [R.7.8.5](#), indicating that
26 aquatic toxicity is unlikely to occur, studies on the growth inhibition of algae/aquatic
27 plants or the short-term toxicity in invertebrates or fish do not need to be conducted
28 (column 2, Annex VII and VIII).

29 5. If no adequate data is available and there are no mitigating factors indicating
30 that aquatic toxicity is unlikely to occur perform a growth inhibition study on
31 algae according to OECD 201 and a short-term toxicity study on *Daphnia* sp.
32 according to OECD 202 or a long-term toxicity study according to OECD 211
33 (According to column 2, Annex VII, a long-term study shall be considered if
34 the substance is poorly water soluble, i.e. solubility <1 mg/L, TGD 2003).
35 Alternatively risk management measures reducing exposure and hence risk
36 sufficiently might be considered.

37 6. Fish: Check availability of accepted alternative methods

38 If there is a need to generate new data on the toxicity in fish and an accepted alternative
39 method is available instead of *in vivo* fish testing perform the alternative test. At the

1 time of writing (2006) no alternative methods have been accepted as an alternative to
2 the *in vivo* fish study. A possible alternative, the fish embryo toxicity test, is currently
3 under evaluation in the OECD Guideline program (see Section [R.7.8.3.1](#) and [Appendix](#)
4 [R.7.8–2](#)).

5 7. Fish: Determine relative sensitivity

6 If there is no alternative to generating new toxicity data from *in vivo* fish testing a limit
7 test should be performed as described in OECD 203 using the lowest EC₅₀ from
8 invertebrates or algae. If no mortality occurs in the limit test that indicates that fish are
9 less sensitive than invertebrates or algae there are no further requirements for short-
10 term fish testing.

11 8. Fish: If mortality occurs in the limit test, perform a short-term toxicity study
12 in fish according to OECD 203 or a long-term toxicity study as appropriate (for
13 detailed guidance see below long-term toxicity testing) (according to column
14 2, Annex VIII, a long-term study shall be considered if the substance is poorly
15 water soluble, i.e. solubility <1 mg/L, TGD 2003). Alternatively risk
16 management measures reducing exposure and hence risk sufficiently might
17 be considered.

18 Normally a Fish Early Life Stage test (OECD 210) would be considered appropriate for
19 examining long-term fish toxicity. However, the fish, juvenile growth test (OECD 215)
20 (for substances with log K_{ow} <5) or egg and sac-fry stage test (EU Annex V C., OECD
21 212) (for substances with log K_{ow} <4) may also be considered. Specific guidance on the
22 consideration of available data on developmental or reproductive effects from non-
23 standard tests is provided in Chapter R.7.

24 9. Using the data specified in the preceding steps, the PNEC value can be
25 derived considering the results from all three trophic levels.

26

27 If the substance meets the criteria for classification into any⁷ of the hazard classes or
28 categories listed in Article 14(4) of the REACH Regulation, namely:

- 29
- 30 • hazard classes 2.1 to 2.4, 2.6 and 2.7, 2.8 types A and B, 2.9, 2.10,
31 2.12, 2.13 categories 1 and 2, 2.14 categories 1 and 2, 2.15 types A to
32 F;
 - 33 • hazard classes 3.1 to 3.6, 3.7 adverse effects on sexual function and
34 fertility or on development, 3.8 effects other than narcotic effects, 3.9
35 and 3.10;
 - 36 • hazard class 4.1;
 - hazard class 5.1;

⁷ Please see Part B, Chapter 8 on Scope of Exposure Assessment for hazard class(es) relevant for the environment.

- 1 • or is assessed to be a PBT or vPvB,
- 2 the chemical safety assessment must include an exposure assessment and a risk
3 characterisation.
- 4 These classes, categories and properties will henceforth be described as “Article 14(4)
5 hazard classes, categories or properties⁸”.
- 6 If the CSA indicates no risk, there are no further requirements for aquatic toxicity
7 testing. If the CSA indicates a need to investigate further effects on aquatic organisms
8 long-term toxicity testing shall be considered. These considerations apply in the same
9 way to all substances in quantities > 10 t.
- 10 A risk from CSA is indicated
- 11 • If $PEC/PNEC > 1$
- 12 • For substances with $\log K_{ow} > 3$ (or $BCF > 100$) and a PEC_{local} or
13 $PEC_{regional} > 1/100$ th of the water solubility.

14 **Long Term Testing**

- 15 1. Check available data from standard long-term testing:

16 **Invertebrates:** If a NOEC value from long-term toxicity testing on *Daphnia sp.*
17 according to OECD 211 or results from tests using equivalent test guidelines are
18 available these can be used directly for the refinement of the PNEC value.

19 **Fish:** If a NOEC value from long-term toxicity testing on fish according to OECD 215 or
20 210 or 212 or results from tests using equivalent test guidelines are available these can
21 be used directly for the refinement of the PNEC value.

- 22 2. Check other available data:

23 Standard testing data might be substituted by one of the following:

- 24 • Other reliable experimental data on aquatic invertebrates or fish (e.g.
25 data from non-standard studies or for non-standard organisms)
- 26 • Reliable QSAR results⁹
- 27 • Reliable read-across from available experimental data on a structurally
28 related substance

29 An adequate value for long-term toxicity in invertebrates or fish from any of the sources
30 listed above may be used directly for the refinement of the PNEC value.

- 31 3. Check possibilities for the prediction of relative species sensitivities:

⁸ In this context “properties” refers to PBT and vPvB.

⁹ Currently reliable QSAR models for chronic toxicity are rare and thus reliable QSAR results will be seldom available. However if QSAR models for chronic toxicity will be available in future they need to be evaluated equivalent to acute toxicity QSAR models as described in Section [R.7.8.4.1](#).

1 The sensitivity of fish relative to algae and invertebrates might be predicted from one of
2 the following:

- 3 • Experimental data from standard studies
- 4 • Other experimental data (e.g. data from non-standard studies or for
5 non-standard organisms)
- 6 • Data generated with QSAR models
- 7 • Read-across from available experimental data on a structurally related
8 substance.

9 If there is compelling evidence, using these methods, to suggest that the fish value is
10 likely to be at least a factor of about 10 less sensitive than invertebrates or algae there
11 are no further requirements for fish testing. There may be other considerations for
12 testing, e.g. if a test result would help to build or improve a data base for a chemical
13 category.

14 The same considerations as detailed above apply to the sensitivity of invertebrates
15 relative to algae and fish, i.e. if there is compelling evidence to suggest that the
16 invertebrate value is likely to be at least a factor of about 10 less sensitive than algae or
17 fish there are no further requirements for invertebrate testing.

18 4. If invertebrates are likely to be more sensitive than fish and algae or the
19 relative sensitivity of invertebrates cannot be predicted prepare a testing
20 proposal for a long-term toxicity study on *Daphnia sp.* according to OECD 211
21 for submission to the Agency. Alternatively risk management measures might
22 be considered.

23 5. If fish are likely to be more sensitive than invertebrates and algae or the
24 relative sensitivity of fish cannot be predicted prepare a testing proposal for a
25 long-term toxicity study on fish according to one of the below listed OECD
26 testing guidelines for submission to the Agency. Alternatively risk
27 management measures reducing exposure and hence risk sufficiently might
28 be considered.

29 Normally a Fish Early Life Stage test (OECD 210) would be considered
30 appropriate for examining fish toxicity. However, the fish, juvenile growth test
31 (OECD 215) (for substances with log Kow <5) or egg and sac-fry stage test
32 (EU Annex V C.) (for substances with log Kow <4) may also be considered.
33 Specific guidance on the consideration of available data on developmental or
34 reproductive effects from non-standard tests is provided in Chapter R.7.

35 Further possible methods for the refinement of the risk assessment, e.g. the
36 use of Species Sensitivity Distributions may be considered.

37

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1 **R.7.8.5.4 Overall conclusion**

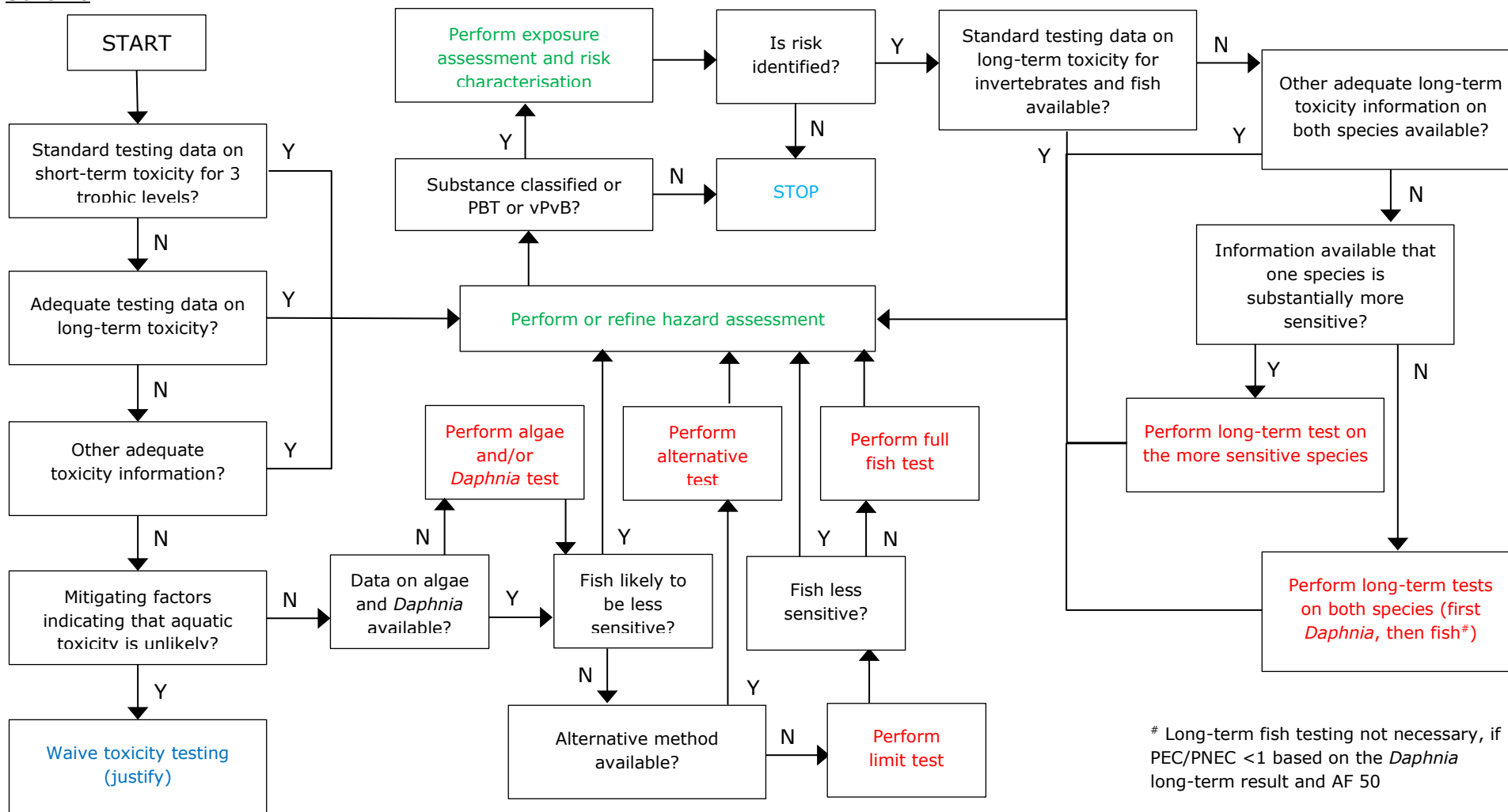
2 In Section [R.7.8.5](#) guidance is given on how to combine all gathered information in order
3 to understand the toxicity pattern of the substance and how to draw overall conclusions
4 on the different regulatory endpoints, Classification and Labelling, PBT/vPvB Assessment
5 as well as PNEC derivation. A major feature of these assessments will be flexibility and
6 expert judgement. The results have to be substantiated thoroughly and communicated.

7

8

Figure R.7.8—4 Decision scheme for the conclusion on chemical safety assessment (PNEC Derivation)

Scheme



1 For the conclusions on the different endpoints often variable amounts of information are
2 required with the consequence that the testing strategies proposed may differ
3 accordingly; e.g. for classification and labelling a limit test may be sufficient, whereas
4 the CSA assessment for the same substance requires a chronic fish test.

5 Therefore, to avoid unnecessary testing the different strategies should be compared
6 critically at the end of the exercise. Moreover, a few rules have to be followed:

7 PBT/vPvB assessment: chronic fish toxicity testing is generally only necessary, when the
8 P and B criteria are fulfilled (see further information in Chapter R.11 Chapter R.11 of the
9 *Guidance on IR&CSA*).

10 **Priorities for future research**

11 To perform substantiated conclusions on the different endpoints the available tools have
12 to be developed further. The following items among others should be considered for
13 further research:

- 14 1. Mechanistic approaches
 - 15 a. Develop knowledge of modes of action so that future CSAs can be
16 adapted to technical progress.
 - 17 b. Sub-lethal acute endpoints as predictors. Better use of information
18 from chronic toxicity tests as well as toxicokinetics to make predictions
19 of Mode of Action. Use data acquired to increase knowledge of
20 structural alerts.
- 21 2. Development, including validation and applicability domain description, of
22 QSAR models for chronic toxicity to pelagic and sediment organisms
- 23 3. Develop validated Test Guidelines for feeding studies on pelagic organisms
- 24 4. Improve knowledge of critical body burdens and compile databases and
25 establish and improve links to various classes of modes of action.
- 26 5. Improve read-across for freshwater to marine organism toxicity and increase
27 database for marine Phyla.
- 28 6. Improve understanding of how to read-across from Human Health and, if
29 possible, biodegradation data to environmental risk assessment (e.g. to
30 increase understanding of biotransformation and identification of relevant
31 metabolites).
- 32 7. Improve predictive techniques for extrapolating from laboratory to field
33 studies.
- 34 8. Consider how population dynamics can be included into ecotoxicology.
- 35 9. Develop & validate *in vitro* tests and based on this develop guidance how to
36 use in-vitro tests.
- 37 10. Develop Guidance how to use genomic information ("omics")
- 38 11. Develop guidance for multi-criteria assessment, that means how to use all
39 available information on derivation of a PNEC, including flexibility of
40 assessment factors.

41

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Appendices to Section R.7.8

- Appendix R.7.8—1** **Critical parameters for aquatic toxicity testing**
- Appendix R.7.8—2** Information and its sources: in vivo
- Appendix R.7.8—3** **Methodology for body burden approaches in aquatic effects assessment**
- Appendix R.7.8—4** **Assessment of available information on endocrine and other related effects**

1 **Appendix R.7.8—1 Critical parameters for aquatic toxicity testing**
2 **(Properties of substances and (tests) systems and other factors**
3 **influencing evaluation of aquatic toxicity)**

4

5 The following table summarizes the critical parameters that influence toxicity testing and
6 potentially testing strategy in the aquatic environment. The table is divided into two
7 main headings, Test related parameters, and Substance related parameters. Both are
8 useful for evaluating the validity of existing studies however, the Substance related
9 parameters also concern information that should be acquired prior to initiating new
10 studies. For more detailed information the reader is referred to OECD (2000) and
11 (ECETOC, 2003). This document gives some first guidance for inorganic compounds and
12 metals. More extensive guidance can be found in Van Gheluwe (2006).

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Table R.7.8—2 Critical parameters for aquatic toxicity testing

Parameter	Sub-parameter	Issue	Recommendation
Test related parameters			
General		Water quality	<p>All ecotoxicological tests should include information on key parameters influencing general water quality, indicating the fitness of the medium to support the organisms being tested and the likelihood that the exposure of the test substances occurred in a way that resembles the conditions in the environment. Frequency of measurement should also be indicated.</p> <p>Any single parameter which was out of the range indicated by the test method should trigger an in depth inquiry into the validity of the study and careful consideration of the relevance of the results.</p>
Oxygen			<p>Oxygen requirements depend on the organism with e.g. rainbow trout requiring very high levels (less than 50% could result in mortality) and certain benthic dwelling organisms capable of survival with almost negligible oxygen availability. However, in sediment tests, oxygen should always be measured close to the sediment as there may be much lower concentrations in the peribenthic layer than in the water column.</p> <p>In certain cases, (e.g. if biodegradation of the test substance or tertiary solvent is high) with non-volatile chemicals, aeration may be provided directly in the test system to increase oxygen concentration but for some species, (e.g. daphnids) this may lead to physical damage of the organisms and significant stress and should be avoided.</p>
pH			<p><i>Pelagic</i> – pH is generally acceptable within the range of 6.5 – 9 but this depends on the organism. Algae tests, for example, may reach a pH of 10 without any notable effect on the growth rate. However, in certain cases, notably ionising organics and metals, pH has an impact on speciation and thus toxicity. In such cases a decision needs to be made on the test strategy to be employed and the acceptable range of pH in the tests. Use of buffers or modified test strategies (e.g. reduction of initial cell numbers) can help to prevent major modifications of pH during the test.</p> <p><i>Sediment</i> – The pH of sediments may vary during the study. This may have an impact on the sediment dwelling organisms but also, for ionising substances, may change the ion exchange capacity of the substrate, increasing or decreasing bioavailability of the test substance and the pore water</p>

Parameter	Sub-parameter	Issue	Recommendation
			concentrations. Such changes should be monitored and controlled if possible.
Temperature			<p>Most Guidelines include temperature as a standard physical parameter as the organisms may be stressed or the validity of the results may not be achievable outside of the recommended limits (e.g. at less than 18°C it may be difficult to achieve the validity criterion of >60 juvenile daphnids per surviving adult within 21 days recommended in OECD 211). However, the change in temperature during the test is just as important. Fish are particularly sensitive to temporal temperature variations which can lead to temperature shock.</p> <p>In any test, spatial variation in temperature is also critical, and as climate rooms are often inconsistent, comprising both hot and cold spots, ideally oxygen should at least be measured in test systems with the greatest spatial separation. Any suggestion that systematic differences in temperature occurred between groups should lead to consideration of the validity of the study.</p>
Hardness/ Conductivity			The optimal ion requirement and composition varies from species to species and these are generally indicated in the test method. Hardness may influence the bioavailability of certain test substances (such as metals and metal compounds) and in these cases measurement of this parameter is relevant. For example, hardness is used in bioavailability models such as Biotic Ligand models (BLM) to describe competition effects for metals.
Alkalinity			Carbonate ions may alter speciation of metals. Hence for a proper understanding of metal speciation in the test medium knowledge on the alkalinity may improve our understanding of the test results.
Chlorides/ salinity			Salt effects may have a pronounced influence on test results. Most organisms tolerate chloride levels up to 500 mg/L. Above this threshold, depending on the organisms tested, osmotic stress may occur and bias the test results. For some metals like Ag the formation of chloride complexes may also influence the bioavailability.
NH ₃ /NH ₄			Ammonia is highly toxic and in dynamic equilibrium with the less toxic ammonium ion, is thus influenced by pH and to some extent temperature. Many species, including fish, directly excrete ammonia via the gills and faeces into the water and in static systems, or in high stock density tests, the ammonia concentration is likely to increase during the study. This may be a particular problem for sediment based systems which may be static for long periods of time. In studies where ammonia can cause a problem,

Parameter	Sub-parameter	Issue	Recommendation
DOC			measurements are generally included in the methodology, however for less validated methods it is worth considering whether the ammonia concentration is likely to have influenced the results. Dissolved organic carbon may be present in some studies, particularly those where natural water has been used. In such cases, DOC measurement is needed. Many adsorbing substances bind to DOC either ionically or hydrophobically and this may increase or decrease the bioavailability of the test substance. DOC is also a key parameter which is incorporated for most bioavailability models for metals. E.g. Biotic Ligand models using speciation models like WHAM VI.
TOC			<i>Sediment:</i> Total Organic Carbon (TOC) of sediments will vary depending on the type of sediment used in the study. This may have an impact on the sediment dwelling organisms but influence the bioavailability of both organic substances and metals/metal compounds..
AVS			<i>Sediment-metals:</i> Acid Volatile Sulfides (AVS) may influence the bioavailability of metals and metal compounds. AVS concentrations in artificial sediments are very low and quite often below detection limit. However, when field sediments are used AVS concentrations can be measured in order to allow a proper interpretation of test results of metal sediment toxicity data.
Substance related parameters			
Molecular weight and size			Molecular weight and size might influence the bioavailability and the uptake of the substance
Water solubility		General	Water solubility is an essential parameter in ecotoxicological testing and data should be available prior to any aquatic effects testing. Failure to do so could result in testing above the solubility limit leading to misinterpretation of the results. Poorly soluble substances are defined by OECD (2000) as substances with a limit of solubility <100 mg/l although technical problems are more likely to occur at <1mg/l as defined in TGD (1996). Very low water solubility (i.e. in the low µg/l range) could be used as a reason to significantly modify a standard test or to test non-pelagic organisms preferentially (see Table R.7.8–3 for more information). Whenever possible pelagic tests should be performed at or below the water solubility of the test

Parameter	Sub-parameter	Issue	Recommendation
			<p>substance in that medium.</p> <p>Tertiary solvents are often used in order to prepare stock solutions so that they can be further diluted to provide test solutions. Solvents used at the maximum allowed concentration (100 mg/l) will rarely increase the solubility of the test substance significantly but may lead to emulsion formation which could cause physical effects. Solvents should be avoided when possible for pelagic tests and if employed, care should be taken that they do not lead to an increase in biochemical oxygen demand BOD due to their (in some cases) rapid degradation. They are also employed to spike sediment and in such cases they are generally removed by air drying prior to use. However, traces of contaminants they contain may remain and furthermore, the organic solvent may have a negative effect on the sediment being used by redistributing or changing the organic carbon fraction. Typically solvents distribute the test substance onto the substrate in a way that does not occur in the environment and therefore the technique should be used with care.</p> <p>Dispersants have been employed in a similar way to solvents but are used more to achieve a stable dispersion than to dissolve the substance in the stock solution. OECD (2000) does not generally advocate the testing of dispersants unless they are natural properties of the substances under scrutiny (e.g. detergents or oil dispersing agents).</p> <p>OECD recommends the use of the column generator method for poorly soluble, solids which do not contain impurities with higher solubility than the test substance itself.</p>
		Multi-constituent substances (UVCBs)	Multi-constituent substances comprise a complex mix of individual substances with different solubilities and physico-chemical properties. In most cases, they can be characterised as a homologous series of substances with a certain range of carbon chain length/number or degree of substitution. Typically it is difficult to test and evaluate these substances. For further information see Table R.7.8–3
	Freshwater		Natural freshwater contains inorganic ions and DOC as well as suspended matter. Synthetic media contain many of the compounds found in natural freshwater but sometimes also other substances are employed to help buffer or maintain bioavailability of certain micronutrients. Standard solubility tests on the other hand are usually performed in deionised water. It is not unusual for measured values at maximum solubility in aquatic tests to differ from the solubility test result. Usually, the maximum solubility of a substance in synthetic medium is lower than the solubility test result indicates but this is

Parameter	Sub-parameter	Issue	Recommendation
			not always the case. This should be taken into account generally when testing is proposed close to the limit of solubility of the test substance but may be exacerbated for certain groups of chemicals e.g. chelates. For strongly adsorbing chemicals adsorption to suspended solids (SS) and for ionised organics such as surfactants, also binding to DOC may occur and the truly dissolved fraction may be difficult to evaluate. In such cases total load may be reported or used as a more applicable endpoint. In such cases it is important that the DOC and SS concentrations are known. For more information see Table R.7.8–3
	Marine		In the marine environment the salinity is so high that the solubility of most substances decreases and precipitation may occur by a process known as salting out. The decrease in solubility has been calculated as approximately 10-50% for neutral non-polar substances. A simple correlation for the salting out factor in seawater as a function of organic solute molar volume is to consider a reduction in solubility by a factor of 1.36 (ECETOC, 2001). For ionising substances, pH dependency should be known when the pH of seawater (approximately 8) is close to the pKa value. Testing considerations should be taken into account as above (freshwater).
	Poorly soluble	Physical effects	These usually apply only to difficult substances with very low solubilities. Certain substances may form mycelles when mixed with water even at very low concentrations (100 µg/l or less) or form a surface film covering aquatic organisms and potentially smothering them. Signs of these effects can be considered likely when daphnids are trapped at the surface in the test solutions (not always reported) or when there is a great variation in effect between replicates of the same concentration
Coloured substances			See Table R.7.8–3 for difficult substances
Sorption	General		Sorption/desorption tests provide information on K _{oc} (organic carbon normalised adsorption coefficient) and K _d (distribution coefficient) to the appropriate compartment. For many chemicals, such studies (or values of K _{oc} derived from K _{ow} QSARs) provide useful information on their likely partitioning behaviour in aquatic studies although it should be noted that for certain chemicals (notably surfactants and metals) the standard Freundlich isotherms derived from such studies are inappropriate.
	Neutral (hydrophobic) (expressed as log	Loss of substance from the test	Highly lipophilic substances (log K _{ow} >4, OECD 2000) are likely to pose problems during testing due to their expected low water solubility and tendency to stick to hydrophobic surfaces such as glassware, tubing, food and test organisms binding by van der Waals forces. Loss from the test solution may also

Parameter	Sub-parameter	Issue	Recommendation
	K _{ow})	system	be expected due to bioconcentration in the test organisms. For these reasons the organism stocking density should be low enough and the test system volume should be high enough so that the concentration of the test substance can be maintained throughout the studies. Naturally, static systems tend not to be appropriate for such substances. Flow-through is preferred when possible but achieving an adequate stock solution under such circumstances may be a challenge.
	Ionic	Loss of substance from the test system	May be positively or negatively charged organic or inorganic chemicals which bind to substrates of opposite charge e.g. cationically charged substances bind to negatively charged humic acids, clay, glassware, microorganisms etc; anionic compounds bind to positively charged Si, Al or Fe oxides). Adsorption mainly becomes an issue when test concentrations are below 1 mg/l. Attempts should be made to minimise binding sites and to saturate them if possible by pre-exposing them to similar concentrations of test chemical as those to be used in the study.
Surface active		Loss of substance from the test system	Surface active substances are a sub-set of the ionic substances mentioned previously and may be cationic, anionic, non-ionic or amphoteric. In all cases supplementary difficulties in estimating K _{oc} arise and the K _{ow} method cannot be used.
Ionising		Change of bioavailability with pH	Knowledge of the PK _a will allow prediction of the extent of ionisation of such substances in test water. As unionised organic species tend to be more hydrophobic than the ionised forms, the solubility and bioavailability of the substance may vary dramatically even between environmental extremes in pH. Consideration should be given to appropriate pHs (to be) used in the test as, solubility may be lower but toxicity may be higher in the unionised form than in the ionized form.
Degradation			OECD recommends testing parent compound for Disappearance Time 50 (DT50 >3) days, breakdown products for DT50 <1h and case-by-case basis for anything in between. A flow-through test is recommended for substances with a DT50 of 4 h as 50% of the nominal parent substance concentration can be maintained with 6 volume renewals per day. ECETOC (2003) and the TGD recommend to test parent substance with a DT50 as low as 12 h, as based on maximum half life allowing 80% maintenance of parent compound in flow-through system and >1% in short term test. However, this should be considered on a case-by-case basis depending on the

Parameter	Sub-parameter	Issue	Recommendation
			technical feasibility of performing such a study.
	Photodegradation		Photodegradation is the reaction of a chemical after absorption of light leads to an electronically excited state with increased reactivity and subsequent transformation. Photodegradation may be either direct (transformation of the substance by direct excitation) or indirect (transformation of another chemical due to transfer of energy from another photosensitive molecule. Kinetic photodegradation is determined experimentally.
	Hydrolysis		Hydrolysis is a common degradation route in the environment, where reaction of a substance with water with a net exchange of the X group with an OH at the reaction centre such that $RX + H_2O \rightarrow ROH + HX$. Hydrolysis is often dependent upon pH as the reaction is commonly catalysed by hydrogen or hydroxide ions. Hydrolysis kinetics are usually determined experimentally and should be used to consider the test type and whether parent or degradation product should be tested.
	Biodegradation		In the cases of readily biodegradable substances, biodegradation may be so fast that it is difficult to maintain test concentrations throughout the study. If such situations are likely then consideration should be given to regular cleaning or replacement of the test vessels during testing and preparation of stock solutions under sterile or near sterile conditions.
Volatility			<p>Vapour pressure is a measure of the equilibrium between the condensed and vapour phases of a substance.</p> <p>The Henry's law constant (H) for a substance is a measure of its equilibrium between an ideal solution phase and the vapour phase. As such it is a measure of the potential for a substance to be lost from solution by evaporation. As an approximation, if H is greater than $100 \text{ Pa}\cdot\text{m}^3/\text{mol}$, more than 50% of the substance could be lost from the water phase-in 3-4 hours (Mackay, 1992). If there is evidence that the substance may volatilise from the test solution during the study, steps should be taken to reduce the loss by using closed systems or reducing headspace.</p>

1 **References:**

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8 Difficult Substances

9 Valid aquatic toxicity tests require the test substance to be dissolved in the water
10 medium under the conditions recommended by the guideline, and the maintenance of a
11 bioavailable exposure concentration for the duration of the test. One or both of these
12 requirements may be difficult to achieve or measure in practice for some types of
13 substance – collectively referred to as *difficult substances*. This can affect both the
14 performance and interpretation of tests, and can be especially problematic when
15 considering existing data from older studies. Such data typically require expert
16 judgement to determine whether there is sufficient information in a test report for a
17 decision to be made on its validity, and also whether the result is suitable for regulatory
18 use.

19 [Figure R.7.8–5](#) indicates the thought processes that must be followed when considering
20 a difficult substance. In general, it is important that the composition of the substance is
21 as well-defined as possible. In some cases, it may be relatively straightforward to make
22 a decision on the use of the data. It should be remembered, however, that a substance
23 may be 'difficult' in several ways (e.g., it might be both a multi-constituent substance
24 and unstable), and each property can present complex challenges, even for experts. It is
25 therefore impossible to provide simple advice that can apply in every situation.
26 Nevertheless, the OECD has produced detailed guidance on how to adjust standard
27 methods for such substances (OECD, 2000) and guidance on data interpretation for
28 classification (OECD, 2001). [Table R.7.8–3](#) presents a summary of the main issues
29 identified in these important sources, which should be consulted for more detailed
30 information.

31 One of the key issues for difficult substances is the ability to quantify actual exposure of
32 the test organisms to the test substance. In general, test results should be expressed in
33 terms of mean measured concentrations as far as possible (though it is often useful to
34 quote both the measured and nominal effect concentrations). The following general
35 principles apply:

- 36 • For static, semi-static and flow-through tests, where the concentrations
37 remain within 80-120% of nominal, the effect concentrations can be
38 expressed relative to nominal or measured concentrations.
- 39 • For static tests, where the concentrations do not remain within 80-120% of
40 nominal, the effect concentrations should be expressed relative to the
41 geometric mean of the measured concentrations at the start and end of the
42 test.

- 1 • For semi-static tests, where the concentrations do not remain within 80-120%
2 of nominal, the effect concentrations should be expressed relative to the
3 mean concentration over the whole exposure period, calculated from the
4 geometric mean of the measured concentrations at the start and end of each
5 media renewal period.
- 6 • For flow-through tests, where the concentrations do not remain within 80-
7 120% of nominal, the effect concentrations should be determined and
8 expressed relative to the arithmetic mean concentration.
- 9 • For tests with chemicals that cannot be quantified by analytical methods at
10 the concentrations causing effects, the effect concentration can be expressed
11 based on the nominal concentrations. However this might result in an
12 underestimation of the toxicity and it should be justified why no quantification
13 by analytical methods is possible.

14 Where loss processes are very fast, the median of the concentrations that are measured
15 after the decline would be more appropriate as a surrogate for the mean exposure
16 concentration. In the absence of a suitable analytical method, a semi-static renewal or
17 flow-through regime may be necessary to ensure that exposure concentrations are in
18 line with target values.

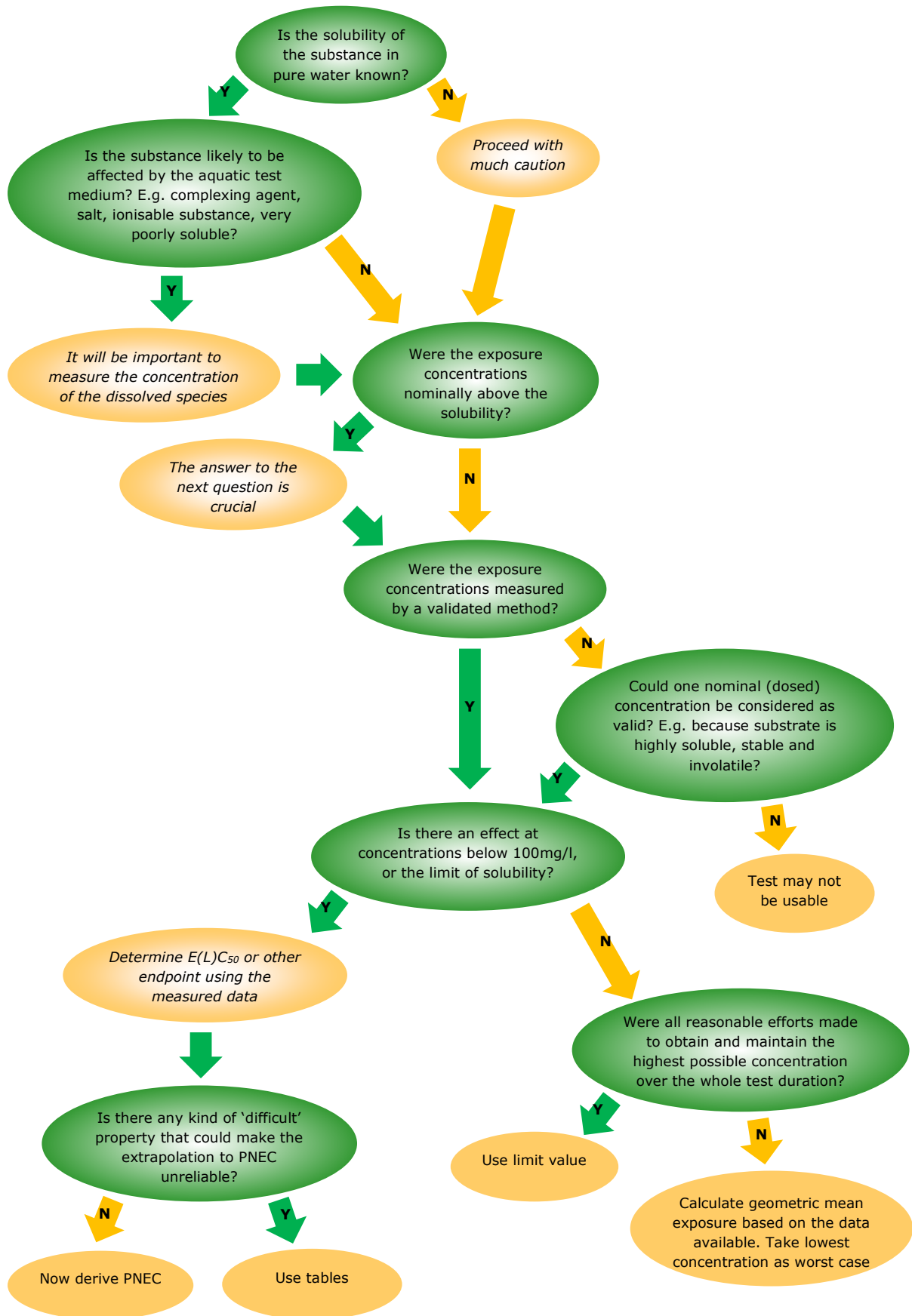
19 Where a measured concentration at the end of the exposure period is absent or where it
20 indicates that the substance is not detected, the validity of the test should be
21 reconfirmed. In order to calculate a mean exposure concentration, the final
22 concentration may be taken as the limit of detection for the method if the substance is
23 not detected. When the substance is detected but not quantified, it is good practice to
24 use half of the limit of quantification. Since there may be various methods for
25 determining that, the method selected to determine mean measured concentrations
26 should be made explicit in the reporting of test results.

27



- a. Polymers are not considered either, because they do not require registration in the initial phases of REACH implementation.
- b. Finally, some substances can contain impurities that can change in proportion and/or chemical nature between production batches. Interpretational problems can arise where either or both the toxicity and water solubility of the impurities are greater than the parent substance. This is not currently considered in this document, but is closely linked with the identity of the registered substance.

1 **Figure R.7.8—5 Considerations for difficult substances**



1 **Table R.7.8—3 Summary of difficult substance testing issues**

Difficult property	Potential problems with standard test procedures	Advice on interpretation	Possible refinements
The substance contains many constituents	<p>Multiple constituents may make analytical monitoring impossible.</p> <p>Differences in partitioning behaviour and water solubility between constituents can make it difficult to achieve a homogeneous solution by direct addition to the test medium (e.g. if some constituents are highly insoluble).</p> <p>This can also present interpretational problems. For example, it might not be possible to know which constituents have caused any observed adverse effects.</p>	<p>Figure R.7.8—6 presents a general pathway for considering such substances.</p> <p>If all the constituents of the substance are fully soluble in the medium across the range of test concentrations, standard test methods are appropriate. Some constituents may have individual properties (e.g. degradability, volatility, etc.) that require steps to be taken to control losses (see below).</p> <p>If the substance is only partially soluble, the constituents should be identified and the toxicity estimated using available information on them. For example, constituents that have structural and physico-chemical similarities should be grouped and treated as if the whole 'block' were one single compound. This approach has been developed for petroleum hydrocarbons in particular, and is known as the 'hydrocarbon block method'. (see draft ESR risk assessment for gasoline, and guidance from CONCAWE) Each 'block' is assembled on the basis of those properties that will influence the outcome of the PEC and PNEC calculations, i.e. usually octanol-water partition coefficient, Henry's Law constant, biodegradability and toxicity. The properties of each block may be estimated using a combination of non-testing methods for representative structures and the available measured data.</p> <p>If this is not possible, tests using water-accommodated fractions (WAFs) may be performed. The method used to prepare the WAF should be fully described in the test report, with evidence provided</p>	It maybe possible to analyse for one of the constituents during the test This approach was used in the UK CCRMP assessment of tetrapropenyl phenol, for one of the long-term aquatic studies.

Difficult property	Potential problems with standard test procedures	Advice on interpretation	Possible refinements
		<p>of attainment of equilibrium and its compositional stability over time if possible. WAFs are prepared individually and not by serial dilution of a single stock WAF. Solvents should also be avoided, and generator systems are not appropriate.</p> <p>Test data obtained with WAFs apply to the multi-constituent substance as an entity. The exposure is generally expressed as the 'loading rate' (mass to volume ratio of the substance to medium) used to prepare the WAF. The measured mass of test substance in the WAF can also be used (as a concentration).</p> <p>For test data obtained with WAFs the following apply if the substance contains constituents with a large range in water solubility: acute test data will correspond to the toxicity of the more soluble constituents, whereas chronic tests will reflect toxicity of the less soluble constituents.</p> <p>The acute lethal loading level (typically expressed as the E(L)L50) is comparable to L(E)C50 values determined for pure substances tested within their solubility range. It may therefore be used directly for classification. However, it cannot be used to derive a PNEC, since partitioning in the environment will make the comparison with a PEC meaningless. No Observable Effect Loading Rate (NOELR) values from chronic tests may be sufficiently low to be of the same order as the level at which most constituents are dissolved (or the PEC value), in which case they can be used for PNEC derivation. For PBT/vPvB assessment, results from WAF tests can only be used in a weight of evidence approach, e.g. in</p>	

Difficult property	Potential problems with standard test procedures	Advice on interpretation	Possible refinements
		<p>combination with modelling.</p> <p>If direct dosing of the medium can be achieved, e.g. by use of solvents within the limits allowed by the test guideline, the data will represent the hazard of the sum of the constituents and the E(L)C50 can be used to obtain a PNEC (though it will still not be known which constituents caused the effects).</p>	
<p>The substance is poorly soluble in the test medium (water solubility typically <1 mg/L)</p> <p>[similar problems can apply if the substance is simply difficult to analyse in the test medium]</p>	<p>Solubility may be difficult to determine and is frequently recorded as less than the analytical detection limit.</p> <p>It may be difficult to dissolve the substance in a test solution, and to maintain and verify concentrations.</p> <p>Toxicity may be observed at concentrations below the lowest measurable concentration.</p> <p>Results may be expressed in terms of nominal concentration, which might exceed the true dissolved concentration of the</p>	<p>Ideally, tests using appropriate dissolution techniques and with accurately measured concentrations within the range of water solubility should be used. Where such test data are available, they should be used in preference to other data. However, some techniques may present certain drawbacks, which must be taken into account. For example, the effect of any solvent needs to be determined, and solvents are not appropriate for mixtures where the use of the solvent can give preferential dissolution of one or more constituents (this may also apply to impurities). OECD (2000) provides more examples.</p> <p>The study report should be read carefully for indications of the presence of undissolved test material (e.g. droplets or surface layer). If this is the case and effects are observed, the results should be treated as invalid.</p> <p>Toxicity may be observed at concentrations nominally in excess of water solubility, or below the detection limit of the analytical method. Such data are not automatically invalid since the original</p>	<p>If the PNEC represents an upper limit, further testing may be required following risk assessment. This may require a more appropriate method or sensitive analysis (e.g. using radio-labelled test compound).</p> <p>For substances that are not acutely toxic at their limit of water solubility, the need for chronic testing has to be addressed if required by the risk assessment (provided the solubility is less than 100 mg/L).</p>

Difficult property	Potential problems with standard test procedures	Advice on interpretation	Possible refinements
	<p>substance in the test medium. This is a particular problem for older studies.</p> <p>Physical effects (e.g. entrapment) may occur if the test concentration is significantly above water solubility.</p> <p>Interpretation of partitioning behaviour can also be problematic where poor solubility in water and octanol may be compounded by insufficient sensitivity in the analytical method.</p>	<p>solubility estimate may be uncertain, and the solution may have been prepared appropriately (e.g. provided any undissolved substance is removed prior to testing). If physical effects are not obvious, then as a realistic worst case, the lowest effect concentration may be based on either the water solubility limit or detection limit of the analytical method, whichever is the lower.</p> <p>If no toxicity is expressed at concentrations up to the water solubility limit, judgement must be applied as to whether the result can be considered valid. The hazard should not be underestimated, and interpretation should stress the side of caution. Due account should be taken of the techniques used to achieve the maximum dissolved concentration. Where these are inadequate, the test should be considered invalid.</p>	<p>Substances that are not chronically toxic to aquatic organisms at their limit of solubility rarely need further consideration.</p> <p>If the substance to be tested is a member of a chemical category or if there are analogue substances, a possibility is to test the analogue substance that has a higher solubility and to extrapolate the results from this test to the substance in question. See ESR on Decabromodiphenylether and MCCP.</p>
The substance is ionisable or is a salt	The extent of ionisation may vary according to pH or the level of counter ions in the media, and relatively small changes may significantly alter the equilibrium between dissociated and non-dissociated	For hazard and risk assessment, the data must be obtained under environmentally relevant conditions. If the relevant dissociation constant (pKa value) for the ionisation process is available (required for substances supplied at 100 t/y), it should be compared with the pH reported in the test report to determine which chemical species were present. It may also be important to check which chemical species are monitored by any analytical	If the test substance ionises to a significant extent, it may be necessary to determine the toxicity of both anionic and cationic species.

Difficult property	Potential problems with standard test procedures	Advice on interpretation	Possible refinements
	<p>species.</p> <p>The dissociated and non-dissociated species may have different water solubilities and partition coefficients, and therefore bioavailability and toxicity. This in turn may cause the expression of different toxicities in freshwater and marine environments. For salts, both the anionic and cationic parts need to be considered.</p> <p>Solubility measurements for regulatory purposes are usually made in distilled water (pH 6-9), whereas the pH of test media is usually 7-8. This may significantly affect solubility, especially for substances with a pKa between 5 and 9.</p>	<p>method used. The absence of this information may make it impossible to interpret the results.</p> <p>The definitive test should be conducted at a pH consistent with the more toxic form of the substance whilst remaining within the range required to maintain the health of the control organisms. A stable pH is important to ensure that the balance between dissociated and non-dissociated forms of the substance is maintained.</p> <p>If no data are available on a salt, effects may be read-across from the anion or cation, whichever has the most toxic effect. If the effect is related to only one of the ions, the classification of the salt should use the effect concentration multiplied by the salt:ion molecular weight ratio.</p> <p>Where a substance causes a change in pH of the test medium (e.g. strong acids and bases), the pH should be adjusted to lie within the specified range for the test using a suitable technique. Care should be taken that this does not lead to removal of the substance (e.g. via sedimentation and/or degradation). The use of buffers can affect the test result, particularly for algae.</p> <p>Growth of algal test cultures can cause an increase of pH due to consumption of bicarbonate ions. Strategies for maintaining the concentration of these ions and therefore reducing pH shifts are</p>	<p>The solubility at different relevant pH should be determined, and pH and substance concentration should be analysed during the test. An example where this issue has been considered is in the ESR assessment of tetrabromo-bisphenol A.</p>

Difficult property	Potential problems with standard test procedures	Advice on interpretation	Possible refinements
<p>The substance is a complexing agent</p>	<p>Speciation may change in the presence of cations (e.g. Ca, Mg) and anions (e.g. SO₄, PO₄), co-complexing agents and other properties of the medium such as pH. This can influence solubility, bioavailability and toxicity of the substance. It may also reduce the availability of essential nutrients (which is only a secondary effect, not direct chemical toxicity).</p> <p>Adsorption to sediments is not easily predicted – adsorption is often strong for these types of substance.</p>	<p>discussed in OECD (2000).</p> <p>This issue is generally of most significance for aquatic plant growth tests. It is important to distinguish between chelated and non-chelated fractions in the test medium if possible, and the extent to which effects are a direct consequence of chemical toxicity (based on the bioavailable fraction). Speciation models may be helpful for this purpose.</p> <p>Data from tests in which complexation is judged to have had a significant bearing on the result are likely to be of questionable value for regulatory use.</p> <p>Compensatory adjustment to water quality parameters (e.g. the concentration of the essential ions) or the testing of an appropriate salt of the test substance may help to achieve a valid test result but protocols incorporating modifications to standard procedures should be validated and approved for use by the regulatory authority.</p> <p>The issue has arisen in the ESR assessment of EDTA, as well as for other complexing agents for the interpretation of algal studies.</p> <p>One approach used has been to run additional tests using enriched nutrient media, reduced substance concentration or addition of extra nutrients at test completion, and then extending the study. This is described in a paper presented at the 24th North American</p>	<p>If toxic effects are believed to be due to complexation, then this could be substantiated by measuring the complexation stability constants. Tests with provision of additional nutrient (to compensate for the complexed fraction) may be helpful in some cases. OECD (2000) suggests testing the substance in both standard algal growth medium and in a modified medium with a higher hardness, as well as the calcium salt. See UBA guidance too.</p>

Difficult property	Potential problems with standard test procedures	Advice on interpretation	Possible refinements
<p>The substance is surface active</p>	<p>Surfactants and detergents can form dispersions or emulsions in which the bioavailability is difficult to ascertain, even with careful solution preparation.</p> <p>Micelle formation can result in an overestimation of the bioavailable fraction even when "solutions" are apparently formed. This presents significant problems of interpretation.</p> <p>QSAR modelling is potentially very difficult since the Kow cannot usually be measured.</p>	<p>SETAC meeting: PW070 Effects of Iron and Micronutrient Metals on Algal Growth in the Presence of Chelators</p> <p>Toxic effect concentrations for dispersions and emulsions should be compared with the dispersibility limit (i.e., the limit at which phase separation takes place) or the critical micelle concentration (CMC) for a substance in water rather than with its water solubility limit. The bioavailable concentration does not change above the CMC, even at higher dosing levels. The highest test concentration should either be 1000 mg active ingredient/litre or the dispersibility limit/CMC, whichever is lower. In the ESR programme, a number of surfactants have been assessed - DODMAC and the alkylamines. For these, one of the main difficult properties was the strong tendency to adsorb on surfaces such as test vessels or organic material.</p> <p>If the E(L)C50 or NOEC(L) is below the CMC then the data can be treated in the usual way for classification and to derive a PNEC. If the substance is not toxic at the CMC, the CMC may be used as a NOEC to derive a precautionary PNEC. If a test has been conducted at concentrations above the CMC and shows effects, the effect concentration should be set as the CMC as a precautionary worst case, unless it is clear that physical effects have occurred.</p> <p>For sediments, it is very important to know the adsorption coefficient, preferably by measurement. An estimated Kow value, though of low reliability for surfactants, may be helpful. Guidance for the selection of appropriate methods of Kow measurement is provided in Section R.7.1.8 in Chapter R.7a of the Guidance on</p>	<p>Techniques for physically separating the test organisms from non-dissolved material, whilst maintaining contact with the water column, should be considered where physical effects are likely to be significant.</p>

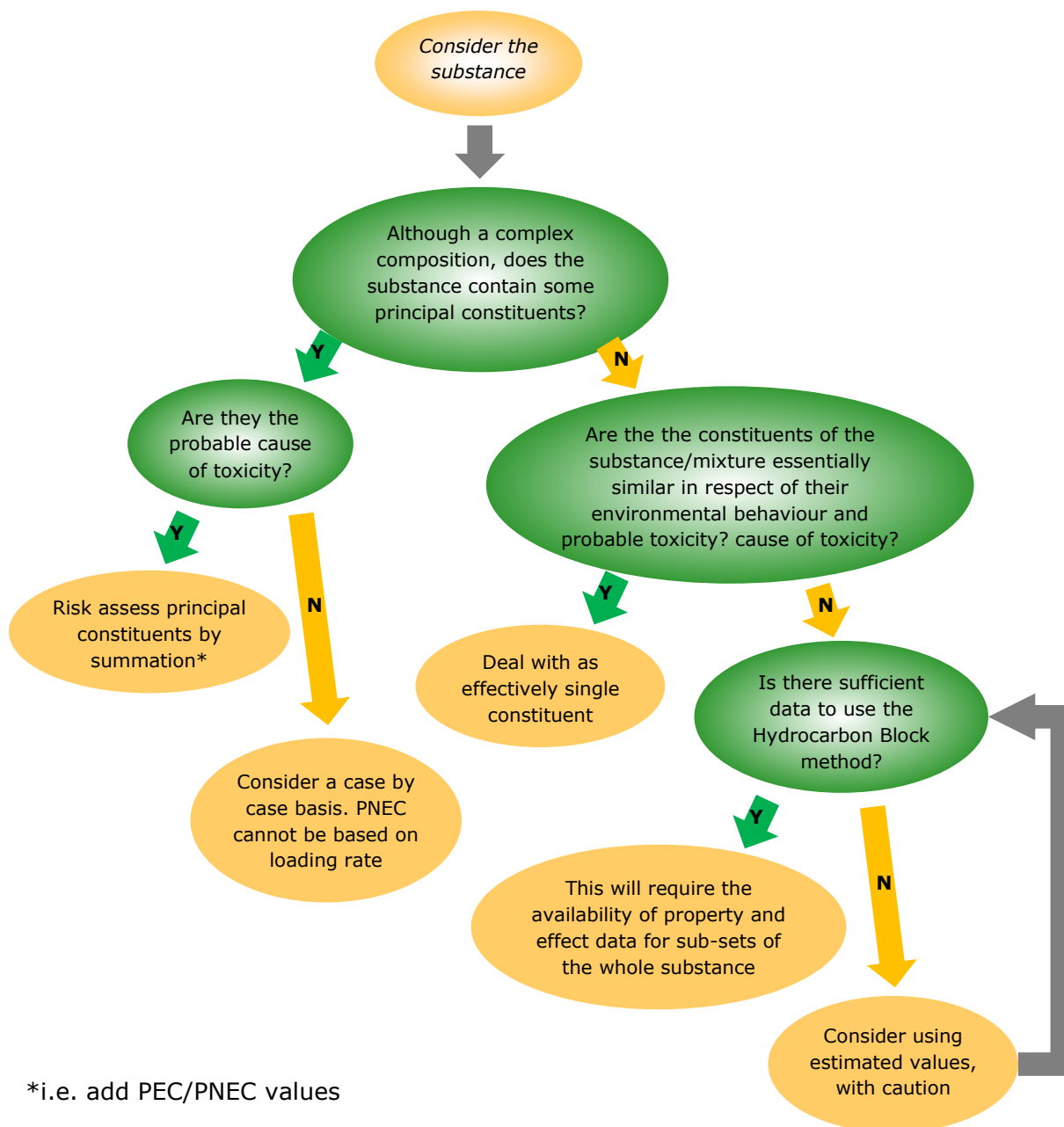
Difficult property	Potential problems with standard test procedures	Advice on interpretation	Possible refinements
<p>The substance is coloured</p>	<p>Absorption of light at relevant wavelengths may cause an indirect effect on aquatic plant growth by inhibiting photosynthesis.</p> <p>Strongly coloured solutions might make it difficult to observe effects in animals.</p>	<p>IR&CSA .</p> <p>Since the amount of light absorbed will vary with solution concentration, effects seen at high concentration are not necessarily environmentally relevant. The endpoint for regulatory use should therefore be based on direct toxic effects. If the test has not been designed to indicate whether any observed effects are caused by light limitation, then the results cannot be used.</p> <p>Early algal studies may not have considered the effect of light absorption, and therefore all observed inhibition was assumed to be inherent toxicity. In the late 90s an approach known as the ETAD method was used. This attempted to compare direct and indirect contact of the test substance with the algae, with the indirect contact used to evaluate light inhibition only. If the results of each experiment comparable, it was interpreted that effects were only due to light inhibition. Such a result could be used to justify not using the algae results for classification or PNEC derivation. More recently, the ETAD method has been thought to be too simplistic for this evaluation, and instead the Manual of Decisions has been updated with the modified algae / Lemna approach as detailed below:</p> <p>The following adjustments to the standard algae growth inhibition test, Annex V method C.3 (or OECD guideline 201) have to be applied:</p> <ul style="list-style-type: none"> • The irradiation (light intensity) should be in the highest end of the range prescribed in the method C.3 (or (draft revised) OECD guideline 201): $120\mu\text{E m}^{-2} \text{ s}^{-1}$ or higher. • The light path should be shortened by reduction of the volume of 	<p>OECD (2000) provides a number of options for performing algal tests with coloured substances. See latest MoD decision, left.</p> <p>The 7-d Lemna growth test avoids the problem since the fronds grow at the water surface.</p>

Difficult property	Potential problems with standard test procedures	Advice on interpretation	Possible refinements
		<p>the test solutions (in the range of 5 - 25 ml).</p> <ul style="list-style-type: none"> • Sufficient agitation (for example by moderate shaking) should be performed in order to obtain a high frequency of exposure of the algae to high irradiation at the surface of the culture. 	
<p>The substance is likely to be lost from the water column</p>	<p>The substance is volatile; losses may be particularly significant if the test is conducted using an open system. Vapour pressure, and more specifically the Henry's Law constant (H), are indicative of potential problems. If H is > 100 Pa.m³/mol, > 50% of the substance could be lost from the water phase-in 3-4 hours. Other factors in the test system may affect the rate of loss (e.g. vessel shape, aeration rate, etc). Volatilisation losses may also be significant for substances with H in the range of 1-10 Pa.m³/mol under vigorous mixing conditions. As a general rule vessels should be sealed during preparation and exposure and the headspace kept to a minimum. Problems with using sealed vessels are outlined in OECD (2000).). Within the ESR programme, two volatile substances styrene and 1,3 butadiene have been assessed. For the latter a combination of QSARs and read-across were used to provide environmental data; 1,3 butadiene was also a known CMR, so avoiding exposure of the substance to laboratory workers was an additional consideration. For styrene, due to it being readily biodegradable, an additional problem was degradation in ecotoxicity test media lowering oxygen levels for test organisms. Normally this could be mitigated providing additional oxidation, however due to the volatility this was likely to increase substance loss. In the studies steps were taken to minimise degradation (e.g. vessel sterilisation), as well using a flow-through system supported by analysis throughout the test. QSARs were also used to support the test results.</p> <p>The substance is adsorptive to glassware, food and/or test organisms. This property often accompanies low water solubility, since hydrophobic chemicals usually prefer to partition to organic phases (i.e. substances with a log K_{ow} >4 or bioconcentration factor >500). Where this occurs, the loss of concentration is usually rapid and exposure may best be characterised by the concentration at the end of the test. Other reasons for adsorption may be formation of ionic or hydrogen bonds negatively charged surfaces of the test vessel or the biological material. . The ESR assessments of tetrapropenylphenol and tris[2-chloro-1-(chloromethyl)ethyl] phosphate (TDCP) provide good examples where substance absorption was considered.</p> <p>The substance is unstable (i.e. degrades - abiotically, biotically or photolytically - or reacts) over the test duration. The loss may be so rapid that the substance itself cannot be tested, and/or specific degradation products may be formed that need consideration. See notes below on interpretation of exposure concentrations.</p> <p>The substance precipitates (e.g. because it has not truly dissolved despite the apparent absence of particulates, and agglomeration occurs during the test). In these circumstances, the L(E)C50 may be considered to be based on the concentration at the end of the</p>		

Difficult property	Potential problems with standard test procedures	Advice on interpretation	Possible refinements
	<p>test for classification purposes. Precipitation may occur as a result of degradation, e.g. an insoluble hydrolysis product or oxidation of test substance, other causes include complexation with media salts, pH change, oxidation. Note some substance may form an emulsion/dispersion, which can be tested as such – see surfactants discussion above.</p> <p>The substance bioaccumulates in the test organisms. This may be particularly important where the water solubility is low. The L(E)C50 may be calculated based on the geometric mean of the start- and end-of-test concentrations for classification purposes.</p> <p>It is necessary to determine whether appropriate methodology has been used (OECD (2000) describes a variety of methods to minimise the impact of these properties). In general, if test concentrations fall below 80% of nominal, measures should have been taken to reduce the decline for the test to be considered valid. This may require exposure regimes that provide for renewal of the test material (semi-static or flow-through conditions are preferred), and it is desirable that test concentrations are measured analytically at suitable time points throughout the test (for volatile, adsorptive unstable substances the latter is essential). These factors should be taken into account in deciding on the test data validity. It should be noted that semi-static and flow-through regimes may lead to an accumulation of organic debris and the development of excessive microbial populations. Test organisms may be stressed by cleaning. Special problems arise with respect to algal tests, which are generally static tests. Data providing an indication of the stability of the test substance under the test conditions may be derived from a review of existing data on the physical and chemical properties of the substance, or from a preliminary stability study (see OECD (2000) for further details). In the absence of analytically measured concentrations at least at the start and end of the test, no valid interpretation can be made and the test should be considered as invalid.</p> <p>Classification should account for the loss of the substance during the test, if relevant and possible. For example, if degradation occurs, it is necessary to determine whether it is the substance or the degradate that has been tested, and whether the data produced are relevant to the classification of the parent substance. Measured concentrations of the parent material and all significant toxic degradates are desirable.</p> <p>Where degradation is rapid (e.g. half-life < 1 hour), the available test data will frequently define the hazard of the degradation products since it will be these that have been tested. These data may be used to classify the parent substance in the normal way.</p> <p>Where degradation is slower (e.g. half-life > 3 days), it may be possible to test the parent substance and thus generate hazard data in the normal manner using a suitable renewal regime. The subsequent degradation may then be considered in determining whether an acute or chronic hazard class should apply.</p> <p>Where degradation rates fall between these two, testing of either parent and/or degradates should be considered on a case-by-</p>		

Difficult property	Potential problems with Advice on interpretation standard test procedures	Possible refinements
	<p>case basis.</p> <p>There may be occasions when a substance may degrade to give rise to a more hazardous or persistent product (this may be determined from preliminary tests or non-testing methods). Leaving a stock or test solution of the parent substance for a period equal to 6 half-lives of the substance will generally be sufficient to ensure that the medium contains only degradation products, which can then be used for toxicity testing. In these circumstances, the classification of the parent should take due account of the hazard of the degradation product, and the rate at which it can be formed under normal environmental conditions.</p> <p>For risk assessment, PECs and PNECs should relate to the same compound(s). For example, the degradation half-life should be compared with the duration of the emission and the time taken for the emission to reach the receiving water. If degradation is rapid, only the degradation product(s) are important. If the substance degrades slowly, the degradation products may be irrelevant for the risk assessment if they are less hazardous than the parent. Between these two extremes, the substance effectively becomes a multi-constituent mixture. Interpretation of the available data will need to carefully assign effects and properties between the original substance and the degradation products. Non-testing approaches may help this decision, especially where the properties of the products have not been measured separately. In some cases, two risk assessments might be needed to explore the significance of the possible extremes (i.e. 'no degradation' and 'complete degradation'). Such analysis can guide which further measurements are needed to understand the significance of the properties and the extent of risk.</p> <p>Some substances adsorb to organic matter more strongly than might be expected from Kow (e.g. aniline reacts irreversibly with sediment components). In addition, adsorption to inorganic matter (which is the major soil and sediment component) is important for several substance types, including metals, dyestuffs, cationic substances, complexing agents and surfactants.</p>	

1 **Figure R.7.8—6 Considerations for multi-constituent substances and**
 2 **mixtures**



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4 **References**

5 OECD (2000). Environmental Health and Safety Publications, Series on Testing and
 6 Assessment No. 23, Guidance Document on Aquatic Toxicity Testing of Difficult
 7 Substances and Mixtures, Environment Directorate, Organisation for Economic Co-
 8 operation and Development, Paris, September 2000.

9 OECD (2001). Environmental Health and Safety Publications, Series on Testing and
 10 Assessment No. 27, Guidance Document on the Use of the Harmonised System for the
 11 Classification of Chemicals Which are Hazardous for the Aquatic Environment,
 12 Environment Directorate, Organisation for Economic Co-operation and Development,
 13 Paris, March 2001.

1 **Appendix R.7.8—2 Information and its sources: *in vivo***
2 **Test guidelines**

3 a. Adopted OECD test guidelines for aquatic pelagic toxicity

Organism	F/S	Type of test	Test guideline (Year)	Exposure
Algae	F	Growth inhibition	201 (2006)	72 h
Lemna sp	F	Growth inhibition	221 (2006)	Up to 14 days
<i>Daphnia sp.</i>	F	Acute immobilisation	202 (2004)	48 h
<i>Daphnia</i>	F	Reproduction	211 (1998)	21 days
Fish	F	Acute toxicity	203 (1992)	96 h
Fish	F	Prolonged toxicity	204 (1984)	14 days
Fish	F/S	Early-life stage toxicity (FELS)	210 (1992)	30-60 days, species dependent
Fish	F/S	Short-term toxicity test on embryo and sac-fry stages	212 (1998)	Species dependent
Fish	F	Juvenile growth	215 (2000)	28 days

4 b. Proposed OECD test guidelines for pelagic aquatic toxicity

Organism	F/S	Type of test	Project nr	Exposure	Additional
<i>Daphnia</i>	F	Enhanced reproduction	2.8	21 days	Endocrine endpoints
Copepod	S	Reproduction and development	2.1	20-26 days	
Mysid	S	Life cycle toxicity	2.13	60 days or longer	Endocrine endpoints
Amphibian	F	Thyroid toxicity	2.19	21 days	Endocrine endpoints
Fish	F	Fish embryo toxicity	2.7	Up to 6 days	
Fish	F/S	Life-cycle toxicity	2.12	Species dependent	Endocrine endpoints
Fish	F	Sexual development	2.14	60-90 days	Endocrine endpoints
Fish	F	Screening	2.18	21 days	Endocrine endpoints

5 F = Freshwater organism S = Saltwater organism

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1 Project 2.1 Copepod Reproduction and Development

2 The test assesses the effect of chemicals on the development and reproduction of the
3 harpacticoid copepods *Nitocra spinipes*, *Tisbe battagliai*, *Amphiascus tenuiremis* and the
4 calanoid copepod *Acartia tonsa*. Newly hatched larvae (termed nauplia/metanauplia), are
5 exposed to the test substance added to water at a range of concentrations. The test
6 duration is usually 21 days, which is sufficient time for the control animals to reach
7 adulthood, first egg sac females to be isolated individually and produce 2 or 3 broods of
8 offspring. Effects on copepod development are measured by the time taken for nauplii to
9 attain the first copepodite stage. At the end of the test, the total number of living
10 offspring produced per parent animal alive at the end of the test is assessed. The
11 survival of the parent animals and time to production of first brood may also be
12 reported. Other substance-related effects on reproduction (e.g. brood size, time interval
13 between successive broods), and possibly intrinsic rate of increase, may also be
14 examined.

15 Project 2.7 Fish Embryo Toxicity test

16 Newly fertilised eggs of zebra fish (*Danio rerio*), fathead minnow (*Pimephales promelas*)
17 or Japanese medaka (*Oryzias latipes*) are exposed to chemicals for up to 48 hours. In
18 case of any evidence of delayed toxicity, the test duration should be extended to a total
19 of 6 days (for zebra fish), i.e. 2 days post hatch. The test is conducted in 24-well multi-
20 plates, 10 embryos/test concentration and at least 5 concentrations. 2 to 3 independent
21 runs per substance are recommended. After 24 and 48 hours incubation, four apical
22 endpoints are recorded as indicators of acute lethal toxicity: coagulation of fertilised
23 eggs, lack of somite formation, detachment of the tail bud from the yolk and lack of
24 heart beat. Embryos are considered dead, if one of these endpoints is recorded as
25 positive.

26 A comparable test was standardised (DIN 38415/A1; DIN 2001) in Germany and has
27 replaced the conventional fish test for routine whole effluent testing. An ISO guideline is
28 in the pipeline.

29 Project 2.8 Enhanced *Daphnia magna* Reproduction

30 This is an enhanced version of the “*Daphnia magna* Reproduction Test” (TG 211; OECD
31 1998). Offspring sex ratio and molt inhibition are evaluated as new endpoints. Sex of
32 neonates can be differentiated under a stereo microscope by the length and morphology
33 of the first antennae. Inhibition of molting can be examined by direct observation under
34 a stereo microscope, as well as by comparing number of molts and/or duration of inter-
35 molt period with that in control group(s).

36 Project 2.12 Fish Life-Cycle Test

37 A comparison of a proposed fish full-life cycle test (FLCT) and a proposed fish two-
38 generation test (TGT) is being conducted. This guideline is intended to be applicable to
39 the fathead minnow (*Pimephales promelas*), medaka (*Oryzias latipes*), sheepshead
40 minnow (*Cyprinodon variegatus*) and zebrafish (*Danio rerio*). The fish FLCT is initiated
41 with fertilized eggs (P generation or F0) and the fish are continuously exposed through
42 reproductive maturity, followed by assessment of the early development of the F1
43 generation. In contrast, in the fish TGT exposure is initiated with the mature male and

1 female fish (P generation or F0): eggs are collected and the F1 generation is evaluated
2 for embryo fertility, development, sexual maturation and reproduction.

3 Viability of F2 is also assessed. The main difference between FLCT and TGT is their
4 relative potential for evaluation of the effect of maternal transfer of chemicals, which is
5 evaluated once in FLCT and twice in TGT. Measurements are made of a number of
6 endpoints in both P and F1 generations reflective of the status of the reproductive
7 endocrine system, including the gonadal-somatic index (GSI), gonadal histology and
8 plasma or whole body concentrations of vitellogenin. Additionally, plasma sex steroids
9 (17 β -estradiol, testosterone, 11-ketotestosterone) and thyroid hormones (T3/T4) may
10 also be measured.

11 Project 2.13 Mysid Life Cycle Toxicity Test

12 This test evaluates reproductive fitness in two consecutive generations of mysids
13 (preferably *Americamysis bahia*), starting with newly-released (< 24 h) individuals of the
14 F0 generations and continuing until the first two broods (F2 generation) of the F1
15 generation. The overall test duration is normally 60 days or longer. Observational
16 endpoints include growth, time to maturity, time to first brood release, interbrood
17 duration, number and sex ratio of offspring.

18 Project 2.14 Fish Sexual Development Test

19 This method is an extension of the existing OECD Test Guideline 210 (1992) Fish, Early-
20 Life Stage (FELS) Toxicity Test, focusing on vitellogenin production and sexual
21 development, i.e. sex ratio as determined via histological examination of the gonads.
22 The test aims at detecting substances acting as estrogens, androgens or aromatase
23 inhibitors in organisms at a very sensitive stage of their life-cycle. The test starts with
24 fertilised eggs and lasts until sexual differentiation is completed (e.g. 60 to 90 days post
25 hatch, depending on the fish species).

26 Project 2.18 Fish-Screening Tests

27 Reproductively active male and female fish of fathead minnow (*Pimephales promelas*),
28 medaka (*Oryzias latipes*) and zebrafish (*Danio rerio*) are housed in groups of 5 males
29 and 5 females and exposed to test chemical for 21 days. Core endpoints as indicators of
30 endocrine disrupter activity are gross morphology (i.e., secondary sexual characteristics)
31 in sexually dimorphic species and vitellogenin levels in the serum or liver. Additionally
32 the spawning status is checked daily in all groups, and quantified in some. Examination
33 of gonadal histology is optional but will not be included as validated endpoint in the first
34 draft TG.

35 Project 2.19 Methods in Amphibians

36 The primary objective of the Amphibian Metamorphosis Assay is the evaluation of thyroid
37 system disrupting activities of the individual test compound. The post-embryonic
38 development (metamorphosis) of *Xenopus laevis* and the regulatory role played by
39 thyroid hormones (TH) during this process are well characterised. In the assay, exposure
40 of *X. laevis* tadpoles is initiated at developmental stage 51 and is continued for a total of
41 21 days. A sub-sampling of 5 tadpoles per treatment tank is performed at exposure day
42 7 for hind-limb length measurement. Tadpoles are exposed to 4 different concentrations
43 of a test substance and a dilution water control. During the exposure period, apical

1 morphological endpoints (developmental stage, hind limb length, whole body length) are
2 assessed for treatment-related deviations from normal development and histological
3 analysis of thyroid gland tissue is conducted with head tissue samples taken from test
4 organisms. Chemical exposure is via the aqueous route achieved using a flow-through
5 exposure regime.

1 **Other test guidelines - National and International standard methods and their publishers**

2 Acceptable alternatives to the OECD tests (described above) are also published by the OPPTS, EU (Official Journal), U.S. EPA and
 3 organisations such as ISO and ASTM:

Standard	Publisher	Web	Address
OECD	Organisation for Economic Co-operation and Development	http://www.oecd.org	OECD 2, rue André Pascal F-75775 Paris Cedex 16, France
EU	Official Journal of the European Communities. Annex V	http://ec.europa.eu/environment/arc_hives/dansub/annex_v_table_default_en.htm	European Chemicals Bureau TP582 Institute for Health and Consumer Protection Joint Research Centre, Ispra Site European Commission Via fermi 1 I-21020 Ispra (VA), Italy
ISO	International Organization for Standardization.	http://www.iso.org	ISO Central Secretariat: International Organization for Standardization (ISO) 1, rue de Varembé, Case postale 56 CH-1211 Geneva 20, Switzerland
AFNOR	Association Française de Normalisation	http://www.afnor.fr	AFNOR Association Française de Normalisation 11, rue Francis de Pressensé 93571 La Plaine Saint-Denis Cedex, France
ASTM	American Society for Testing and Materials	http://www.astm.org	ASTM International, 100 Barr Harbor Drive, PO Box C700, West Conshohocken, PA, 19428-2959 USA

Standard	Publisher	Web	Address
BSI	British Standards Institution	http://www.bsi-global.com	BSI British Standards 389 Chiswick High Road London W4 4AL, United Kingdom
CAN	Environment Canada, Environmental Protection Series	http://www.ec.gc.ca	Environment Canada, Inquiry Centre 70 Crémazie St. Gatineau, Quebec K1A 0H3, Canada
DIN	Deutsches Institut für Normung	http://www.din.de	DIN Deutsches Institut für Normung e.V. Stabsstelle Kommunikation Burggrafenstraße 6 10787 Berlin, Germany
DS	Dansk Standard (Danish Standard Association)	http://www.ds.dk	Dansk Standard Kollegievej 6 2920 Charlottenlund, Denmark
NEN	Nederlands Normalisatie-instituut	http://www.nen.nl/	NEN Postbus 5059 2600 GB Delft, The Netherlands
NS	Norges Standardiseringsforbund	http://www.standard.no	Standard Norge Postboks 242 1326 Lysaker, Norway
ÖNORM	Österreichisches Normungsinstitut	http://www.on-norm.at	ON Österreichisches Normungsinstitut Heinestraße 38 1020 Wien, Austria

Chapter R.7b: Endpoint specific guidance
Draft Version 4.0 (Public) – March 2017

Standard	Publisher	Web	Address
OPPTS	US-EPA Office of Prevention, Pesticides and Toxic Substances	http://www.epa.gov/oppts/index.htm	US-EPA Office of Prevention, Pesticides, and Toxic Substances MC 7101M 1200 Pennsylvania Avenue, N.W. Washington, DC 20460, USA
SFS	Suomen (Finland) Standardisoimisliitto	http://www.sfs.fi	Suomen Standardisoimisliitto SFS PL 116, 00241 HELSINKI, Finland
SIS	Standardiseringskommissionen i Sverige	http://www.sis.se	SIS, Swedish Standards Institute Sankt Paulsgatan 6 118 80 Stockholm, Sweden

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1 National and international standard methods / Guidelines (OECD, 1998):

Taxonomic group	Fresh/Salt	Species	Exposure time / endpoint	Guideline
Algae	F	<i>Selenastrum capricornutum</i>	Short-term / Growth rate (Chronic)	US-EPA 1994 (40 CFR 797.1060, 40 CFR 797.1075, 40 CFR 797.1050)
		<i>Scenedesmus subspicatus</i>		
		<i>Chlorella vulgaris</i>		
	S	<i>Skeletonema costatum</i>		
		<i>Thalassiosira pseudonana</i>		
		<i>Isochrysis galbana</i>		
	F	<i>Selenastrum capricornutum</i>	Short-term / Growth rate (Chronic)	ASTM (E 1218-90), FIFRA (§122-2), OECD (201), ISO (8692), NF (T90-304), DIN (38412 Teil 33), BS (6068: Section 5.10:1990), NEN (6506), SFS (5072), CAN (1/RM/25, 1992), EU (L 384 A Vol. 35 C.3)
		<i>Scenedesmus subspicatus</i>		
		<i>Chorella vulgaris</i>		
	S	<i>Skeletonema costatum</i> <i>Phaeodactylum tricornutum</i>	Short-term / Growth rate (Chronic)	ISO (10253), BS (91/56211 DC), NEN (6506), SFS (5072)
Macrophytes	S	<i>Champia parvula</i>	Short-term / Reproduction (Chronic)	US-EPA (EPA/600/4-87/028)
Plants	F	<i>Lemna gibba</i>	Short-term / EC50 (Acute)	ASTM (E-1415-91), FIFRA (§123-2), US-EPA (1994)(40 CFR 797.1160)

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Taxonomic group	Fresh/ Salt	Species	Exposure time / endpoint	Guideline
Crustaceans	S	<i>Mysidopsis bahia</i>	Short-term / LC50 (Acute)	ASTM (E 1463-92), FIFRA (§72-3 c), US-EPA (EPA/600/4-90/027), US-EPA (1994): 40 CFR 797.1930)
	S	<i>Artemia salina</i>	Short-term / LC50 (Acute)	US-EPA (EPA/600/4-90/027)
	S	<i>Penaeus aztecus</i> <i>Penaeus duorarum</i> <i>Penaeus setiferus</i>	Short-term / LC50 (Acute)	US-EPA (1994) 40 CFR Ch. 1 (7-1-92) Part 797.1970)
	S	<i>Nitocra spinipes</i>	Short-term / LC50 (Acute)	SS (028106), DS (2209), ISO/TC 147/SC 5/WG 2N56
	S	<i>Acartia tonsa</i>	Short-term / LC50 (Acute)	ISO/TC 147/SC 5/WG 2N56
	S	<i>Tisbe battagliai</i>	Short-term / LC50 (Acute)	ISO/TC 147/SC 5WR 2N56
	F	<i>Daphnia magna</i> <i>Daphnia pulex</i>	Short-term / LC50 (Acute)	US-EPA (EPA/600/4-90/027), OECD (202), ASTM (E 729-88a), FIFRA (§72-2), ISO (6341), NF (T90-301), DIN (38412 Teil 11), BS (6068: Section 5,1:1990), NEN (6501), ONORM (M 6264), SFS (5052), SS (028180), DS (ISO 6341), CAN (EPS 1/RM/11, 1990), US-EPA (1994) (40 CFR 797-1300), EU (L 384 A vol. 35 C.2)
	F	<i>Ceriodaphnia dubia</i>	Short-term / LC50 (Acute)	ASTM (E 1295-89), US-EPA (EPA/600/4-90/027)
	S/F	<i>Gammarus fasciatus</i>	Short-term / LC50 (Acute)	US-EPA (1994) (40CFR 795.120), CAN (EPS1/-

Taxonomic group	Fresh/ Salt	Species	Exposure time / endpoint	Guideline
		<i>Gammarus pseudolimnaeus</i> <i>Gammarus lacustris</i>		RM/26, 1992)
	S	<i>Mysidopsis bahia</i>	Long-term / survival, growth, fecundity (Subchronic)	US-EPA (EPA/600/4-87/028)
	S	<i>Mysidopsis bahia</i> <i>Mysidopsis bigelowi</i> <i>Mysidopsis almyra</i>	Long-term / life cycle (Chronic)	ASTM (E-1191-90), US-EPA (1994) (40 CFR 797.1950)
	F	<i>Daphnia magna</i>	Short-term / reproduction (Subchronic)	US-EPA (1994) (40 CFR 797.1330), OECD (202), NEN (6502)
	F	<i>Daphnia magna</i>	Long-term / life cycle (Chronic)	ASTM (E-1193-87), FIFRA (§72-4 C), US-EPA (1994) (40 CFR 797.1350)
	F	<i>Ceriodaphnia dubia</i>	Short-term / reproduction (Subchronic)	CAN (EPS 1/RM/21, 1992), US-EPA (EPA/600/4-89/001)
Insects (mosquito)	F	<i>Wyemyia Smithii</i>	Short-term / LC50 (Acute)	ASTM (E-1365-90), FIFRA (§142-1)
Rotifers	F	<i>Brachyonus</i>	Short-term / LC50 (Acute)	ASTM (E-1440-91)
Bacteria	S	<i>Photobacterium phosphoreum</i>	Short-term / Light emission (Acute)	NF (T90-320), DIN (38412 Teil 34), ONORM (M 6609), ISO/TC 147/SC 5/WG 1, CAN (EPS/1/RM/24,

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Taxonomic group	Fresh/ Salt	Species	Exposure time / endpoint	Guideline
	F	<i>Pseudomonas</i>	Short-term / Growth (Chronic)	1992) DIN (38412 Teil 8), NEN (6509 2e Ont w) ISO (DIS 10712. N133)
	F	<i>Activated sludge</i>	Short-term / respiration Inhibition (Acute)	OECD (209), EU (L 133 vol 31 p. 118), ISO 9509
Amphibians	F	<i>Xenopus</i>	Short-term / teratogenesis (Subchronic)	
Fish	F	<i>Brachydanio rerio</i> <i>Oncorhynchus mykiss</i> <i>Pimephals promelas</i> <i>Cyprinus carpio</i> <i>Oryzias latipes</i> <i>Poecilia reticulata</i> <i>Lepomis macrochirus</i> <i>Lepomis cyanellus</i> <i>Salmo gairdneri</i> <i>Oncorhynchus kistutch</i> <i>Salvelinus fontinalis</i>	Short-term / LC50 (Acute)	ASTM (E-729-88a), FIFRA (§ 72-1), US-EPA (EPA/600/4-90/027 + US-EPA (1994) 40 CFR 797.1440), OECD (203), ISO (7346-1-3), NF (T90-303+305), DIN (38412 Teil 15+20), BS (6068: Section 5,2; 5,3; 5,4:1985), SFS (3035+5073), DS (ISO 7346/1-3), CAN (EPS 1/RM/9), EU (L 383 A vol. 35 C.1)

Taxonomic group	Fresh/ Salt	Species	Exposure time / endpoint	Guideline
		<i>Carassius auratus</i> <i>Ictalurus punctatus</i> <i>Leuciscus idus</i>		
	F	<i>Poecilia reticulata</i>	Short-term / LC 50 (Acute)	NEN (6504)
	F	<i>Abassis macleayi</i>	Short-term / LC 50 (Acute)	OFR 54
	S	<i>Sheepshead minnow</i> <i>Fundulus heteroclitus</i> <i>Menidia sp.</i> <i>Gasterosteus aculeatus</i> <i>Lagodon rhomboides</i> <i>Leiostomus xanthurus</i> <i>Cymatogaster aggregata</i> <i>Oligocottus maculosus</i> <i>Citharichthys stigmaeus</i> <i>Paralichthys dentatus</i> <i>Paralichthys lethostigma</i> <i>Platichthys stellatus</i> <i>Parophrys vetulus</i>	Short-term / LC50 (Acute)	ASTM (E729-88a), FIFRA (§72-3 a), US-EPA (EPA/600/4-90/027), SS (028189), CAN (EPS 1/RM/10)

Taxonomic group	Fresh/ Salt	Species	Exposure time / endpoint	Guideline
Fish (cont)	F	<i>Clupea harengus</i> <i>Brachydanio rerio</i> <i>Pimephals promelas</i> <i>Cyprinus carpio</i> <i>Oryzias latipes</i> <i>Poecilia reticulata</i> <i>Lepomis macrochirus</i> <i>Salmo gairdneri</i> <i>(Oncorhynchus mykiss)</i>	Long-term / growth (Subchronic)	OECD (204), ISO (10229-1), BS (93/500175 DC)
	F	<i>Brachydanio rerio</i> <i>Oncorhynchus mykiss</i> <i>Cyprinus carpio</i> <i>Oryzias latipes</i> <i>Carassius auratus</i> <i>Lepomis macrochirus</i> <i>Pimephales promelas</i>	Short-term / egg and sac-fry stages (Subchronic)	OECD (212)
	S	<i>Menidia peninsulae</i>		

Taxonomic group	Fresh/ Salt	Species	Exposure time / endpoint	Guideline
		<i>Clupea harengus</i> <i>Gadus morhua</i>		
	F	<i>Pimphales promelas</i>	Short-term / early life stage test (Subchronic)	CAN (EPS 1/RM/22, 1992, US-EPA (600/4-89/001)
	F	<i>Oncorhynchus mykiss</i> <i>Salmo gairdneri</i> <i>Salvelinus fontinalis</i> <i>Esox lucius</i> <i>Pimephales promelas</i> <i>Catostomus commersoni</i> <i>Ictalurus punctatus</i> <i>Lepomis macrochirus</i> <i>Morone saxatilis</i>	Long-term / early life-stage test (Subchronic)	ASTM (E-1241-92), FIFRA (§72-4 a), US-EPA (1994) (40 CFR 797.1600), SS (SS 028193), NS (4763), SFS (5501), CAN (EPS 1/RM/28, 1992)
	S	<i>Opsanus beta</i> <i>Cyprinodon variegatus</i> <i>Menidia menidia</i>		
Fish (cont.)	F	<i>Mogunda mogunda</i>	Long-term / early life stage test (Subchronic)	OFR 52

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Taxonomic group	Fresh/ Salt	Species	Exposure time / endpoint	Guideline
	S	<i>Cyprinodon variegatus</i>	Long-term / survival, teratogenicity (Subchronic)	US-EPA (EPA/600/4-87/028)
	S	<i>Cyprinodum variegatus</i> <i>Menidia beryllina</i>	Long-term / survival, growth (Subchronic)	US-EPA (EPA/600/4-87/028)
	F	<i>Salmo gairdneri</i> <i>Pimephales promelas</i> <i>Brachydanio rerio</i> <i>Oryzias latipes</i> <i>Oncorhynchus kisutch</i> <i>Oncorhynchus tshawytscha</i> <i>Salmo trutta</i> <i>Salvelinus fontinalis</i> <i>Salvelinus namaycush</i> <i>Esox lucius</i> <i>Catostomus commersoni</i> <i>Lepomis macrochirus</i> <i>Ictalurus punctatus</i>	Long-term / hatching, survival, growth, malformations, behaviour (Subchronic)	OECD (210)
	S	<i>Jordanella floridae</i>		

Taxonomic group	Fresh/ Salt	Species	Exposure time / endpoint	Guideline
		<i>Gasterosteus aculeatus</i> <i>Cyprinodon variegatus</i> <i>Menidia menidia</i> <i>Menidia peninsulae</i>		
Echinoderms	S	<i>Arbacia punctulata</i>	Short-term / fertilization (Subchronic)	US-EPA (EPA/600/4-87/038), CAN (EPS1/RM/27, 1992)
Mussels	S	not specified	Short-term / LC50 (Acute)	ASTM (E-724-89), FIFRA (§72-3 b)
	S	<i>Crassostrea virginica</i>	Short-term / shell growth (Acute)	US-EPA (1994)(40 CFR 797.1800)

1 * Short-term < 14 days, Long-term > 14 days

2

1 Databases

2 For the endpoint of aquatic toxicity Ecotoxdatabase, IUCLID, ECETOC database and N-
3 class database may be useful sources of information. Other useful sources of information
4 can be found through existing risk assessment or data evaluation programs such as
5 ESIS, HERA and the OECD HPV program (SIDS). It is recommended that you consult the
6 original scientific paper to ensure an understanding of the context of the data retrieved
7 from the databases.

8 EAT (European Centre for Ecotoxicology and Toxicology of Chemicals (ECETOC) Aquatic
9 Toxicity database (<http://www.ecetoc.org>)

10 The ECETOC Aquatic Toxicity (EAT) database (ECETOC, 1993) contains more than 5450
11 entries on almost 600 chemicals, provides the most comprehensive compilation of highly
12 reliable ecotoxicity data published in the scientific press in the period 1970 - 2000. The
13 EAT 3 database is available as an Excel spreadsheet. For each entry there are 32 fields
14 of information on the substance, test species, test conditions, test description, endpoint,
15 results and source references. All the references are held at ECETOC; ECETOC AISBL,
16 Avenue Edmond Van Nieuwenhuysse 4 Bte 6, B-1160 Brussels, Belgium.

17 Ecotoxdatabase (<http://www.epa.gov/ecotox/>)

18 The database is maintained by the US-EPA and provides single chemical toxicity
19 information on aquatic and terrestrial life for about 8400 chemicals. Peer-reviewed
20 literature is the primary source of information encoded in the database. Pertinent
21 information on the species, chemical, test methods, and results presented by the
22 author(s) are abstracted and entered into the database. Another source of test results is
23 independently compiled data files provided by various United States and International
24 government agencies. Prior to using ECOTOX, you should visit the "About ECOTOX/Help"
25 section of this Web Site.

26 ESIS (European chemical Substances Information System)
27 (<http://esis.jrc.ec.europa.eu/>)

28 ESIS is an IT System which provides you with information on chemicals, related to:

- 29 • EINECS (European Inventory of Existing Commercial chemical Substances),
- 30 • ELINCS (European List of Notified Chemical Substances),
- 31 • NLP (No-Longer Polymers),
- 32 • HPVCs (High Production Volume Chemicals) and LPVCs (Low Production
33 Volume Chemicals), including EU Producers/Importers lists,
- 34 • C&L (Classification and Labelling), Risk and Safety Phrases, Danger etc...,
- 35 • IUCLID (International Uniform Chemical Database) containing information on
36 approx. 10 500 different substances on the effects on human health and the
37 environment.
- 38 • Priority Lists, Risk Assessment process and tracking system in relation to
39 Council Regulation (EEC) 793/93 also known as Existing Substances
40 Regulation (ESR).

1 HERA (Human and Environmental Risk Assessment) (<http://www.heraproject.com>)

2 HERA is a voluntary industry programme initiated by A.I.S.E. and CEFIC to carry out
3 focused risk assessments of the ingredients of household cleaning and detergent
4 products.

5 HSDB (Hazardous Substances Data Bank) (<http://toxnet.nlm.nih.gov>)

6 This is a toxicology data file on the National Library of Medicine's (NLM) Toxicology Data
7 Network (TOXNET®). It focuses on the toxicology of potentially hazardous chemicals. It
8 is enhanced with information on human exposure, industrial hygiene, emergency
9 handling procedures, environmental fate, regulatory requirements, and related areas. All
10 data are referenced and derived from a core set of books, government documents,
11 technical reports and selected primary journal literature. HSDB is peer-reviewed by the
12 Scientific Review Panel (SRP), a committee of experts in the major subject areas within
13 the data bank's scope. HSDB is organized into over 5000 individual chemical records.

14 N-class database (<http://www.kemi.se/en/Content/Databases/>)

15 The steering group for the Nordic Council of Ministers project on Environmental Hazard
16 Classification is responsible for the continuous updating of the N-Class database. The
17 database contains substances that have been discussed by the EC-Commission on the
18 Classification and Labelling for environmental effects. Substance specific data, gathered
19 from various documents that have been discussed at Commission working group
20 meetings on environmental effects (mainly covering ecotoxicity), may be found in the N-
21 Class database.

22 OECD Integrated HPV database (<http://webnet.oecd.org/hpv/ui/Default.aspx>)

23 This database tracks all High Production Volume (HPV) chemicals through the process of
24 investigation in the OECD programme on the Investigation of Existing Chemicals. Once
25 agreed in the OECD, it shows the results of assessments as well as the actual reports
26 and background information behind them. The database contains the list of HPV
27 chemicals together with any annotations on each chemical provided to the Secretariat by
28 Member countries, there are links to relevant documents.

29 When making the first evaluation of an existing chemical, a minimum set of data is
30 necessary to determine its potential hazards. To ensure that such data are available,
31 OECD developed the SIDS (Screening Information Data Set). The SIDS outlines the
32 minimum data elements essential for determining whether or not a chemical requires
33 further investigation

34 The database has a comprehensive search facility allowing searches to be made in a
35 number of categories: e.g., chemical name, CAS number, sponsoring country, stage of
36 investigation.

37 Members of the general public have "read only" access to the database and so can follow
38 the progress of a chemical both through and after its assessment. They can also obtain
39 completed assessments on individual chemicals once these have been agreed in the
40 OECD.

41

1 OHMTADS (<http://www.nisc.com/cis/details/ohm-tads.htm>)

2 The Oil and Hazardous Materials/Technical Assistance Data System includes 1,402
3 MSDS-like fact sheets prepared by the US Environmental Protection Agency in the 1970s
4 and 1980s. Each fact sheet deals with one chemical substance. The database is no
5 longer updated, and some material in the database has been rendered incorrect over
6 time by changes in regulatory requirements. However, the database still contains a
7 wealth of still-useful data and references. Consequently, each record is presented with a
8 warning about the age of the database and the need to verify critical information
9 through more current sources. Users can retrieve records by CAS Registry Number (the
10 preferred method), chemical name, and/or subject terms/phrases.

11 Riskline (<http://apps.kemi.se/riskline/>)

12 Riskline contains peer reviewed information on both environment and health. The
13 database is produced by the Swedish Chemicals Inspectorate, Sweden. Each reference in
14 Riskline is furnished with a critical evaluation. It represents the unanimous opinion of a
15 group of toxicological experts in the value of the research that is presented in the
16 document. The evaluation might vary depending on the organization that reviewed the
17 literature. All documents center around one chemical element of family of elements.
18 Abstracts from the original documents are added to the unit record. All items are
19 indexed and the chemical substances identified by CAS numbers.

20 Japanese Ministry of the Environment (<http://www.env.go.jp/en/chemi/>)

21 The Ministry has conducted numerous aquatic toxicity tests in accordance with OECD
22 TGs and GLP for many chemicals. The results from these tests are available on the
23 indicated website.

24

25 **Literature sources**

26 Environmental Risk Limits in the Netherlands, reports 601640001 Part I, II and III
27 (1999)

28 This report, produced by the National Institute of Public Health and the Environment
29 (RIVM), documents risk limits, i.e. Maximum Permissible Concentrations (MPCs) and
30 Negligible Concentrations (NCs) for approximately 200 substances in water, soil,
31 sediment and air from the last decade in the framework of the project, 'Setting
32 Integrated Environmental Quality Standards'. The objective was to present the
33 procedures to derive the environmental risk limits to interested parties involved in
34 environmental policy or environmental risk assessment of chemical substances. These
35 risk limits are the none-regulatory standards used in the Dutch environmental policy.
36 The reports include aquatic toxicity data on a number of chemicals. The quality of data
37 has been assessed and ranked.

38 Canadian Environmental Quality Guidelines (1999) issued by Canadian Council of
39 Ministers of the Environment.

40 Canadian Water Quality Guidelines for the Protection of Aquatic Life help to protect all
41 plants and animals that live in lakes, rivers, and oceans by establishing acceptable levels
42 for substances or conditions that affect water quality such as toxic chemicals,

1 temperature and acidity. The guidelines are based on toxicity data on the most sensitive
2 species of plants and animals found in Canadian waters and act as science-based
3 benchmarks for the protection of 100% of the aquatic life species in Canada, 100% of
4 the time. The guidelines are available on CD-ROM and can be purchased from Canadian
5 Council of Ministers of the Environment (<http://www.ccme.org>).

6 US-EPA Water Quality Criteria for Aquatic life

7 The Aquatic life criteria provide protection for plants and animals that are found in
8 surface waters. The US-EPA develops these criteria as numeric limits on the amounts of
9 chemicals that can be present in river, lake, or stream water without harm to aquatic
10 life. Aquatic life criteria are designed to provide protection for both freshwater and
11 saltwater aquatic organisms from the effects of acute (short term) and chronic (long
12 term) exposure to potentially harmful chemicals. Aquatic life criteria are based on
13 toxicity information and are developed to protect aquatic organisms from death, slower
14 growth, reduced reproduction, and the accumulation of harmful levels of toxic chemicals
15 in their tissues that may adversely affect consumers of such organisms. Developed
16 criteria can be found at <http://epa.gov/waterscience/criteria/aqlife.html>.

17

18 **References**

19 OECD, 1998, Detailed Review Paper on Aquatic Toxicity Methods for Pesticides and
20 Industrial Chemicals, OECD SERIES ON TESTING AND ASSESSMENT, Number 11,
21 NV/MC/CHEM(98)19/PART

22 ECETOC, 1993. Aquatic Toxicity Data Evaluation. ECETOC technical report number 56.
23 European Centre for Ecotoxicology and Toxicology of Chemicals, Brussels.

24

1 **Appendix R.7.8—3 Methodology for body burden approaches in** 2 **aquatic effects assessment**

3

4 The tests described in the TGD divide data collection into discrete compartments which
5 can be classified as acute and chronic toxicity and bioaccumulation. In practice the data
6 compilations are often obtained from different sources using different species or strains
7 and from different media. The classical approach to risk assessment then compiles these
8 data to arrive at an overall interpretation. In certain cases, there may be benefits in
9 measuring, for example, bioconcentration and toxicity on the same species in the same
10 experiment and in many cases standard tests can be ameliorated by addition of
11 analytical measurement of the internal metric.

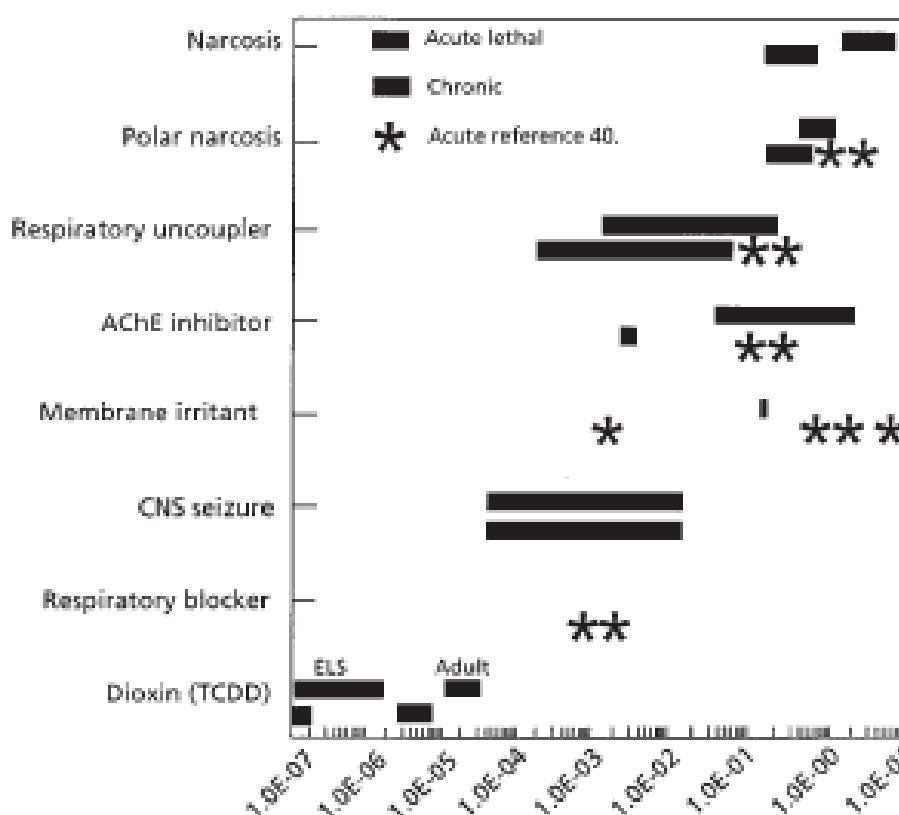
12 The major drawback of relating ecotoxicological effects to *external* concentrations only is
13 in the cases where chemicals do not show (acute) toxic effects at aqueous
14 concentrations below their aqueous solubility, while chronic effects; food-web cascading
15 effects, or aggregate and mixture effects in combination with other non-chemical and
16 chemical stressors may occur. Moreover, measuring external concentrations for low
17 solubility substances is often extremely difficult. For this reason it may be preferable to
18 use an alternative metric for measuring effects: internal body burden. The body burden
19 at which mortality occurs is known as the Lethal Body Burden (LBB) and for sub-lethal
20 endpoints Critical Body Burden (CBB).

21 This concept of critical body burdens (CBBs) is reasonably well-established, particularly
22 for acute effects ((McCarty and Mackay 1993);(McCarty 1986)) of chemicals that act via
23 a narcosis mode of action. A number of reviews have been made on this concept,
24 (Barron *et al.*, 1997; Barron *et al.*, 2002), (Sijm and Hermens 2000) and Thompson
25 and Stewart (2003). (McCarty 1991) recommended merging acute, chronic and
26 bioaccumulation tests into one to greatly increase the information that could be obtained
27 from a single test. This approach, although having a number of practical difficulties,
28 could provide a more robust method for collating lethal concentration, BCF and chronic
29 effects while adhering to the principle of validated guideline studies rather than
30 performing three standard tests under subtly different conditions and trying to combine
31 the results of the studies.

32 McCarty and Mackay (1993) were amongst the first to propose that the internal
33 concentration of a chemical that is related to a biological effect is a more accurate and
34 technically correct basis for comparing and ranking toxicity amongst chemicals and this
35 was supported in later publications (Gobas *et al.*, 2001) and Mackay, 2001).

36

- 1 The following [Figure R.7.8—7](#) gives the range of body burdens originally tabulated in
 2 McCarty and Mackay (1993).



3

4 **Figure R.7.8—7 Calculated body burdens (in mmol.l⁻¹) associated with**
 5 **different acute and chronic toxicity endpoints for fish exposed to eight**
 6 **categories of organic chemicals.**

7

8 Similar ranges of L/CBB have also been published (Thompson and Stewart 2003) and
 9 shown to be relatively consistent with the Figure:

10 MoA I (acute = 1 to 10 mmol.kg⁻¹, chronic = 0.1 to 1 mmol.kg⁻¹) and

11 MoA II (acute = 0.5 to 2 mmol.kg⁻¹, chronic = 0.05 to 0.1 mmol.kg⁻¹).

12 Other MoAs tend to be lower but typically more variable (depending on species and
 13 whether LBB or CBB is considered (see [Figure R.7.8—7](#))).

14

15 **Advantages and disadvantages of the body burden approach**

16 A LBB or CBB can either be measured directly during a study in which biological effects
 17 and chemical body burdens are measured in the same test organisms, or estimated
 18 indirectly. Indirect estimates can be on the basis of measured bioconcentration and
 19 critical external concentrations from different studies, so that LBB = LC50 × BCF and CBB
 20 = NOEC × BCF. Alternatively, indirect estimates can be made on the basis of data
 21 predicted by QSARs although the domain of applicability of the QSAR should be clearly

- 1 demonstrated. This approach has been demonstrated for non-polar (Type I) narcotic
2 substances (baseline toxicity) and polar (Type II) narcotic substances (McCarty 1986,
3 McCarty *et al.*, 1992, 1993).
- 4 The advantages of using the body burden are:
- 5 Knowledge of the CBB should reduce uncertainty in risk assessment as CBB can be used
6 as a tool to help classify the known modes of action of chemicals.
- 7 Toxic effects should be additive within a MoA class because the CBB is independent of
8 chemical structure, so mixture toxicity can be estimated more readily. Moreover, there is
9 evidence that all chemicals have narcotic MoA below the level at which their toxic action
10 is exerted (Dyer *et al.*, 2000).
- 11 QSARs based on Kow can be used to estimate CBBs for MoA I and II (McCarty 1986).
12 Therefore, CBB can be used as a basis for building category approaches for classes of
13 chemicals.
- 14 Data compilations are becoming available that allow theoretical aspects of the body
15 burden approach to be explored and tested empirically, particularly for acute lethal
16 effects caused by chemicals with MoA I and II.
- 17 Potentially, body burdens are a technically easier metric to measure than external
18 concentrations for very poorly soluble or highly adsorbing and bioaccumulable
19 substances.
- 20 Naturally, the CBB approach currently also has shortcomings however, the following
21 shortcomings are common to both CBB and classical (external concentration)
22 approaches:
- 23 1. a value for LBB cannot automatically be used to predict a CBB as the MoA
24 may change from narcotic to non-narcotic for certain chemicals over the long
25 term
 - 26 2. The critical body burden of a chemical may differ between species, however
27 the use of lipid normalisation may decrease. According to Sijm & Hermens
28 (2000), it can be argued that, on a wet weight basis, fatter individuals may
29 accumulate higher body burdens of toxicants before being affected. Lipid
30 normalisation should, in this case, diminish intraspecies variation but
31 according to the literature only reduces variation by 50%.
 - 32 3. Other factors may influence CBB such as the sex, life-stage etc.
 - 33 4. The CBB is usually measured in the whole body of a test organism, although
34 effects may be expected to occur in specific target organs due to high
35 concentrations causing severe damage in particular tissues (e.g., gill).
36 However, this depends on the rate of movement of the chemical in the body.
- 37 There are also technical problems associated with precise measurement of CBB:

1 Body burden data in organisms that die early in a test may be lower than those in
2 organisms that survive to the end of a test. However, there is a similar issue for classical
3 tests where LC₁₀ occurs at an earlier stage than LC₅₀ due to inter-individual variability.

4 Tests on body burden will also include the gut content and, in the case of invertebrates,
5 cuticular adsorption of substance which cannot easily be subtracted to determine true
6 body burden. However, the same applies to standard BCF and BAF tests and while these
7 issues can interfere with the approaches used for CBB determination, they can generally
8 be avoided with careful aforethought.

9 For classically tested invertebrates (e.g. *Lumbriculus* or *Daphnia*) it may be difficult to
10 provide sufficient biomass to achieve quality analytical results. Biomass is an important
11 consideration to take into account prior to conducting the experiment particularly when
12 bioaccumulation is low.

13 Use of total radioactivity to measure body burden, without measuring parent compound
14 specifically, does not take into account biotransformation and potential incorporation of
15 the metabolites into the biomass. This can lead to gross overestimations of the body
16 burden.

17 No normalised studies exist today which take body burdens into account. However,
18 experienced ecotoxicologists should be capable of modifying existing tests to include
19 both bioaccumulation and toxicity in the same design. While any single study would use
20 more animals than a study not including body burden, collectively there are possibilities
21 for reducing the total number of animals used.

22 Some data indicate that the body burden technique may not be suitable for substances
23 with a low log K_{ow} (<1). More evidence for this is needed, however, it should be
24 recognised that most applications for the CBB approach really become useful at higher
25 values of log K_{ow}.

26

27 **Use of body burden data in risk assessment**

28 There are many areas where the generation of body burden data can provide results
29 which can be used in risk assessment: in helping to clarify or form chemical groups and
30 to identify MoA; increasing confidence in data; potential simultaneous provision of BCF
31 and toxicity reducing animal use, for example. Especially, when testing difficult
32 substances it may not even be possible to use standard testing techniques based on
33 aquatic toxicity. In such cases L/CBBs, used in conjunction with QSARs and/or read-
34 across from less difficult substances and quality physico-chemical data, may provide a
35 more reliable data set than standard techniques. The use of such an approach should be
36 reviewed on a case-by-case basis also taking into account the level of technical input
37 required to achieve a suitable result.

38

39 **Conclusion on body burden techniques**

40 The document provides an overview of the current state of the science for body burden
41 methodology, advantages and disadvantages. There is good experimental evidence to
42 support the hypothesis that Critical Body Burden (CBB), at least for acute lethal toxicity

1 is relatively constant for substances with narcotic mode of action. The CBB approach has
2 been recommended for use in risk assessment (Gobas *et al.* (2001) and Mackay (2001))
3 for single substances and could help in category approaches. It could also be used to
4 help assess risk of multiple constituent compounds.

5 If there is information on the critical body burden of a substance in an (aquatic)
6 organism this information could help to identify whether or not the chemical is a baseline
7 narcotic chemical or has a more specific mode of action and thus would provide an
8 indication of its aquatic toxicity.

9

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41

1 **Appendix R.7.8—4 Assessment of available information on** 2 **endocrine and other related effects**

3

4 This chapter is appended to the main guidance document on aquatic toxicity testing. It
5 provides guidance for the evaluation of information relating to (potential) endocrine
6 activity of a substance or long-term adverse effects on development and/or reproduction
7 in aquatic organisms. As this kind of information is not part of the standard information
8 requirements set out in REACH Annexes VII-X (see below), this part of the guidance is
9 based on the evaluation of available information and none of the screening and testing
10 methods discussed has been fully validated or approved as OECD Test Guideline (Status
11 January 2007). Relevant information on the assessment of (potential) endocrine activity
12 in aquatic organisms may also be derived from *in vitro* studies, mammalian screening
13 assays for endocrine activity and other human health endpoints from repeated-dose
14 toxicity, carcinogenicity and reproductive toxicity studies.

15

16 **Endocrine disruption guidance**

17 Definition

18 According to a widely accepted consensus reached at an international workshop in
19 Weybridge, UK, in 1996 (which was later also adopted by OECD expert groups) "*an*
20 *endocrine disruptor is an exogenous agent that causes adverse health effects in an intact*
21 *organism, or its progeny, consequent to changes in endocrine function.*"

22 "Endocrine disruption" is not a toxicological endpoint *per se* but a functional change of
23 the endocrine system which may involve a variety of molecular mechanisms and which
24 may result in adverse health effects in an organism or its progeny. This guidance
25 document distinguishes between the identification of an endocrine mode of action and
26 the characterisation of sub-lethal chronic and adverse effects on development and
27 reproduction, which may also arise from other mechanisms of toxicity; the causal link
28 between an endocrine mode of action and an adverse effect should be established to
29 meet the Weybridge/OECD definition of an endocrine disruptor.

30 Objective of the guidance

31 Endocrine disruption is the occurrence of adverse effects on development or reproduction
32 of aquatic organisms due to a substance's endocrine activity. Such adverse effects,
33 particularly involving reproduction and development, are of high relevance for the
34 assessment of the potential hazards a substance may pose to the aquatic environment.

35 The guidance in this chapter is supposed to cover the following cases of available
36 information beyond the standard information requirements:

- 37 • information indicating potential endocrine activity in aquatic organisms (from
38 human health endpoints, molecular structure, or non-standard *in vitro* assays)
- 39 • information on an endocrine mode of action in aquatic organisms

- 1 • information on adverse effects on reproduction or development of aquatic
2 organisms

3 Available information on adverse effects on development or reproduction should be
4 considered for use in classification, the chemical safety assessment, and the PBT
5 assessment in regards to the toxicity properties of a substance.

6 Furthermore, if a clear link between serious adverse effects and an endocrine mode of
7 action can be established, the substance may fall under the provisions of Article 56 f),
8 which specifies that *substances - such as those having endocrine disrupting properties*
9 *(...) – for which there is scientific evidence of probable serious effects to human health or*
10 *the environment which give rise to an equivalent level of concern to those of CMR, PBT*
11 *or vPvB substances may be included in Annex XIV of substances subject to the*
12 *authorisation procedure. The inclusion will be decided on a case-by-case basis following*
13 *the preparation of an Annex XV dossier by the Competent Authorities.*

14

15 **Information requirements**

16 As indicated above, for registration of a chemical, there is no requirement set out in
17 REACH Annexes VII to X to provide information on the endocrine activity of a substance
18 or on a substance's reproductive or specific developmental toxicity in aquatic organisms.

19 However, according to Article 12, the information specified in Annexes VII-X is to be
20 seen as a minimum requirement. The technical dossier shall include all physico-chemical,
21 toxicological and ecotoxicological information that is relevant and available to the
22 registrant. This general requirement is confirmed with regard to the chemical safety
23 report and the safety data sheets in REACH Annexes I, II, and VI.

24 If, in the course of evaluation of available information, it is indicated that a substance
25 displays an endocrine mode of action in aquatic organisms, this may constitute a concern
26 that requires further investigation regarding potential adverse effects on development or
27 reproduction. Such investigations may be requested on a case-by-case basis by a
28 Member State, when performing the substance evaluation of a registration dossier
29 (Article 45). This provision includes the request of specialised studies not covered by the
30 REACH Annexes VII-X, such as the endocrine-specific studies described in this Appendix.

31

32 **Information and its sources**

33 Non-testing data

34 Non-testing data include information derived from SARs, QSARs, read-across and
35 chemical categories. The general principles how to generate information by these
36 methods are explained in the main part of this guidance document. Models are under
37 development under the umbrella of OECD and ECB programmes for specific endocrine-
38 related mechanisms, in particular for estrogen and androgen receptor binding (see
39 Netzeva *et al.*, 2006; Saliner *et al.*, 2006; for a recent overview of models see Devillers
40 *et al.*, 2006; for structural requirements specific for ER binding see Fang *et al.*, 2001; for
41 structural requirements specific for AR binding see Fang *et al.*, 2003; Tamura *et al.*,
42 2006).

1 Due to availability and quality of experimental data, more SAR and QSAR models are
2 available for mechanism-related endpoints than for endocrine activity in intact organisms
3 and for long-term adverse effects. However, the development of models that can predict
4 *in vivo* effects, in view of their saving potential, may become more important in the
5 future. Among the models (SARs and QSARs) that predict mechanism-related endpoints,
6 more models were developed for estrogenic activity compared to androgenic activity.

7 Along with the classical SAR and QSAR models, a number of 3-dimensional QSARs (3D
8 QSARs, derived from Comparative Molecular Field Analysis, CoMFA) and docking studies
9 were published in the literature. There is a good scientific basis for the development of
10 the latter models since most of the endocrine disrupting effects are provoked by binding
11 of chemicals to specific receptors (i.e. interactions, suitable for molecular modelling).
12 However, there are still technical constraints in the transferability of such models for
13 quantitative application unless the result of them is presented in different form (e.g.
14 translated into structural alerts).

15 There is a large range of computational models that have been successfully applied to
16 model endpoints, related to endocrine disruption. These range from structural features
17 and structural alerts¹⁰ (e.g. the presence of steroid skeleton, diethylstilbestrol skeleton
18 or phenolic ring increase the probability of a chemical to be a binder to the estrogen
19 receptor), to pharmacophore queries, to different discriminant models for assignment to
20 an activity class (e.g. derived from linear discriminant analysis, k-Nearest neighbour
21 modelling, decision tree analysis, biophore-type analysis, common reactivity pattern
22 analysis etc.) to various quantitative models for prediction of potency, derived from local
23 (e.g. congeneric) or global (diverse) data sets. The descriptors in the models also vary
24 from structural fragments, through various hydrophobic, steric and electrostatic
25 descriptors, to steric and electrostatic fields in CoMFA analysis and energies in docking
26 studies. The choice of descriptors and modelling technique is largely dependent on the
27 purpose and data series and no single recommendation can be given but rather critical
28 and realistic evaluation of the models and underlying data is required depending on the
29 problem to be solved.

30 Testing data

31 Throughout this Appendix, laboratory (experimental) methods are further divided into
32 *screening assays* and (confirmatory) tests. In this sense, *screening assays* are lower tier
33 *in vitro* or *in vivo* investigations which allow the identification of a potential endocrine
34 mode of action of a substance, while definitive or confirmatory tests are higher tier *in*
35 *vivo* methods to confirm the screening results and to characterise any adverse effects
36 that may result from such a mode of action. Note should be taken that the term
37 *screening assay*, in this context, does not relate to a blind screening of large numbers of
38 chemicals. All of the methods described below are endocrine-specific studies that will
39 only be relevant for a limited number of substances.

¹⁰ A discrimination between structural feature and structural alert could be done. For example, a tert-butyl moiety and phenol group are structural features associated with high potential for estrogen binding. However, the combination is viewed as a structural alert for estrogenicity only if the two functional groups are in p-position to each other, while, for example, o-position is not linked to a receptor-mediated gene activation.

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In vitro screening data

At present, validated *in vitro* assays and internationally accepted Test Guidelines for regulatory purposes are not yet available. However, molecular mechanisms of the endocrine system, especially of the sexual hormone system of vertebrates, are well characterised and a large number of *in vitro* assays are used in scientific research. Although the basic principles have been applied to biological material from a variety of species, including aquatic vertebrates, assays based on mammalian systems are usually in the most advanced stage of development as expressed by their validation status. Given the similarity of endocrine systems across vertebrate taxa, these assays may also provide valuable information on the assessment of potential endocrine activity of chemicals in aquatic organisms, in particular fish.

The following *in vitro* assays for the detection of possible endocrine activity of substances were selected for further development with the aim of validation for regulatory use. They are at different stages of development, validation and regulatory acceptance; their status in 2006 is indicated below.

Estrogen and Androgen Receptor Binding Assays

Principle: Binding of a hormone to its receptor in the cytosol is an early event in the pathway of hormonal regulation. Assays that study the capacity of xenobiotic substances to compete with natural hormones from their binding sites have been developed with estrogen and androgen receptors from several species in different cellular systems. This type of assay cannot predict whether the binding of a substance to a hormone receptor will result in its activation (agonistic activity) or inhibition (antagonistic activity).

Status: Prevalidation of two receptor binding assays within the integrated project ReProTect funded by the 6th Framework Programme of the European Commission is now continuing under the umbrella of the OECD into validation led by the US-EPA and in collaboration with Japan. The US-EPA has completed validation of an assay based on the androgen receptor from rat prostate cytosol and conducted studies on the nature of binding interaction for 50 structurally diverse chemicals with the estrogen receptor from rat uterine cytosol (Laws *et al.*, 2006).

Transcriptional Activation (Reporter Gene) Assays

Principle: The active ligand-receptor complex translocates into the cell nucleus, where it aligns to specific DNA sequences and induces gene transcription. Incorporation of recombinant hormone-responsive gene elements and their promoters together with elements encoding easily detectable proteins into suitable host cells allows the detection of hormone receptor activation by visualising the response at the gene transcription level. As these assays can only show receptor activation, while antagonistic receptor interactions remain undetected, a positive test result does not always mean that exposure to the substance would result in an agonistic effect *in vivo*. The relevance of these genetically engineered systems to *in vivo* dose response of endogenous receptor and target genes has been evaluated in the Japanese Report in peer review at the OECD (see below).

1 *Status:* Validation of the Stably Transfected Transcriptional Activation (TA) Assay to
2 Detect Estrogenic Activity was performed in Japan for ER agonists and is at the stage of
3 peer-review within the OECD Test Guidelines programme. Prevalidation of four
4 transcriptional activation assays for ER and AR (anti)agonists detection has been carried
5 out within the integrated project ReProTect funded by the 6th Framework Programme of
6 the European Commission and these are now progressing to validation.

7 *Vitellogenin Assays*

8 *Principle:* Activation of the estrogen receptor in the liver of fish induces the biosynthesis
9 of the egg yolk protein vitellogenin (VTG). Based on this principle, assays have been
10 developed using primary cultured hepatocytes (e.g. from medaka or rainbow trout) to
11 assess the influence of substances on VTG production via estrogenic or anti-estrogenic
12 activity.

13 *Status:* This assay has been studied in several common fish species, with most data
14 available for mature male rainbow trout and carp. The sensitivity of the cell cultures and
15 the methods of detection of VTG protein by ELISA are being validated while those
16 measuring VTG mRNA, using RT-PCR, still need to be validated.

17 *Steroidogenesis Assays*

18 *Principle:* Certain cell cultures express the enzymatic systems to metabolise cholesterol
19 via native biosynthetic pathways into the final active steroid hormones such as
20 androgens and estrogens in sufficient quantities for analytical determination of the rate
21 of steroid synthesis. This provides a basis to develop an *in vitro* assay for stimulators
22 and inhibitors of steroidogenic pathways relevant to vertebrates (see OECD Draft
23 Detailed Review Paper on Steroidogenesis, May 2002). A particular focus of
24 investigations is placed on the enzyme aromatase, which converts androgens into
25 estrogens (see OECD Draft Detailed Review Paper on Aromatase, February 2002).

26 *Status:* Pre-validation work within the OECD framework is in progress for an assay based
27 on the H295 human adrenocortical carcinoma cell line that has been shown to express all
28 of the key enzymes necessary for steroidogenesis. The US-EPA is conducting
29 prevalidation studies on human recombinant aromatase.

30 The latest information on the status of *in vitro* methods that are under development can
31 be obtained from the ECVAM website (current address: <http://ecvam.jrc.it>).

32 *In vivo screening and testing data*

33 *Principle:* Intact organisms are exposed through the water to the chemical in a range of
34 sub-lethal concentrations for a period of a few weeks at minimum. Males and females
35 are tested and a number of endpoints are measured to either trigger further
36 investigation or conclude on the absence of concern. Biomarker endpoints will play an
37 important role in screening whereas reproductive and developmental landmarks will be
38 assessed in long-term toxicity testing.

39 *Status:* At present, there are no validated *in vivo* screening assays for the identification
40 of substances with potential endocrine activity in aquatic organisms or test methods for
41 the investigation whether a substance with endocrine activity has adverse impact in
42 aquatic organisms. However, a number of methods are used in scientific research (see

1 monographs No. 21, 55, and 57 in the OECD Series on Testing and Assessment). The
2 performance of such methods is not included in the minimum requirement by REACH but
3 for some substances relevant information may be available, e.g. from the scientific
4 literature. For these cases, the compilation of available methods is given below as an
5 orientation about the current state of development in the field of endocrine screening
6 and testing and as references for the evaluation of older studies. The following methods
7 were selected for further development with the aim of validation for regulatory use for
8 the detection of endocrine activity or the characterisation of chronic effects on the
9 development and reproduction of aquatic organisms. They are at different stages of
10 development, validation and regulatory acceptance; their status in 2006 is indicated
11 below.

12

13 *Vertebrates*

14 In relation to the sexual hormone system of fish, a range of methods is under
15 development and validation, covering different levels of biological complexity.

16 • **Screening Assays**

- 17 - 21-Day Fish Screening Assay, draft TG proposal (OECD, 2004)

18 This assay is proposed for the detection of estrogenic, androgenic or aromatase
19 inhibiting substances in adult organisms which have reached sexual maturity. It can be
20 run with several common fish species: zebrafish, fathead minnow, medaka and possibly
21 the three spined-stickleback. The assay lasts over a period of 21 days. Core endpoints
22 are VTG levels in the serum or liver (medaka), which indicate disturbances of the
23 estrogenic balance, and secondary sex characteristics in sexually dimorphic species (not
24 in zebrafish), which are liable to disturbances of the androgenic balance. The OECD
25 validation studies are completed and the peer-review will take place early 2007 (see
26 monographs No. 47, 60, and 61 in the OECD Series on Testing and Assessment).

27 • **Confirmatory Tests**

- 28 - Fish Sexual Development Test, draft TG proposal (OECD, 2006)

29 This method has been proposed as an extension of the existing OECD Test Guideline 210
30 (1992) Fish, Early-Life Stage (FELS) Toxicity Test. The enhancements focus on sexual
31 development, i.e. sex ratio as determined via histological examination of the gonads,
32 and on VTG production. The test aims at investigating the impact of substances acting as
33 estrogens, androgens or aromatase inhibitors in organisms at a very sensitive stage of
34 their life to endocrine activity. It can be run with several common test species: zebrafish,
35 fathead minnow, medaka and possibly the three-spined stickleback. The test starts with
36 fertilised eggs and lasts until sexual differentiation is completed (e.g. 60 to 90 days post
37 hatch, depending on the fish species). After test development work in Denmark, the
38 initial OECD validation study for fathead minnow and zebrafish has recently been
39 initiated.

- 40 - Fathead Minnow Reproduction Test, draft TG proposal (US-EPA, 2001):

41 A draft proposal for a fathead minnow reproduction test, including vitellogenin,
42 secondary sex characteristics, gonad histopathology, fecundity and fertility assessments,

1 is being validated in the United States. The test duration is 42 days, with 21 days of pre-
2 exposure where fecundity is recorded daily, and 21 days of chemical exposure. The US-
3 EPA validation programme is in progress and guidance documents should be developed
4 for the interpretation of gonad histopathology.

5 - Fish Full Life Cycle / 2-Generation Test

6 These tests allow an assessment of chronic effects on developmental and reproductive
7 endpoints (see OECD Draft Detailed Review Paper on Fish Two-Generation Toxicity Test
8 and Proposal for a Fish Two-Generation Test Guideline, March 2003). The most complete
9 test design, which allows assessment of trans-generational transfer of effects, begins
10 with exposure of adult, reproducing fish (F0 generation) and continues until in-life
11 biological effects of the F2 generation can be determined. This time point as well as the
12 total test duration may vary considerably depending upon the species used.

13 Measurements include developmental and reproductive endpoints (hatching, sex ratio,
14 survival, growth, fecundity, fertility and behaviour) as well as biochemical, histological
15 and morphological markers that are indicative of specific mechanism of endocrine
16 disruption. The validation is under preparation. Results from such tests have already
17 been used in risk assessments of specific substances of concern within the EU priority
18 existing substances programme and in the authorisation of pesticides.

19 - 21-Day Amphibian Metamorphosis Assay, draft TG proposal (OECD,
20 2005)

21 This assay was developed for the detection of chemicals affecting the thyroid hormone
22 system in amphibian species (see monograph No. 46 in the OECD Series on Testing and
23 Assessment). The metamorphosis of amphibians, and in particular *Xenopus laevis*, the
24 test species in this assay, is a well-studied phenomenon under the dependence of
25 thyroid hormone signalling. Development stage, whole body length, hind-limb length and
26 thyroid histology are the endpoints measured during the assay. The assay lasts for 21
27 days; hind-limb length is measured after 7 days and other endpoints are measured at
28 termination of the assay. The test allows the characterisation of adverse effects on
29 amphibian metamorphosis and growth as well as the identification of a thyroid disruptive
30 mode of action, which may also be of relevance for other vertebrate species. Validation
31 of this test method is ongoing.

32

33 *Invertebrates*

34 The endocrine systems of aquatic invertebrates differ considerably from those of
35 vertebrates and the knowledge in this field is less advanced. Consequently, consideration
36 of specific endocrine-related endpoints in long-term invertebrate testing is only at the
37 beginning (see also monograph No. 55 in the OECD Series on Testing and Assessment)
38 of its development and its status and implication should be checked carefully:

39 • **Confirmatory Tests**

40 - Enhanced Test Guideline 211, *Daphnia magna* Reproduction Test, (OECD,
41 2006)

1 *Principle:* This method is an enhancement of TG 211 which is intended to detect
2 chemicals interacting with the hormone system of aquatic arthropod species, i.e.
3 chemicals acting like the juvenile hormone or like ecdysteroids. In addition to the
4 traditional endpoints measured in the existing *Daphnia* reproduction test, the new
5 endpoints are offspring sex ratio and molt inhibition. This enhanced version has the
6 same exposure duration as the existing TG 211, but additional technical efforts and time
7 are required for the microscopic evaluation of the endpoints.

8 *Status:* The validation study is on-going in the OECD TG programme with Japan as lead
9 country.

10 Other Test Guideline projects are currently in progress for marine or estuarine species,
11 where development and reproductive endpoints are assessed. These assays are not
12 intended to specifically identify endocrine modes of action:

13 - Copepod Development and Reproduction Test, draft TG proposal (OECD,
14 2005)

15 This test examines the development and reproduction of marine harpacticoid and
16 calanoid copepod species. Eggs or newly hatched larvae (< 24 h) are exposed for 20-26
17 days. Endpoints are larval mortality, larval development rate and reproductive success.
18 The validation study is in progress in the OECD TG programme with Sweden as lead
19 country.

20 - Mysid 2-Generation Test, draft TG proposal

21 This test evaluates reproductive fitness in two consecutive generations of mysids
22 (preferably *Americamysis bahia*), starting with newly-released (<24 h) individuals of the
23 F0 generations and continuing until the first two broods (F2 generation) of the F1
24 generation. The overall test duration is normally 60 days or longer. Observational
25 endpoints include growth, time to maturity, time to first brood release, interbrood
26 duration, number and sex ratio of offspring. The pre-validation is ongoing in the United
27 States under OECD auspices.

28

29 **Evaluation of information**

30 This section attempts to assist the user (e.g. registrant) in judging and ranking the
31 adequacy (i.e. reliability and relevance) of information related to (potential) endocrine
32 activity of a substance or its reproductive and developmental toxicity towards aquatic
33 organisms. Since information of this kind is not part of the REACH information
34 requirements, the following considerations are supposed to apply to those cases where
35 this information is already available, e.g. from the scientific literature, or where it is
36 specifically requested by a CA, e.g. in the course of substance evaluation. This is a
37 relatively new area of testing and assessment where information needs to be evaluated
38 carefully on a case-by-case basis.

39 Non-testing data

40 The evaluation of QSAR results consists of 1) evaluation of the validity of the model and
41 2) evaluation of the reliability of the individual model prediction. Guiding principles are
42 explained in the general introduction to the TGD as well as in the main text on aquatic

1 toxicity. Guidance on the application of grouping approaches (read-across and chemical
2 categories) is given in the general introduction.

3 A special attention deserves the way, in which the activity class is assigned for
4 development of the model, if it is intended to discriminate between active and inactive
5 chemicals. The cut off, if such utilized to obtain binary classification from continuous
6 data, should be clearly described when arguing the validity of the model prediction.
7 Generally, the classification models tend to demonstrate higher accuracy than those
8 predicting continuous values but the borderline predictions will need additional
9 consideration. Nevertheless, both types of models should be evaluated according to the
10 OECD principles and commonly encountered pitfalls (e.g. over-fitted models), described
11 in the cross-cutting guidance on (Q)SAR, should be avoided. The global models, derived
12 on diverse data sets, have generally larger domains of applicability but local models can
13 be preferred if available for a specific chemical of interest. An understanding of structural
14 features that form structural alerts is highly desirable and mechanistic interpretation of
15 models and descriptor combinations should be looked for. Finally, the use of several
16 models is expected to increase the confidence in the prediction but expert judgment
17 might be required in case of contradicting results (e.g. the chemical is predicted active in
18 classification model but with extremely low activity from a potency model, or vice versa).

19

20 Screening and testing data

21 *In vitro* screening data

22 Guiding principles to judge the adequacy of information obtained from *in vitro* assays are
23 explained in the general introduction to the TGD as well as in the main text on aquatic
24 toxicity (it should be noted that for the assessment of potential endocrine activity, data
25 from mammalian systems may also provide information of relevance to aquatic
26 organisms).

27 *In vivo* screening data

28 Guiding principles of evaluating the reliability and relevance of *in vivo* data are explained
29 in general parts of this guidance document. In addition, many of the specific
30 considerations for aquatic test systems and organisms detailed in the main text on
31 aquatic toxicity apply.

32 The purpose of *in vivo* studies for the investigation of endocrine activity of chemicals is
33 to determine 1) whether the chemical is active on the endocrine system of aquatic
34 organisms (e.g. vitellogenin induction as indicator of estrogenic activity), and 2) whether
35 this mechanism induces adverse effects in long-term studies (e.g. decrease in the
36 number of offspring, effect on sex ratio in developing organisms).

37 - 21-Day Fish Screening Assay, draft TG proposal (OECD, 2004)

38 For the results to be meaningful, the vitellogenin data in control males and females
39 should be within the range reported in the literature and indicated in the draft test
40 guideline. For test results to be considered positive, significant responses should be
41 observed at sub-lethal concentration (e.g. 0.5 or 0.1 times the LC₅₀; this value would
42 need further discussion and agreement). Importantly, a homologous ELISA method

1 (using standard VTG from the same species and homologous antibodies) should be used.
2 Any loss of biological sample and any deviation from the protocol should be reported. As
3 experience with compounds that are negative for estrogenic modes of action and
4 experience with the rate of false positives for the VTG endpoint is limited, some caution
5 with positive results is currently necessary.

6 For the evaluation of androgenic substances, a fish species should be used, which
7 possesses the necessary characteristics to determine an endpoint relevant for
8 androgenic stimulation, for instance secondary sex characteristics or an androgen-
9 sensitive biochemical marker such as spiggin induction in the stickleback. In the case of
10 suspected androgen activity fathead minnow, medaka, or stickleback are therefore the
11 only recommended test species in a fish screening assay. Zebrafish is not suitable for the
12 evaluation of androgenic substances in this assay.

13 No response on the endpoints measured in this assay indicates that the substance does
14 not act as estrogen or androgen agonist or aromatase inhibitor/estrogen antagonist in
15 fish *in vivo*. However, such a test compound may still have endocrine activity mediated
16 through other, non-investigated mechanisms. Together with partial and full-life cycle
17 studies that include developmental and reproductive parameters, these data can be used
18 in a *Weight of Evidence* assessment whether adverse effects may be occurring through
19 the covered endocrine modes of action.

20 *In vivo testing data*

21 - Fish Sexual Development Test, draft TG proposal (OECD, 2006)

22 The current TG210 is suitable for the characterisation of a substance's adverse effects on
23 fish survival, growth and development. The proposed extension, whether an enhanced or
24 separate Test Guideline, focuses on a more detailed evaluation of sexual development,
25 where the sex ratio and the production of vitellogenin are the main core endpoints. The
26 discussion and attention for the evaluation of data should be focused on the statistical
27 analysis and interpretation of the sex ratio endpoint. There may be concerns on the
28 interpretation of results, due to a natural high variability in the sex ratio (i.e. male to
29 female ratio can naturally vary between 35-65%) in control populations. Consequently,
30 the value of "x" in EC_x currently poses question for a regression analysis (i.e. x=10 is not
31 realistic, x=25 may be possible). Alternatively, if the LOEC/NOEC determination is the
32 objective of the assay, a large number of replicate tanks (> 4) is necessary to level off
33 the between-replicate variability and maintain sufficient power of the assay. Solutions to
34 level-off the variability of the sex ratio exist, like the increase of the number of egg
35 clutches (minimum of 5) used at the start of the test. When evaluating data from this
36 test, attention should be paid to such test parameters and adherence to validity criteria
37 specified in the test guideline.

38 - Fathead Minnow Reproduction Test, draft TG proposal (US-EPA, 2001):

39 Care should be exercised in the evaluation of fecundity and gonad histopathological
40 findings to differentiate toxic response which may not always be indicative of specific
41 reproductive toxicity. An analysis of the data in a *Weight of Evidence* approach is
42 foreseen and should be documented. The data should be transparently reported,
43 especially for gonad histopathology, so that a transparent judgement can be made of the
44 nature and reliability of the responses observed and whether the results are sufficient to

1 conclude on the cause of the effects on reproductive capacity. Guidance documents are
2 in preparation in the US and the OECD to assist pathologists in preparing the samples
3 and evaluation the slides in a standardised fashion.

4 - Fish Full Life Cycle / 2-Generation Test

5 These tests allow an assessment of apical developmental and reproductive endpoints.
6 Effects observed in these studies are of high relevance for the assessment of chronic
7 toxicity to aquatic vertebrates. The inherent assumption is that effect levels derived from
8 these endpoints are relevant to protect populations. However, the endpoints are not
9 indicative or specific to any particular endocrine mode of action.

10 - 21-Day Amphibian Metamorphosis Assay, draft TG proposal (OECD,
11 2005)

12 This test allows the detection of interaction of a substance with the thyroid system. This
13 test may be used when there is some indication that the substance may disturb growth
14 and development, essentially for confirming the mode of action (i.e. thyroid). As thyroid
15 is heavily conserved in vertebrates, a negative response in the 21-Day Amphibian
16 Metamorphosis Assay indicates that the substance does not impact the thyroid system in
17 any vertebrate taxa. A positive response may be used in conjunction with chronic tests
18 to conclude on the hazard and the derivation of effect levels.

19 - Invertebrate life cycle tests, including developmental and reproductive
20 endpoints

21 The life cycle of invertebrates is controlled by distinct and different endocrine systems
22 than vertebrates. In some cases (e.g., mollusks), the hormones may be similar to the
23 steroids found in vertebrates, while in other cases (e.g., aquatic arthropods) the
24 hormones are specific to certain invertebrate groups, such as juvenile hormone or
25 ecdysteroids.

26 Test methods for invertebrates, such as life cycle or multi-generation studies, focus on
27 non-specific population-relevant endpoints of reproduction and development, rather than
28 identifying any specific endocrine mode of action for particular invertebrate groups
29 (except for the proposed enhancement to the existing *Daphnia* reproduction test).

30 - Enhanced OECD TG 211 on *Daphnia magna* Reproduction Test, draft TG
31 proposal, 2005;

32 The evaluation of test results is not any different from the existing OECD TG 211. The
33 evaluation of additional endpoints provides a mechanistic insight into the effects
34 observed on development and reproduction. Care should be exercised in the
35 interpretation of changes in the sex ratio in the daphnids as this is not specific for an
36 endocrine mode of action in these parthenogenic organisms where several test
37 conditions (e.g. temperature, food abundance) can affect the sex ratio of the offspring.
38 The regulatory interpretation of changes in the sex ratio endpoint is still new and
39 requires further discussion.

40 Several new reproductive and developmental assays have been recently proposed for
41 aquatic invertebrates and are listed in Section 3. These proposals are based on
42 endpoints relevant for reproduction and development, and do not include additional

1 markers to indicate any endocrine mode of action. None of these tests have advanced to
2 the stage of regulatory guidelines, and none are currently required by Annexes VII to X
3 in the REACH legislation.

4 - Mammalian toxicity data

5 Results from mammalian *in vitro* and *in vivo* screening assays should provide both
6 positive and negative indications of endocrine modes of action which are also relevant
7 for aquatic vertebrate species.

8 Studies on repeated dose toxicity, long-term toxicity and carcinogenicity, reproductive or
9 developmental toxicity in mammals may provide both positive and negative indications
10 of endocrine modes of action which are also relevant for aquatic vertebrate species.

11 For detailed guidance on the evaluation of such data the relevant sections of the chapter
12 on Human Health Hazard Assessment should be consulted.

13 Interpretation and use of this data within an integrated assessment of endocrine activity
14 in aquatic organisms is outlined in section 6 of this Appendix.

15

16 **Conclusions on endocrine activity**

17 The purpose of this section is to give guidance if and how information relating to
18 endocrine activity of a substance and to the adverse effects that may arise from such
19 activity should be considered for conclusions on the regulatory endpoints classification &
20 labelling, PBT assessment and chemical safety assessment and on the assessment of
21 endocrine disrupting properties as referred to in Article 57 f).

22 Suitability of information on Classification and Labelling

23 Disruption of the endocrine activity, which may result in long-term toxicity, is usually not
24 of relevance for classification according to the current EU system, which is based on
25 information from short-term and chronic toxicity testing. A basis for exceptions is
26 provided by the 'safety net' categories for substances, which do not fall under the 'core
27 set of criteria' (Aquatic acute 1; H400, Aquatic Chronic 1; H410, Aquatic Chronic 2;
28 H411, Aquatic Chronic 3; H412 according to CLP Regulation).

29 According to the CLP Hazard statement H413 could be assigned (under the safety net
30 classification)¹¹. There are no defined criteria for these classifications but both have been
31 proposed and argued for in the course of the classification of bisphenol A, in order to
32 take account of its endocrine disrupting properties. In any case, such a decision should
33 be based on available information that a substance causes adverse effects on
34 development or reproduction of aquatic organisms which should be derived not from
35 screening assays, but from suitable long-term confirmatory tests, such as those detailed
36 in sections 3 and 4.

¹¹ In accordance to section 4.1.2.4 of Annex I to the CLP Regulation, a "safety net" classification (referred to as Chronic Category 4) for use when the data available do not allow classification under the formal criteria for acute 1 or chronic 1 to 3 but there are nevertheless some grounds for concern.

1 Suitability of information on PBT/vPvB assessment

2 The assessment of whether a substance fulfills the T criterion with respect to freshwater
3 or marine organisms (long-term NOEC/EC10 < 0.01 mg/l) is usually based on results
4 from standard long-term toxicity testing of the kind that is specified in REACH Annexes
5 VII-X to REACH. Standard toxicity testing in fish is based on the assessment of growth
6 and mortality. Some substances, however, may cause sublethal chronic effects in
7 concentrations below those affecting growth or survival, which may also be of serious
8 concern for the aquatic environment, such as an impairment of sexual development or
9 reproductive performance.

10 Information on reproductive or developmental effects in fish is not part of the
11 requirements of REACH Annexes VII-X to REACH but may be available for some
12 substances, e.g. from the scientific literature. Suitable long-term studies are those
13 studies which are designed to investigate specific toxicity on reproduction or sexual
14 development as in the Fish Sexual Development Test, the Reproduction Test or the Full
15 Life-Cycle / Two-Generation Test that are described in sections 3 and 4. Parameters
16 derived from such studies with a widely accepted relevance for reproduction, which may
17 have an impact on population level, are egg numbers, fertilization rate, time to hatch,
18 hatching rate and sex ratio. This information should be considered for use in the
19 assessment of chronic toxicity as part of PBT assessment if it is derived from a suitable
20 long-term study and judged as adequate according to the principles outlined in section 4.

21 The relevance of changes in fish gonad histology or spermatogenesis and whether these
22 should be considered adverse effects is controversial. Changes to secondary sex
23 characteristics or biochemical parameters such as vitellogenin or spiggin are regarded as
24 evidence that a substance acts via a specific endocrine mode of action, which may or
25 may not result in long-term adverse effects. In itself, information on such parameters is
26 not suitable for use in PBT/vPvB assessment, but it may be the basis for a CA to request
27 further investigations of potential long-term adverse effect in the course of substance
28 evaluation.

29 Suitability of information on Chemical Safety Assessment

30 The use of information on sub-lethal long-term effects in Chemical Safety Assessment
31 (CSA) should generally be considered according to the same principles as outlined above
32 for PBT assessment.

33 It is subject to a controversial debate whether the conclusion that an adverse effect is
34 elicited by an endocrine mode of action justifies a modification of the assessment factor
35 used in risk assessment. For the further progress of this debate it might be helpful to
36 bear in mind the provision contained in the TGD 2003 with regard to this issue: *In*
37 *general, justification for changing the assessment factor could include one or more of the*
38 *following: (...) Knowledge of the mode of action including endocrine disrupting effects (p*
39 *100).*

40 More guidance on the selection of the appropriate assessment factor is given in guidance
41 provided by [Chapter R.10](#).

42 Suitability of information on assessment in relation to Article 57 (f)

1 According to Article 57 (f), the list of substances subject to authorisation (Annex XIV),
2 may include "*substances – such as those having endocrine disrupting properties (...) –*
3 *for which there is scientific evidence of probable serious effects to human health and the*
4 *environment which give rise to an equivalent level of concern to those of other*
5 *substances listed in points (a) to (e) and which are identified on a case-by-case basis*
6 *(...)*".

7 While the identification of such substances is a responsibility of the Member States,
8 executed by the preparation of an Annex XV dossier, which should justify the proposal
9 and specify the concern, the evaluation of environmental hazard information will form
10 the basis for it. In accordance with the principles outlined in the previous sections,
11 available information on a accordance with the principles outlined in the previous
12 sections, available information on a substance can be evaluated for its suitability to
13 support a conclusion that:

- 14 • there is an indication or evidence of *endocrine disrupting properties* (instead
15 of this wording, which is a direct quote from the REACH regulation, the more
16 fitting term *endocrine activity or mode of action* is used throughout this
17 Appendix)
- 18 • there is *scientific evidence of probable serious effects* to the aquatic
19 environment due to these properties (i.e. within the terminology of this
20 Appendix "adverse effects on development and/or reproduction")

21 Indication of potential endocrine activity in aquatic organisms may be provided by
22 considerations relating to the molecular structure, available information from endocrine-
23 specific *in vitro* screening assays, such as those outlined in sections 3 and 4, or available
24 information from mammalian toxicity studies. However, structural data *alone* should be
25 regarded as an insufficient basis at this time.

26 Evidence of an endocrine mode of action in aquatic organisms may be provided by
27 information on biochemical, histological or morphological changes measured in
28 endocrine-specific studies. Generation of this kind of information is not a standard
29 requirement under REACH but may be requested by a CA in specific cases during
30 substance evaluation, e.g. on the basis of available alerts such as those listed above.

31 Evidence of *probable serious effects* to the aquatic environment due to *endocrine*
32 *disrupting properties* may encompass information regarding adverse effects on
33 development or reproduction, which can be obtained from suitable long-term studies
34 such as those outlined in sections 3 and 4. However, reproductive or developmental
35 toxicity can also be caused by other toxicological mechanisms and a case-by-case
36 decision must be reached based on *Weight of Evidence* considering all available
37 information on adverse effects in conjunction with information on specific endocrine
38 modes of action. Again, it should be noted that this kind of information is not a standard
39 requirement.

40 It may be available in some cases, e.g. from the scientific literature, and it may also be
41 requested by a Competent Authority under substance evaluation in specific cases, e.g.
42 on the basis of available information that a substance acts via an endocrine mode of
43 action.

1 The overall conclusion should be on the presence or not of endocrine disrupting
2 properties of the substance and the characterisation of adverse effects, based on
3 existing information or information that is generated on specific request by the
4 Competent Authority under substance evaluation. It is not the responsibility of the
5 registrant to conclude on an *equivalent level of concern*, as specified under Article 54 (f).
6 This task is the responsibility of the Competent Authority or the Agency, who prepare a
7 dossier according to Annex XV for the identification of substances of very high concern
8 and for their eventual inclusion in Annex XIV.

9 **Integrated assessment of potential endocrine activity**

10 In the following, a strategy for an integrated assessment of all available information on
11 potential endocrine activity of a substance is proposed (see scheme). It takes up
12 concepts developed by the OECD in its conceptual framework for endocrine disrupter
13 testing and assessment, which provides a toolbox with methods categorised according to
14 levels of increasing biological complexity (OECD, 2002).

15 This section is intended to summarise what has been outlined before about how to
16 gather and evaluate existing information on endocrine activity and how this may relate
17 to the purposes and requirements of REACH.

18 Most of the presently available knowledge, experience and methodology relates to the
19 system of sexual hormones (estrogens/androgens) of vertebrates, with fish as the most
20 extensively studied aquatic species. Progress is also being made with regard to the
21 thyroid system in amphibians. Coverage of invertebrate species and their distinct
22 endocrine systems, such as those of juvenile or ecdysteroid hormones, remains sparse.

23 In the proposed assessment strategy, three types of information are distinguished:
24 preliminary information that indicates potential endocrine activity in aquatic organisms;
25 information that indicates a specific endocrine mode of action in an intact aquatic
26 organism; information that allows the characterisation of long-term adverse effects,
27 which may be caused by endocrine activity but also by other mechanisms of toxicity.

28 **1. Preliminary indication of potential endocrine activity in aquatic** 29 **organisms**

30 Preliminary indications of potential endocrine activity that might be of relevance for the
31 aquatic environment but are derived from information sources outside aquatic toxicity
32 testing include considerations of the molecular structure, which will apply to all
33 substances, and results from *in vitro* screening assays, which are not part of the
34 standard information requirements but may be available in certain cases, e.g. from
35 scientific research. Preliminary indications applicable to vertebrate species may also
36 come from results from mammalian toxicity testing, which may to a certain extent be
37 part of the standard information requirements.

38 Non-testing information (molecular structure):

39 The different approaches of generating information by non-testing methods have been
40 outlined in sections 3 and 4. In relation to the steroid sexual hormone system of
41 vertebrates, a number of QSAR models based on experimental data are available resp.
42 under development. Qualitative approaches, such as SAR, read-across or

1 categorisations, may consider similarities with natural hormones or xenobiotic
2 substances of confirmed hormonal activity with regard to all known endocrine systems.

3 Within the domain of non-testing data, a sensible tiered approach can be applied for
4 screening and prioritization purposes (Tong *et al.*, 2003). Such approach can start with
5 rejection filters (e.g. molecular weight lower than 94 or higher than 1000 is not likely to
6 be associated with estrogen binding affinity), include models for qualitative assignment
7 of activity (e.g. classification as active or inactive compounds) and then applying models
8 for quantitative estimation of the potency in case that the chemical is predicted active as
9 a result of the previous step. The last step includes incorporation of human knowledge
10 and expertise in the evaluation of the results of the previous steps and additional rules
11 for refinement can be applied.

12 With regard to the endpoint under prediction, a differentiation is to be made between
13 mechanistic endpoints, i.e. mainly interactions with a defined molecular target,
14 endpoints relating to biochemical responses (screening assays) or adverse effects
15 (definitive tests) *in vivo*. Among these, endpoints that derive from methods which are
16 included in this document are to be considered with priority since there is an intensive
17 research ongoing in the field of test methods for endocrine disruption. As is generally the
18 case in the evaluation of the non-testing data, the quality of experimental data they are
19 based on might also be important (e.g. does it come from a single source or it is
20 compilation from different sources).

21 Information from *in vitro* screening assays:

22 Although there are principally *in vitro* systems for the study of all kinds of endocrine
23 systems and mechanisms in use in scientific research, the most relevant methods to
24 date are those related to the sexual steroid hormones, which are described in section 3.
25 Other types of assays, e.g. *in vitro* thyroid receptor binding assays, may become more
26 important in the future.

27 Given the high degree of conservation of the molecular components of endocrine
28 systems across vertebrate taxa, the ability of a substance to bind to a mammalian
29 hormone receptor, activate transcription of hormone-responsive genes or interfere with
30 steroid hormone biosynthesis in a mammalian cell line may suggest similar activity in
31 aquatic vertebrates.

32 Regarding the relevance of test results, the usual limitations of *in vitro* methods apply:
33 focus on a single mechanism of action *in vitro* vs. the diversity and complexity of
34 molecular structures and regulatory pathways *in vivo*; lacking or limited metabolic
35 capacity of some test systems; disregard of complex physiological processes, such as the
36 toxicokinetic distribution of a substance, the organ- or tissue-specific expression of its
37 molecular targets, feedback regulations or mechanisms of adaptation.

38 Information from mammalian toxicity testing:

39 Standard studies on repeated dose toxicity, long-term toxicity and carcinogenicity,
40 reproductive and developmental toxicity or non-standard studies on specific endocrine
41 mechanisms in mammals can provide indications of endocrine activity that might also be
42 of relevance for aquatic vertebrates.

1 With respect to the sexual hormone system, this includes changes in endocrine-
2 responsive tissues (gonads, secondary sex organs), reproductive functions (estrous
3 cycling, spermatogenesis, mating behaviour, fertility, gestation, parturition or lactation)
4 or developmental landmarks (e.g. anogenital distance, vaginal opening, preputial
5 separation). All of these changes might be caused by impact on molecular pathways that
6 are also present in aquatic vertebrates such as interactions with steroid hormone
7 receptors or biosynthesis, transport and metabolism of steroid hormones.

8 Indications of thyroid activity include developmental impairments, histopathological
9 changes of the thyroid gland or (not routinely investigated) thyroid hormone levels.

10 Weight of Evidence:

11 If there is information available for the same chemical from different sources, the
12 following questions should be considered for the overall conclusion: Is the information
13 consistent or is it in conflict with each other? In the case of conflicting data, the quality
14 of each piece of information should be evaluated in accordance with the principles
15 described in section 4, as should its biological relevance with respect to aquatic
16 organisms, and, finally, the potential impact of such information on the overall
17 regulatory decision.

18 **2. Indication of specific endocrine activity in intact aquatic organisms**

19 Evidence that a substance can operate by a specific endocrine mode of action in aquatic
20 organisms can only be derived from the investigation of specific, endocrine-responsive
21 endpoints. None of these are covered by standard aquatic toxicity testing. Endocrine-
22 specific screening assays are, however, under development and validation for both
23 mammalian rodents (uterotrophic and Hershberger assays) and for aquatic vertebrates
24 (21-day fish screening assay and amphibian metamorphosis assay).

25 In the endocrine specific aquatic assays, vitellogenin in fish responds to estrogens
26 (induction in males) and aromatase inhibitors (suppression in females), and secondary
27 sexual characteristics in fish respond to androgens (induction in females). Specifically for
28 the stickleback, spiggin may also provide the means to specifically characterise (anti-
29)androgenic modes of action. Specificity and significance of other endpoints such as other
30 biochemical parameters (e.g. hormone levels) or histopathological changes of the
31 gonads, including impairment of spermatogenesis, are under debate. The specific
32 endpoints which are included in the 21d-Fish Screening Assay can also be assessed in
33 conjunction with higher tier chronic tests. As isolated information, biomarker responses
34 cannot be used for regulatory conclusions. They may raise a strong concern that the
35 substance in question might cause serious long-term adverse effects, in particular if
36 environmental exposure, persistence and/or bioaccumulation are high. Such a concern
37 may lead to a specific request for further investigations by a Competent Authority in the
38 course of dossier or substance evaluation.

39 Evidence of thyroid activity is provided by histopathological changes to the thyroid gland,
40 which can be observed in the Amphibian Metamorphosis Assay or similar test systems. If
41 a protocol was used in accordance to the current OECD test guideline development,
42 effects information on the progress of metamorphosis will be available from the same
43 study and can be considered for use in regulatory decisions as outlined below. Thyroid
44 histology reported as isolated information may not be suitable for use in regulatory

1 decisions. It may support the interpretation of other toxicity data, also from mammalian
2 toxicity studies. It may also raise a strong concern that the substance in question might
3 cause serious long-term adverse effects, in particular if environmental exposure,
4 persistence and/or bioaccumulation are high. Such a concern may lead to a specific
5 request for further investigations by a Competent Authority in the course of dossier or
6 substance evaluation.

7 Evidence of specific endocrine mode of action in invertebrates as isolated information will
8 only be found in very rare cases and no general guidance can be given for its use.

9 **3. Characterisation of long-term adverse effects**

10 The reproductive capacity of fish can be adversely affected by a number of mechanisms
11 of toxicity. Observation of such effects, which can threaten fish populations, can be
12 made during studies that cover a distinct sensitive life stage such as sexual development
13 or active reproduction or studies that cover a complete life-cycle or even two or more
14 consecutive generations. Only the latter allow the identification of delayed reproductive
15 effects through endocrine disruption during early life stages. Information on sublethal
16 adverse effects, if judged as adequate, should be considered for use in PBT assessment or
17 Chemical Safety Assessment/PNEC derivation. Classification as R52 or R53 (CLP: Aquatic
18 Chronic4: H413) according to the safety net criteria might be proposed. A causal link
19 between a reproductive adverse effect and an endocrine mode of action might prompt a
20 proposal for identifying the substance as a substance of very high concern (Annex XV)
21 by a Competent Authority. If the adverse effects information is provided by a
22 reproductive and developmental study similar to those currently under development in
23 the OECD TG programme, information on endocrine-specific endpoints will be available
24 from the same study and assessment of a causal link may be possible based on similar
25 dose responses.

26 Long-term toxicity caused by chemicals with thyroid activity can be manifest as
27 developmental disturbance, e.g. promotion or inhibition of amphibian metamorphosis.
28 Similar considerations apply as outlined above for adverse effects in fish.

29 Adverse effects on development or reproduction of invertebrates may be reported from
30 non-standard studies and, if rated adequate, should be considered for use in the
31 assessment of chronic toxicity. A causal link to a specific endocrine mode of action will
32 only be found in rare cases.

33

1 **Table R.7.8—4 Integrated assessment of potential endocrine activity in aquatic organisms; based on the evaluation of available information which is**
 2 **not part of the REACH requirements**
 3

1. Preliminary indication of potential endocrine activity in aquatic organisms		
<i>Estrogen/androgen axis:</i>	<i>Thyroid:</i>	<i>Invertebrate systems:</i>
- molecular structure - mammalian toxicity - <i>in vitro</i> screening	- molecular structure - mammalian toxicity	- molecular structure
<p>-> determine concern of potential endocrine mode of action of the substance using Weight of Evidence of all available information, including environmental fate and exposure</p> <p>-> strong concern may prompt a proposal by the Competent Authority to include the substance in the Community rolling action plan in order to perform a substance evaluation</p>		
2. Indication of specific endocrine modes of action in intact aquatic organisms		
<i>Estrogen/androgen axis:</i>	<i>Thyroid:</i>	<i>Invertebrate systems:</i>
- biochemical markers - morphological changes (- gonad histopathology)	- thyroid histopathology	- rare individual cases
<i>Study type:</i>	<i>Study type:</i>	
Fish Screening Assay Fish Sexual Develpt. Test Fish Reproduction Test Fish Full Life-Cycle Test	Amphibian Metamorphosis Assay	
<p>-> determine concern of potential endocrine mode of action in intact aquatic organisms using Weight of Evidence of all available information, including environmental fate and exposure</p> <p>-> strong concern may prompt a proposal by the Competent Authority to include the substance in the Community rolling action plan in order to perform a substance evaluation</p>		
3. Characterisation of long-term adverse effects [#]		
<i>Estrogen/androgen axis:</i>	<i>Thyroid:</i>	<i>Invertebrate systems:</i>
- fish (sexual) development - fish reproduction	- amphibian development	- development - reproduction
<i>Study type:</i>	<i>Study type:</i>	<i>Study type:</i>
Fish Sexual Develpt. Test Fish Reproduction Test Fish Full Life-Cycle Test	Amphibian Metamorphosis Assay	Invertebrate Reproduction or Life-Cycle Tests
<p>-> consider use of chronic NOEC/EC10 for PBT assessment and Chemical Safety Assessment</p> <p>-> consider classification and labelling according to safety net categories (R52, R53 or H413 according to CLP)</p> <p>-> causal link of adverse effect with an endocrine mode of action may prompt consideration for Annex XV by CA</p>		

4 [#]It should be noted that the listed adverse effects, which may occur as a result of endocrine
 5 activity of a substance, may also be caused by other mechanisms of toxicity

6

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8

9 **R.7.8.7 Introduction to sediment organisms' toxicity**

10 Substances that are potentially capable of depositing on or sorbing to sediments to a
11 significant extent have to be assessed for toxicity to sediment-dwelling organisms. In
12 addition, marine sediment effects assessment is necessary for substances that are
13 known to be persistent in marine waters and may accumulate in sediments over time. In
14 general substances with a $K_{oc} < 500 - 1000$ l/kg are not likely sorbed to sediment
15 (SETAC 1993). According to this, a $\log K_{oc}$ or $\log K_{ow}$ of ≥ 3 is used as a trigger value for
16 sediment effects assessment although other considerations or combinations of triggers
17 might be important as well (e.g. binding to sediment particles that is not K_{ow}/K_{oc}
18 driven, but where for instance the distribution coefficient K_d is important, persistence in
19 the sediment compartment).

20 **R.7.8.7.1 Definition of toxicity to sediment organisms**

21 Sediments may act as both a sink for chemicals through sorption of contaminants to
22 particulate matter, and a source of chemicals through resuspension. Sediments integrate
23 the effects of surface water contamination over time and space and may thus present a
24 hazard to aquatic communities (both pelagic and benthic) which is not directly
25 predictable from concentrations in the water column.

26 The sorption or binding behaviour of chemicals to sediment is determined by certain
27 properties. Especially substances with high $\log K_{ow}$ or $\log K_{oc}$ values adsorb to the
28 organic fraction of the sediment. In addition, substances that bind to components of the
29 sediment via chemical reactions or substances that ionically bind to inorganic as well as
30 organic fractions may accumulate in the sediment.

31 Effects on benthic organisms are of concern because they constitute an important link in
32 the aquatic food chain and play an important role in the recycling of detritus material.
33 Whole-sediment tests using benthic organisms are most suitable for a risk assessment
34 for the sediment compartment. By using such tests it is possible to adequately address
35 all routes of exposure. Due to the generally long-term exposure of benthic organisms to
36 sediment-bound substances, long-term tests with sublethal endpoints like reproduction,
37 growth or emergence are most relevant. Field and mesocosm studies should be
38 considered to validate results of laboratory studies, particularly for substances where
39 sediment ageing processes have been shown to occur (e.g. like for nickel, as shown in
40 Costello *et al.*, 2011).

1 **R.7.8.7.2 Objective of the guidance on toxicity to sediment** 2 **organisms**

3 The main objective is to provide guidance to registrants on sediment toxicity testing and
4 to allow registrants to develop an Integrated Testing Strategy (ITS) for sediment toxicity
5 (defined in details in section 7.8.14).

6 The aim of sediment toxicity tests is to find out at which concentrations a substance
7 adsorbed or bound to sediment exhibit toxic effects to benthic organisms. Special
8 attention should be given to the pathways by which the test organisms are exposed to
9 the substance. In particular spiking methodology should be considered in detail and be
10 performed in the most realistic way possible (e.g. Brumbaugh *et al.*, 2013).

11 The determination of the concentration-response relationship should lead to the
12 identification of the No Observed Effects Concentration NOEC or EC₁₀ from long-term
13 tests (or median lethal concentration LC₅₀ from acute tests in some cases). This
14 NOEC/EC₁₀ (or LC₅₀) is subsequently used for deriving a Predicted No Effect
15 Concentration for the sediment (PNEC_{sediment}). In general, EC₁₀ values are preferred as
16 these are statistically derived from the entire dataset, and less dependent on test design
17 considerations than the NOEC. The use of acute studies is not recommended and
18 preference should be given to the use of chronic data. This PNEC_{sediment} is compared with
19 the Predicted Environmental Concentration in the sediment (PEC_{sediment}) to decide
20 whether there is a risk to sediment organisms from the exposure to the substance (see
21 Part E of the Guidance on IR&CSA on risk characterisation).

22 **R.7.8.8 Information requirements for toxicity to sediment organisms**

23 The information requirements for sediment toxicity are described by REACH Annexes VII
24 to XI, that specify the information that shall be submitted for registration and evaluation
25 purposes.

26 For this endpoint information requirements are formulated for substances produced or
27 imported in quantities of ≥ 1000 t/y (Annex X to REACH).

Column 1	Column 2
Standard information required	Specific rules for adaptation from column 1
9.5.1 Long-term toxicity to sediment organisms	9.5.1 Long-term toxicity testing shall be proposed by the registrant if the results of the chemical safety assessment indicate the need to investigate further the effects of the substance and/or relevant degradation products on sediment organisms. The choice of the appropriate test(s) depends on the result of the chemical safety assessment.

28 **R.7.8.9 Information on toxicity to sediment organisms and its sources**

29 For most substances uptake from water (bioconcentration, defined as the net result of
30 uptake, transformation, and elimination of a substance in an organism due to
31 waterborne exposure) is believed to be the predominant route of exposure for aquatic

1 organisms. For organic substances and metals pore water is one of the primary exposure
2 routes for benthic organisms (Di Toro *et al.*, 1991; Ankley *et al.*, 1991). However, for
3 highly lipophilic compounds or other substances that adsorb to particles (e.g. metals),
4 uptake from food or sediment may contribute to the overall exposure, depending on the
5 living and feeding strategy of the exposed organisms. Dietary exposure is important for
6 explaining substantial proportions of steady state tissue concentrations for exposed
7 organisms. The importance of dietary exposure relative to water exposure as a cause of
8 toxicity is currently not fully understood. In summary, factors that influence adsorption
9 and thus distribution between sediment and water influence also toxicity to aquatic
10 (pelagic and benthic) species. A compilation of such factors is given in [Appendix R.7.8–](#)
11 [1](#).

12 **R.7.8.9.1 Data on toxicity to sediment organisms – Information** 13 **sources**

14 **Testing data on toxicity to sediment organisms**

15 Numerous standardised test methods for sediment tests are available and many different
16 benthic organisms are proposed in these guidelines. Registrants should clearly report
17 and justify deviations from guidelines. Hereinafter an overview of the available
18 standardised (short- and long-term) test methods for sediment with benthic organisms is
19 given. In Table R.7.8–5 different test species are further characterised in terms of the
20 taxonomic group, habitat and feeding mode.

21 Whenever new sediment toxicity data is generated, accepted long-term guideline studies
22 are preferred. For existing studies, non-standard, non-guideline studies may be
23 acceptable if these are well documented, relevant and of high quality. Often such studies
24 are used in weight-of-evidence approaches.

25 OECD test guidelines exist for insects and midge larvae *Chironomus sp.* (OECD 218 and
26 233), oligochaetes *Lumbriculus sp.* (OECD 225), and *Myriophyllum spicatum* (OECD
27 239). The three OECD guidelines that are most relevant when generating new data for
28 REACH purposes are OECD 218, 225 and 233. Each of these guidelines covers
29 ecologically relevant long-term toxicity endpoints and thus generates information
30 appropriate for the fulfilment of the information requirements of REACH Annex X 9.5.1
31 (*Long-term toxicity testing on sediment organisms*). Nevertheless OECD 233 is the most
32 comprehensive as it covers all relevant reproductive endpoints and offers a more
33 complete level of information. The relative sensitivity of OECD 218 and 225 is substance
34 dependent. As an example, OECD 218 (or OECD 233) is more relevant than OECD 225 if
35 arthropods are suspected to be particularly sensitive or if toxicity is due to metabolic
36 activation (see for instance Nowell *et al.*, 1999). A guideline for rooted plants
37 (*Myriophyllum spicatum*) is also available (OECD 239). Registrants should choose the
38 most appropriate and sensitive test protocol(s) based on, for example, substance
39 properties/uses and provide a justification for the choice. The proceedings of the ECHA
40 topical scientific workshop on sediment risk assessment offer additional information on
41 the relevance of the different taxonomic groups and exposure groups that should be
42 considered in the selection of the test species (ECHA 2014).

43 Standardised tests from ASTM, US EPA and ISO are also available with other fresh- and
44 marine water species, such as crustacean amphipods *Hyalella sp.*, *Gammarus sp.* and

1 nematodes e.g. *Caenorhabditis elegans*. Nematodes are commonly found in the
2 sediment compartment and are thus biologically relevant species to be studied. The
3 feeding strategy of the nematode species should be considered in connection with the
4 binding process of the chemical to sediment particles, as in general nematodes are
5 selective feeders and do not ingest the sediment particles; a justification for the
6 selection of the species should be provided. Polychaetes, amphipods, molluscs such as
7 bivalves are recognised test species for the estuarine and marine environment. Test
8 methods are available for *Arenicola marina*, *Corophium volutator*, *Leptocheirus*
9 *plumulosus*, and *Amphiascus tenuramis*, and tests with early life stages of sea urchins or
10 bivalves that would be more representative of the sediment-water interface.

11 Details of the most common guidelines for sediment toxicity testing are given in the
12 sections below.

13 OECD Test Guidelines

14 Test No 218: Sediment-water chironomid toxicity using spiked sediment¹²

15 Test No 219: Sediment-water chironomid toxicity using spiked water¹³

16 Both guidelines are designed for studying long-term toxicity (28d exposure) of
17 substances to the sediment-dwelling larvae of the freshwater midge *Chironomus* sp.
18 Measured endpoints are total number of adults emerged and time to emergence. Spiking
19 the sediment (OECD 218) is recommended for continuous and intermittent release of
20 substances while spiking the water phase (OECD 219) was initially developed for
21 pesticide specific exposure situations. Therefore, OECD TG 219 is in principle not
22 acceptable unless a case-by-case justification for its suitability, e.g. related to the
23 expected environmental release conditions, is provided.

24 Test No 233: Sediment-water chironomid life-cycle test using spiked water or spiked
25 sediment¹⁴

26 This test is an extension of the OECD test guideline 219 (spiked water) or 218 (spiked
27 sediment). The guideline is designed to assess the effects of prolonged exposure of
28 *Chironomus* sp. to substances. The sediment-dwelling freshwater dipteran *Chironomus*
29 sp. is exposed to throughout its life-cycle to water- or sediment-spiked substances.

30 The complete exposure duration is circa 44 days for *Chironomus riparius* and *C.*
31 *yoshimatsui*, and circa 100 days for *C. dilutus*. Chironomid emergence, time to
32 emergence, and sex ratio of the fully emerged and living midges are assessed.

33 Test No 225: Sediment-water *Lumbriculus* toxicity test using spiked sediment¹⁵

¹² See OECD library at http://www.oecd-ilibrary.org/environment/test-no-218-sediment-water-chironomid-toxicity-using-spiked-sediment_9789264070264-en.

¹³ See OECD library at http://www.oecd-ilibrary.org/environment/test-no-219-sediment-water-chironomid-toxicity-using-spiked-water_9789264070288-en.

¹⁴ See OECD library at http://www.oecd-ilibrary.org/environment/test-no-233-sediment-water-chironomid-life-cycle-toxicity-test-using-spiked-water-or-spiked-sediment_9789264090910-en.

- 1 This Test Guideline is designed to assess the effects of prolonged exposure (28 days) to
2 sediment-associated substances on the reproduction and the biomass of the endobenthic
3 oligochaete *Lumbriculus variegatus* (Müller).
- 4 The measured endpoints are reproduction and biomass (ECx and/or NOEC/LOEC).
- 5 Test No 239: Water-Sediment *Myriophyllum spicatum* toxicity test¹⁶
- 6 This test guideline is designed to assess the toxicity of substances on the growth of
7 rooted aquatic plants (*Myriophyllum spicatum*) growing in a water-sediment system (in
8 particular situations the test guideline can also be adapted for use with other species
9 such as the reed *Glyceria maxima*).
- 10 Shoot apices of healthy and non-flowering plants are exposed over a period of 14 days.
11 The measured quantitative variables include assessment of shoot growth expressed as
12 both weight (fresh and dry) and length (fresh). The measured qualitative variables
13 include presence or not of chlorosis and necrosis or growth deformities. Normally,
14 exposure via sediment is the relevant route of exposure for sediment risk assessment.
- 15 Test No 235: *Chironomus sp.*, acute immobilisation test¹⁷
- 16 This Test Guideline describes an acute immobilisation assay on chironomids and is
17 designed to complement the existing Test Guidelines for chironomid chronic toxicity
18 assays (OECD 218, 219 and 233).
- 19 The test method is based on OECD 202: *Daphnia sp.* Acute Immobilisation Test. First
20 instar *Chironomus sp.* larvae are exposed to a range of concentrations of the test
21 substance in water-only vessels for a period of 48 hours. *C. riparius* is the preferred
22 species but *C. dilutus* or *C. yoshimatsui* may also be used for the test. Immobilisation is
23 recorded at 24 and 48 hours, and if data allow, the EC50 is calculated at 24 and 48
24 hours. A limit test with a single concentration may also be performed at 100 mg/L of test
25 substance or up to the practical limit of solubility (whichever is lowest) in order to
26 demonstrate that the EC50 is greater than this concentration.
- 27 ASTM Test Guidelines
- 28 A number of ASTM guidelines with different species are available¹⁸. Most of the cited
29 ASTM guidelines are designed to be short-term tests (10-d exposure) with mortality as
30 endpoint. However, for some of these species (*Hyalella azteca*, *Chironomus sp.*,
31 *Leptocheirus plumulosus*, *Neanthes arenaceodentata*) also long-term toxicity tests (28d
32 exposure) with sublethal endpoints are recommended by the guidelines.

¹⁵ See OECD ilibrary at http://www.oecd-ilibrary.org/environment/test-no-225-sediment-water-lumbriculus-toxicity-test-using-spiked-sediment_9789264067356-en

¹⁶ See OECD ilibrary at http://www.oecd-ilibrary.org/environment/test-no-239-water-sediment-myriophyllum-spicatum-toxicity-test_9789264224155-en.

¹⁷ See OECD ilibrary at http://www.oecd-ilibrary.org/environment/test-no-235-chironomus-sp-acute-immobilisation-test_9789264122383-en.

¹⁸ ASTM test guidelines: <http://www.astm.org/Standard/standards-and-publications.html>

- 1 E1706-05. Standard test method for measuring the toxicity of sediment-associated
2 contaminants with freshwater invertebrates: a short- or long-term test described for
3 *Chironomus sp.*, *Hyalella azteca*, *Hexagenia spp.*, *Tubifex tubifex*, or *Diporeia sp.*
- 4 E1611-00. Standard guide for conducting sediment toxicity tests with marine and
5 estuarine polychaetous annelids: a short- or long-term test described for *Neanthes*
6 *arenaceodentata* or *Neanthes virens*.
- 7 E1367-03e1. Standard test method for measuring the toxicity of sediment-associated
8 contaminants with marine and estuarine invertebrates: a short-term test described for
9 *Leptocheirus plumulosus*, *Ampelisca abdita*, *Eohaustorius esturarius*, *Rhepoxynius*
10 *abronius*.
- 11 E2591-07. Standard guide for conducting whole sediment toxicity tests with amphibians:
12 a short-term test described for *Rana pipiens*, *Rana clamitans*, *Rana sylvatica*, *Bufo*
13 *americanus*.
- 14 The general procedures described in the standards E1611-00 and E1367-03e1 might also
15 be useful for conducting tests with other estuarine or marine invertebrates.
- 16 US-EPA Test Guidelines
- 17 **EPA 600/R-99/064** Methods for measuring the toxicity and bioaccumulation of
18 sediment-associated contaminants with freshwater invertebrates.
- 19 • 100.1: *Hyalella azteca* 10-d survival and growth test for sediments (short-
20 term)
 - 21 • 100.2: *Chironomus dilutus* (previously named *C. tentans*): 10-d survival and
22 growth test for sediments (short-term)
 - 23 • 100.4: *Hyalella azteca*: 42-d test for measuring the effects of sediment-
24 associated contaminants on survival, growth and reproduction (long-term)
 - 25 • 100.5: 50 – 65-d life-cycle test for measuring the effects of sediment-
26 associated contaminants to *Chironomus dilutus* (long-term)
- 27 **EPA 600/R-94/025** Methods for assessing the toxicity of sediment-associated
28 contaminants with estuarine and marine amphipods.
- 29 • 100.4: 10-d test for measuring the effects of sediment-associated
30 contaminants on survival with *Ampelisca abdita*, *eohaustorius estuaries*,
31 *Leptocheirus plumulosus*, or *Rhepoxynius abronius*. Reburial of surviving
32 amphipods in control sediment is an additional measurement that can be used
33 as an endpoint.
- 34 EPA 600/R-01/020 Method for assessing the chronic toxicity of marine and estuarine
35 sediment-associated contaminants with the amphipod *Leptocheirus plumulosus*. 28-d
36 test with survival, growth and reproduction as endpoints (long-term).
- 37 ISO test guidelines
- 38 **ISO 16712:2005** Water quality - Determination of acute toxicity of marine or estuarine
39 sediment to amphipods. Method for the determination of acute toxicity to amphipods

1 (e.g. *Gammarus sp*, *Corophium sp*), including a scenario for exposure over a period of
2 10-d to substances or preparations spiked into clean sediment, samples of contaminated
3 marine or estuarine sediments or substance, industrial or municipal sludge, or other solid
4 wastes that may combine with marine or estuarine sediments (short-term).

5 **ISO 10872:2010:** Water quality - Determination of the toxic effect of sediment and soil
6 samples on growth, fertility and reproduction of *Caenorhabditis elegans* (*Nematoda*)
7 Method for the determination of toxicity of environmental samples on growth, fertility
8 and reproduction of *Caenorhabditis elegans*, a bacterivorous nematode found primarily in
9 terrestrial soils but also in aquatic sediments of polysaprobial fresh-water systems. The
10 method is applicable to contaminated whole fresh-water sediment (maximum salinity 5
11 %), soil and waste, as well as elutriates and aqueous extracts thereof, and to pore
12 water. This test has a duration of only 72 h, but as it measures both growth and
13 reproduction endpoints it can be considered as a long-term test. However, the result
14 from this test alone cannot be used alone for the derivation of the PNEC_{sediment}.

15 **ISO 14371:2012:** Water quality - Determination of fresh water sediment toxicity to
16 *Heterocypris incongruens* (*Crustacea, Ostracoda*)

17 A direct contact test for the determination of the percentage mortality and/or growth
18 inhibition on the cosmopolitan freshwater ostracod *Heterocypris incongruens* (Ramdohr,
19 1808) after a 6-d exposure to whole sediment. This is a short-term test

20 **ISO 16191:2013:** Water quality - Determination of the toxic effect of sediment on the
21 growth behaviour of *Myriophyllum aquaticum*

22 A method for determining the toxicity of environmental samples on the growth of the
23 macrophyte plant *Myriophyllum aquaticum*. The method is applicable to natural
24 freshwater sediment and to artificial sediment. The endpoint measured is inhibition of
25 growth (short-term).

26 **ISO 16303:2013:** Water quality - Determination of toxicity of fresh water sediments
27 using *Hyalella azteca*

28 A method for the determination of toxicity to young *Hyalella azteca* in whole sediment
29 (freshwater or brackish) based on survival and growth inhibition after 14 d and/or 28 d
30 (short-term/long-term).

31 OSPAR Guideline

32 (OSPAR 2005): A Sediment Bioassay using an Amphipod *Corophium sp*. – Marine
33 sediment toxicity test. Either *Corophium volutator* or *Corophium arenarium* may be
34 used. In the test adult *Corophium* are exposed to spiked sediments for 10 days.
35 Endpoints are survival and burrowing activity (short-term).

36 Note that, in addition to the guidelines described above, also Environment Canada
37 (1997a, 1997b) for instance has a collection of biological test methods for testing

1 freshwater sediment species *Hyalella azteca*, *Chironomus dilutus* or *Chironomus riparius*
2 and marine or estuarine amphipods or luminescent bacteria¹⁹.

3 Non-standard test methods

4 There are many non-standard methods available for the testing of effects of substances
5 on sediment organisms. An overview of available non-standard test methods can be
6 found in OECD (1998). To ensure a transparent assessment of the data adequacy,
7 relevance and reliability, detailed reporting of a study is especially important for
8 acceptability of data obtained from non-standard methods. Information on what should
9 be reported in a robust study summary (RSS) or study summary (SS) is given in the
10 ECHA Practical Guide on *How to report robust study summaries*²⁰.

11 Information obtained from non-standard methods may best be used in a *Weight of*
12 *Evidence* (WoE) approach: using this approach, several lines of evidence that would not
13 be sufficient as stand-alone information to fulfil the endpoint may be combined to reach
14 a conclusion on a property of a substance. More information on WoE approaches is given
15 in *Chapter R.4 Evaluation of available information* of the REACH *Guidance on IR&CSA*.
16 Any WoE approach submitted should fulfil the criteria set in REACH Annex XI section 1.2.
17 Acceptability of such approaches is always case specific.

18 Tests performed without sediment

19 There are several non-standard tests available in which benthic organisms are exposed
20 in a water-only test system to the substance in question. Such tests do not take into
21 account the different routes of exposure that may occur under environmental conditions.
22 Therefore, for the derivation of the PNEC_{sediment}, such tests can only be used for
23 screening purposes in combination with the equilibrium partitioning method. In addition,
24 if compared with sediment tests on the same species in the presence of sediment such
25 tests may provide information on the importance of sediment ingestion.

26 **R.7.8.10 Evaluation of available information on toxicity to sediment** 27 **organisms**

28 A general overview of the properties of substances and test systems that influence the
29 evaluation of aquatic toxicity tests are described in section 7.8.4 and [Appendix R.7.8–1](#).
30 Some of these properties are also related to sediment toxicity.

31 **R.7.8.10.1 Data on toxicity to sediment organisms – Evaluation of** 32 **information**

33 **Non-testing data on toxicity to sediment organisms**

34 For most substances the availability of experimental data on sediment organisms is
35 limited. In the absence of such data, a read-across from pelagic effect values is possible

¹⁹ Biological Test methods Series are published at: <http://ec.gc.ca/faunescience-wildlifescience/default.asp?lang=En&n=0BB80E7B-1>.

²⁰ Practical Guides are available on the ECHA website at: <http://www.echa.europa.eu/practical-guides>.

1 as a screening approach (equilibrium partitioning method, EPM) (for more information
2 see Chapter R.10). It has to be considered that the equilibrium partitioning method may
3 result in either an overestimation or underestimation of the toxicity to benthic organisms
4 (Di Toro *et al.*, 2005). Therefore, this method can only be used as a rough screening to
5 help determine whether sediment toxicity tests with benthic organisms are required.

6 General guidance on how to extrapolate via read-across or substance categories is given
7 in Section R.6.2. There is currently not enough sediment toxicity data to validate
8 Quantitative Structure Activity Relationships (QSAR) models for sediment toxicity. Their
9 use for sediment toxicity assessment is hence limited.

10 Equilibrium partitioning method

11 In the absence of any ecotoxicological data for sediment-dwelling organisms, the $PNEC_{sed}$
12 may be provisionally calculated using the equilibrium partitioning method (EPM). This
13 method uses the $PNEC_{water}$ for aquatic organisms and the suspended matter/water
14 partitioning coefficient as inputs (e.g. Di Toro *et al.*, 1991). For advice on the actual
15 calculation of the $PNEC_{sediment}$ using the EPM ($PNEC_{sediment\ screen}$), please refer to Chapter
16 R.10 of the Guidance on IR&CSA (section 10.5). Normally, EPM can only be applied to
17 neutral organic chemicals.

18 Several factors have to be considered when using this method. To increase the
19 reliability of $PNEC_{sediment\ screen}$ derived using the EPM, it is imperative that a conservative
20 but realistic partitioning coefficient (e.g. K_d , K_{oc} , K_{ow}) is chosen. A clear justification
21 must be given for the chosen coefficient and any uncertainty should be described in a
22 transparent way.

23 The EPM takes into account only uptake via the water phase and includes a
24 normalisation to 5% organic matter (OC)²¹. However, uptake may also occur via other
25 exposure pathways like via ingestion of and direct contact with sediment depending on
26 the organism used for testing. Especially for highly adsorbing substances these
27 additional uptake routes may be important. Therefore, in order to account for the
28 increased importance of uptake via the gut with increasing adsorption, for compounds
29 with a log K_{ow} greater than 5, the EPM can only be used in a modified way. For such
30 substances, an additional factor of 10 is applied to the PEC/PNEC ratio. As already
31 highlighted, the EPM is considered only as a screening tool for assessing the level of risk
32 to sediment-dwelling organisms. If with this method a PEC/PNEC ratio >1 is derived,
33 then data improvement is necessary either by refining the exposure assessment or by
34 performing tests with benthic organisms, preferably using spiked sediment, to support a
35 refined risk assessment for the sediment compartment.

36 EPM is based on sorption to organic matter. Therefore, it cannot be used for some
37 classes of substances, e.g. when binding behaviour is not driven by lipophilicity (e.g.
38 aromatic amines forming covalent bonds to sediment components, ionisable

²¹ To be noted that EUSES calculated PECs regional are also normalised to 5% OC while PECslocal are normalised to 10%.

1 substances²², surface active substances). Substances that do not exhibit a toxic effect
2 when tested in water-only test systems, for example because equilibrium was not
3 reached during exposure phase due to low water solubility, may nevertheless exert
4 significant toxic effects in sediment tests as these substances may accumulate in
5 sediments. As no real PNEC_{aquatic} has been derived, the EPM cannot be used to derive the
6 PNEC_{sediment screen}. The EPM is thus not applicable for instance with poorly water soluble
7 substances for which no effects are observed in aquatic studies. For such substances, at
8 least one sediment study has to be performed for a more realistic sediment risk
9 assessment.

10 The testing strategy developed for sediment toxicity assessment is explained in Section
11 R.7.8.12 of this Guidance.

12 **Testing data on toxicity to sediment organisms**

13 The effects of sediment-bound substances on benthic organisms can be best assessed by
14 performing long-term whole-sediment tests that take into account all possible routes of
15 exposure (overlying water, pore water, ingestion of sediment, direct contact with
16 sediment) that may occur in the environment. In general, sediment tests with water-
17 only systems may only be used for screening purposes in combination with the EPM. If
18 EPM does not indicate a risk and a water-only study also indicates a high NOEC/EC10,
19 the confidence in the EPM result could in some cases be high. Bioaccumulation studies
20 can be instructive to decide on the need for sediment testing or on the species to be
21 tested. For instance, a very poorly water soluble substance that does not exert effects in
22 aquatic studies, but shows a relatively high bioaccumulation potential very likely needs a
23 sediment risk assessment.

24 In general, for tests that have been performed according to standard test guidelines, the
25 validity criteria or acceptability requirements specified in these guidelines have to be
26 fulfilled for acceptance of the study. Due to the complex test system, results from whole-
27 sediment tests may be influenced by several parameters (e.g. sediment composition,
28 spiking method, feeding mode of exposed organisms). Critical factors that are important
29 for evaluating sediment toxicity tests (standard and non-standard tests) are discussed
30 below. It is important that the registrant clearly justifies his choices, e.g. test system,
31 test species, method of spiking etc. as outlined below.

32 Test organisms and species selection

33 Only species that act as ecological representatives for the sediment compartment are
34 acceptable as test organisms. The available test methods (see Section [R.7.8.9](#)) refer
35 mostly to invertebrates of the trophic level *primary consumer* or *decomposer*. The
36 number and types of species presently used in (standard) test protocols may be
37 insufficient to reflect all of the ecological/physiological aspects (and possibly the
38 sensitivity) of benthic communities. For example, rooted aquatic plants and
39 microorganisms are currently poorly covered. The OECD 239 test (with the rooted plant
40 *Myriophyllum spicatum*), for instance, was only adopted in September 2014. Efforts are

²² In this context are considered as ionisable those substances which present that characteristic at environmental pH (4-9).

1 being made to extend the knowledge to cover more ecological/physiological aspects (see
2 for instance Diepens *et al.*, 2014a; Diepens *et al.*, 2014b). Therefore, the concept of
3 covering several trophic levels which has been applied for the pelagic compartment
4 cannot be followed for the sediment. Instead, the test species should cover different
5 habitats and feeding strategies in the sediment. Further, different taxonomic groups
6 (normally species from different phyla, subphyla, or in case of *Arthropoda* classes)
7 should be represented. Usually, a distinction is made between epibenthic species (living
8 on or slightly above the sediment surface) and endobenthic species (burrowing in the
9 sediment). Regarding invertebrates, different exposure conditions and feeding strategies
10 should be represented by species representing a variety of life strategies, where
11 possible: (1) surface deposit and/or filter feeders; (2) sub-surface feeders; (3)
12 burrowing species with a combined surface and sub-surface feeding behaviour. These
13 different exposure routes and feeding behaviours imply differences in sediment ingestion
14 rates, in the degree of contact with the sediment and in the exposure through pore
15 water and overlying water. Each group represents different energy pathways and
16 different trophic levels in aquatic food webs and hence may express different responses
17 to substance exposures. If there are indications that plants are a sensitive group, tests
18 with (rooted) plant should be considered. However, in many cases there will not be a
19 large data set for the sediment compartment. The integrated testing strategy outlined in
20 Section R.7.8.14 below explains the minimum data set needed for sediment risk
21 assessment.

22 Substance properties and mode of action are also important parameters to consider
23 when selecting appropriate test organisms. Especially for strongly adsorbing or binding
24 substances (e.g. $\log K_{ow} > 5$) sediment-dwelling organisms that feed on sediment
25 particles (e.g. *Lumbriculus variegatus*, *Tubifex tubifex*) are usually most relevant.
26 However, also a specific mode of action that is known for a given substance may
27 influence the choice of the test species (e.g. for substances suspected of having specific
28 effects on arthropods a test with *Chironomus* is more appropriate than tests on other
29 *Phyla*). Knowledge about a (potential) mode of action similar to that of an insecticide or
30 fungicide (e.g. based on structural similarity) for substances registered under REACH can
31 be used to determine the species to be tested for fulfilling REACH requirements. Data on
32 pelagic species could highlight whether invertebrates or plants/algae are substantively
33 more sensitive; any data on terrestrial species could also highlight whether for instance
34 oligochaetes, arthropods, nematodes or plants are likely to be more sensitive. Similarly,
35 data from analogues can inform on the most relevant sediment species to be tested.

36 Additional species/groups might be added if a specific mode of action is observed or
37 predicted, such as endocrine disruption. In the latter case molluscs might for instance be
38 selected. Another example where alternative species should be additionally tested is
39 where echinoderms (only present in the marine compartment) are deemed important as
40 these may not be sufficiently protected using test data on the traditional invertebrates
41 given above (ECHA, 2014).

42 Endpoints

43 Endpoints studied in sediment toxicity tests should be of ecological relevance, i.e. where
44 possible showing effects relevant at the population level. For long-term tests the sub-
45 lethal endpoints reproduction, growth and (insect) emergence are most relevant.
46 Behavioural endpoints like sediment avoidance or burrowing activity have not been

1 standardised. Such endpoints can give indications on toxic effects but should not be
2 interpreted in isolation. For short-term tests survival is the normal endpoint to be
3 considered.

4 Some endpoints, particularly the reproduction ones, show a high variability which makes
5 a reliable evaluation of test outcome difficult. Further guidance can for instance be found
6 in OECD document on "Current approaches in the statistical analysis of ecotoxicity data:
7 a guidance to application" (OECD 2006).

8 Exposure pathways

9 Once substances have reached the benthic sediment compartment, there are three
10 possible exposure routes: (1) the sediment pore water (for benthic organisms that
11 burrow in the sediment); (2) the water overlying the sediment water interface (for
12 epibenthic organisms and for benthic organisms that burrow in the sediment and create
13 burrows that connect with the overlying water, and through which the overlying water
14 circulates); and (3) the ingestion and/or contact with sediment particles (for sediment-
15 ingesting organisms). For some species different routes of exposure could be relevant
16 according to the situation, depending on the food availability in the substrate (this is
17 particularly true for species subject to alternations between immersion and emersion
18 phases). Sediment organisms can thus be exposed via their body surfaces to substances
19 in solution in the overlying water and in the pore water and to bound substances by
20 direct contact or via ingestion of contaminated sediment particles. The exposure route
21 that is most important is strongly influenced by species-specific feeding mechanisms, gut
22 retention time and the behaviour of the organisms in or on the sediment. The dominant
23 exposure route may change in different life stages or due to different activities of a life
24 stage. For the evaluation of available sediment tests it has to be assessed which
25 exposure routes are covered by the test design and the test organisms used. For
26 strongly adsorbing or binding substances (e.g. $\log K_{ow} > 5$ or $\log K_{oc} > 3$), uptake from
27 food or sediment may contribute to overall exposure. For such substances preference
28 should be given to test designs and test organisms that cover the exposure via sediment
29 ingestion, as this is the most relevant exposure route for such substances. Care should
30 be taken to use the same metric in both effects (PNEC) and exposure assessment (PEC).
31 Concentrations in bulk sediment/overlying water/pore water/... must be measurable in
32 the test system(s) and matched by an exposure prediction (PEC) using the same metric.

33 Composition of sediment, artificial vs natural sediment

34 Both artificial and natural sediments have advantages and drawbacks.

35 Natural sediment could be considered of greater representativity and ecological
36 relevance. But commonly characterised natural sediments are not available on the open
37 market and they present the disadvantage of a more complicated collection,
38 characterisation, inter-study comparisons. Furthermore the residual contaminants that
39 may be found in natural sediment may make interpretation of results more complicated
40 (even if corrected for by the controls).

41 Many of the standard test methods advocate the use of artificial sediment as the solid
42 matrix for benthic effects assessment, on the basis of the assumption that results will be
43 more standardised if sediment components are well controlled, even if this approach
44 may entail decrease in environmental realism. Furthermore, the constituents of artificial

1 sediment are generally well characterised. However artificial sediment may separate into
2 layers according to particle size with the clay particles settling at the surface. Such
3 layering may prevent penetration of certain species into the sediment layer (Wiegelhofer
4 *et al.*, 2003). Furthermore, due to lack of significant microbial flora, results derived with
5 artificial sediment may not be the same as those derived with natural sediment.

6 On the whole, due to the level of characterisation and reproducibility possible, artificial
7 sediment is generally preferred over natural substrate (OECD 2004a and b) unless
8 effects at a specific local site are being considered. The use of standardised sediments is
9 also useful for quality control purposes. Nevertheless there are some exceptions where
10 natural sediments can be more useful (e.g. data rich metals requiring more realistic
11 equilibration in natural sediments).

12 Artificial sediment may be conditioned by continued mixing of the components for days
13 or even weeks prior to spiking to improve the homogeneity, increase the microbial flora
14 and transform the organic matter into a more environmentally realistic form. However,
15 such mixing may dramatically increase the Biochemical Oxygen Demand (BOD) of the
16 sediment-water system leading to a need for supplementary aeration to prevent
17 suffocation of test organisms.

18 In addition to the requirements outlined in the different guidelines, sediments used in
19 studies should be characterised by for example determining the particle size, organic
20 matter (OM) content, cation exchange capacity (CEC)/anion exchange capacity (AEC).
21 Usually, at least a normalisation to 5% OM content should take place, unless the
22 substance does not bind to the organic fraction of the sediment, but rather to the
23 inorganic fraction. Further, the sediments should preferably be characterised by origin
24 (natural sediments), pH and ammonium content of pore water, total organic carbon and
25 nitrogen content, particle size distribution and percent water content. When testing
26 metals, SEM (Simultaneously Extracted Metals) and AVS (Acid Volatile Sulfides)
27 concentrations should be measured as well as Fe and Mn (ICMM, 2002).

28 Grain size of the sediment used in the test may influence the bioavailability of the test
29 substance. It may also be an important factor in tests for other reasons. For example,
30 the extent to which bacteria can be adsorbed onto the sediment depends on particle
31 size. Likewise, different species of amphipods prefer sediments of different particle size
32 distributions. One should thus consider the tolerance of a given species with regard to
33 the grain size distribution of the sediments in question. Some further information can be
34 found in DeWitt *et al.* (1988) and Burton *et al.* (1991).

35 Method of spiking

36 There are two methods to spike a test substance into a test system: one method is to
37 spike the water phase, the other to spike the sediment phase. The selection of the
38 appropriate method depends on the intended application of the test. However, in
39 general, spiking of the sediment is preferred over spiking of the water phase. For both
40 methods an equilibration time without presence of the test organisms is necessary to
41 enable the distribution of the test substance between the water and sediment phases to
42 equilibrate according to the distribution behaviour of the substance, as explained below.

43 In some guidelines, such as the OECD 233, both water and sediment spike scenarios are
44 described. In OECD 233, the water exposure scenario is intended to simulate a pesticide

1 spray drift event to cover the initial peak concentration in surface waters. Water spiking
2 may also be useful for evaluating other types of exposure (including chemical spills), but
3 does not accurately represent accumulation processes within the sediment lasting longer
4 than the test period. If spiking via the water phase has been performed for a study, it
5 must be carefully considered whether an exposure via the sediment has also taken
6 place. If possible and relevant (e.g. in the absence of analytical measurements in
7 existing studies) sediment concentration should be calculated from the water
8 concentration using the equilibrium partitioning method (see Chapter R.10, section
9 10.5).

10 The scenario of spiking the sediment is intended to simulate accumulated levels of
11 substance persisting in the sediment. For industrial substances with continuous and
12 intermittent release, spiking the sediment is recommended. Spiking a sediment-water
13 test system can be difficult for poorly soluble substances. The standard approach is to
14 dissolve the test substance in a solvent and then to spike sand, blow-off the solvent and
15 then mix sediment with the remaining sand at various concentrations. The drawback
16 with this technique is that even after hours or sometimes days of mixing, the substance
17 may not be homogeneously mixed to the sediment but still present as solid particles on
18 the original sand and for some substances evaporation losses could occur. Roughly, a
19 Henry's law constant of 1-10 Pa.m³/mol can be used as an indication when issues with
20 volatility could become important. Use of an organic solvent added to wet sediment is
21 not recommended as this may have irreversible effects on the organic matter fraction of
22 the sediment (U.S. EPA 2000). Direct addition can in some cases be a viable alternative,
23 but has to be performed with care (e.g. achieving homogeneity can be very challenging).

24 Equilibrium between water-phase and sediment-phase

25 After spiking the water-sediment system with the test substance, an equilibration period
26 is necessary to ensure partitioning of the substance between the water-phase and solid-
27 phase according to the substance-specific distribution characteristics. This partitioning
28 should take place under the temperature and aeration conditions used during the
29 exposure phase. Appropriate equilibration time is sediment and substance specific and
30 can be in the order of hours to days and in some cases up to several weeks and might
31 require taking into account several considerations. In some cases a balance between
32 equilibration and degradation/hydrolysis might need to be found. This is for instance
33 acknowledged in the proposed guidance on a sediment-water *Lumbriculus* toxicity test
34 using spiked sediments (OECD 2007). Results of higher tier environmental fate studies
35 (e.g. degradation simulation testing, bioaccumulation) can inform on the appropriate
36 equilibration time.

37 For metals and inorganic metal compounds both short equilibration times and high
38 spiked metal concentrations in sediments will accentuate partitioning of metals to the
39 dissolved phase and increase the probability of exposure and/or toxicity via dissolved
40 metals (Lee *et al.*, 2004, Simpson *et al.*, 2004, Hutchins *et al.*, 2008, Brumbaugh *et al.*,
41 2013). As a consequence, for static and semi-static tests it is recommended that the
42 concentration of the test substance be measured in the overlying water, solid sediment
43 phase and pore water, and that testing be initiated only when the overlying water, solid
44 sediment and pore water concentrations reach steady state concentrations. Aging and
45 weathering processes may have an impact on sediment toxicity. Aging may involve the
46 redistribution of some metals from one solid phase to another, and this redistribution

1 can result in decreases in toxicity to benthic organisms (e.g. as shown in Costello *et al.*
2 (2011) for nickel). The rate at which these changes occur may be longer than the
3 duration of many chronic sediment toxicity tests, which suggests that laboratory tests
4 performed with metals spiked into natural sediments will be conservative, as they will
5 usually be too short in duration to capture ageing processes. Therefore, the influence of
6 ageing processes should be considered in a *Weight of Evidence* based analysis of
7 uncertainties that are applied to laboratory-derived PNEC values. However, currently
8 there are no agreed methods available to take these phenomena into account in
9 standard sediment test protocols and standardised test methods with artificial sediment
10 take little account of the impact of sediment aging processes occurring in the
11 environment.

12 Aging might also be relevant for some organic substances and is linked to bioavailability
13 (discussed under [R.7.8.10.3](#)), but less knowledge is available compared with metals.

14 Feeding

15 In long-term tests, especially with reproduction or growth as endpoint, feeding of the
16 test organisms is necessary. When possible according to the guideline, the tests should
17 be designed in such a way that the food necessary for the test organisms during the
18 study is added to the sediment prior to spiking with the test substance, especially for
19 strongly adsorbing substances (see for instance paragraph 31 of OECD TG 218 and 233).
20 Thereby, it is ensured that the food taken up by the test organisms is also contaminated
21 with the test substance comparable to environmental conditions. Food types are diverse
22 depending on the study, varying from ground, flaked fish food to plant material (e.g.
23 *Urtica* powder, ground sphagnum peat or alpha cellulose) to cultured *E. coli* cells at
24 known concentration. It has to be considered that any food added to the test system
25 either periodically or only at test initiation may influence water quality due to
26 degradation (see section on water quality below).

27 Duration of exposure

28 Most guidelines have clearly defined test durations or critical milestones (e.g. chironomid
29 emergence) that need to be achieved. A consideration in the selection of test guidelines
30 is the duration of exposure in a sediment test: it should be long enough to ascertain that
31 the test substance is really taken up by the test organisms. Especially for strongly
32 adsorbing substances it may take some time to reach equilibrium between the sediment
33 concentration in the test system and in the test organisms. It is recommended that a
34 sediment test should have a duration of at least 10 days. Most standardised test
35 methods (see Section R.7.8.9.1) include an exposure period of at least 10 days for
36 short-term and 28 days for long-term tests. However, there are other methods available
37 in which the exposure period is much shorter (e.g. *Caenorhabditis elegans* 72 h). The
38 short duration of exposure in such a test can be regarded as an advantage, as it is both
39 cost- and time-efficient as it reduces the total test time. However, if only a short-term
40 test is available (e.g. 72 h study), the result from this test cannot be used alone for the
41 derivation of the PNEC_{sediment}.

42 Water and sediment quality parameters

43 Quality parameters like oxygen content, pH, ammonium concentration, temperature and
44 water hardness should be measured in both pore water and overlying water, usually at

1 regular intervals during a test. The results should be reported in the study report.
2 Monitoring and reporting of these parameters is important for the evaluation of sediment
3 studies, as these water quality parameters may have an influence on the results of the
4 toxicity study. The standard guidelines also often specify which parameter should be
5 measured at what frequency and with which intervals, and how the results should be
6 reported.

7 Ideally, the oxygen content in the overlying water should not fall below 60% of
8 saturation at test temperature, as limited oxygen availability may result in adverse
9 effects on the test organisms. This should be measured as close to the sediment layer as
10 possible. However, a temporary shortfall below this value may not automatically mean
11 that a test is not valid. In this case it should be checked that the control response is
12 within the normal range. Many sediment dwelling species are capable of surviving at
13 oxygen concentrations as low as 2 mg/L.

14 The pH of the overlying water should be in a range between 6 and 9. However, it has to
15 be considered that a pH value above 8 may enhance the formation of toxic NH_3 from
16 NH_4^+ . Ammonium may be formed during the study e.g. from the food added to the test
17 system and certain species excrete ammonia directly. As NH_3 that is built up at pH
18 values above 8 is toxic to most aquatic organisms, it has to be verified that toxic effects
19 observed during the study are not caused by high ammonium concentrations (typically
20 $<1 \mu\text{g/l}$ is recommended in the guidelines).

21 Also sediment parameters should be measured, especially in case natural sediments are
22 used. Important parameters are for example the redox potential, the cation exchange
23 capacity (CEC), particle size distribution, total organic carbon content.

24 Test system

25 The overlying water systems in sediment tests may be static, semi-static or flow-
26 through. Semi-static or flow-through systems may contribute to good water quality in
27 terms of e.g. oxygen content or ammonium concentration thus limiting the influence of
28 such factors on the test results. However, as regular renewal of overlying water is
29 expected to affect chemical equilibrium resulting in losses of test substance from the
30 system, static systems are usually recommended. As a general rule OECD test guidelines
31 on sediment toxicity require analytical determinations of the test concentrations,
32 although in some guidelines some exceptions to this are allowed. In any case, sufficient
33 evidence of test concentration maintenance throughout the study should be given and
34 the registrant should justify his selection of overlying water renewal.

35 Test design

36 The following guidance should be applied when evaluating non-standard tests. Tests
37 performed according to standard guidelines should follow the guidance given in those
38 standard guidelines.

39 For a proper statistical evaluation of the test results, the number of test concentrations
40 and replicates per concentration are critical factors and are described in the guidelines. If
41 a solvent is used for the application of the test substance, a solvent control is necessary.
42 Estimations of the number of replicates should be based on the statistical power required
43 for the test and therefore the coefficient of variation of the parameter under review.

1 A limit test using only one test concentration and a control (and solvent control) may be
2 performed.

3 According to a number of OECD guidelines samples for chemical analysis of the test
4 substance should be taken at least from the control, lowest and highest concentrations,
5 at least at the end of the equilibration phase (start of exposure) and at the end of the
6 test. If samples are only taken at the beginning and end of the study, it is very difficult
7 to properly assess the exposure conditions. Therefore, it is important to sample at
8 appropriate frequency for the study length in the relevant matrices, e.g. water column
9 (to document the lack of exposure via this route), and bulk sediment and pore water to
10 document the potential exposure via these routes. This is depending on the guideline
11 and substance tested.

12 At least the sediment and the overlying water should be sampled for analysis. If possible
13 pore water concentrations can be analysed, as this will provide a more accurate
14 determination of the concentration to which the sediment dwelling organisms were
15 actually exposed. As conventional pore water measurements may lead to results that
16 cannot be interpreted, the use of Passive Sampling Devices (PSDs) to estimate the
17 "freely dissolved concentrations" may be a good alternative. PSDs work best for non-
18 polar organic chemicals while they are more difficult to be implemented for polar
19 compounds. However, PSDs have important limitations. Passive sampling for example,
20 cannot account for dietary uptake. Additionally, most of the PSDs experiments
21 performed in the laboratory not always reflect the actual situation in the field as
22 equilibrium conditions may never be obtained under realistic field conditions. For metals
23 the free ion and its potential to complex/compete/internal distribution with other organic
24 and inorganic ligands for the available biological binding sites is key to understand metal
25 bioavailability. Further studies are necessary to fully evaluate the potential of passive
26 sampling devices for metals. Equilibrium devices such as pore water "peepers" are
27 providing promising results with a view to be used for those benthic species that are
28 exposed to metals primarily through contact with the porewater. Diffuse gradient in thin
29 films (DGT) (i.e. non-equilibrium devices to measure metal flux) have been less
30 evaluated for assessing the bioavailability of metals in superficial sediments with regard
31 to predicting benthic organism bioaccumulation/toxicity.

32 Effect values should be preferably based on initial measured concentrations. However,
33 this approach should only be followed if analysis shows that the substance being tested
34 has been satisfactorily maintained within ± 20 % of the nominal or measured initial
35 concentration throughout the test.

36 If the deviation from the nominal or measured initial concentration is greater than ± 20
37 %, the reason of the variation should be investigated and the analysis of the results
38 should be based on the geometric mean of concentration during exposure. For some
39 substances complete recovery of irreversibly bound substance may not be technically
40 possible (e.g. aromatic amines). In this case, if clearly explained and justified, nominal
41 concentrations can be used provided that the substance is stable in the test system, i.e.
42 no biotic or abiotic degradation or removal from the test system is expected to occur.

1 **R.7.8.10.2 Field data, monitoring and mesocosm data on sediment** 2 **organisms**

3 For the purposes of prospective risk assessment when evaluating single substances in a
4 regulatory context, such as under REACH, field and monitoring data should preferably be
5 used in a *Weight of Evidence* approach. Experimental ecosystem studies and mesocosm
6 studies examine the effect of substances on aquatic (model) ecosystems. These studies
7 generally study both the effects of substances on pelagic organisms via the water phase
8 and on benthic organisms via the sediment. Some further information on ecosystem
9 studies can be found in Section R.7.8.3.1 under the subheading *In vivo – multiple*
10 *species (field data)*. Such ecosystem field data should normally only be used in a *Weight*
11 *of Evidence* approach together with other information.

12

13 **R.7.8.10.3 Rules according to Annexes to REACH and related** 14 **considerations for toxicity to sediment organisms**

15 **The rule in Column 2 of Annex X to REACH**

16 According to Annex X, section 9.5.1., column 2, to REACH long-term toxicity tests for
17 sediment organisms shall be proposed if the result of the chemical safety assessment
18 indicates the need to investigate further the effects of the substance and/or relevant
19 degradation products on sediment organisms. The need to conduct testing may be
20 triggered by the following cases, e.g.:

- 21 i. PEC/PNEC >1 based on Equilibrium Partitioning Method (EPM)
- 22 ii. PEC/PNEC >1 based on available sediment studies (short/long term)
- 23 iii. Information on degradation of the parent compound in the water column
24 showing formation of relevant degradation/transformation products (see
25 Section R.7.1) that will be distributed to the sediment
- 26 iv. Information on degradation of the parent compound in the sediment showing
27 formation of relevant degradation/transformation products exclusively in this
28 compartment (i.e. indications of anaerobic/aerobic degradation in the
29 sediment of the parent compound to relevant degradation/transformation
30 products)
- 31 v. Monitoring data showing occurrence of the substance or relevant
32 degradation/transformation products in sediment at ecologically relevant
33 concentrations
- 34 vi. Results from a PBT/vPvB assessment that further information is needed (see
35 Chapter R.11 of the *Guidance on IR&CSA*).

36 **General rules in Annexes VI and XI to REACH**

37 In Annex VI it is stated that, in some cases, the rules set out in Annexes VIII to X to
38 REACH may require certain tests to be undertaken earlier than or in addition to the
39 tonnage-triggered requirements.

1 For substances that strongly adsorb or bind to sediment, uptake from sediment or food
2 may become more important than uptake from water. Compounds that do not adsorb to
3 particles are covered by the pelagic tests. On the other hand, substances with a high
4 potential to adsorb onto sediment (e.g. $\log K_{ow} > 5$ or $\log K_{oc} > 3$) require sediment
5 assessment even at tonnages below 1000 t/y. Therefore, at least a screening
6 assessment using the equilibrium partitioning method (EPM) has to be performed for
7 such substances. If this screening assessment results in a PEC/PNEC value above 1, data
8 improvement is necessary independent on the tonnage of the substance either by
9 performing further long-term testing with sediment organisms or by refining the
10 exposure assessment. The same approach also applies to substances with intermittent
11 release to the aquatic environment that adsorb onto particles and that do not degrade
12 rapidly. Substances with tonnages below 1000 t/y and a not having a high potential for
13 adsorption (e.g. $\log K_{ow} < 5$ or $\log K_{oc} < 3$) do not normally need a sediment risk
14 assessment.

15 Furthermore, it has to be considered that substances that do not exhibit a toxic effect
16 when tested in water-only test systems because equilibrium was not reached during the
17 exposure phase may nevertheless exert significant toxic effects in sediment tests. This
18 may be especially true for poor water soluble substances with high adsorption potential.
19 The exposure duration in aquatic studies can in some cases be too short to reach steady
20 state conditions for such substances. Therefore, if no effects are observed in pelagic
21 tests, extrapolation from pelagic data to sediment data is not possible. In such cases,
22 performing a toxicity test on sediment organisms (whole sediment tests) at lower
23 tonnage levels (in accordance with Annex VI to REACH) may also be necessary.

24 **Bioavailability considerations for metals and inorganic metal compounds**

25 Metal bioavailability in freshwater and marine sediments is governed by different
26 ligands/processes (e.g. organic carbon, sulfides, iron and manganese oxy hydroxide and
27 redox potential) and the relative importance of these binding phases may differ
28 depending on the metals binding capacity and general behaviour.

29 It is recommended to make a clear differentiation between for example metal/inorganic
30 metal compounds that are susceptible for binding with sulfides and those metals that are
31 not sulfide binders, but where the use of partitioning to Fe-Mn (oxy)hydroxides,
32 speciation calculations (reduced forms under anoxic conditions) and organic carbon
33 normalisation may be more appropriate.

34 If it is relevant to take bioavailability of metals/inorganic metal compounds in sediments
35 into account in the CSR, such as SEM/AVS for metals, then it is recommended this
36 correction be performed for both the effect data and exposure data. Further information
37 about metals can be found in chapter 3.5.2 of Appendix R.7.13-2 on SEM-AVS
38 normalization²³.

39 **Bioavailability considerations for organic substances**

²³ SEM = Simultaneously Extracted Metals; AVS = Acid Volatile Sulfides.

1 Also for organic substances bioavailability corrections are – at least theoretically –
2 possible. The term bioavailability is defined in many different ways. According to the
3 proceedings of the topical scientific workshop (ECHA 2013) the following is proposed.
4 The total concentration of a chemical in a sediment can be divided into an irreversibly
5 bound pool (i.e. non-extractable, bound residues), reversibly bound, and freely dissolved
6 pool. The reversibly bound and the freely dissolved pool constitute the (bio-)accessible
7 pool. Accessibility is operationally defined. The accessible pool defines the fraction of the
8 total concentration that can undergo degradation, be mobilised or taken up by
9 organisms. However, it is a poor measure for the actual diffusion, partitioning or uptake
10 process, which is rather driven by the freely dissolved concentration or the chemical
11 activity. The chemical activity, as well as the freely dissolved concentration, can be
12 measured by passive sampling devices. Bioavailability is linked to (bio-)accessibility and
13 to the freely dissolved concentration (or the chemical activity). Bioavailability also
14 includes the uptake of a chemical by the organisms. Although recent developments in
15 the scientific community suggest using bioavailability concepts in risk assessment (e.g.
16 Ortega-Calvo *et al.*, 2015), there is relatively little experience applying these concepts in
17 a regulatory context in prospective risk assessment and the uncertainty when using
18 bioavailability corrections can be relatively large. Proper justifications are a prerequisite
19 when using bioavailability concepts.

20 **Degradation products**

21 For substances that degrade (biotically or abiotically) in the environment (but are not
22 readily biodegradable) it might be necessary to test the degradation products, instead of
23 or in addition to the parent substance. Generally, degradation products tend to be less
24 hydrophobic than the parent substance and therefore have a lower adsorption potential,
25 thus the relevance of the degradation products for the sediment compartment is
26 normally lower than that of the parent compound. The same triggers as for parent
27 compounds (e.g. $\log K_{oc} > 3$) can be applied to degradation products. If it is foreseeable
28 that degradation products accumulate in the sediment compartment, testing of
29 degradation products might be necessary. It should be noted that degradation of
30 substances that have a low bioavailability due to a very high $\log K_{ow}/\log K_{oc}$ might be
31 (much) more bioavailable than the parent compound.

32 **R.7.8.11 Species Sensitivity Distributions**

33 The Species Sensitivity Distribution (SSD) approach used for setting environmental
34 protection values (e.g. PNECs) for the pelagic compartment has only rarely been applied
35 to the sediment compartment. This is mainly due to the lack of toxicity data for a
36 sufficient number of distinct species that would fully reflect the complexity of the benthic
37 community. Furthermore, currently there is no scientific agreement on the number and
38 type of data to be used in a sediment SSD.

39 The SSD approach is protective for a community only if the species within the SSD are
40 representative of that community. With a limited suite of organisms for which data exists
41 for a given substance it is unlikely that those organisms are a good representation of the
42 community which is the protection target. In any case the usability of the SSD approach
43 for deriving sediment reference values is limited to data rich substances. For most
44 substances, there is not enough data to employ the SSD approach. If used, the
45 justification provided for an SSD would need to be evaluated on a case-by-case basis.

1 The EFSA PPR Opinion (2015) provides some scientific principles to be considered when
2 using the SSD approach when assessing sediment organisms exposed to active
3 substances of pesticides and transformations products from these substances. These
4 considerations can help to build a justification for SSD approaches under REACH.

5 **R.7.8.12 Remaining uncertainty**

6 Compared to the case for the pelagic compartment, there are fewer tests on different
7 organism groups or trophic levels available that examine the effects of industrial
8 substances on sediment organisms. Thus, experience with these tests and with the
9 assessment concept is still limited. For some metals more work is available, e.g. on SEM-
10 AVS, including field studies (see e.g. Nguyen *et al.*, 2011).

11 The majority of the available experimental studies with standardised test methods deal
12 with benthic invertebrates. Therefore, specific effects of substances on plants (that root
13 in the sediment) or microorganisms are seldom covered by the available experimental
14 studies. Recently, a standardised test with rooted aquatic plants has been developed and
15 adopted by OECD (OECD 239, adopted in 2014). Both rooted aquatic plants and
16 microorganisms also play an important role in benthic communities. Therefore, studies
17 according to OECD 239 should be considered if there are indications that these organism
18 groups are relevant for a given substance, especially in cases of higher tier sediment
19 assessments (e.g. when considering the use of an SSD). The OECD 239 rooted
20 macrophyte test can also be adapted for use with other species such as the reed *Glyceria*
21 *maxima*. This species may be most relevant when other information on the substance
22 (e.g. on its mode of action or data from terrestrial plant testing) indicates that the
23 substance shows particular phytotoxicity to monocotyledonous plants rather than to
24 dicotyledonous plants. Currently, standardised studies with microorganisms relevant for
25 the sediment compartment are not available.

26 In the absence of any sediment tests, the equilibrium partitioning method can be used
27 for neutral organic substances as a screening method to decide whether sediment tests
28 are necessary. This gives rise to a further uncertainty as the EPM may over- or
29 underestimate the toxicity of substances on sediment organisms. The additional factor of
30 10 on the PEC/PNEC ratio for highly adsorbing/ binding substances is meant to account
31 for the possibility of uptake via sediment ingestion and so take account of this
32 uncertainty. It should, however, be emphasised that this is only a screening approach.
33 The EPM approach was discussed in more detail in section R.7.8.10. When the
34 information requirement in REACH is applicable it is intended to cover long-term toxicity
35 to sediment organisms. Therefore, if new data are to be generated following the EPM
36 assessment, the testing strategy would normally already start with long-term tests but
37 without having information on the relative sensitivity of the test organisms to the
38 substance under consideration. Thus, there is the uncertainty that if only one long-term
39 test is being performed, the employed species may not be the most chronically sensitive.
40 This uncertainty is only partly covered by the assessment factor of 100 and the result
41 from this approach should therefore be treated with some caution.

42 Column 2 of the standard information requirement for sediment long-term testing in
43 REACH Annex X, sub-section 9.5.1. deals with the choice of the most appropriate test(s)
44 – thereby implying that more than one test could be carried out and may be needed to
45 fulfil the information requirement. Therefore, it is possible to carry out more than one

1 sediment test. This also allows for carrying out further testing, for example to lower the
2 assessment factor used for PNEC derivation. The guidance on the use of assessment
3 factors (provided in Chapter R.10) for the derivation of $PNEC_{\text{sediment}}$ foresees the use of
4 AF 1000 if only short-term sediment data are used. The Guidance specifies further that
5 $PNEC_{\text{sediment}}$ derived from short-term data may only be used as part of a screening
6 approach in combination of the EPM.

7 **R.7.8.13 Conclusions for toxicity to sediment organisms**

8 **R.7.8.13.1 Concluding on suitability for Classification and Labelling**

9 Whole sediment tests with benthic organism are not standard tests for classification and
10 labelling, as only exposure via the water phase is normally considered for deciding on
11 the classification. If available, tests with sediment organisms performed without
12 sediment can be useful for classification and labelling.

13 **R.7.8.13.2 Concluding on suitability for PBT/vPvB assessment**

14 Concerning the PBT assessment, there are no direct T criteria for sediment studies, but
15 long-term sediment toxicity tests may be appropriate to decide whether a substance
16 fulfils the T criterion. Full guidance on the suitability for PBT/vPvB assessment is given in
17 Chapter R.11 of the *Guidance on IR&CSA*.

18 **R.7.8.13.3 Concluding on suitability for use in Chemical Safety** 19 **Assessment**

20 The available data on sediment toxicity have to be evaluated for their adequacy for use
21 in effect assessment and PNEC derivation according to the criteria described in Section
22 R.7.8.10. Normally, little if any data will be available for sediment toxicity. In this case
23 the equilibrium partitioning method can be used as a first screening approach to decide
24 whether experimental data on toxicity to sediment organisms are necessary. For
25 substances with a $\log K_{ow} > 5$ an additional factor of 10 has to be applied on the
26 PEC/PNEC ratio, to take into account exposure of the benthic organisms via sediment
27 ingestion. The EPM can, for instance, normally not be used for substances that are
28 poorly water soluble and for which no effects are observed in acute and/or chronic
29 aquatic studies or for substances with a high adsorption or binding behaviour that is not
30 driven by lipophilicity (e.g. ionisable substances, surface active substances, substances
31 forming covalent bound with sediment particles like e.g. aromatic amines). For such
32 substances at least one sediment study has to be performed.

33 If sediment tests are available in which the test substance was applied to the test
34 system via spiking of the water phase, the effect values given in mg/L have to be
35 converted into a sediment concentration (mg/kg) using the substance-specific
36 partitioning coefficient or if available, measured sediment concentrations can be used.

37 If only one long-term sediment test is available, it should preferably be for an
38 endobenthic, sediment-ingesting species and the exposure time should be long enough
39 to enable adequate uptake of the sediment-associated substance by the test organism.
40 E.g. if only a 72 h test with the bacterivorous nematode *Caenorhabditis elegans* is

1 available (is considered as long-term test as growth inhibition and egg production are
2 measured), the result from this test cannot be used alone for the derivation of the
3 PNEC_{sediment}. However, such a test can be used as 2nd or 3rd test to lower the assessment
4 factor if (a) long-term test(s) with other benthic species like *Lumbriculus* or *Chironomus*
5 are already available. In general, results from short-term tests may only be used for
6 deriving a PNEC_{sediment screen} in combination with the EPM.

7 **R.7.8.14 Integrated Testing Strategy (ITS) for toxicity to sediment** 8 **organisms**

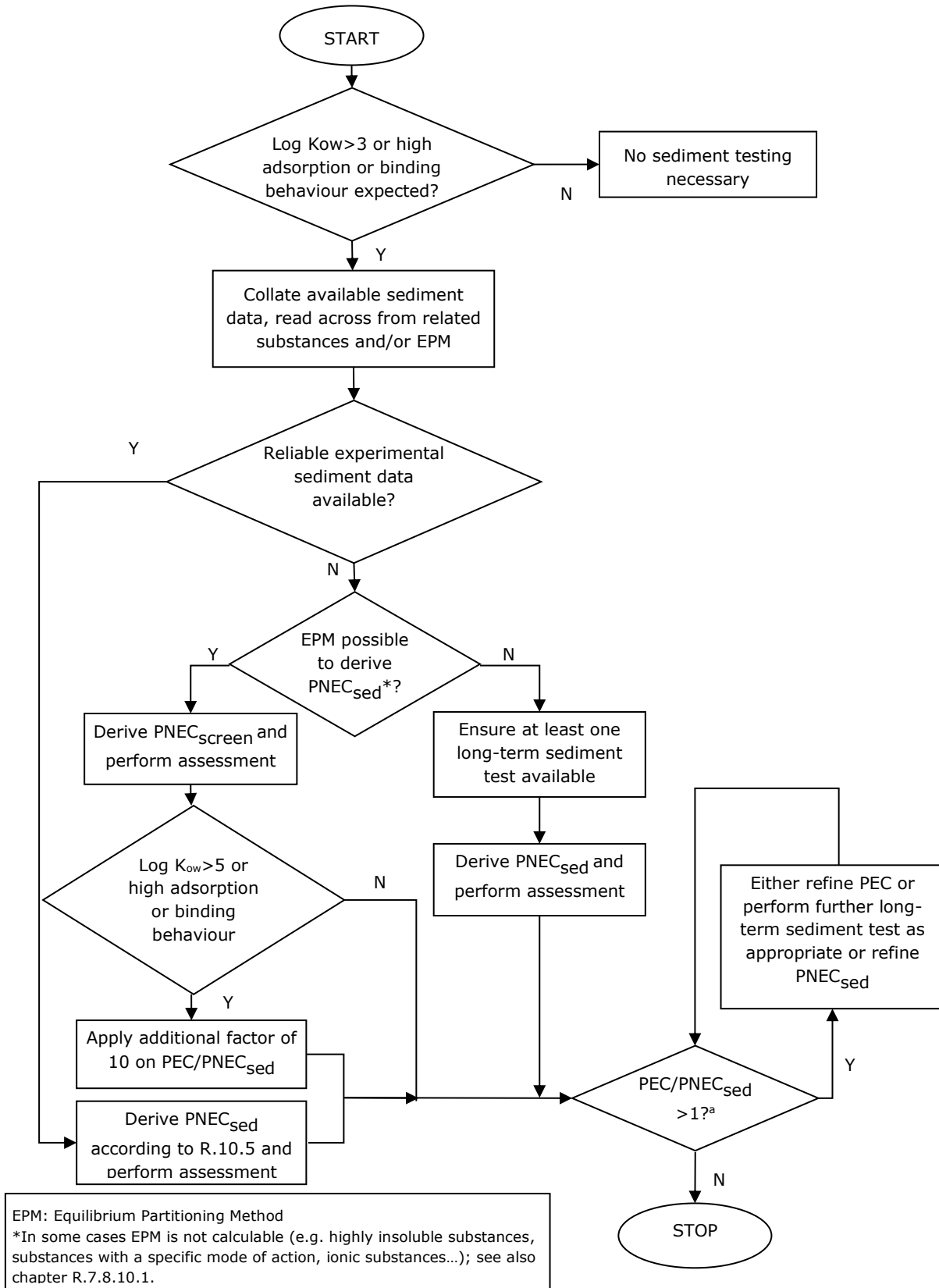
9 **R.7.8.14.1 Objective / General principles**

10 An integrated testing strategy for the sediment compartment is necessary primarily for
11 the use in chemical safety assessment, i.e. for the derivation of a PNEC_{sediment}.

12 The testing strategy visualised in [Figure R.7.8–8](#) described below has the objective to
13 give guidance on a stepwise approach to fulfil the regulatory demand.

14

1 **Figure R.7.8—8 Integrated Testing Strategy (ITS) for toxicity to sediment**
2 **organisms**



3

1 ^aNote: in case no further risk refinements are possible, then apply appropriate risk reduction
2 measures (e.g. minimizing exposure sufficiently so that $RCR < 1$).

3 **R.7.8.14.2 Testing strategy for toxicity to sediment organisms**

4 The main property of a substance that triggers the assessment for the sediment
5 compartment is the potential to adsorb or bind onto sediment. Further triggers for a
6 sediment assessment are also given in R.7.8.7. A log K_{ow} of 3 should be used as trigger
7 value for a sediment assessment. For substances exceeding this trigger value, the
8 availability of existing sediment toxicity data should be checked. In the absence of any
9 (acceptable) sediment tests, the equilibrium partitioning method can be applied as a first
10 screen.

11 For substances with a log K_{ow} between 3 and 5 this screening assessment results in the
12 same risk characterisation ratio for sediment as for the pelagic compartment, as both
13 PEC_{sediment} and $PNEC_{\text{sediment screening}}$ are modelled from the corresponding pelagic data using
14 the same partitioning coefficient.

15 Special attention should be given to substances with a log $K_{ow} > 5$. The same attention
16 should be given to substances with a correspondingly high adsorption or binding
17 behaviour when adsorption is not triggered by the lipophilicity but by other mechanisms
18 (e.g. ionising substances, surface active substances, substances that bind chemically
19 with sediment components, substances where K_d predicts high binding potential). To
20 take into account uptake of sediment-bound substance by benthic species, this
21 $PEC/PNEC$ ratio derived according to the rules outlined in R.10.5 is increased by a factor
22 of 10 for all such substances, unless scientific evidence can be provided that the extra
23 factor is not applicable for that specific group of substances. In the latter case the non-
24 application of this additional factor has to be substantiated in detail. If the $PEC/PNEC$
25 ratio is below one, no risk for the sediment compartment is indicated for the substance
26 under consideration and further tests are not needed.

27 If the $PEC/PNEC$ ratio is above one, there is a need to perform long-term sediment tests
28 with benthic species.

29 For substances that are poorly water soluble and for which no effects are observed in
30 aquatic studies, the application of the equilibrium partitioning method is not possible. For
31 such substances at least one sediment test has to be performed.

32 If there is already one or more (acceptable) acute or long-term sediment test(s)
33 available, a $PNEC_{\text{sediment}}$ is derived from these tests using an appropriate assessment
34 factor (as described in the Guidance on IR&CSA, chapter R.10). In general, results from
35 short-term tests may only be used for deriving a $PNEC_{\text{sediment,screen}}$ in combination with
36 the EPM. If long-term sediment tests with more than one benthic species are available, it
37 has to be considered whether these organisms represent different habitats and feeding
38 strategies and are thus exposed via different exposure pathways. Only in this case, a
39 reduction of the assessment factor is possible. If the $PEC/PNEC$ ratio is below one, no
40 risk for the sediment compartment is indicated and further tests are not needed. If the
41 $PEC/PNEC$ ratio is above one, there is a need to perform (further) long-term sediment
42 tests with benthic species.

1 If there are no adequate long-term sediment tests available, a test with preferably either
2 *Lumbriculus variegatus* or *Chironomus* sp.. using spiked sediment should be performed,
3 unless there are specific reasons to select another guideline/other species as explained
4 above. Proper justification of species selection needs to be given in the dossier. A
5 PNEC_{sediment} has to be derived from the (lowest available) NOEC/EC₁₀ using an
6 appropriate assessment factor.

7 It should be noted that both PEC_{sediment} and PNEC_{sediment} should be normalised to the
8 same OM content²⁴.

9 If the PEC/PNEC ratio is below 1, no risk for the sediment compartment is indicated and
10 there is no need to perform further tests. If the PEC/PNEC ratio is still above 1, the
11 uncertainty can be reduced either by refinement of PEC or by performing another long-
12 term sediment test with species representing different habitats and feeding strategies.

13 Toxicity data selection and compilation should not solely represent an array of taxonomic
14 groups but should also aim for a balanced and realistic representation of functional
15 attributes, including – but not limited to – functional traits. More precisely, regarding
16 invertebrates different exposure conditions and feeding strategies should be represented
17 by a variety of life strategies. Table R.7.8-5 can be used as a starting point to determine
18 differences in taxonomic group, habitat and feeding strategy.

19 The following benthic species (from different taxonomic groups) are usually
20 recommended for testing:

- 21 • *Lumbriculus variegatus*, in long-term test using spiked sediment
- 22 • *Chironomus* sp., in long-term test using spiked sediment
- 23 • a further benthic species in long-term tests using spiked sediment. Selection
24 of 3rd species should supplement the first 2 species in terms of habitat,
25 feeding strategy, life-stage. This could be e.g. *Hyalella azteca*.

26 Some long-term guideline studies have a longer duration than others. Studies with
27 longer duration are usually preferred for substances that have an equilibration time
28 (time to reach steady state in the body) that is anticipated to be very long. Information
29 on equilibration times can come from different sources, such as the logKow and/or
30 logKoc value, (aquatic) bioconcentration studies, ecotoxicity data. For example, a
31 *Hyalella azteca* 28-d study (e.g. ISO 16303:2013) might not be a good option for a
32 substance with a very long equilibration time, in which case a 42-d study with *H. azteca*
33 (e.g. EPA 600/R-99/064, 100.4) is a better choice.

34 New studies should normally be performed with non-vertebrate species. They should
35 follow internationally accepted guidelines and should be performed under Good
36 Laboratory Practices (GLP). Any testing with for instance amphibians (ASTM guideline
37 E2591-07) should be very well justified by registrants.

²⁴ See footnote 21.

1 However, if there is in addition to the risk for the sediment compartment also a risk for
 2 the pelagic compartment and the PEC/PNEC for the pelagic compartment is higher than
 3 the PEC/PNEC for the sediment compartment, any risk reduction measures applied to
 4 reduce the exposure of the aquatic compartment will also influence/cover the sediment
 5 compartment. In such a case the need to perform further sediment tests may be
 6 postponed to await the outcome of the emission reducing measures.

7 If the $PNEC_{\text{sediment}}$ is derived from the lowest NOEC/EC10 from three long-term sediment
 8 tests covering different exposure pathways and the PEC/PNEC ratio for the sediment
 9 compartment is still above one, further action must be taken to reduce the PEC.

10 In order to reduce testing, group approaches and read-across methods should be
 11 considered to partially or completely waive sediment studies. There should be sufficient
 12 studies available that further toxicity values can be reasonably predicted.

13 Examples: if for a certain chemical category clear evidence exists that the additional
 14 factor of 10 significantly overestimates the toxicity to sediment organisms, the EPM can
 15 be used without this additional factor. This must be substantiated in detail. In other
 16 cases it may be sufficient to perform only one (long-term) sediment test, if for another
 17 substance from which read-across is possible, it can be deduced which is the most
 18 sensitive test species / test system in order to attain the lowest assessment factor.

19 A general guidance on how to extrapolate via read-across or chemical categories is given
 20 in Section R.6.2.

21 For the marine compartment, the same testing strategy is followed. Most of the existing
 22 marine whole sediment tests measure acute toxicity; only a few measure long-term,
 23 sub-lethal, endpoints. A higher assessment factor is generally applied to the marine
 24 environment than to the freshwater environment.

25 Comprehensive guidance on establishing the size of the assessment factors is given in
 26 Section R.10.5 in Chapter R.7c of the [Guidance on IR&CSA](#).

27 **Table R.7.8—5 Characterisation of the most common benthic test species from**
 28 **OECD, ISO, USEPA, ASTM and OSPAR guidelines**

Species	Taxonomic group	Habitat	Feeding mode	Relevant guideline(s)
<i>Myriophyllum spicatum</i>	rooted dicotyledonous macrophyte plant	Freshwater, rooted	Rooted plant	OECD 239
<i>Chironomus</i> sp.	insect	freshwater , endobenthic	Suspension and deposit feeder	OECD 218/219/233/235 ASTM E1706-05 US-EPA 100.2/100.5
<i>Lumbriculus</i>	oligochaete	freshwater,	Sediment ingestor	OECD 225

<i>variegatus</i>		endobenthic		
<i>Hyalella azteca</i>	amphipod	Freshwater, Epibenthic	Detritivore, some subsurface deposit feeding	ASTM E1706-05 US-EPA 100.1/100.4 ISO 16303:2013
<i>Hexagenia</i> sp.	insect	freshwater, endobenthic	Surface particle collector	ASTM E1706-05
<i>Tubifex tubifex</i>	oligochaete	freshwater, endobenthic	Sediment ingestor	ASTM E1706-05
<i>Diporeia</i> spec.	amphipod	freshwater, endobenthic	Deposit feeder	ASTM E1706-05
<i>Caenorhabditis elegans</i>	nematode	freshwater, endobenthic	bacterial ingestor	ISO 10872:2010
<i>Leptocheirus plumulosus</i>	amphipod	estuarine, endobenthic	Suspension and deposit feeder	US-EPA 600/R- 01/020 ASTM E1367- 03e1
<i>Ampelisca abdita</i>	amphipod	marine, endobenthic	Suspension and deposit feeder	ASTM E1367- 03e1
<i>Eohaustorius esturarius</i>	amphipod	estuarine, endobenthic	Deposit feeder	ASTM E1367- 03e1
<i>Rhepoxynius abronius</i>	amphipod	marine endobenthic	Meiofaunal predator, deposit feeder	ASTM E1367- 03e1
<i>Neanthes arenaceodentata</i> <i>Neanthes virens</i>	polychaete	marine, endobenthic	Omnivorous deposit feeder	ASTM E1611-00
<i>Corophium volutator</i>	amphipod	marine, endobenthic	Suspension and deposit feeder	OSPAR (2005)
<i>Gammarus</i> sp.	amphipod	Freshwater estuarine	Grazer; detritivore	ISO 16712:2005
<i>Heterocypris</i>	Ostracod	Freshwater,	Omnivorous	ISO 14371:2012

<i>incongruens</i>		epibenthic		
<i>Rana pipiens</i>	amphibian	Freshwater, Epibenthic/pelagic	Suspension feeder	ASTM E2591-07
<i>Rana clamitans</i>	amphibian	Freshwater, Epibenthic/pelagic	Benthic feeder	US-EPA 100
<i>Rana sylvatica</i>	amphibian	Freshwater, Epibenthic/pelagic	Deposit feeder	US-EPA 100
<i>Bufo americanus</i>	amphibian	Freshwater, Epibenthic/pelagic	Suspension and detritus feeder	US-EPA 100

1

2 **R.7.8.15 References on sediment organisms toxicity**

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1

2 **R.7.8.16 Introduction to stp microorganisms' toxicity**

3 **R.7.8.16.1 Definition of toxicity to STP microorganisms**

4 Adequate functioning of a STP (Sewage Treatment Plant) is essential to protect the
5 downstream aquatic environment and to minimize operational costs. The endpoint of STP
6 toxicity, as part of environmental risk assessment, was also included in the EU TGD
7 (CEC, 2003). The aim of the assessment is the protection of the biodegradation and
8 nutrient removal functions, and process performance in general, of municipal and
9 industrial STPs.

10 Since chemicals may cause adverse effects on microbial activity in STPs, it is necessary
11 to derive a PNEC_{micro-organisms} (here called PNEC_{stp}). The PNEC_{stp} will be used as
12 toxicity measure for the calculation of the risk quotient ($PEC_{stp}/PNEC_{stp}$) for microbial
13 activity in STPs.

14 **R.7.8.16.2 Objective of the guidance on toxicity to STP** 15 **microorganisms**

16 PNEC_{stp} is determined by means of microbial toxicity tests. Currently used test systems
17 for measuring the effect of chemicals on microbial activity have different endpoints and
18 different levels of sensitivity. A number of internationally accepted test systems have
19 been proposed in the past and their recommended use under REACH will be discussed
20 further in this document.

21 For the engineered environment of a STP, *functional* endpoints (i.e. good and stable
22 functioning) take precedence over *structural* endpoints (i.e. microbial population
23 composition).

24 If the substance under consideration is released to both industrial- (i.e. production site)
25 and municipal STPs, the toxicity assessment should be conducted separately for both
26 types of STPs, with parameters relevant to the respective systems (see higher)²⁵.

27 **R.7.8.17 Information requirements for toxicity to STP microorganisms**

28 The assessment of PNEC_{stp} is a requirement as of volumes of 10 tonne/year and above
29 (REACH Annex VIII test requirement 9.1.4.). The type of test specified under 9.1.4 of
30 REACH is an activated sludge respiration test (e.g OECD 209). Respiration inhibition is
31 only one of many possible test approaches for measuring effects on microbes, but it is
32 the most widely accepted indicator of the combined activity of sludge microorganisms.
33 As such, the respiration inhibition test is preferred for the generation of new microbial

²⁵ In practice, many STPs treating domestic sewage also receive a fraction of industrial effluents, and a clear separation can not always be made. Municipal/domestic STPs are defined here as those plants of which the load predominantly consists of domestic waste waters.

1 toxicity data. This test can be substituted by a nitrification inhibition test if there are
2 indications that the substance may be toxic to nitrifying bacteria.

3 Good quality data obtained with other types of microbial inhibition test methods,
4 degradation- or sewage treatment simulation tests, can be also used to meet the REACH
5 requirements, in particular if these studies were already existing (ITS scheme see
6 Section R.7.8.21).

7 Column 2 of Annex VIII in REACH indicates that STP toxicity testing is not needed in the
8 following cases:

- 9 • no emissions to STP (PEC = 0)
- 10 • the compound is readily biodegradable and PEC below test concentration
11 applied
- 12 • there are mitigating factors, such as a very low solubility that would limit the
13 exposure.

14 **R.7.8.18 Information on toxicity to STP microorganisms and its sources**

15 **R.7.8.18.1 Laboratory data on toxicity to STP microorganisms and its** 16 **sources**

17 **Non-testing data on toxicity to STP microorganisms**

18 The practical use of QSARs for predicting STP toxicity is still limited. Although there are
19 some QSARs for toxicity to microorganisms published (e.g. Blum & Speece 1990; Ren &
20 Frymier 2002b; Redman *et al.*, 2005; Schulz *et al.*, 2005), this is not a very well
21 developed science domain today. The existing microbial toxicity QSARs are mainly
22 developed for baseline toxicity towards individual species of microorganisms, such as the
23 ciliate *Tetrahymena pyriformis* (see work of T. Schulz and colleagues), and the
24 bioluminescent *Vibrio fischeri*, formerly known as *Photobacterium phosphoreum* in the
25 Microtox® test. On top of models for non-polar narcotics, some additional models
26 specific to a particular class of chemicals are available. Since conceptual consistency is to
27 be achieved between the experimental and QSAR approach for protecting
28 microorganisms in STPs, the use of QSAR models developed for ciliates and individual
29 species of bacteria not indigenous to STPs is to be excluded, however.

30 Preliminary QSAR models for baseline toxicity to *P. putida* and for activated sludge
31 respiration inhibition are reported in Redman *et al.* (2005). The reported models are
32 based on a limited number of observations and have not been published yet in the peer
33 reviewed literature. More validation work is needed here.

34 No QSAR models exist that accurately predict and protect nitrification inhibition. This is
35 a significant outage, since nitrification can be the most sensitive endpoint – as illustrated
36 in the experience of the EU existing chemicals programme.

37 The ProperEst website developed by the Fraunhofer Institute, to be publicly released,
38 intends to provide a comprehensive compilation and documentation of microbial QSAR
39 models

40 (http://www.ime.fraunhofer.de/en/business_areas_AE/ChemicalSafety/Ersatz_Tierversu)

1 [che1.html](#)). In a *Weight of Evidence* context, consideration can be given to the use of
2 read-across instead of testing, in particular for series of close chemical homologues for
3 which there exist experimental data on some of the individual homologues.

4

5 **Testing data on toxicity to STP microorganisms**

6 Information from subcellular microbial systems:

7 A number of microbial inhibition test approaches exist which are based on subcellular
8 systems, e.g. the Triphenyl Tetrazoliumchloride (TTC) Dehydrogenase assay (Ryssov-
9 Nielsen 1975), β -galactosidase activity (Katayama-Hirayama 1986). Such in-vitro
10 systems based on a single reaction have not been sufficiently validated in the context of
11 STP risk assessment, and their use is therefore not accepted.

12 Information from microbial inhibition tests:

13 PNEC_{stp} is routinely determined by means of microbial toxicity tests. This section
14 provides an overview of the most commonly used microbial toxicity tests and their
15 underlying concept. The toxicological endpoints are: respiration (i.e. O₂ uptake)
16 inhibition, nitrification (i.e. ammonia conversion) inhibition, growth inhibition and
17 bioluminescence. The list in this section is not aimed to be exhaustive, as many
18 methodological variations and a suite of different test organisms have been proposed in
19 the literature.

20 Literature information on the toxicity for microorganisms has to be assessed for its
21 relevance with regard to the endpoint considered, i.e. microbial processes in a STP. In
22 general, short-term measurements in the order of hours are preferred, in accordance
23 with the hydraulic retention time in a STP (e.g. 10 h). Data on microbial toxicity from
24 standard- and non-standard test methods is available for some compounds in the open
25 literature (e.g. Blum & Speece 1991), in handbooks (e.g. Verschueren 2001), and in
26 various databases (e.g. TETRATOX (www.vet.utk.edu), IUCLID).

27 Data from ciliate growth inhibition tests, preferably with the species *Tetrahymena* (OECD
28 1998; Pauli & Poka 2005), are also relevant for the risk assessment for STPs²⁶. Ciliated
29 protozoa, constituting the most important class of protozoa in STPs are, except for
30 certain industrial plants, important for their functioning (NB: mainly for floc formation
31 and settling properties, rather than for degradation processes). Toxicity data on ciliates
32 are considered to be supplementary to the data on activated sludge or specific bacterial
33 strains, i.e. no correlation exists between activated sludge and ciliate test results,
34 neither are ciliates consistently more sensitive.

35 Tests using other characteristics (e.g. ciliary motion, cell movement, etc.) should not
36 serve as a basis for the PNEC-derivation. For *Tetrahymena sp.* growth inhibition there
37 exists a very large single endpoint database TETRATOX (www.vet.utk.edu). More than

²⁶ Following an international pilot ring test, a growth test with the ciliate *Tetrahymena pyriformis* was recommended for ecotoxicological risk assessment by the German Federal Environmental Agency. A full validation study to establish an internationally recognized Test Guideline has been conducted in the years 2000-2003. The resulting draft for an OECD protozoan test Guideline is currently under review.

1 2400 industrial organic compounds - of which more than 1,600 are published - have
2 been tested at the University of Tennessee.

3

4 Information from biodegradation- and simulation tests

5 Absence of microbial toxicity can often be inferred from biodegradation studies in the
6 laboratory. The information content of ready biodegradability tests (available as of 1 t/y)
7 can under certain conditions also be used to derive a NOEC. This can be used to avoid
8 new testing. The assumption that the substance under investigation is not inhibitory to
9 the micro-organisms when dosed in the test system is implicit in ready biodegradability
10 testing (i.e., EC C.4A-F, OECD 301A-F (OECD, 1992) and OECD 310 (2006)). If a
11 compound degrades well in a ready biodegradability test, or does not inhibit the
12 degradation of a positive control at a certain concentration, this concentration can be
13 used as a NOEC value.

14 Any Ready Biodegradability Test relying on continuous monitoring, e.g. the MITI I test
15 (EC C.4F; OECD 301C) or the Manometric Respirometry test (EC C.4D; OECD 301F) is
16 considered more reliable for observing the effects of a chemical on the inoculum. A
17 partial or transient toxic effect often results in a delayed mineralisation of the test
18 substance and/or the positive control.

19 Data from biodegradation/removal studies using either inherent degradability tests
20 (OECD 302A-C), or the laboratory/pilot scale Activated Sludge Simulation test
21 (Continuous Activated Sludge (CAS) – OECD 303A and ISO-11733) may also be
22 acceptable to derive a $PNEC_{stp}$ (OECD 1981; OECD 2001). The latter are laboratory scale
23 models for simulation of activated sludge, representing realistic approximation to actual
24 conditions in full scale STPs. The $PEC_{effluent}$ (or in the absence of that value the $PEC_{influent}$)
25 from well-conducted simulation studies using domestic activated sludge would
26 correspond to the concentration of the chemical substance that does not perturb the
27 proper functioning of the CAS unit with regard to performance parameters such as test
28 substance elimination, BOD/COD removal, nitrification, etc., when compared to a parallel
29 non-dosed control.

30 **R.7.8.18.2 Field data on toxicity to STP microorganisms and its** 31 **sources**

32 Absence of toxicity of a chemical can in a number of cases also be inferred from
33 observations made at full scale plants. In particular for industrial STPs, the operators
34 may have plant performance data in combination with chemical emission/exposure
35 information, which can potentially be used to justify a $PNEC_{stp}$.

36 In addition, many full scale STPs are monitored on-line by commercial respirometer
37 apparatus. A variety of commercial respirometers for activated sludge are available on
38 the market (e.g. Strathtox, RODTOX, Oxitop, etc.). These systems monitor the Oxygen
39 Uptake Rate (OUR) of the plant and can be used to derive a NOEC for respiration
40 inhibition similar to laboratory tests and equipment. Some apparatus can also measure
41 nitrification inhibition.

1 **R.7.8.19 Evaluation of available information on toxicity to STP micro-** 2 **organisms**

3 **R.7.8.19.1 Laboratory data on toxicity on STP microorganisms**

4 **Non-testing data on toxicity on STP microorganisms**

5 Use of non-testing data (QSARs) for STP Toxicity is not generally recommended given
6 the limited availability of validated models relevant to STP organisms, and because an
7 activated sludge respiration inhibition test is not particularly costly, complex or time-
8 consuming to perform. Actual experimental data will typically overwrite calculated data,
9 but QSARs may be useful to provide a preliminary estimate of toxicity for difficult-to-test
10 substances.

11 In cases where relevant and well validated (Q)SARs for microbial toxicity would be
12 developed in the future, this information could be fitted into the ITS to estimate
13 PNEC_{stp}. Sound scientific judgement is needed to evaluate whether this information can
14 replace the need for laboratory testing.

15 **Testing data on toxicity on STP microorganisms**

16 Information derived from sub-cellular microbial test systems (e.g. enzyme activity) as
17 indicator of STP toxicity cannot be used.

18 The core microbial functions of a STP that need to be protected include carbon
19 (BOD/COD) removal and nitrification. For some installations it is also important to
20 protect other processes such as denitrification and biological P removal. Since there are
21 no standardized test protocols for the latter endpoints, an assessment factor approach is
22 routinely used to provide an adequate level of protection. There exists an anaerobic
23 toxicity test ISO 13641 (2003) based on inhibition of biogas production, but its use to
24 estimate the risk to STPs with biological nutrient removal would require further study.

25 Toxicity tests with bacteria

26 In general, preference is given to tests with a mixed inoculum that assess the
27 functioning of the entire microbial community in an STP, rather than tests based on
28 single species or even microbial sub-systems. Respirometry is generally considered as an
29 approach that will integrate the functioning of all organisms in an STP. The respiration
30 inhibition test is generally positioned as a screening-level test (Painter 1986).

31 Nitrification inhibition tests, which assess the functioning of the sub-population of
32 nitrifying organisms, are also amongst the preferred tests.

33 Not all microbial test systems are equally sensitive, however. Umweltbundesamt (UBA
34 1993) and Reynolds *et al.* (1987) suggest the following order of increasing sensitivities
35 among particular test systems: respiration inhibition test < inhibition control in base-set
36 tests < growth inhibition test with *P. putida* < inhibition of nitrification. Ren & Frymier
37 (2003b) showed that nitrifying bacteria have a different, and generally higher sensitivity
38 to toxicants, than other test systems. The response of the respiration-, *Tetrahymena*-
39 and Shk1-assay clustered quite closely together in terms of sensitivity.

1 If activated sludge from an industrial sewage treatment plant is used as inoculum for a
2 respiration or nitrification test, it is assumed that the microorganisms are adapted to the
3 substance. Therefore, the test results cannot be extrapolated to municipal sewage
4 treatment plants, since in municipal plants the bacteria may not be as adapted to the
5 substance as the industrial sludge.

6 Often inhibition test data on individual bacterial species may be available. Results of the
7 cell multiplication inhibition test with *P. putida* (Bringmann and Kühn 1980) should be
8 used for calculation of the PNEC_{micro-organisms} only in cases where no other test
9 results are available. A similar recommendation is made for the Shk1 assay, which is
10 based on a constructed bioluminescent *Pseudomonas sp.* originally isolated from
11 activated sludge (Kelly *et al.*, 1999; Ren & Frymier 2002a; Ren & Frymier 2003a).

12 Other single species tests with e.g. *Vibrio fischeri* (used in the MICROTOX® test),
13 *Pseudomonas fluorescens* or *Escherichia coli* should be considered of low relevance for
14 STPs. The tests with *P. fluorescens* and *E. coli* (Bringmann and Kühn 1960) cannot be
15 used for determination of the PNEC_{stp} as they use glucose as a substrate (nor is *E. coli* a
16 bacterium that will tend to multiply in an activated sludge environment). Likewise, *Vibrio*
17 *fischeri* requires a high salinity environment. The information from such single-species
18 screening tests may eventually be considered together with other existing data in a
19 *Weight of Evidence* approach.

20 Biodegradation and sewage treatment simulation tests:

21 The information content of ready or inherent biodegradability tests can also be used to
22 derive a NOEC under the following conditions:

- 23 • when in a ready or inherent biodegradability test the compound is found to be
24 respectively readily or inherently biodegradable,
- 25 • when in a ready or inherent biodegradability test a toxicity control has been
26 included that shows good degradation of a positive control substance (e.g.
27 glucose, sodium acetate) in the presence of the test substance.

28 Subject to expert judgement, data from biodegradation/removal studies using the
29 laboratory/pilot scale Activated Sludge Simulation, Continuous Activated Sludge (CAS -
30 OECD303A and ISO-11733) may also be acceptable to derive a PNEC_{stp}. In such tests it
31 will be needed to monitor parameters such as BOD/COD removal, N-removal, sludge
32 settling, etc., as compared to a parallel non-dosed control. Measuring chemical removal
33 in such tests is optional, but can provide valuable additional information.

34 It should be noted that laboratory or field results obtained with an industrial sludge
35 should be seen as plant-specific and cannot be extrapolated. Results for a municipal
36 sludge can be extrapolated to other municipal installations provided that the emission
37 pattern of the chemical is similar.

38 Protozoa toxicity tests

39 Ciliate-based test data can be used for deriving a PNEC_{stp} in case these are the sole data
40 available, or in multiple-data situations where the ciliates have the lowest NOEC.

41

1 Substances difficult to test for STP toxicity:

2 Volatile and semi-volatile substances should not be tested in an open test system, e.g.
3 the activated sludge respiration inhibition or nitrification inhibition test, since the
4 chemical may be stripped from the system by the aeration. In such case, the
5 recommended approach is to use a closed system, such as in OECD301F (Manometric
6 Respiratory test) or OECD 310 (CO₂ headspace test).

7 **R.7.8.19.2 Field data on toxicity on STP microorganisms**

8 Also subject to expert judgement, data from full scale domestic or industrial STP that
9 have received a certain chemical for prolonged periods can provide information useful to
10 derive a PNEC_{stp}. This information can be used to avoid the need for additional laboratory
11 testing. It would require that the concentrations of the chemical in the effluent or
12 influent are well known, and the stable and efficient operation of the plant in the
13 presence of the chemical has been confirmed (as e.g. indicated by prolonged BOD/COD-
14 and N-removal performance, sludge settling, etc.).

15 **R.7.8.19.3 Exposure considerations for toxicity on STP microorganisms**

16 The paragraph below provides some guidance on exposure considerations for deriving a
17 PNEC_{stp}:

18 Microbial toxicity testing above the solubility limit of a chemical is to be avoided, similar
19 to toxicity test with higher organisms. It is also unrealistic because insoluble chemicals
20 will be removed in the primary settling tank or fat trap of full scale installations, and thus
21 will not reach the activated sludge.

22 However, data from existing tests where the experimentally derived NOEC is higher than
23 the aqueous solubility can still be used as valid information to derive a PNEC_{stp}. This can
24 be justified because it is a conservative estimate unlikely to occur in practice, and
25 because undissolved test substance is found to be less confounding in microbial tests
26 than in tests with higher organisms.

27 In the case of the respirometric method OECD 209, the test duration is very short; 30 or
28 180 minutes exposure to the chemical, followed by the measurement of oxygen uptake
29 rate over 5-10 minutes. For chemicals with a low solubility, a contact time of 180
30 minutes (3 h) is to be used to ensure sufficient exposure. Some authors have proposed
31 even longer exposure in respiration tests to lower the variability of the results (e.g.
32 Gendig *et al.*, 2003).

33 Keeping exposure constant during microbial toxicity tests: In batch microbial tests, the
34 exposure is often not constant due to degradation, adsorption and other loss processes.
35 It is generally assumed that the microorganisms have been exposed at the maximum
36 level at the onset of the test and that the toxic effect, if any, has taken place at that
37 point. Observation of degradation is further evidence of the detoxification ability of the
38 microbes. For very unstable or sorptive chemicals, the need for a simulation test with
39 continuous dosing such as the OECD 303A test may be considered if a batch test is
40 deemed unreliable. This is not recommended as a routine procedure, however. The
41 reader is also referred to OECD (2000) on testing of difficult substances.

1 **R.7.8.19.4 Remaining uncertainty for toxicity on STP microorganisms**

2 The choice of assessment factors to derive PNEC from microbial tests in the past has
3 been rather empirical/arbitrary, and is not based on the same amount of comparative
4 research as e.g. for the acute/chronic ratio for higher organisms (Table R.10-6 and
5 Section R.10.4). One of the reasons that tests with single species of microorganisms
6 have a lower assessment factor as compared to the recommended activated sludge
7 respiration test, is that the latter is short term screening-type test, while former
8 measure a chronic-type endpoint (growth).

9 Another aspect which requires consideration is that microbial toxicity results (e.g.
10 respiration inhibition) tend to be proportional to the density of the culture, i.e. the test
11 substance/biomass ratio. In other words, *dose* rather than *concentration* will determine
12 the toxicity. This aspect is often overlooked in STP toxicity testing but can explain part of
13 the differences in sensitivity sometimes noted between microbial inhibition tests
14 (Elnabarawy *et al.*, 1988).

15 The OECD 209 method operates at 1.6 g SS/l. The SimpleTreat Model version 3
16 (implemented in EUSES) uses 4 g SS/l in the aeration vessel as a default model value.
17 When comparing microbial inhibition data from different test systems and origins it is
18 good practice to verify if biomass levels are comparable. As a rule of thumb, deviations
19 in biomass larger than a factor 10 are not suitable for direct cross-comparison. Inhibition
20 tests executed at typical SS levels (1–4 g/l) should be considered as more reliable (nb:
21 this guidance does not apply to nitrifying organisms for which levels in sludge are always
22 much lower).

23 **R.7.8.20 Conclusions for toxicity to sewage treatment plant** 24 **microorganisms**

25 Microbial toxicity tests on STP organisms are not required for Classification & Labelling,
26 nor do they qualify for PBT assessment. Therefore the test data will only find application
27 in Chemical Safety Assessment.

28 Mainly experimentally-derived microbial inhibition data will be used to derive a $PNEC_{stp}$ in
29 the absence of well-established QSARs. As a general rule, data generated according to
30 international standard guidelines and to GLP are to be preferred over other types of
31 data.

32 Equally, however, it is important to appreciate that conclusions are to be based on the
33 best available data, and that GLP studies can sometimes be flawed in other aspects.
34 Thus, also available non-standard tests can be used, provided the data are considered
35 scientifically valid.

36 In case of multiple microbial inhibition data, the $PNEC_{stp}$ is usually derived from results
37 obtained for the most sensitive test system available, regardless of whether this is a test
38 with activated sludge, relevant single bacterial species or ciliated protozoa. If there is
39 considerable uncertainty around individual datapoints or questionable outliers, a *Weight*
40 *of Evidence* approach can be followed.

1 **R.7.8.21 Integrated Testing Strategy (ITS) for toxicity to STP micro-** 2 **organisms**

3 **R.7.8.21.1 Objective / General principles**

4 The main objective of an ITS for STP Toxicity is to ensure that all available relevant
5 exposure and effects information can be used before any new testing is initiated. This
6 way, time and financial investment can be minimized, but without compromising on the
7 quality of the assessment. On the other hand, the ITS should also allow to refine
8 unfavourable screening data by means of higher tier testing. In the case of STP toxicity,
9 the most realistic and highest tier test is a sewage treatment plant simulation test
10 (OECD303A or equivalent).

11 The proposed scheme is to be followed for both industrial and/or domestic (i.e.
12 municipal) sewage treatment plants, as applicable from the chemical's release pattern.

13 **R.7.8.21.2 Preliminary considerations**

14 In accordance with REACH Annex VI, the preliminary step of the ITS consists of a
15 collection and critical evaluation of all (public) data that may be available for the STP
16 Toxicity endpoint.

17 It should be noted that based on the test requirements in Annex VII for most substances
18 a Ready Biodegradability test will be available. As such, there may be some relevant –
19 but not necessarily fully conclusive- STP toxicity data available (except for inorganic
20 chemicals which cannot be tested for degradability). The principle followed in the ITS is
21 that existing data from short term tests can be retested/overwritten by more
22 realistic/higher tier data, except if the existing data already come from simulation or
23 field testing.

24 Step 1 covers calculation of exposure (PEC_{stp}) in both domestic and industrial plants, as
25 applicable; this information will be needed to calculate the PEC/PNEC ratio and decide on
26 need for more data/higher tier testing. Guidance on the PEC_{stp} calculation is provided by
27 Chapter R.16.

28 Steps 2-4 cover evaluation of existing hazard information and the strategy to make
29 optimal use of existing information, and avoid the need for new testing where possible.

30 Step 5 covers the execution of an activated sludge respiration test; i.e. first tier of STP
31 toxicity testing (short term test).

32 Step 5* covers the retesting option for short term tests for industrial plants, based on
33 sludge from that plant. These results are only relevant for this single plant, and cannot
34 be extrapolated to other industrial or domestic plants.

1 Step 6 covers the execution of a confirmatory, longer term simulation test, i.e. the
2 highest possible tier of STP toxicity testing. This is the test level with the highest real
3 world relevance²⁷.

4 **R.7.8.21.3 Testing strategy for toxicity to STP microorganisms**

5 **Stage 1.** Calculation of exposure. Outcome: PEC_{stp} or $PEC_{influent}$ (calculate for both
6 domestic and industrial STP, as applicable).

7 **Stage 2.** Assessment of information from existing and quality-assured microbial
8 inhibition tests to derive a $PNEC_{stp}$ (i.e. data from respiration inhibition,
9 nitrification inhibition, ciliate growth, sludge growth inhibition, *P. putida*, Shk1
10 assay).

11 Stage 2.1. IF adequate data are available, THEN derive $PNEC_{stp}$.

12 IF $PEC/PNEC < 1$, THEN stop.

13 IF $PEC/PNEC > 1$ for domestic plants, THEN move to stage 6, confirmatory
14 testing

15 IF $PEC/PNEC > 1$ for industrial plants, THEN move to stage 5* (nb: for
16 industrial plants, there is the possibility to perform an activated sludge
17 respiration test (or nitrification inhibition test) test with sludge from the
18 specific installation)

19 Stage 2.2. IF no data are available, or the data are considered inadequate, THEN
20 move to stage 3.

21 **Stage 3.** Assessment of information from Ready Biodegradation tests to derive a
22 $PNEC_{stp}$.

23 Stage 3.1. IF the chemical is readily biodegradable, or if there is evidence of good
24 degradation of a positive control in the presence of the test substance,
25 THEN derive $PNEC_{stp}$.

26 IF $PEC/PNEC < 1$, THEN stop.

27 IF $PEC/PNEC > 1$, THEN go to stage 5 (nb: a respiration inhibition test can
28 be used, if needed, to refine/overwrite the information inferred from a
29 ready test. The respiration inhibition test may need to be done for both
30 domestic and industrial sludge, as applicable).

31 Stage 3.2. IF no data are available from a Ready tests, or for all other situations not
32 falling under stage 3.1 (e.g. not readily biodegradable and no information
33 on inhibition), THEN go to stage 4.

34 **Stage 4.** Assessment of existing and quality-assured information from inherent
35 biodegradability tests, simulation tests, and/or field data.

²⁷ Based on the experience with the existing high production volume chemicals programme in the EU (ca. 150 chemicals), it is expected that this approach will be seldom needed. For the large majority of chemicals, a lower tier assessment based on a short term tests will suffice.

- 1 Stage 4.1. IF adequate data are available, THEN derive $PNEC_{stp}$.
2 IF $PEC/PNEC < 1$, THEN stop.
3 IF $PEC/PNEC > 1$, THEN risk reduction needs to be considered (no further
4 refinement testing possible).
- 5 Stage 4.2. IF no data are available, or data are inadequate, THEN move to stage 5.
- 6 **Stage 5.** Execution of an activated sludge respiration inhibition test (OECD 209). (NB:
7 this test can also be substituted by a nitrification inhibition test)
- 8 Stage 5.1. IF $PEC/PNEC < 1$, THEN stop.
- 9 Stage 5.2. IF $PEC/PNEC > 1$ for domestic and/or industrial plants, THEN move to step
10 6
- 11 **Stage 5.** * Refinement test for industrial plants only: a test resulting in $PEC/PNEC > 1$
12 can be repeated with sludge from the industrial plant of interest. This results
13 can not be extrapolated to other plants
- 14 Stage 5.1. * If on the basis of a test with nitrifying bacteria (existing data), a
15 $PEC/PNEC$ ratio above 1 is derived for an industrial STP, a revised $PNEC_{stp}$
16 for a specific industrial site can be derived from a nitrification inhibition
17 test using the sludge from this site's STP. (NB: For domestic STPs a
18 revision of the $PNEC$ is not possible in this way, since sludge from one
19 single STP can not be regarded as being representative of all domestic
20 STPs with respect to their nitrifying activity).
21 IF $PEC/PNEC_{revised} < 1$, THEN stop.
22 IF $PEC/PNEC_{revised} > 1$, THEN proceed to stage 6 (simulation tests with
23 investigation of nitrification performance)
- 24 Stage 5.2. * If on the basis of a standard respiration inhibition-, standardised
25 biodegradation- or an activated sludge growth inhibition test (existing
26 data), a $PEC/PNEC$ ratio above 1 is derived for an industrial STP, a revised
27 $PNEC_{stp}$ for can be derived from a respiration inhibition test using sludge
28 from the site's specific STP.
29 IF $PEC/PNEC_{revised} < 1$, THEN stop.
30 IF $PEC/PNEC_{revised} > 1$, THEN move to stage 6.
- 31 Stage 5.3. * If on the basis of a single species test with ciliated protozoa a $PEC/PNEC$
32 ratio above 1 is derived for domestic or industrial sewage treatment
33 plants, a test reflecting the integrity of the native ciliate population is
34 necessary (except if it can be shown that protozoa are not relevant in the
35 system under consideration²⁸). It is recommended here to move to stage
36 6, simulation testing, with investigation of settling performance.
- 37 **Stage 6.** Confirmatory simulation testing: an pilot scale simulation test, using activated
38 sludge from the STP of interest (domestic or industrial) as a source of

²⁸ At present a standard protocol for a test on ciliated protozoa which can provide data on revising the $PNEC_{stp}$ (based on ciliates) is not available. However, additional research results are underway and will be presented in 2007 by UBA.

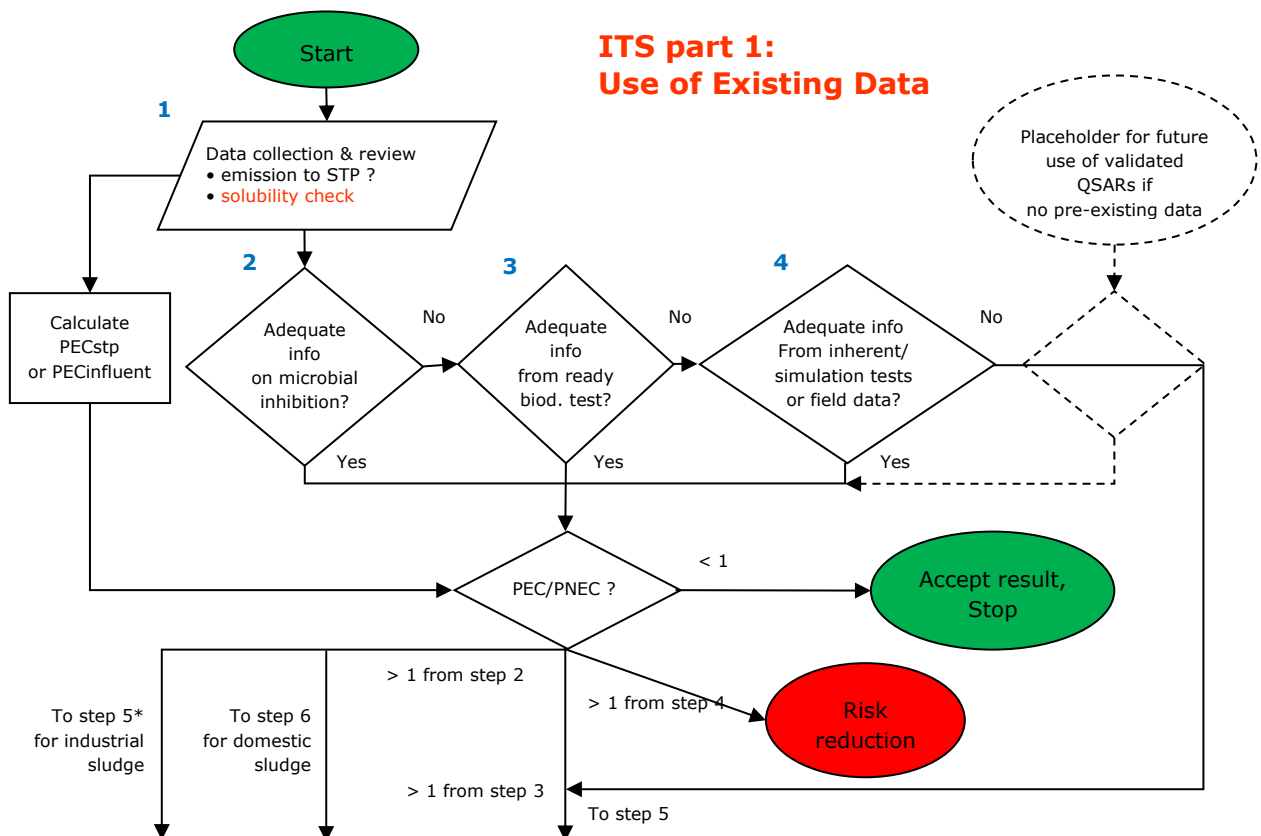
inoculum can be used as a highly realistic test to refine the $PNEC_{STP}$ derived from any short term microbial inhibition test. The stability and performance of the plant should be monitored over a somewhat longer period (e.g. 2 weeks, following a 2 week start-up period). The test should monitor critical performance parameters such as BOD/COD removal, N-removal (nitrification), and the evolution of the sludge volume index (SVI) parameter, versus an undosed control.

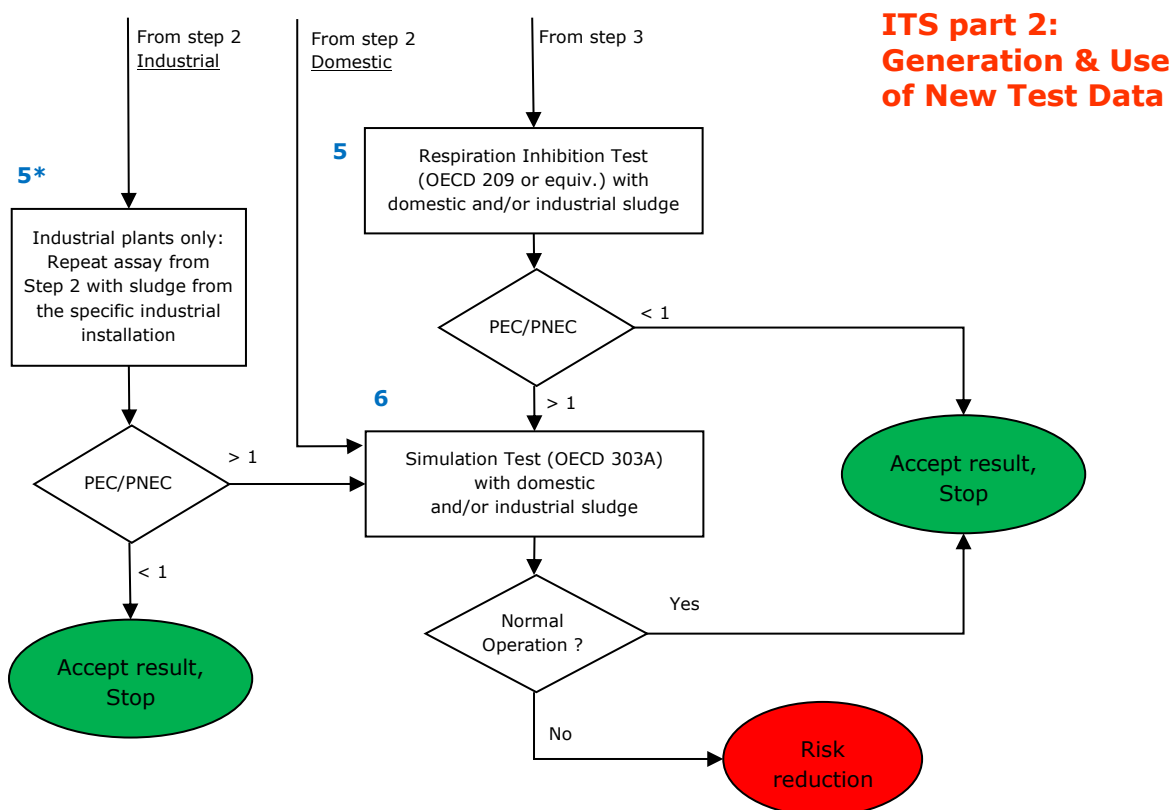
Stage 6.1. IF good and stable reactor performance, THEN stop (i.e. $PEC/PNEC < 1$)

Stage 6.2. IF signs of inhibition or operational issues versus an undosed control unit, THEN $PEC/PNEC > 1$, and risk management (emission reduction at source) is required.

(NB: for situations of intermittent release, a simulation test can be more difficult to perform; it would require a realistic dosing regime, which simulates the situation for the emission to the full scale plant).

Figure R.7.8—9 Integrated Testing Strategy for toxicity on STP microorganisms





1

2

3 R.7.8.22 References on toxicity to STP microorganisms

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11 **R.7.9 Degradation/biodegradation**

12 **R.7.9.1 Introduction**

13 Degradation is an important process that can result in the loss or transformation of a
14 chemical substance in the environment. Degradation of organic substances in the
15 environment influences exposure and, hence, it is a key parameter for estimating the
16 risk of long-term adverse effects on biota. Degradation rates, or half-lives, are
17 determined in, or default rates assigned from, laboratory-based degradation tests. These
18 tests can be simple screening tests (e.g. the OECD 301 ready biodegradability tests and
19 the OECD 111 hydrolysis as a function of pH test), or relatively complex higher tiered
20 simulation types of tests (e.g. the OECD 308 aerobic and anaerobic transformation in
21 aquatic sediment systems, OECD 309 aerobic and anaerobic transformation in surface
22 water and the OECD 303 aerobic sewage treatment).

23 Information on the degradability of substances may be used for hazard assessment (e.g.
24 for classification and labelling), risk assessment (for chemical safety assessment) and
25 persistence assessments (for PBT/vPvB assessment). Hazard and persistence
26 assessments, or risk in general, and aquatic hazard classification in particular, are
27 normally based on data obtained in standardised tests for ready biodegradability and
28 hydrolysis. Results of tests simulating the biodegradation in water, aquatic sediment and
29 soil may also be used for these purposes. Other types of test data that may be
30 considered in an assessment of the potential environmental hazard or risk include
31 sewage treatment plant (STP) simulation data, inherent biodegradability, anaerobic
32 biodegradability, biodegradability in seawater and abiotic transformation (OECD, 2006b).
33 In determining which higher tiered or simulation degradation data are required
34 consideration should be given to the partitioning behaviour of the substance and its
35 release or emission pattern. This may be useful for prioritising testing requirements to
36 those environmental compartments that are the most relevant. Consideration should be
37 given to whether the substance being assessed can be degraded to give stable and/or
38 toxic degradation products. Where such degradation can occur, the assessment should
39 give due consideration to the properties (including toxic effects and bioaccumulation
40 potential) of the products that might arise.

1 **R.7.9.1.1 Definition of degradation/biodegradation**

2 Degradation can result in the loss or transformation of a chemical substance in the
3 environment. Degradation processes can be abiotic or biotic. Abiotic or non-biological
4 degradation can occur by physico-chemical processes such as hydrolysis, oxidation and
5 photolysis. Removal due to biotic or biological degradation is commonly known as
6 biodegradation. Biodegradation can proceed in the presence of oxygen (aerobic
7 biodegradation) or in the absence of oxygen (anaerobic biodegradation).

8 Biodegradation is often preceded by the terms primary or ultimate. Primary
9 biodegradation describes the initial transformation of a substance by microorganisms to
10 another organic substance, a transformation product or metabolite; ultimate
11 biodegradation describes the (multistep) degradation process leading to inorganic end
12 products and biomass.

13 There are numerous terms and phrases associated with assessing degradation. Some of
14 the commonly used terms are defined in [Table R.7.9–1](#).

15 **Table R.7.9–1 Glossary of terms associated with degradation**

Term	Definition
Fate	Distribution of a substance in various environmental compartments (e.g. soil or sediment, water, air, biota) as a result of transport, partitioning, transformation, and degradation.
Biodegradation	The biologically mediated degradation or transformation of substances usually carried out by microorganisms.
Primary biodegradation	The structural change (transformation) of a chemical substance by microorganisms resulting in the loss of the original chemical identity.
Ultimate aerobic biodegradation	The breakdown of a substance by microorganisms in the presence of oxygen resulting in the formation of carbon dioxide, sulphate, nitrate and new biomass
Ultimate anaerobic biodegradation	The breakdown of a substance in absence of oxygen resulting in the formation of carbon dioxide and final reduction products like methane, H ₂ S, or NH ₃ , mineral salts and new biomass.

Term	Definition
Ready biodegradability tests	Stringent screening tests, conducted under aerobic conditions, in which a high concentration of the test substance (in the range of 2 to 100 mg/L) is used and ultimate biodegradation is measured by non-specific parameters like Dissolved Organic Carbon (DOC), Biochemical Oxygen Demand (BOD) and CO ₂ production. Small amounts of domestic sewage, activated sludge or secondary effluent form the microbial inoculum in tests for ready biodegradability. The inoculum should not have been artificially pre-adapted to the test substance through previous exposure to either the test substance or structurally related substances. The test substance is provided as the sole source of carbon for energy and growth. A positive result in a test for ready biodegradability can be considered indicative of rapid and ultimate degradation in most environments including biological STPs
Inherent biodegradability tests	Tests inoculated with a high concentration of microorganisms carried out under aerobic conditions in which biodegradation rate and/ or extent are measured. The test procedures offer a higher chance of detecting biodegradation compared to tests for ready biodegradability and therefore if an inherent test is negative this could indicate the potential for environmental persistence.
Simulation tests	Aerobic and anaerobic tests that provide data on biodegradation under specified environmentally relevant conditions. These tests attempt to simulate degradation in a specific environment by use of indigenous biomass, media, relevant solids (i.e. soil, sediment, activated sludge or other surfaces) to allow sorption of the substance, and a typical temperature that represents the particular environment. A representative and low concentration of test substance is used in tests designed to determine the biodegradation rate constant whereas higher concentrations for analytical reasons are normally used for identification and quantification of major transformation products.
Persistence	A substance that resists degradation processes and is present in the environment for a long time. Specific criteria have been established in Persistent Organic Pollutant (POP) protocols, in the TGD and in REACH (PBT/vPvB; see sections 1.1.1 and 1.2.1 of Annex XIII to REACH). In the latter persistent (P) and very persistent (vP) refers to substances that have degradation half-lives above certain trigger values in surface water, sediment or soil.
Abiotic degradation	Degradation mediated through processes other than biodegradation such as hydrolysis, photolysis and interactions with other substances (e.g. oxidation). Abiotic degradation studies typically provide a measure of primary degradation.
Hydrolysis	Decomposition or degradation of a substance by reaction with water

Term	Definition
Photolysis	Chemical decomposition or degradation induced by light or other radiant energy. Direct photolysis in natural water involves the transformation of a substance resulting from the direct absorption of a solar photon. Indirect photolysis in natural water sometimes involves the transformation of a substance due to energy transfer from naturally occurring photosensitizers in electronically excited triplet states. However, indirect photolysis more often involves the transformation of a substance due to reactions with transient oxidants such as hydroxyl radicals, molecular oxygen in a singlet electronic state, and peroxy radicals. Indirect photolysis is an important transformation process for substances in the gaseous state in air.
Oxidation	A substance may undergo oxidation/reduction or other transformation reactions (under storage, use etc.). These reactions may be slow and initiated for instance by the atmospheric oxygen or the presence of other oxidising agents.
Degradation rate constant	Typically a first order or pseudo first order kinetic rate constant, k (d^{-1}), which indicates the rate of the degradation processes. However, depending upon the ratio of the substance to degrader biomass, the rate constants may be Monod constants reflecting growth processes.
Half-life, $t_{1/2}$	Term used to characterise the rate of a first or pseudo-first order reaction. It is the time interval that corresponds to a concentration decrease by a factor 2. The half-life and the degradation rate constant are related by the equation $t_{1/2} = -\ln 2/k$. Half-lives are usually expressed in hours or days and can be assigned to either primary degradation or ultimate biodegradation (mineralisation).
DT50	(Disappearance Time 50) is the empirically measured time within which the initial concentration of the test substance is reduced by 50%. It should be stated whether the DT50 refers to primary degradation or mineralisation (ultimate biodegradation)
DT90	(Disappearance Time 90) is the time within which the initial concentration of the test substance is reduced by 90%. In the case of a first-order reaction, this time would be slightly longer than 3 half-lives
Degradation product(s)	The substances produced as a result of degradation processes. For aerobic ultimate degradation, or mineralisation, these are carbon dioxide, water and mineral salts.
Field Data	Measured concentrations of a substance in an environmental compartment, which can be related to loading, partitioning, dilution and degradation.

1 **R.7.9.1.2 Objective of the guidance on degradation/biodegradation**

2 The purpose of this report is to define an integrated testing strategy (ITS) that would
3 help collect information on substances, within the context of REACH, i.e. to enable the
4 hazard and risk assessment of substances to be performed. This information should form
5 the basis for classification, PBT- and vPvB-assessment, and exposure assessment for use
6 in risk characterisation. To do this all degradation data sources, including non-testing
7 data, simulation testing data, field data, and exposure data will be taken into account.

8 Degradation is an important endpoint against the following *regulatory* needs:

- 9 • Identifying whether a substance fulfils the P or vP criteria within the PBT/vPvB
10 assessment and determining whether a substance has the potential to cause
11 long-term adverse effects in the environment in environmental hazard
12 classification
- 13 • Determining the Predicted Environment Concentration (PEC) of a substance in
14 environmental exposure assessment for use in risk characterisation

15 The general process of information collection will be a step-wise process. The following
16 four processes are foreseen for collection of information on substance properties by a
17 potential registrant according to the Guidance Note in Annex IV on the information
18 requirements referred to in Article 9:

- 19 • Gather and share existing information
- 20 • Consider information needs
- 21 • Identify information gaps
- 22 • Generate new data/propose testing strategy

23 Within the report the proposed general ITS will be tested against selected substances.
24 For exploration of elements of the strategy, fractions of the data of data-rich substances
25 will be used to test the strategy i.e. different tonnage levels, different levels of available
26 data etc.

27 **R.7.9.2 Information requirements for degradation/biodegradation**

28 Article 10 of REACH presents the information that should be submitted for registration
29 and evaluation of substances. In Article 12 of REACH the dependence of the information
30 requirements on production volume (tonnage) is established in a tiered system,
31 reflecting that potential exposure increases with volume. Referring to article 10, Annexes
32 VI to XI to REACH set out the requirements for generating information on the substance
33 to be registered. However, for existing substances all available information should be
34 used independently from the tonnage trigger.

35 In addition, if the registrant cannot derive a definitive conclusion (i) ("The substance
36 does not fulfil the PBT and vPvB criteria") or (ii) ("The substance fulfils the PBT or vPvB
37 criteria") in the PBT/vPvB assessment using the relevant available information, the
38 registrant must, based on section 2.1 of Annex XIII to REACH, generate the necessary
39 information for deriving one of these conclusions, regardless of his tonnage band (for
40 further details, see Chapter R.11 of the *Guidance on IR&CSA*). Alternatively, the

1 substance may be considered and managed “as if it is a PBT or vPvB” (see Chapter R.11
2 for details).

3 **R.7.9.2.1 Annex VII (Registration tonnage >1 t/y -<10 t/y)**

4 Current text regarding degradation in Annex VII to REACH. This information is required if
5 the substance meets the criteria laid down in Annex III:

6 • substances for which it is predicted (i.e.; by the application of (Q)SARs or
7 other evidence) that they are likely to meet the criteria for category 1A or 1B
8 classification in the hazard classes carcinogenicity, germ cell mutagenicity or
9 reproductive toxicity or the criteria in Annex XIII.

10 • substances:

11 - with dispersive or diffuse use(s) particularly where such substances are
12 used in consumer mixtures or incorporated into consumer articles; and

13 - for which it is predicted (i.e. by application of (Q)SARs or other
14 evidence) that they are likely to meet the classification criteria for any
15 health or environmental hazard classes or differentiations under
16 Regulation (EC) No 1272/2008.

17

Column 1 Standard Information Required	Column 2 Specific rules for adaptation from Column 1
9.2. Degradation 9.2.1. Biotic 9.2.1.1. Ready biodegradability	7.2.1.1. The study does not need to be conducted if the substance is inorganic

18

19 Ready Biodegradation Test:

20 The waiving of the requirements for the following tests should be considered in the
21 following circumstance:

22 **Column 2:** “The study does not need to be conducted if the substance is inorganic.”

23 Inorganic substances cannot be tested for ready biodegradability.

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1 **R.7.9.2.2 Annex VIII (Registration tonnage \geq 10 t/y)**

2 Current text regarding degradation in Annex VIII to REACH

Column 1	Column 2
Standard Information Required	Specific rules for adaptation from Column 1
9.2. Degradation	9.2. Further degradation testing shall be considered if the chemical safety assessment according to Annex I indicates the need to investigate further the degradation of the substance. The choice of the appropriate test(s) will depend on the results of the chemical safety assessment.
9.2.2. Abiotic 9.2.2.1. Hydrolysis as a function of pH	9.2.2.1. The study does not need to be conducted if: – the substance is readily biodegradable; or– the substance is highly insoluble in water;

3 The requirements at this supply tonnage are for data on ready biodegradation (as
 4 defined in Annex VII to REACH) and for hydrolysis data at pHs 4, 7 and 9. Normally, a
 5 test for ready biodegradability would be required, although it may be possible to provide
 6 a valid QSAR as described in Section R.6.1.

7 Hydrolysis Test

8 This test is designed to provide information on abiotic degradation that can help in
 9 classification, persistence testing and in determining the fate of a substance in
 10 environmental surface waters. The test may be waived under the following
 11 circumstances.

12 **Column 2:** "The substance is readily biodegradable."

13 In these circumstances, the hydrolysis test will provide little additional information since
 14 rapid mineralisation in the environment is already assumed.

15 **Column 2:** "The substance is highly insoluble in water"

16 In these circumstances, the test will be practically very difficult to conduct without
 17 special analytical techniques. In addition, it is likely that the aqueous environment may
 18 not be the principal environmental compartment of concern (see Section [R.7.9.6](#)). The
 19 test may still be important in certain circumstances however, for example where
 20 hydrolysis occurs at the surface of particles of the undissolved substance leading to more
 21 soluble products, but may be considered on a case-by-case basis if needed for risk
 22 assessment purposes.

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1 **R.7.9.2.3 Annex IX (Registration tonnage \geq 100 t/y)**

2 Current text regarding degradation in Annex IX to REACH:

Column 1	Column 2
Standard Information Required	Specific rules for adaptation from Column 1
9.2. Degradation	9.2. Further biotic degradation testing shall be proposed by the registrant if the chemical safety assessment according to Annex I indicates the need to investigate further the degradation of the substance and its degradation products. The choice of the appropriate test(s) depends on the results of the chemical safety assessment and may include simulation testing in appropriate media (e.g. water, sediment or soil).
9.2.1. Biotic	
9.2.1.2. Simulation testing on ultimate degradation in surface water	9.2.1.2. The study need not be conducted if: <ul style="list-style-type: none"> – the substance is highly insoluble in water; or – the substance is readily biodegradable.
9.2.1.3. Soil simulation testing (for substances with a high potential for adsorption to soil)	9.2.1.3. The study need not be conducted: <ul style="list-style-type: none"> – if the substance is readily biodegradable; or – if direct and indirect exposure of soils is unlikely.
9.2.1.4. Sediment simulation testing (for substances with a high potential for adsorption to sediment)	9.2.1.4. The study need not be conducted: <ul style="list-style-type: none"> – if the substance is readily biodegradable; or – if direct and indirect exposure of sediment is unlikely.
9.2.3. Identification of degradation products	9.2.3. Unless the substance is readily biodegradable

3

4 Additional biodegradation testing may be required at this tonnage depending on the
5 relevant environmental exposure considerations.6 *Further biotic degradation testing shall be proposed by the registrant if the chemical*
7 *safety assessment according to Annex I indicates the need to investigate further the*
8 *degradation of the substance and its degradation products. The choice of the appropriate*
9 *test(s) depends on the results of the chemical safety assessment and may include*
10 *simulation testing in appropriate media (e.g. water, sediment or soil).*11 This may be taken as providing a general framework by which the exclusion of certain
12 testing may be justified by the need to clarify or revise the conclusions of the CSA.13 Simulation testing of ultimate degradation in surface water14 **Column 2:** "The substance is readily degradable."

1 In these circumstances, the simulation test will provide little additional information since
2 rapid mineralisation in the environment is already assumed. This will be so unless a
3 refinement of the estimated environmental half-life is needed to aid the risk
4 characterisation at the regional scale.

5 **Column 2:** "The substance is highly insoluble in water"

6 The solubility in water may be so low that the test may be practically difficult or
7 impossible to conduct at concentrations below the water solubility limit of the substance.
8 It is also likely that the surface water environment will not be the principal environment
9 of concern and, if a simulation test is required, consideration should be given to a test in
10 a different environmental media (e.g. soil, sediment). If the substance is considered to
11 be a potential PBT/vPvB, e.g. by fulfilling screening criteria on persistence, then it is
12 necessary to consider additional information in accordance with section 2.1 of Annex XIII
13 to REACH.

14 Simulation testing on ultimate degradation in soil

15 **Column 2:** "The substance is readily degradable."

16 In these circumstances, the simulation test will provide little additional information since
17 rapid mineralisation in the environment is already assumed. This will be so unless a
18 refinement of the estimated soil degradation half-life is needed to aid the risk
19 characterisation.

20 **Column 2:** "If direct and indirect exposure of soil is unlikely."

21 If there is no exposure of the soil, or the exposure is so low that no refinement of the
22 PECs is required, then this test may not be necessary. If the substance is considered a
23 PBT/vPvB candidate, then it may be necessary to consider this test if soil is the
24 environmental compartment of concern (see Section [R.7.9.6](#)).

25 Simulation testing on ultimate degradation in sediment

26 **Column 2:** "The substance is readily degradable"

27 In these circumstances, the simulation test will provide little additional information since
28 rapid mineralisation in the environment is already assumed. This will be so unless a
29 refinement of the estimated sediment degradation half-life is needed to aid the risk
30 characterisation at the regional scale.

31 **Column 2:** "If direct and indirect exposure of sediment is unlikely"

32 If there is no exposure of sediment, or the exposure is so low that no refinement of the
33 PEC_{regional} is required, then this test may not be necessary. If the substance is considered
34 a PBT/vPvB candidate, then it may be necessary to consider this test if sediment is the
35 environmental compartment of concern (see Chapter R.11 of the *Guidance on IR&CSA*).

36 Identification and/or assessment of degradation products

37 These data are required if information on the degradation products following primary
38 degradation is required in order to complete the CSA. This is considered further in
39 Section [R.7.9.4](#).

1 **Column 2:** "The substance is readily degradable"

2 In these circumstances, it may be considered that any degradation products formed
3 during such degradation would themselves be sufficiently rapidly degraded and therefore
4 that no further assessment would be required.

5 **R.7.9.2.4 Annex X (Registration tonnage \geq 1000 t/y)**

6 Current text regarding degradation in Annex VIII to REACH:

Column 1	Column 2
Standard Information Required	Specific rules for adaptation from Column 1
<p>9.2. Degradation</p> <p>9.2.1. Biotic</p>	<p>9.2. Further biotic degradation testing shall be proposed if the chemical safety assessment according to Annex I indicates the need to investigate further the degradation of the substance and its degradation products. The choice of the appropriate test(s) depends on the results of the chemical safety assessment and may include simulation testing in appropriate media (e.g. water, sediment or soil).</p>

7 These data concern further confirmatory testing on biodegradation and are required if
8 information on the degradation products following primary degradation is required in
9 order to complete the CSA including the PBT/vPvB-assessment or if it is felt necessary by
10 the registrant because of implications for the hazard classification.

11 **R.7.9.3 Information on degradation/biodegradation and its sources**

12 This section identifies sources of information, including non-testing and testing data,
13 which are important in the assessment of degradation. An inventory of officially adopted
14 EU and OECD test guidelines and their application domain will be provided.

15 Other information such as the substance physico-chemical properties are also important
16 in identifying appropriate studies to conduct, for example certain biodegradation tests
17 are not applicable for volatile and poorly water-soluble substances. These data can also
18 assist in identifying environmental compartments of concern in order to prioritise higher
19 tiered testing data accordingly.

20 **R.7.9.3.1 Data on degradation/biodegradation**

21 **Non-testing data on degradation/biodegradation**

22 Databases

23 Qualitative information is available for a number of biodegradation pathways, most
24 notably the EAWAG (former University of Minnesota) Biocatalysis/Biodegradation
25 Database (<http://eawag-bbd.ethz.ch/>). This database collates known biodegradation
26 pathways that have been published in the open literature. Many of these experimental
27 studies were designed to determine pathways of biodegradation using pure cultures of
28 microorganisms. Therefore these data can aid in the identification of potential
29 degradation products where analysis of metabolites is warranted. Similar degradation

1 pathways and tools are available on the website for the KEGG databases and tools
2 (<http://www.genome.jp/tools/pathpred/>).

3 The suitability of this data on use in hazard, persistence and risk assessment needs
4 careful consideration and may only contribute as part of a *Weight of Evidence*
5 assessment if other data are available.

6 Another source of empirical information that collates biodegradation, photooxidation and
7 hydrolysis data is the Japanese National Institute of Technology and Evaluation (NITE)
8 database (<http://www.nite.go.jp/en/chem/qsar/evaluation.html>).

9 Quantitative Structure Activity Relationships

10 A variety of models have been developed to predict biodegradation. These include
11 structure biodegradability relationships (SBRs) and quantitative structure
12 biodegradability relationships (QSBRs). SBRs provide qualitative endpoints such a
13 passing or failing a ready biodegradation test. QSBRs provide an estimation of rate or
14 half-life. More information can be found in ECHA Practical Guide "*How to use and report*
15 *(Q)SARs*"²⁹ and in Nendza *et al.*, 2013.

16 Examples of such models include:

- 17 • Syracuse Research Corporation's Estimation software (freely available) that
18 includes packages to determine log octanol-water partition coefficients,
19 Henry's Law constant, indirect photolysis in the atmosphere (by reaction with
20 OH and NO₃), biodegradation and hydrolysis ([http://www.srcinc.com/what-
21 we-do/environmental/tools-and-models.html](http://www.srcinc.com/what-we-do/environmental/tools-and-models.html)). This model is also included in
22 EPI Suite of the US EPA (<https://www.epa.gov/tsca-screening-tools>). The
23 Biodegradation Probability Program for Windows (BIOWIN)) as part of the EPI
24 Suite calculates the probability score that a substance under aerobic
25 conditions with mixed cultures of microorganisms will biodegrade rapidly or
26 slowly in the environment according to expert judgement (BIOWIN 1 & 2),
27 two models (BIOWIN 3 & 4) estimating ultimate and primary biodegradation
28 timeframe (hours, days, weeks, months, years) for environmental
29 degradation half-lives according to a training set of around 200 substances
30 evaluated by expert judgment and two models (BIOWIN 5&6) that have been
31 validated against MITI ready biodegradability results on many hundreds of
32 substances (Loonen *et al.*, 1999). BIOWIN 7 models anaerobic biodegradation
33 instead scoring a substance in "pass" or "fail". In the help files of the
34 software, the training set substances used for development of the BIOWIN
35 models are presented. BIOWIN 7 has been developed based on a data set
36 using a serum bottle anaerobic biodegradation screening test but this data set
37 was not separated into separate training and validation sets. BIOHCWIN (also
38 developed by Syracuse Research Corporation) is a model that predicts the
39 primary degradation half-lives of hydrocarbons in water. It relates to a
40 situation that is similar to heavily polluted sites (i.e. with higher substance

²⁹https://echa.europa.eu/documents/10162/13655/pg_report_qsars_en.pdf/407dff11-aa4a-4eef-a1ce-9300f8460099

1 concentrations than those very low concentrations most frequently observed
2 in the environment). A description of the model and its development is given
3 in Howard *et al.* (2005). It is noted that BIOWIN 5 and 6 QSAR models for
4 biodegradation estimation have been developed based on a training data set
5 consisting of results from ready biodegradability tests, in particular MITI I
6 data, which use a uniquely derived inoculum. The training set for BIOWIN 1,
7 2, 3 (ultimate degradation time frame) and 4 (primary degradation
8 timeframe) on the contrary, was based on the overall conclusions of a panel
9 of US EPA experts for rapid or slow environmental degradation and based on
10 various types of degradation information on the training set substances.
11 Nevertheless also the BIOWIN 1, 2 and 3 models have been investigated in
12 the literature for their predictability concerning non ready and ready
13 biodegradability (Howard *et al.* 1992). A recent study from Prosser *et al.* 2016
14 on 489 data points showed that the primary biodegradation half-life values
15 predicted by BIOHCWIN were within one order of magnitude from
16 experimental values for >87% of the data points. Although this result
17 suggests that the model predictions are accurate for a large set of
18 hydrocarbons, the model should be used with care. Some data used in
19 BIOHCWIN are based partially on half-lives obtained for single compounds
20 studied as multi-constituent substances, e.g. cycloalkanes (Howard *et al.*,
21 2005), thus the predicted half-lives can be affected by co-metabolism and
22 therefore may overestimate the rate of degradation compared to a situation
23 where co-metabolism does not occur. For example, when PBT assessment is
24 conducted for a single monoconstituent hydrocarbon substance, the presence
25 of a hydrocarbon co-substrate should not be assumed. Therefore, in such
26 cases BIOHCWIN half-lives which are below the P/vP criteria should not be
27 used to support a conclusion "not P/vP". Care must also be taken with
28 branched compounds, as it appears that their environmental half-lives may be
29 underestimated by BIOHCWIN (Rorije *et al.*, 2012). It is worth noting that
30 none of the BIOWIN models automatically report whether their predictions are
31 within their applicability domains. The models domains reported in the on-line
32 BIOWIN user's guide should be consulted for further information;

- 33 • The QSAR Toolbox (<https://www.qsartoolbox.org/home>) is a software
34 application to identify and fill data gaps for (eco)toxicological and
35 environmental fate endpoints. The Toolbox can be used to
 - 36 - identify relevant structural characteristics and potential mechanism or
37 mode of action of a target chemical,
 - 38 - identify other chemicals that have the same structural characteristics
39 and/or mechanism or mode of action,
 - 40 - fill the data gap(s) using the experimental results available from the
41 databases contained in the Toolbox.
- 42 • The CATALOGIC software suite (commercial, requires licence) is a platform for
43 models and databases related to the environmental fate of substances such as
44 abiotic and biotic degradation, bioaccumulation and acute aquatic toxicity.
45 Microbial degradation of substances (BOD and CO₂ production) is predicted by
46 CATALOGIC 301B, 301C and 301F models. Biodegradability is estimated
47 based on simulated catabolic pathways, material balance of molecular
48

- 1 transformations and probabilities for their occurrence. The models predict
2 also primary and ultimate half-lives, quantities of metabolites. Physico-
3 chemical properties and acute toxicity to aquatic organisms from the major
4 trophic levels are also predicted (Dimitrov *et al.*, 2011); [http://oasis-](http://oasis-lmc.org/products/models/environmental-fate-and-ecotoxicity.aspx)
5 [lmc.org/products/models/environmental-fate-and-ecotoxicity.aspx](http://oasis-lmc.org/products/models/environmental-fate-and-ecotoxicity.aspx)).
6 Applicability domain, QMRF and QPRF are also provided;
- 7 • The EAWAG Biocatalysis/Biodegradation Database also proposes a Pathway
8 Prediction System (PPS) (<http://eawag-bbd.ethz.ch/predict/>). This model
9 predicts plausible degradation pathways using biotransformation rules
10 established from the reactions compiled in the EAWAG-BBD database.
 - 11 • TOPKAT (commercial, requires a licence) has an aerobic biodegradability
12 module. This module comprises a statistically significant and cross-validated
13 quantitative structure-toxicity relationship (QSTR) model applicable to a
14 specific class of substances, and the data from which these models were
15 derived. A single study that reported the biodegradability of 894 compounds,
16 as assessed by the Japanese Ministry of International Trade and Industry
17 (MITI) I test protocol, was used to develop these models. Molecular structure
18 is the only input required to conduct an assessment of aerobic
19 biodegradability ([http://accelrys.com/products/collaborative-science/biovia-](http://accelrys.com/products/collaborative-science/biovia-discovery-studio/qsar-admet-and-predictive-toxicology.html)
20 [discovery-studio/qsar-admet-and-predictive-toxicology.html](http://accelrys.com/products/collaborative-science/biovia-discovery-studio/qsar-admet-and-predictive-toxicology.html));
 - 21 • Multicase and its newer version Case-Ultra (commercial, requires licence)
22 have a META program to predict metabolic breakdown pathways of
23 substances and a ready biodegradability prediction program. All rules have
24 been determined based on reliable literature sources. In META a tree of
25 predicted transformation products can be generated, saved and analysed for
26 mammalian metabolism, aerobic biodegradation, anaerobic biodegradation
27 and photodegradation (<http://www.multicase.com/meta-pc>);
 - 28 • The Danish QSAR database is freely available at <http://qsar.food.dtu.dk>. It
29 contains, besides QSAR predictions of a range of physical chemical properties,
30 toxicity and ecotoxicity endpoints, also predictions on photolysis, hydrolysis
31 and all the biodegradability QSAR models included in the EPISuite program
32 package (v. 4.1, c.f. also above). In addition this QSAR prediction database
33 contains predictions from three QSAR models developed by the DTU³⁰ QSAR
34 group in collaboration with the Danish EPA. The models were trained on
35 several hundreds of the same MITI I and other available ready
36 biodegradability test data by employing three modelling approaches: Case-
37 Ultra, Leadscope and SciQSAR. A simple yes/ no conclusion relating to
38 whether the individual predictions falls within the applicability domain of each
39 model as defined by the modelling concept is provided. Finally a majority vote
40 prediction of the three lastly mentioned QSAR model predictions is also
41 provided. The database contains predictions for more than 630,000 discrete
42 organic substances including almost all those substances pre-registered for

³⁰ DTU: Danmarks Tekniske Universitet, Technical University of Denmark

1 REACH and require only the CAS number as an input. In addition the database
2 allows complex searches to be made (combined search algorithms concerning
3 the predictions for all endpoints included in the database by use of the
4 following conditional options to be fulfilled by specific searches: "OR", "AND"
5 and "NOT" and conditions such as ">", "<", "=", "contains" plus the option for
6 choice of freely selected sub-structures and in relation to a structural
7 similarity index value). QMRFs on all models developed and validated by the
8 DTU group are provided on the website;

- 9 • VEGA (<http://www.vega-qsar.eu/>) is a freely available platform offering a
10 collection of QSAR models developed by US EPA and by the EU funded
11 CAESAR project for (eco)toxicological and environmental fate endpoints.
- 12 • An approach based on consensus modelling has been used in a Canadian
13 exercise screening the DSL³¹ (Arnot *et al.*, 2005). The approach needs to be
14 further investigated for its usefulness in relation to the (REACH) P assessment
15 and should be used with care and sufficient justification.

16 For specific classes of substances it may also be possible to run specific QSARs. For
17 example BIOHCWIN based on hydrocarbons (Howard *et al.*, 2005), and other models
18 based on alcohols (Yonezawa and Urushigawa, 1979a), *n*-alkyl phthalates (Yonezawa
19 and Urushigawa, 1979b), chlorophenols and chloroanisoles (Banerjee *et al.*, 1984), *para*-
20 substituted phenols (Paris *et al.*, 1983), and *meta*-substituted anilines (Paris *et al.*,
21 1987).

22 The use of QSAR model predictions are of particular relevance and interest when test
23 data are lacking and in addition when assessing multi-constituent substances for which it
24 may often be difficult to find or even to generate test data on relevant individual
25 constituents (including impurities) due to analytical, practical and cost implications.

26 For prediction of hydrolysis there are also some freely available models. The Syracuse
27 Research Corporation's Estimation software (EPISuite) includes also the HYDROWIN
28 program to estimate hydrolysis half-life. Another useful but not freely available program
29 for estimation of hydrolysis is SPARC (<http://www.archemcalc.com/sparc.html>).

30 For prediction of photolysis the Syracuse Research Corporation's Estimation software
31 includes the AOPWIN program, which calculates the indirect photolysis half-life in the
32 atmosphere by reactions with OH⁻ and NO₃⁻ radicals.

33 A photodegradation model is also available in Multicase.

34 Please note that the above list of models is not exhaustive. In any case, the end-user
35 should always assess the validity and the applicability of the models before using them.

36 **Testing data on degradation/biodegradation**

37 Physico-chemical data

³¹ DSL: Domestic Substance List which is a comprehensive inventory of known substances in Canadian commerce (past and current) and currently includes approximately 24,000 substances.

1 The interaction of a substance with the environment is an important consideration. The
2 fate and behaviour of a substance is largely governed by its inherent physico-chemical
3 properties. Knowledge regarding the physico-chemical properties of the substance
4 enables the most appropriate abiotic degradation and biodegradation tests to be
5 identified. These data together with multimedia fate and transport models will also
6 enable higher tiered tests to be prioritized accordingly. Information on the following
7 physico-chemical properties determined using the relevant OECD technical guidelines is
8 desirable: vapour pressure, water solubility, adsorption - desorption using a batch
9 equilibrium method, dissociation constants in water, partition coefficient (n-
10 octanol/water, K_{ow}), and estimation of the adsorption coefficient (K_{oc}). Additional
11 information is provided in Section R.7.1.

12 For substances for which experimental data on partition coefficients ($\log K_{ow}$, $\log K_{oa}$ and
13 $\log K_{aw}$) are not available, estimation methods based on QSAR models based sub-
14 structure fragment methods may be used if the model used can be shown to be valid for
15 the substances. If the substance has (a) functional group(s) or other structural features
16 not represented in the training set of the model, and for which no fragment coefficient
17 was developed, the predictions may be misleading.

18 Abiotic degradation data

19 Abiotic processes such as hydrolysis, oxidation and photolysis may transform substances
20 in aquatic environments, soil and air. Abiotic transformation can be an important step in
21 the pathway for degradation of substances in the environment (OECD, 2006b). The
22 following guideline exists to assess abiotic degradability:

23 OECD TG 111: Hydrolysis as a function of pH

24

25 There are various draft or adopted US EPA and OECD guidelines concerning photolysis.
26 These are (1) Phototransformation of substances on soil surfaces (OECD, 2002a) and (2)
27 Phototransformation of substances in water by indirect photolysis (US EPA OPPTS
28 835.5270) from 1998. There is an additional guideline on how to assess the direct
29 photolysis of substances in water (OECD TG 316 and US EPA OPPTS 835.2210 from 2008
30 and 1998, respectively).

31 For substances for which experimental data on abiotic degradation are not available,
32 QSARs may be considered to derive rates or estimates of degradation (see above).

33

34 Biodegradation data

35 In general, the assessment of degradation processes should be based on data, which
36 reflect the environmental conditions as realistically as possible. Data from studies where
37 degradation rates are measured under conditions that simulate the conditions in various
38 environmental compartments are preferred. The applicability of such data should,
39 however, be judged in the light of any other degradation data including results from
40 screening tests. Most emphasis is put on the simulation test results but in the absence of
41 simulation test data, approximate values for generic degradation rates and half-lives
42 have to be estimated from screening test data, e.g. for calculation of environmental fate

1 and exposure as described in Chapter R.16 of the [Guidance on IR&CSA](#). Listed below are
2 the OECD guidelines to assess biodegradability:

- 3 • OECD TG 301: Ready Biodegradability

4 A: DOC Die-Away Test

5 B: CO₂ Evolution Test

6 C: Modified MITI Test (I)

7 D: Closed Bottle Test

8 E: Modified OECD Screening Test

9 F: Manometric Respirometry Test

- 10 • OECD TG 302: Inherent Biodegradability:

11 A: Modified SCAS Test

12 B: Inherent Biodegradability: Zahn-Wellens/EMPA Test

13 C: Inherent Biodegradability: Modified MITI Test (II)

- 14 • OECD TG 303: Simulation Test - Aerobic Sewage Treatment

15 A: Activated Sludge Units

16 B: Biofilms

- 17 • OECD TG 304A: Inherent Biodegradability in Soil

- 18 • OECD TG 306: Biodegradability in Seawater

- 19 • OECD TG 307: Aerobic and Anaerobic Transformation in Soil

- 20 • OECD TG 308: Aerobic and Anaerobic Transformation in Aquatic Sediment
21 Systems

- 22 • OECD TG 309: Aerobic Mineralisation in Surface Water - Simulation
23 Biodegradation Test

- 24 • OECD TG 310: Ready Biodegradability - CO₂ in sealed vessels (Headspace
25 Test)

- 26 • OECD TG 311: Anaerobic Biodegradation of Organic Compounds in Digested
27 Sludge - Method by Measurement of Gas Production

- 28 • OECD TG 314: Simulation Tests to Assess the Biodegradability of Chemicals
29 Discharged in Wastewater

30 [Appendix R.7.9–1](#) contains a list of the ISO and OPPTS tests that are equivalent to the
31 OECD guidelines listed above. This chapter also lists some of the important attributes of
32 each test.

33 The existing methods for testing ready biodegradability (OECD TG 301 series and OECD
34 TG 310) and the endpoints evaluated are compiled in Section [R.7.9.4](#). It is important to
35 recognise that not all of these test guidelines are equally applicable to all types of
36 substances. Difficulties may especially occur during tests on substances which have low
37 water solubility, high volatility or adsorbing properties. The applicability of the ready

1 biodegradability tests for poorly water soluble, volatile and adsorbing substances has
2 been summarised by the OECD (2006).

3 In 2008, the OECD published OECD TG 314. This test guideline aims to allow checking of
4 the fate of a substance on its way through the sewer system and sewage treatment
5 plant to the mixing zone in surface water. It comprises the following five different
6 component guidelines:

- 7 • Sewer System, OECD 314A
- 8 • Activated Sludge, OECD 314B
- 9 • Anaerobic Digester Sludge, OECD 314C
- 10 • Mixing Zone for Treated Effluent and Surface Water, OECD 314D
- 11 • Mixing Zone for Untreated Wastewater and Surface Water, OECD 314E

12 Up to now OECD TG 314 has seldom been used and so there is little regulatory
13 experience of it under REACH. The applicability of these new proposed guidelines for
14 quantitative environmental exposure assessment requires further discussion. They
15 cannot be used on their own for PBT/vPvB assessment and may only be considered as a
16 part of a weight-of-evidence approach. These studies are neither a screening study nor
17 equivalent to a simulation study on degradation in the environment. They do not employ
18 relevant environmental conditions for assessing the persistence of the substance in the
19 compartments relevant for the PBT/vPvB assessment, i.e.: natural surface water,
20 sediment or soil. Furthermore, they are also not relevant for classification & labelling,
21 because they provide information neither on ready biodegradability nor on degradation
22 rates in individual environmental compartments (i.e. natural surface water, sediment or
23 soil).

24 Non-standard published biodegradation studies

25 In addition to the standardised data described above there is a vast amount of non-
26 standardised biodegradation data that has been published in the scientific literature.
27 Many of these studies share some common principles with the ready biodegradability
28 tests, for example the test substance is usually introduced to the microorganism or
29 microbial community as the sole source of carbon for growth and energy. There is a
30 general reluctance to use these types of data for regulatory purposes. However, they
31 may be valuable, as part of a *Weight of Evidence* assessment, and attempts should be
32 made to gather, evaluate and when appropriate use them.

33 **R.7.9.3.2 Field data on degradation/biodegradation**

34 The ultimate verification for an environmental risk assessment is to measure substance
35 concentrations or removal in the environment (e.g. Fox *et al.*, 2000). Monitoring data
36 can be used directly as exposure data for risk assessment but also to refine input data in
37 the exposure models, e.g. the biodegradation rates. Available information from suitable
38 and reliable field studies or monitoring studies should be considered in a weight-of-
39 evidence approach for the assessment of persistence in the PBT/vPvB assessment.

1 When monitoring data are considered in the risk assessment of substances, the data are
2 often obtained from existing monitoring programmes. In that case the field or monitoring
3 study has not specifically been designed to fulfil regulatory needs. In such cases extra
4 care should be given to the selection of relevant data. When field studies or monitoring
5 campaigns are specially designed to fulfil regulatory needs of REACH the monitoring
6 studies can be designed and implemented accordingly. It must be noted that monitoring
7 data can be required under REACH only as a result of a substance evaluation. For the
8 use of existing and the generation of new field data attention should be given to
9 following aspects:

- 10 • reliable and representative data should be selected by evaluation of the
11 sampling and analytical methods employed and the geographic and time
12 scales of the monitoring campaigns. As sampling and measurements are
13 usually performed at a local geographical area, a justification is required to
14 demonstrate that measured substance concentrations are representative for
15 the risk assessment, particularly if the data are to be used in regional
16 exposure models.
- 17 • the data should be assigned to local or regional scenarios by taking into
18 account the sources of exposure and the environmental fate of the substance.
- 19 • the measured data should be compared to the corresponding calculated PEC.
20 For naturally occurring substances background concentrations have to be
21 taken into account. For risk characterisation a representative PEC should be
22 decided upon based on measured data and a calculated PEC.

23 In the risk assessment of substances a cautious approach is followed. This means that
24 PECs are computed for a relevant scenario that describes usually the worst-case (but still
25 realistic) situation. A common quantification of a vulnerable situation is a combination of
26 geochemical scale and parameters, time scale and climate that results in the 90th
27 percentile PEC. An example of this approach for surfactants in surface water is described
28 by Feijtel *et al.* (1999). This approach is also used in environmental risk assessment for
29 pesticide registrations (European Commission (2014) and EFSA (2015)).

30 Sewage treatment plants

31 Monitoring in sewage treatment plants can be very useful. The endpoint usually is a
32 percentage of removal during the residence time in the sewage treatment plant. Also for
33 the determination of metabolites monitoring the sewage treatment plant (STP) is a good
34 tool. Monitoring in STP's is usually not expressed as a biodegradation rate as removal
35 due to degradation and/ or sorption to sludge solids is usually not resolved. Publications
36 on monitoring in STP's include Morral *et al.* (2006), Eadsforth *et al.* (2006) and Belanger
37 *et al.* (2006).

38 Surface water mesocosms.

39 A mesocosm is a controlled field experiment. Although the primary endpoints of this
40 study are the effects on aquatic organisms, it is possible to obtain information on the
41 fate of substances at the same time. The system is usually closed, and spiked with the
42 substance under realistic outdoor conditions, with representative flora and fauna
43 included. OECD (2006a) provides guidance for the set-up of microcosm and mesocosm
44 experiments.

1 For the marine environment no such guidance document exists, but the IOCCP
2 (International Oceans Carbon Coordination Project) noted that there was an immediate
3 need to develop guidelines and protocols for mesocosm experiments, and is pulling
4 together appropriate scientists from different research programs to develop these.
5 <http://www.unesco.org/new/en/natural-sciences/ioc-oceans/>.htm. The TGD (2003)
6 indicates that the same rules as for fresh surface water should apply for seawater.
7 Relevant literature includes Grice & Reeve (1982), Lauth *et al.* (1996), Culp *et al.* (2000)
8 and Deneer *et al.*, (2015).

9 Large-scale monitoring studies have been performed for surfactants. These monitoring
10 studies are generally focussing on improvement of PNEC's or better estimates of PEC's
11 instead of better estimates of biodegradation rates. An overview of methods, fate and
12 risk assessment for surfactants is given in Knepper *et al.* (2003).

13 Soil and groundwater

14 Three types of field data can be distinguished for soil and groundwater.

- 15 • Lysimeter studies
- 16 • Field studies
- 17 • Monitoring studies

18 Lysimeter studies can be compared with mesocosm studies. They are closed, controlled,
19 outdoor systems, making it possible to use radiolabelled substances and to study the
20 mass-balance. Field studies are semi-controlled, because the system is not closed, the
21 mass-balance cannot be checked, so loss of substance is more undefined than compared
22 to lysimeter studies. In monitoring studies, even more uncertainties arise, because the
23 exposure of the compartment is not under control and the system is not closed.

24 Especially for pesticides many lysimeter, field and monitoring experiments have been
25 performed. Guidance for the performance and evaluation of these studies, aiming at risk
26 assessment in soil and groundwater is given by OECD (2000a), Verchoor *et al.* (2001)
27 and Cornelese *et al.* (2003). The following references may be considered in order to
28 assess dissipation and degradation in the soil compartment: NAFTA (2006), EFSA (2014)
29 and OECD (2016).

30

31 **R.7.9.4 Evaluation of available information on degradation/ biodegra-** 32 **dation**

33 **R.7.9.4.1 Data on degradation/biodegradation**

34 **Non-testing data on degradation/biodegradation**

35 QSAR calculations

36 Chapter R.6 (QSARs and grouping of chemicals) of the [Guidance on IR&CSA](#) provides
37 general recommendations for assessing which QSARs may be suitable for regulatory
38 purposes.

1 Templates for the transparent documentation of the extent of validation of the models
2 (QSAR Model Reporting Format (QMRF)) as well as for reporting information relevant for
3 judging the reliability of predictions for individual substances (QSAR Prediction Reporting
4 Format (QPRF)) have also been developed. A QMRF displays a description of the QSAR
5 model relative to the five OECD QSAR validation principles in a systematic and
6 summarised way (OECD 2006c). A QPRF should show how a prediction of an individual
7 endpoint for a substance relates to the applicability domain of the QSAR model used. It
8 may furthermore contain test data information on the endpoint on close structural
9 analogues to the substance that the prediction is made for. In that case it also describes
10 how closely related the analogues are to the substance that the prediction is made for.

11 A QMRF inventory is available at <http://qsar.db.jrc.it/qmrf/>.

12 QSAR prediction for ready biodegradability

13 An overview of existing validations of a range of the most frequently used QSAR models
14 for prediction of ready/not ready biodegradability is given in Pavan & Worth (2006).

15 One example on the use of QSAR models for predicting ready biodegradability is the
16 BIOWIN models, which estimate biodegradation of discrete organic substances.
17 According to the Biowin helpfile, the criteria for an *overall* YES or NO prediction are as
18 follows: If Biowin3 (ultimate survey model) result is "weeks" or faster (e.g. days or days
19 to weeks) ≥ 2.75 AND Biowin5 (MITI linear model) ≥ 0.5 , then the prediction is YES
20 (readily biodegradable). If this condition is not satisfied, the prediction is NO (not readily
21 biodegradable) according to this proposal for drawing an overall *Weight of Evidence* -
22 based conclusion (EPISuite ver. 3.12, 2004). The acceptability of this generic *Weight of*
23 *Evidence* -based criterion has until now not been considered in the EU working groups
24 dealing with hazard and risk assessment.

25 Another example of a Weight of Evidence procedure that has been used is the TGD
26 (2003) QSAR based screening criterion for identifying substances for persistence (P and
27 vP). BIOWIN 2 < 0.5 or BIOWIN 6 < 0.5 and BIOWIN 3 < 2.25 (- 2.75), i.e. for substances
28 fulfilling this algorithm but BIOWIN 3 indicates a value between 2.25 and 2.75 more
29 degradation relevant information is generally warranted in relation to the PBT testing
30 strategy according to the working practices of the EU PBT Expert Group (cf. TGD (2003)
31 and EU Working Group on Substances of very High Concern (Working document: SHC/TS
32 2-3/029 2002) and Table R.11—4: Screening information for P, vP, B, vB and T in
33 *Chapter R.11* of the [Guidance on IR&CSA](#).

34

35 In general the following freely available BIOWIN models may be used when predicting
36 the ready biodegradability of substances BIOWIN1, 2, 3, 5 and 6. It is noted that
37 according to various validation studies performance of the models seem to differ, but in
38 general predictions about no ready biodegradability seem to be more certain than
39 predictions about ready biodegradability (GHS 2004 & OECD 2004:
40 (ENV/JM/TG/2004)26Rev1 and references therein). However, in some particular cases
41 arguments may be provided for using also ready biodegradability predictions for
42 regulatory decisions (e.g. when many valid individual QSAR model predictions supported
43 by read-across considerations indicate ready biodegradability). The prediction value cut-
44 off points between ready and not ready biodegradability predictions relative to the

1 particular BIOWIN model is indicated in the table. These cut-off points were used in a
2 comparison of 177 high production volume (HPV) chemicals in relation to biodegradation
3 test data compared with model predictions by the shown QSAR models (OECD 2004:
4 (ENV/JM/TG/2004)26Rev1) but the same cut-off points have been used in the past in a
5 range of validations studies ([Table R.7.9–2](#)).

6

7

8 **Table R.7.9–2 QSAR Cut off Points between Ready and Non-Ready**
9 **Biodegradability**

QSAR model	Probability cut off point	Reference:
BPP1 (BIOWIN1, linear) BPP2 (BIOWIN2, non-linear)	0.5	Howard <i>et al.</i> (1992); Boethling <i>et al.</i> (1994); and TemaNord (1995)
BPP3 (BIOWIN3)	2.75	Boethling <i>et al.</i> (2004)
BPP5 (BIOWIN5, linear) BPP6 (BIOWIN6, non-linear)	0.5	Roije <i>et al.</i> (1999); Tunkel <i>et al.</i> (1999); and Boethling <i>et al.</i> (2003)
DK BioDEG (Case Ultra, Leadscope, SciQSAR and majority vote prediction)	yes/no	http://qsar.food.dtu.dk

10

11 Generally it is only recommendable to use single QSAR model predictions when these are
12 clearly within the applicability domain of the model. Whether this is the case may not
13 always be easy to conclude. For BIOWIN models the structural domain can be checked
14 *manually* by checking whether or not a prediction on the individual substance was
15 exclusively based on sub-structures known to the model or whether the substance also
16 contained sub-structures unknown to the model. It is noted that the BIOWIN models will
17 always return predictions even for substances which only contain sub-structures that are
18 unknown to the particular BIOWIN model (i.e. not represented in the training set of
19 substances for the model) but those predictions may be highly unreliable. This is due to
20 the fact that the BIOWIN models then predicts a probability of biodegradability which is
21 solely related to the molecular mass of the substance (i.e. the greater the molecular
22 mass (MM), the lower the value assigned to the probability score, i.e. the larger the MM
23 the higher the predicted likelihood for not being rapidly or readily biodegradable).

24 This implies that checking of whether predictions are within the applicability domain of
25 BIOWIN models may be particularly important. Contrary to this, Multicase, Case Ultra,
26 Leadscope, SciQSAR and CATALOGIC models all include more automated features for
27 checking whether the individual predictions they make are within the applicability
28 domain of the model (it should be noted that each model has its own separate specific
29 way of defining its applicability domain). For Multicase and Case-Ultra models the
30 program contains, for example, possibilities to pre-define the structural domain by use of

1 statistically defined criteria. However, different possibilities exist for defining how
2 stringent such definitions of the applicability domain are (see further information e.g. in
3 the QMRFs available on the DK QSAR prediction database website). The applicability
4 domain of CATALOGIC models is based on the multi-layer concept including general
5 parametric sub-domain, structural sub-domain and mechanistic sub-domain (Dimitrov *et*
6 *al.*, 2005). CATALOGIC software provides also transparent interpretation why a
7 prediction is classified in or out of domain (Dimitrov *et al.*, 2011). QMRF and QPRF are
8 accompanying CATALOGIC predictions.

9 When using model predictions from several QSAR models, e.g. in a *Weight-of-Evidence*
10 approach, it is important to assess the reliability and relevance of each individual model.
11 Another aspect to consider is the extent to which the training sets of the different
12 models do or do not *overlap* (see further details in OECD, 2004 –
13 ENV/JM/TG(2004)26Rev1 where different types of *Weight-of-Evidence* approaches
14 referring to BIOWIN 1, 2, 5 and 6 model predictions have been exemplified and
15 discussed). In this context it has to be considered that training sets for various QSAR
16 models often overlap to some or even significant extent. However, even when the
17 training sets are identical, the application of different modelling approaches may in some
18 cases provide different results (e.g. examples of this in the predictions by the Case Ultra,
19 Leadscope and SciQSAR included in the Danish QSAR database). This is because the
20 different modelling concepts differ in how they integrate the training set information.

21 Borderline predictions which are close to the cut-off between ready and not ready
22 biodegradability should be interpreted with caution. It has for example been proposed to
23 not use BioWIN 1, 2, 5, 4 and 6 model predictions with a biodegradability probability
24 score between 0.4 and 0.6 (because the cut-off point between ready and not ready
25 biodegradability is 0.5). Such a strategy is supported by an analysis done by RIVM on
26 the SIDS data set included in OECD 2004, ENV/JM/TG(2004)26Rev1 to increase the level
27 of predictability (Rorije, 2005).

28 QSARs for abiotic degradation

29 The HYDROWIN model estimates aqueous hydrolysis rate constants for a limited number
30 of different substance classes: esters, carbamates, epoxides, halomethanes and selected
31 alkyl halides (US EPA 2004). The SPARC model can alternatively be used for estimating
32 hydrolysis half-lives.

33 The EPISuite program package includes a model for estimating indirect photo-oxidation.
34 The Atmospheric Oxidation Program for Microsoft Windows (AOPWIN) estimates the rate
35 constant for the atmospheric, gas-phase reaction between photochemically produced
36 hydroxyl radicals and organic substances in the atmosphere (12 hours daylight is
37 assumed). It also estimates the rate constant for the gas-phase reaction between ozone
38 and olefinic/acetylenic compounds. The rate constants estimated by this program can be
39 used to calculate atmospheric half-lives for organic substances based upon average
40 atmospheric concentrations of hydroxyl radicals and ozone. The prediction of the
41 atmospheric degradation half-life of substances in the gaseous phase may be useful for
42 assessing their potential for long-range environmental (primarily atmospheric) transport.

43 CATALOGIC includes 3 models for abiotic degradation. For these 3 models predictions
44 are accompanied with applicability domain, QMRF and QPRF:

- 1 - CATALOGIC Abiotic 301C model simulates aerobic degradation under MITI I
2 (OECD TG 301C) test conditions in the absence of inoculum. The model predicts
3 the quantities (mol/mol parent) of the parent substance and its transformation
4 products at 28th day. Sequence of abiotic transformations, half-lives (primary and
5 ultimate) are also predicted. A training set of 252 experimental data was used to
6 parametrise the model.
- 7 - CATALOGIC Neutral hydrolysis model [2] simulates hydrolysis of discrete organic
8 substances at neutral pH (6.5-7.4), for temperatures of 20-35°C and atmospheric
9 pressure. Kinetic data for 1121 substances is used to parametrise the model. The
10 model predicts quantities (mol/mol parent) of parent and products as a result of
11 hydrolysis and hydrolysis rate constant (d^{-1}).
- 12 - CATALOGIC Acidic hydrolysis model simulates hydrolysis of discrete organic
13 substances at acidic medium (pH 2), temperature 40 °C and atmospheric
14 pressure. Kinetic data for about 500 substances were used to parametrise the
15 model. The model predicts quantities (mol/mol parent) of parent and products as
16 a result of hydrolysis, acid-catalyzed hydrolysis rate constants and half-life.

17 SAR evaluation

18 Besides QSARs, application of other Structure-Activity Relationships (SARs) that are
19 based on qualitative information is also possible.

20 The general characteristics/profile of a substance may give a first indication of the
21 degradation possibilities. A large number of chemical substances are not completely
22 stable, but have certain reactivity potential. By time or by influence of environmental
23 factors, the substance may undergo transformations, which lead to structural changes.
24 In collecting and reviewing existing information on degradation characteristic of the
25 substance, information on possible transformation properties is important.

26 Even if biological processes accelerate the transformation of some simple inorganic
27 substances they may not normally degrade biotically and consequently biodegradability
28 testing of inorganic substances is not worth doing. The inorganic substances may
29 dissociate in the environment (like water soluble salts) or undergo other transformation
30 reactions (atmospheric oxidation, photo-oxidation, hydrolysis, slow biomethylation etc.)
31 that may change the character or magnitude of environmental hazards or risks. The rate
32 of these transformations may be fast, indicating remarkable instability of the substance
33 under certain conditions. For unstable substances, the character of instability and the
34 rate of transformation and transformation products (to other substances) need to be
35 described to estimate hazards and fate of the substance properly. If no test data are
36 available, the rate of transformation needs to be described to some extent, i.e. the
37 expected order of magnitude of rate of transformation at specified conditions ($t_{1/2}$ =
38 minutes, days or weeks?). In addition, one of the key issues is how relevant the
39 qualitative and temporal conditions, in which the substance is unstable, are for typical
40 use and/or emission scenario situations.

41 Organic substances may contain structures that indicate a rapid biotic degradation or on
42 the contrary that the substance is recalcitrant. For example, some organics may often be
43 degradable (e.g. fatty acids), while other types of organics often are recalcitrant (e.g.
44 multi-branched alkyl structures). See further details in OECD (1993).

1 Annex XI, Section 1.5. of the REACH Regulation introduces the concept of read-across
2 (see also [ECHA Read-Across Assessment Framework](#) (RAAF)). This concept is based on
3 the identification of similar substances. Information for one or more reference
4 substances may be used to make a prediction for the target substance. According to that
5 annex, the similarities between the reference substance(s) and the target substance may
6 be based on:

7 (1) *“a common functional group;*

8 (2) *the common precursors and/or the likelihood of common breakdown products via*
9 *physical and biological processes, which result in structurally similar chemicals; or*

10 (3) *a constant pattern in the changing of the potency of the properties across the*
11 *category.”*

12 That annex also specifies that in order to be acceptable, the results derived from a read-
13 across approach should:

14 - *“be adequate for the purpose of classification and labelling and/or risk*
15 *assessment,*

16 - *have adequate and reliable coverage of the key parameters addressed in the*
17 *corresponding test method referred to in Article 13(3),*

18 - *cover an exposure duration comparable to or longer than the corresponding test*
19 *method referred to in Article 13(3) if exposure duration is a relevant parameter, and*

20 - *adequate and reliable documentation of the applied method shall be provided.”*

21 Therefore, if adequate evidence exists, read-across or category approaches may in
22 principle be considered for assessing the environmental fate and pathway properties,
23 including degradability of a substance. This has been the case, e.g, in the PBT/vPvB
24 assessment of several substances included to the Candidate List of Substances of Very
25 High Concern³²

26 More guidance is available in Chapter R.6 (QSARs and grouping of chemicals) of the
27 [Guidance on IR&CSA](#).

28 **Testing data on degradation/biodegradation**

29 Abiotic degradation

30 Hydrolysis

³² See for example the degradation assessments in the Support Documents of UV-327, UV-350 and PFNA:

<https://echa.europa.eu/candidate-list-table/-/dislist/details/0b0236e1808db547>

<https://echa.europa.eu/candidate-list-table/-/dislist/details/0b0236e1808db5e2>

<https://echa.europa.eu/candidate-list-table/-/dislist/details/0b0236e1808db499>

1 Abiotic hydrolytic transformation of substances in aquatic systems may be examined at
2 pH values normally found in the environment (pH 4-9) by use of the guideline:
3 Hydrolysis as a Function of pH (OECD 111). This method is generally applicable to
4 chemical substances (¹⁴C-labelled or non-labelled) for which an analytical method with
5 sufficient accuracy and sensitivity is available. The test is conducted at 3 different pH
6 values (pH 4, 7 and 9). The results of a test of hydrolysis may include (OECD, 2006b):

- 7 • Repeatability and sensitivity of the analytical methods;
- 8 • Recoveries;
- 9 • Mass balance during and at the end of the study (when ¹⁴C-labelled test
10 substance is used);
- 11 • Half-life or DT₅₀.

12 Most hydrolysis reactions follow apparent first order reaction rates and, therefore, half-
13 lives are independent of the concentration. This usually permits the extrapolation of
14 laboratory results determined at high concentrations to low environmentally realistic
15 concentrations. The specific reporting requirements for the hydrolysis test are described
16 below.

17 Temperature dependence of hydrolysis

18 In general, the hydrolysis reactions are relatively sensitive to temperature. Reliable
19 extrapolation of hydrolysis rates from higher to lower temperature (e.g. from 25°C to
20 10°C) may contain remarkable uncertainties (OECD, 2004; Lyman *et al.*, 1990).
21 Temperature dependence of hydrolysis reactions can be reliably determined only by
22 testing the reaction rate at a number of temperatures. The OECD TG 111 on hydrolysis
23 points out that higher tier (tier 2) hydrolysis tests should be carried out with a minimum
24 of three temperatures and preferably at least one temperature below the standard
25 reporting temperature of 25°C. The temperature dependence of hydrolysis reactions
26 reflects to the intrinsic activation energy of the reaction that is taking place. The higher
27 the activation energy is, the slower is the relative rate of hydrolysis at reduced
28 temperature. In practice, temperature dependence of the activation energy is specific for
29 each substance and reaction, leading to variable reaction rates between substances at
30 reduced temperature compared to standard reporting test temperature (25°C).

31 High extrapolation uncertainties can be best avoided by selecting appropriate testing
32 temperatures. For the PBT/vPvB assessment purposes, the testing temperature of 12°C
33 is required for tier 2 testing purposes³³.

34 However, a rough hydrolysis temperature correction estimate may be done by using the
35 following equation: $t_{1/2}(X^{\circ}\text{C}) = t_{1/2} e^{(0.08(T - X))}$. This equation uses "fixed" activation
36 energy (ca. 54 kJ/mol) for all hydrolytic reactions and for all substances. This equation
37 results in a fixed 1.5 fold change in hydrolysis rate per 5°C change in temperature.

³³ Please note that 12°C is at present considered by authorities as the mean temperature of European surface waters and is required by the ECHA Member State Committee to be used as the testing temperature for new simulation degradation tests.

1 Modifications to the hydrolysis test conditions

2 At screening level, priority should be given to test results applying standard test
3 methods. However, quite often modifications to standard methods are needed to
4 overcome testing difficulties, but basically these test modifications should not have
5 influence on the observed degradation rates. For instance in highly modified test
6 systems, surface-controlled reactions can predominate over bulk solution hydrolysis
7 (reflecting rather soil than aquatic environment). The highly modified systems may
8 result in different degradation rates compared to those that would be obtained from
9 standard guidelines using homogeneous solutions.

10 Typically very dilute solutions and a relatively low temperature are the prevailing
11 environmental conditions. Attention is needed to interpret whether the test conditions,
12 e.g. test temperature and test substance concentration have had such an influence on
13 the test results that reliable extrapolation to environmental conditions may be possible.
14 If the abiotic transformation is likely to be reversible in the environmental conditions, the
15 relevance of the transformation observed in an experimental study must be carefully
16 interpreted to determine whether results can be used for the persistence assessment.

17 For example, unnecessarily high temperature, high concentrations of the test substances
18 and buffer should be avoided, as reaction mechanisms may be heavily influenced by the
19 test concentrations, pH and the test medium chemistry, and the temperature. Dissolved
20 organic carbon and adsorption processes could affect hydrolysis rates as well.

21 Phototransformation

22 Phototransformation is not a standard information requirement of REACH. However, the
23 potential effects of solar irradiation on the fate of substances in surface water may be
24 examined by use of OECD guideline 316 Phototransformation of Chemicals in Water –
25 Direct Photolysis (OECD, 2008) and for soil and surface water, respectively, by using the
26 draft guidelines on Phototransformation of Chemicals in Water – Direct and Indirect
27 Photolysis (draft August 2000) and on Phototransformation of Chemicals on Soil Surfaces
28 (draft January 2002). Other guidelines are available for further guidance, e.g. Guideline
29 for Testing of Chemicals, Draft Document, "Phototransformation of Chemicals on Soil
30 Surfaces"; adopted January, OECD, 2002; Indirect Photolysis Screening tests (OPPTS
31 835.5270) pp 24, US EPA, 1998; or Direct Photolysis Rate in Water by Sunlight (OPPTS
32 835. 2210), US EPA, 1998.

33 Two types of phototransformation are distinguished. In direct phototransformation, the
34 reacting substance itself directly absorbs light. In indirect phototransformation, another
35 substance absorbs light and transfers the excess energy to an acceptor substance
36 causing this acceptor substance to react.

37 The direct and indirect phototransformation of substances in natural water bodies is a
38 complex process that depends on a number of factors such as:

- 39 • the chemical structure and electronic absorption spectrum of the substance;
- 40 • the concentration, composition, and absorption spectra of chromophoric
41 dissolved organic matter (CDOM; from which photosensitizers and singlet
42 oxygen arise);

- 1 • the concentration of nitrate (the primary source of hydroxyl radicals); and
2 • the solar photon flux spectrum to which the substance, CDOM and nitrate are
3 exposed.

4 Any data on degradation half-lives or DT₅₀, DT₇₅ and DT₉₀ values should be reported
5 along with calculations associated with these data, and the results of any outdoor
6 experiments, if available. Where possible, information on transformation products should
7 be provided as well (OECD, 2006b).

8 The level of information required in the test report depends on the complexity and
9 purpose of the study. Consequently, OECD has identified a number of tiers for direct and
10 indirect photolysis in water (see the relevant guidelines for details, OECD, 2006b).

11 Phototransformation data may be of use for assessing direct photolysis in air. It may
12 also be of use for assessing photolysis in water when factors such as water depth,
13 suspended matter and latitude are taken into account.

14 According to Castro-Jiménez and van de Meent (2011) light absorption in natural water
15 is significantly slower than measured in laboratory water with direct photo degradation
16 occurring around 30 times more slowly for typical fresh water, 400 times more slowly for
17 typical coastal sea water, and 500 times more slowly for ocean water. They also
18 conclude that the contribution of photodegradation in water to overall degradation is
19 significant only for substances that reside in water to a considerable extent. They
20 highlight that many substances reside in sediment and soil, rather than water. They give
21 as an example bromophenyl ethers, which are “photochemically labile in water”, but only
22 slowly photodegrade in the environment. The authors however have not investigated
23 indirect phototransformation and acknowledge that indirect phototransformation is less
24 understood but could be more important than direct phototransformation. Contrary to
25 direct phototransformation, indirect phototransformation is stimulated in natural
26 environmental waters by the presence of Dissolved Organic Matter (which is not present
27 in pure lab water).

28 Biodegradation

29 Ready biodegradability

30 Ready biodegradability tests must be designed so that positive results are unequivocal.
31 Given a positive result in a test of ready biodegradability, it may be assumed that the
32 substance will undergo rapid and ultimate biodegradation under most environmental
33 conditions. In such cases, no further investigation of the biodegradability of the
34 substance, or of the possible environmental effects of transformation products, is
35 normally required. However, the fact that the substance is found to be readily
36 biodegradable does not exclude a possible need for further information about
37 biodegradation rate constants and the transformation products in cases of high influx
38 into a receiving environment in particular when this significantly influences the risk ratio
39 and this is above 1. Realising that ready biodegradability tests may sometime fail
40 because of the stringent test conditions, positive test results should generally supersede
41 negative test results. However, when conflicting test results are reported, possible
42 differences in the test conditions and design should be investigated. In particular the
43 origin of the inocula should be examined in order to verify whether or not there are
44 differences in the adaptation of the inocula which may explain the differences in the

1 results (OECD, 2006b). Results from tests based on adapted inocula are generally
2 regarded as inappropriate for assessment. An inoculum is considered adapted not only if
3 special arrangements were made with the aim to adapt the inoculum to the substance
4 but also if the inoculum used was previously exposed to the substance or structurally
5 similar substances (e.g. in an industrial STP, in a contaminated site or in municipal waste
6 water treatment plants (WWTPs) receiving releases from sites using the substance). For
7 substances that are widely used and continuously emitted to WWTPs, e.g. if they are
8 ubiquitous in consumer products, it is acknowledged that pre-exposure of the degrading
9 microorganisms may not be avoided. In this case, use of such inocula can be acceptable
10 provided that the level of pre-exposure remains low. However, inocula from WWTPs
11 influenced by point sources must not be used, e.g. if effluents from an industrial site
12 using the substance are connected to the municipal WWTP.

13

14 When faced with conflicting results using different ready biodegradability test methods, it
15 is also important to consider the following.

16 • Test substance concentration:

17 - Very high concentrations (100 mg/L) used for some of the OECD 301
18 tests increase the probability of inhibition or mass transfer issues for
19 test substances with a low water solubility.

20 - Very low concentrations (2-5 mg/L) used for the closed bottle test can
21 sometimes overestimate degradation given the poor signal to noise
22 (theoretical vs. background) ratio in such a test.

23 • Inoculum:

24 The pre-treatment of the inoculum such as in the MITI test (OECD 301C) may
25 seriously lower the diversity and biodegradation capacity of the microbes
26 (Forney *et al.*, 2001; Kayashima *et al.*, 2014).

27 Differing results always have to be assessed considering the test conditions, substance
28 properties and reliability of the data. Good data reliability depends on the test method
29 applied, statistical robustness of the study and its reporting which in turn depend on
30 several factors, e.g. number of replicates, and number of controls.

31 Information on the operational conditions of the sewage treatment plant where the
32 inoculum was sampled (e.g. mass loading rate, sludge retention time) and measurement
33 of additional parameters such as DOC, biomass growth and/or carbon balance would
34 facilitate the interpretation of the results.

35 A negative result in a test for ready biodegradability does not necessarily mean that the
36 substance will persist in the environment and will not be degraded under relevant
37 environmental conditions. A ready biodegradability test is only a screening test, and if
38 that test could not demonstrate that the substance is readily biodegradable then further
39 testing under less stringent test conditions should be considered at the next level.

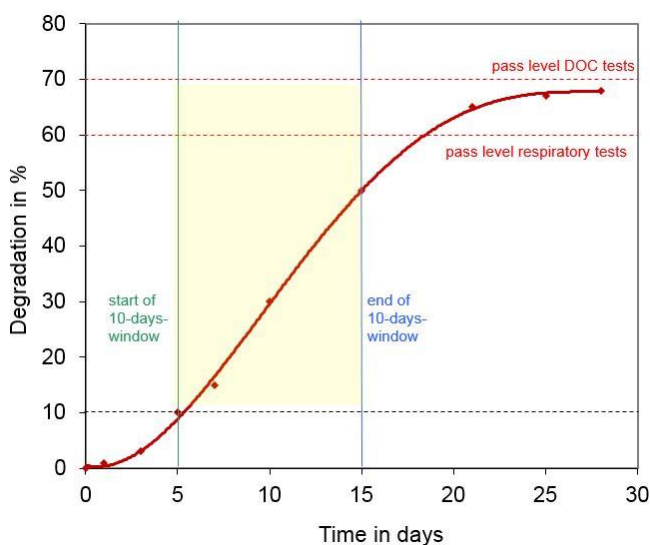
40 The OECD test guidelines which can be used to determine the ready biodegradability of
41 organic substances include the six test methods described in the OECD 301 series of test
42 guidelines and OECD 310. The following pass levels of biodegradation, obtained within

1 28 days, may be regarded as evidence of ready biodegradability: 70% DOC removal (TG
2 301 A and TG 301 E); 60% theoretical carbon dioxide (ThCO₂; TG 301 B); 60%
3 theoretical oxygen demand (ThOD; TG 301 C, TG 301 D and TG 301 F).

4 The pass levels for ready biodegradability mentioned above relate to measured sum
5 parameters for DOC depletion, oxygen use or CO₂ production and implies total
6 degradation (assumes that 30-40 % of the organic carbon of the test substance is either
7 assimilated by the microbial biomass for growth or present as products of biosynthesis).
8 It is worth noting that those assumptions rely on the fact that ready biodegradability
9 tests are conducted with very high concentrations of the test substance providing both a
10 carbon and energy source for the microbes which typically allow them to grow
11 significantly. However, in the environment, the substance concentration would generally
12 be much lower than the concentrations used in ready biodegradability tests and other
13 more easily degradable compounds may be used as a primary carbon and energy source
14 by the microorganisms, in preference to the substance. Thus, a substance which could
15 be degraded under ready biodegradability test conditions may actually not be degraded
16 under environmental conditions as microorganisms may preferentially metabolise
17 compounds on which they can grow faster. The substance will be used as carbon and
18 energy source only after other more easily degradable substrates have been consumed,
19 a phenomenon known as diauxie. Also the opposite situation is possible: substances that
20 do not degrade under ready biodegradability test conditions may be degraded in the
21 presence of another carbon source, via a phenomenon known as co-metabolism. The
22 substance is then a non-growth substrate which is degraded concurrently to another
23 substrate, the primary substrate, which serves as primary carbon and energy source.

24 These pass levels for ready biodegradability have to be reached in a 10-day window
25 within the 28-day period of the test. The 10-day window does not apply to TG 301 C or if
26 the test substance is made of a composition of homologous constituents. The 10-day
27 window begins when the degree of biodegradation has reached 10% DOC removal, ThOD
28 or ThCO₂ and must end before or at day 28 of the test (see [Figure R.7.9–1](#)).

29 The OECD "Guidelines for the Testing of Chemicals, Revised Introduction To The OECD
30 Guidelines For Testing Of Chemicals, Section 3 Part I: Principles And Strategies Related
31 To The Testing Of Degradation Of Organic Chemicals" (OECD, 2006b) indicates that
32 ready biodegradability tests are intended for pure substances and are generally not
33 applicable for complex compositions containing different types of constituents, like
34 UVCB. For an UVCB substance, observed biodegradation may indeed represent the
35 biodegradation of only some constituents. This OECD document indicates that "*it is
36 sometimes relevant to examine the ready biodegradability of mixtures of structurally
37 similar chemicals*". Still "*a case by case evaluation should however take place on
38 whether a biodegradability test on such a complex mixture would give valuable
39 information regarding the biodegradability of the mixture as such (i.e. regarding the
40 degradability of all the constituents) or whether instead an investigation of the
41 degradability of carefully selected individual components of the mixture is required*". This
42 OECD document indicates that the 10-day window need not be applied only if the test is
43 carried out on a mixture of structurally similar constituents and if it is anticipated that a
44 sequential biodegradation of the individual constituents is taking place.



1

2 **Figure R.7.9—1 Pass levels for ready biodegradability**

3

4 Another test for ready biodegradability, which represents an alternative to the CO₂
 5 Evolution Test (OECD 301 B), is the Headspace Test (Ready Biodegradability – CO₂ in
 6 sealed vessels; OECD 310). This test is especially suitable for volatile compounds. In this
 7 test, the CO₂ evolution resulting from the ultimate aerobic biodegradation of the test
 8 substance is determined by measuring the inorganic carbon (IC) produced in sealed test
 9 bottles, and the pass level has been defined as 60% of theoretical maximum IC
 10 production (ThIC).

11 Ready biodegradability tests usually last for 28 days. However, biodegradability tests
 12 may be ended before 28 days, i.e. as soon as the biodegradation curve has reached a
 13 plateau for at least three determinations. Alternatively, they may be prolonged beyond
 14 28 days when the curve shows that biodegradation has started but that the plateau has
 15 not been reached by day 28 (OECD, 1992). Where substances have not achieved the
 16 pass level for ready biodegradability in the 28-day ready biodegradability test duration
 17 the substances are considered to be not readily biodegradable (OECD, 1992).
 18 Substances where mass transfer or substance availability is limited fall into this category
 19 e.g. poorly-water soluble substances. New tests should be conducted in accordance with
 20 the OECD principles for Good Laboratory Practice, and the test report and robust study
 21 summary should include information on how validity criteria were met and the
 22 information identified in Section R.7.9.9 of Chapter R.7c of the [Guidance on IR&CSA](#).

23 There may be a high level of variations in the results for a same substance because of
 24 several criteria imposed in those ready biodegradability tests (low test volumes, lack of
 25 consideration of inoculum quantity and quality, stringent protocol for the preparation of
 26 the inoculum (Goodhead *et al.*, 2014), unrealistic conditions of the test).

27

28 Marine Biodegradability

29 OECD TG 306 series on Biodegradability in Seawater includes seawater variants of the
 30 Closed Bottle Test (OECD 301 D) and of the Modified OECD Screening Test (OECD 301

1 E). Degradation of substances in seawater has generally been found to be slower than
2 that in freshwater tests inoculated with activated sludge or sewage effluent. This is also
3 confirmed in the research program conducted in CEFIC LRI ECO11 project (see report of
4 the workpackage 1), where it was demonstrated that both magnitude and variation in
5 the bacterial diversity were higher in the following order for the different environmental
6 sources: activated sludge > rivers > estuaries > sea water.

7 OECD test guideline 306 explicitly indicates that results of those tests (shake flask and
8 closed bottle) "*are not to be taken as indications of ready biodegradability, but are to be*
9 *used specifically for obtaining information about the biodegradability of chemicals in*
10 *marine environments*". Those tests "*are not tests for ready biodegradability since no*
11 *inoculum is added in addition to the micro-organisms already present in the seawater.*
12 *Neither do the tests simulate the marine environment since nutrients are added and the*
13 *concentration of test substance is very much higher than would be present in the sea*".

14 However, it is acknowledged that biodegradation in seawater is generally slower.
15 Therefore >60% ThOD or >70% DOC removal in a Biodegradability in Seawater test
16 (OECD 306) obtained after 28 day (Closed Bottle Method) or 60 day (Shake Flask
17 Method) is indicative of potential for ultimate biodegradation in the marine environment
18 and can also be regarded as a piece of evidence that the substance is likely to fulfil the
19 criteria for ready biodegradability. For example, a positive OECD 306 test is regarded as
20 an indication of rapid degradation for classification and labelling.

21 A result of >20% ThOD or DOC removal is indicative of a potential for primary
22 biodegradation in the marine environment.

23 Modified Ready Biodegradability Tests

24 Two modifications to the standard ready biodegradability and marine biodegradability
25 tests are identified below. These consider biodegradability testing at low test substance
26 concentrations and assessing the biodegradation of poorly water soluble substances.
27 Provided that all other conditions in the Ready Biodegradability Tests are fulfilled, these
28 tests are regarded as Ready Biodegradability Tests and the results can be used directly
29 for hazard classification.

30 Testing at low test substance concentrations due to inoculum toxicity

31 For substances that are known or expected to exert toxicity to the microbial inoculum at
32 the test substance concentration normally employed in most tests for ready
33 biodegradability a lower test substance concentration should be used, e.g. by selecting
34 the Closed Bottle Test (OECD 301 D). The toxicity of the test substance to
35 microorganisms can be determined using one of a number of microbial toxicity tests e.g.
36 the activated sludge respiration inhibition test (OECD 209). Where possible a lower test
37 substance concentration than is generally recommended by the test guideline/method
38 should still allow the assessment of biodegradability to be determined reliably through
39 the measurement of carbon dioxide evolution, oxygen demand or dissolved organic
40 carbon removal. Reduction in the toxicity in the ready biodegradability tests may also be
41 achieved by the introduction of carriers allowing the 'slow-release' of the test substance
42 during the test period.

43 Conducting studies at low concentrations may only be possible if the test substance is
44 radiolabelled. If this is not possible then the primary biodegradability of the test

1 substance should be measured using specific chemical analysis. If primary degradation is
2 being measured then an attempt should be made to identify any major degradation
3 products.

4 Biodegradability assessments of poorly water-soluble substances

5 The standardised ready biodegradation test methods adopted by the OECD that are
6 listed above were initially developed to evaluate the biodegradability of test substances
7 which are soluble in water to at least 100 mg/L provided they are non-volatile and non-
8 adsorbing. For substances that are poorly soluble in water, volatile or adsorbing the
9 OECD concluded that only a subset of the ready biodegradability test guidelines were
10 applicable ([Appendix R.7.9–1](#)).

11 For poorly-water soluble or adsorptive substances these are the OECD 301B, 301C, 301D
12 and 301F test series and the OECD 310 test. For volatile substances they are the OECD
13 301C, 301D and 301F test series and the OECD 310 test.

14 Tests using DOC analysis cannot be used to assess the biodegradability of poorly water
15 soluble substances unless it is measured in addition to another parameter. Specific
16 chemical analysis can also be used to assess primary degradation of the test substance
17 and to determine the concentration of any intermediate substances formed. Specific
18 chemical analysis is obligatory in the MITI method (OECD 301C; OECD, 1992).

19 Several experimental means for improving the bioavailability of poorly water soluble
20 substances are proposed in Annex III of the OECD TG 301. The use of silica gel matrices
21 is generally seen as the preferred option. Solid carriers are not recommended for solid
22 test substances but may be suitable for oily substances. Emulsifiers or solvent which
23 give a stable dispersion of the test substance may be used, but it should be verified that
24 they are not toxic to bacteria and must not be biodegraded or cause foaming under test
25 conditions. Therefore if solvents or emulsifiers are used, careful consideration of their
26 properties is needed beforehand as well as an additional control. Other strategies to
27 determine the biodegradability of poorly water-soluble substances are described in
28 [Appendix R.7.9–3](#) as well as in ECETOC (2013) and OECD (2000b).

29

30 Enhanced Biodegradation Screening Tests

31 If a substance does fail to reach the pass level for the ready biodegradability, results
32 from other screening tests (enhanced ready tests or tests on inherent biodegradation)
33 may be useful additional testing to show that a substance is not persistent. In some
34 cases it may be justifiable to go directly to an enhanced test design, e.g. if the substance
35 is poorly soluble. Different methods for the enhancement or modification of standard
36 biodegradation tests are available to demonstrate that a substance is not persistent.

37 A number of potential enhancements to the ready biodegradation test have been
38 identified. These enhancements have been proposed for the determination of persistence
39 in vPvB/PBT assessment only but are not to be used for Classification and Labelling and
40 quantitative exposure and risk assessment.

41 These enhancements are designed to help to improve the environmental relevance of
42 biodegradability assessments for persistence assessment only without the immediate

1 requirement for simulation level testing. Details on the potential enhancements
2 described below have been discussed (Gartiser *et al.*, 2016a, 2016b; Koalczyk *et al.*,
3 2014; ECETOC, 2013) and should allow adequate results to be obtained for assessment
4 of persistence. However, they would benefit from being ring-tested by appropriate
5 international standards bodies. Test substances that degrade in these enhanced
6 biodegradation screening tests must not be considered readily biodegradable (unless
7 ready biodegradability in a standard, i.e. without enhancements, ready biodegradation
8 test is shown).

9 With the exception of the MITI I test (OECD 301C), for the current ready biodegradation
10 tests, the inoculum can be obtained from a number of sources as long as it has not been
11 significantly pre-exposed to the test substance, e.g. it is not from a site which was
12 exposed to industrial chemicals. The current ready biodegradability testing approach
13 includes use of inoculum from e.g. municipal STP pre-exposed to substances which are
14 generally continuously emitted to municipal STPs.

15 Inocula from municipal STPs can also be used in enhanced tests.

16 For both ready biodegradability and simulation degradation tests biodegradation depends
17 upon one or more competent micro-organism(s) being introduced into the test flask and
18 these microorganisms being able to establish themselves (and in a ready
19 biodegradability test, grow significantly) under the conditions of the test. For many
20 substances the use of replicate flasks may give rise to high levels of variability and
21 several studies for an identical substance can give different results. The variability of
22 these results is (based on general experience) largely due to differences in the
23 composition of the microbial inoculum introduced into the test flask on day zero.

24 Therefore a test strategy employing enhanced ready biodegradability testing is
25 warranted that can represent a relevant microbial diversity in the test system as long as
26 the microbes do not induce pre-adaptation of the inoculum. It should be noted that the
27 purpose of using enhanced biodegradation screening tests is to confirm a potential for
28 degradation, which can be considered for the assessment of persistence (i.e. PBT and
29 vPvB assessment; see Chapter R.11 of the [Guidance on IR&CSA](#)). These tests, however,
30 do not provide information on ready biodegradability. The purpose of those
31 enhancements should only be to compensate the poor bioavailability to the degrading
32 microorganisms of poorly soluble and/or adsorptive substances, but should not be used
33 to induce additional adaptation of the inoculum.

34 Test approaches in enhanced biodegradation screening tests could include:

- 35 • Test duration – Experimental adaptations described in the above paragraphs
36 on “Ready biodegradability tests” and “Biodegradability assessments of poorly
37 water-soluble substances” are generally regarded as preferable for
38 investigating substances of low bioavailability. However, the prolongation of
39 the test duration can sometimes constitute another possible option, e.g. if
40 during a regular ready biodegradability test degradation is observed but has
41 not reached a plateau within 28 days. For poorly water soluble substance, the
42 poor bioavailability of the substance can indeed limit the degradation rate.
43 The prolongation of the test duration may thus give the microorganisms more
44 time for accessing and degrading the substance. The prolongation of the test
45 duration should only be considered if some initial, slow but steady,

1 biodegradation was observed but did not reach a plateau by the end of the
2 ready biodegradability test, i.e. after 28 days. However a late acceleration of
3 biodegradation is likely to reflect an adaptation of the microorganisms and in
4 that case the prolongation of the test duration should not be regarded as
5 adequate for the P/vP assessment. Furthermore, the test must in any case be
6 terminated within 60 days since it becomes ever more probable that the test
7 system will deteriorate the longer the test lasts. For interpretation of the test
8 results see Section [R.7.9.5](#) and Chapter R.11 of the [Guidance on IR&CSA](#).

- 9
- 10 • Testing in larger vessels – the drive to generate tests that allow rapid and
11 small-scale chemical assessments does not work for biodegradability
12 assessments. At very small test volumes the total number of and the number
13 of different types of microorganisms introduced into the test flask decreases.
14 Conducting biodegradation tests using larger volumes of environmental
15 waters increases the total number of microorganisms introduced into the test,
16 and the number of different types, without changing the density of
17 microorganisms introduced (Ingerslev *et al.*, 2001). This will increase the
probability of introducing a competent microorganism into the test vessel.

18 The following test approaches are not deemed acceptable from the regulatory
19 perspective:

- 20
- 21 • Increasing the biomass concentration - there is already some flexibility of the
22 inoculum concentration given in Ready Biodegradability Tests. Going beyond
23 the limits defined will change the ratio of substance to inoculum in a way that
is deemed to be too favourable.
 - 24 • Pre-adaptation – the use of inoculum from contaminated sites or sites pre-
25 exposed to the test substance before testing starts is not a modification
26 finding regulatory acceptance for persistence testing. This applies also to the
27 practice of conducting a second ready biodegradability test using the inoculum
28 derived from the initial study.
 - 29 • Semi-continuous assessments - which also favour an artificial optimal
30 microbial adaptation to the substance, are not consistent either with the aims
31 of a reasonable worst case assessment of persistence.
 - 32 • Addition of co-substrate(s) – as for the other screening tests, the test
33 substance should be the only carbon source. Therefore, the addition of a co-
34 substrate may cause additional uncertainty due to unspecific sum parameters
35 and an increase of inoculum blanks and is therefore not acceptable.

36

37 Inherent Biodegradability

38 Tests from the OECD 302 series have been developed to determine the inherent
39 biodegradability of organic substances and include three methods: the Modified SCAS
40 Test (OECD 302 A), the Zahn-Wellens/EMPA Test (OECD 302 B) and the Modified MITI
41 Test (II) (OECD 302 C).

1 Biodegradation above 70% of theoretical (measured as BOD, DOC removal or COD) may
2 be regarded as evidence of inherent, ultimate, biodegradability. Inherent
3 biodegradability data may be used directly for the assessment of environmental
4 persistence of the substance (see section [R.7.9.5.2](#) and Chapter R.11 of the [Guidance on](#)
5 [IR&CSA](#)) and for extrapolation to a rate constant in models for estimation of the
6 elimination of substances in STP. However, this extrapolation is only allowed:

- 7 - if the pass level of 70% degradation in the Zahn-Wellens/EMPA Test is
8 reached within seven days, including the lag-phase and the
9 degradation-phase, i.e. the exponential growth phase of the
10 microorganisms. The degradation-phase should be no longer than
11 three days, and the percentage removal in the test before
12 biodegradation occurs should be below 15%. Or
- 13 - if the pass level of 70% in the Modified MITI Test (II) is reached within
14 14 days, including the lag-phase and the degradation-phase, and the
15 degradation-phase is no longer than three days.

16 Biodegradation above 20% of theoretical may be regarded as evidence of inherent,
17 primary, biodegradability and suggests that stable degradation products are likely to be
18 formed. Further testing should then be considered to conclude on the persistence of the
19 substance and of its degradation products.

20 Careful interpretation of data must be performed when considering the use of DOC
21 removal as a degradation sum parameter to ensure that elimination did not occur due to
22 adsorption or volatilisation (both of which are physical removal processes which should
23 not be misinterpreted as transformation or biodegradation). The shape of the
24 degradation curve may give an indication of whether or not a biological degradation
25 process occurred (e.g. very fast initial disappearance of the test substance may indicate
26 that the process was due to volatilisation or adsorption). In certain cases when results of
27 ready biodegradability tests (i.e. OECD 301 series or OECD 310) indicate that the pass
28 level criterion is almost fulfilled (i.e. ThOD or DOC slightly below 60% or 70%
29 respectively) such results can be used as evidence for inherent biodegradability. This is
30 also the case when the pass level criterion is fulfilled but the 10-day window criterion is
31 not.

32

33 Simulation tests

34 Simulation tests aim at assessing the rate and extent of biodegradation in a laboratory
35 system designed to represent either the aerobic treatment stage of STP or
36 environmental compartments, such as fresh or marine surface water or sediments, or
37 soil. (OECD, 2006b).

38 Simulation testing on sewage treatment

39 The fate of substances in STPs can be studied in the laboratory by using the Simulation
40 Tests OECD 303 or OECD 314B. Such tests may sometimes be warranted when an
41 environmental risk assessment of substance needs to be refined and when the exposure
42 scenario indicates that emission to STPs take place. However, they cannot be used on
43 their own for PBT/vPvB assessment and may only be considered as a part of a weight-of-

1 evidence approach. In particular, the half-lives determined from those tests are not
2 suitable for comparison with the REACH Annex XIII criteria for persistence. These studies
3 indeed do not employ relevant environmental conditions for assessing the persistence of
4 the substance in the compartments relevant for the PBT/vPvB assessment, i.e.: natural
5 surface water, sediment or soil. For the PBT/vPvB assessment it has to be demonstrated
6 that the substance will indeed not persist in any of the environmental compartments.
7 Therefore, not only exposure to natural water from STP effluents but also other
8 possibilities of exposure (including indirect exposure and redistribution between
9 environmental compartments) need to be taken into account for the PBT/vPvB
10 assessment.

11 Aerobic Sewage Treatment: Activated Sludge Units (OECD 303 A) and Biofilms (OECD
12 303 B). The removal of the test substance is determined by monitoring the concentration
13 of DOC and/or Chemical Oxygen Demand (COD) in the influent and effluent. The test
14 recommends addition of the test substance at a concentration of DOC between 10 mg/L
15 and 20 mg/L. However, many substances are normally present at very low
16 concentrations, even in waste water, and procedures for testing the biodegradation at
17 suitably low concentrations (<100 µg/L) are presented in Annex 7 to the TG 303 A.

18 Another test usable for this purpose is OECD 314B. The test conditions of OECD 303 and
19 OECD 314B differ and thus results are expected not to be comparable between these two
20 tests. For instance reproducibility of test results is clearly defined in OECD 303 but no
21 such specifications are given in OECD 314B. Another example is the test duration which
22 is limited to 28 days in OECD 314B whereas it is flexible in OECD 303, in which it
23 depends on both the stabilisation phase (it ends if the inoculum removes DOC of the
24 organic medium efficiently) and plateau phase (at least 21 days). If the intended use is
25 to refine the exposure and risk assessments, then these results have to be evaluated
26 case by case to determine whether and for which exposure scenarios they may be
27 relevant and sufficiently representative and reliable.

28 Biodegradation in a DOC based Semi-Continuous Activated Sludge (SCAS) test can
29 normally only be determined when the substance is non-sorptive and non-volatile.
30 However, if a SCAS test is performed with a radiolabelled substance and a mass balance
31 is done on the effluent and solids, then it is possible to determine biodegradation for any
32 type of non-volatile substance. The value of an SCAS test for estimating biodegradation
33 increases when off-gases are trapped for CO₂ and other organic volatiles. However, its
34 value for assessment purposes is low because of the strong potential for adaptation of
35 micro-organisms to the substance in this kind of test.

36 No specific pass levels have been defined for the elimination of substances in aerobic
37 sewage treatment simulation tests (OECD 303). The test results may be used to
38 estimate the removal in STPs and the resulting effluent concentrations for predicting the
39 concentration in the treatment plant and the receiving aquatic environment.

40 The assessment of biodegradability and/or removal in sewage treatment plants should
41 preferably be based on results from tests simulating the conditions in treatment plants.
42 Such a test may be the OECD 303 A test. Data from non-standardised tests and/or tests
43 not performed according to the principles of GLP may be used if expert judgement has
44 confirmed them to be equivalent to results from the standardised degradation tests on
45 which the calculation models, e.g. SimpleTreat, are based. The same applies to STP
46 monitoring data, i.e. in-situ influent/effluent measurements.

1 There is separate endpoint specific guidance for toxic effects of substances on STPs (see
2 Section [R.7.8.20](#)).

3 Simulation testing on soil, sediment and water

4 Simulation studies may be required to refine the persistence assessment for a
5 substance. These studies are considered to be more environmentally realistic than the
6 screening studies.

7 The following tests can be used to simulate the biodegradation of organic substances
8 under environmentally realistic conditions in soil, sediment or surface water: Aerobic and
9 Anaerobic Transformation in Soil (OECD 307); Aerobic and Anaerobic Transformation in
10 Aquatic Sediment Systems (OECD 308); and Aerobic Mineralisation in Surface Water –
11 Simulation Biodegradation Test (OECD 309).

12 Aerated soils are aerobic, whereas water-saturated or water-logged soils are frequently
13 dominated by anaerobic conditions. The surface layer of aquatic sediments can be either
14 aerobic or anaerobic, whereas the deeper sediment is usually anaerobic. These
15 conditions in soil or sediment may be simulated by using aerobic or anaerobic tests
16 described in the test guidelines (OECD 307 and OECD 308).

17 The simulation degradation studies include two types of investigations: a) a degradation
18 pathway study where degradation products (i.e. degradation metabolites) are identified
19 and quantified, b) a kinetic study where the degradation rate constants (and degradation
20 half-lives) of the parent substance and, if applicable, of the degradation products are
21 experimentally determined.

22 Generally, a low concentration of the test substance is used for simulation testing (e.g.
23 from 1 µg/L to 100 µg/L and preferably between 1 and 10 µg/L in TG 309). The test
24 concentration should indeed be low enough to ensure that the biodegradation kinetics
25 (first order or pseudo-first order) obtained in the test reflect those expected in the
26 environment. In simulation tests, the degradation is generally assumed to occur *via* co-
27 metabolism and even if microbial growth occurs it is generally assumed to be limited
28 because the concentration of the test substance will not be high enough for serving as a
29 primary source of energy and carbon in the same way as in the ready biodegradability
30 tests. Higher concentrations of the test substance (e.g., >100 µg/L) can be used only to
31 overcome potential analytical limitations when identifying and quantifying the
32 transformation products.

33 Both radiolabelled and non-labelled test substances can be used. For assessing total
34 mineralisation, a ¹⁴C –labelled test substance is typically used and ¹⁴CO₂ evolution is
35 measured. One should ensure that the ¹⁴C label is located in the most recalcitrant part of
36 the molecule. If a sensitive specific analytical method is available, the primary
37 biodegradation can be assessed by measuring the total residual concentration of the test
38 substance. Disappearance of the parent substance however does not necessarily imply
39 its mineralisation. Chemical analyses can be used in parallel with radiolabelling
40 techniques. Specific chemical analyses can also be used to identify and quantify
41 transformation products.

42 The soil simulation degradation test according to OECD TG 307 includes the
43 determination of the degradation half-lives in 4 different types of soils. The sediment
44 degradation test according to OECD TG 308 includes the determination of the

1 degradation half-lives in 2 different types of sediment. The surface water degradation
2 test according to OECD TG 309 includes the determination of the degradation half-life in
3 at least one surface water sample and at two different concentrations of the test
4 substance. These concentrations should differ from each other by a factor of 5 to 10 and
5 should represent the expected range of concentrations in the environment. They both
6 should be low enough to be below the water solubility limit of the test substance and to
7 ensure that the biodegradation follows first order kinetics.

8 New kinetic simulation studies should be conducted at environmentally relevant
9 temperatures, by default at 12°C, which is regarded as a reasonable alleged average
10 temperature for the European Union. If information on degradation half-life is already
11 available from existing simulation degradation tests performed at a higher temperature,
12 they should be normalised to a half-life corresponding to 12°C by using the Arrhenius
13 equation (see paragraph below named "Temperature correction"). In every case, kinetic
14 results such as the degradation rates and degradation half-lives should correspond to an
15 environmentally relevant temperature, i.e. by default 12°C. For the purpose of
16 identifying degradation products, a higher test temperature (but within the frame
17 provided by the study guideline) could be used to overcome potential analytical
18 limitations for the identification and quantification of those degradation products.

19 The results of simulation tests may include:

- 20 • First order or pseudo-first order rate constant;
- 21 • Degradation half-life or DT50
- 22 • Length of the lag phase
- 23 • Half-saturation constant;
- 24 • Maximum specific growth rate;
- 25 • Fraction of mineralised label, and, if specific analyses are used, the final level
26 of primary degradation;
- 27 • The fraction of non-extractable residues and a justification for the chosen
28 extraction procedure;
- 29 • Mass balance during and at the end of the study;
- 30 • Identification and concentration of major transformation products, where
31 appropriate;
- 32 • A proposed pathway of transformation, where appropriate;
- 33 • Rate of elimination (e.g. for risk assessment purposes)

34 Non-standard published biodegradation studies

35 When judging poorly reported or non-standard data then the following minimum
36 information needs to be available in order to make any use of the data:

- 1 • The source and density of the inoculum, this should not be taken from an
2 industrial site and the density should be equivalent to that of a ready
3 biodegradation test
- 4 • Any pre-treatment of inoculum including pre-exposure to the test substance
- 5 • The test substance, its purity and the concentration that is used in the test
- 6 • The motivation for the study
- 7 • The analyte being measured (parent compound, DOC, BOD or CO₂ evolution)
- 8 • Details regarding the biochemical pathway for degradation if available
- 9 • Either a degradation rate or a removal percentage; in the latter case it needs
10 to be considered whether or not the removal only reflects distribution
11 processes like adsorption or volatilisation.

12 Reporting biodegradation studies

13 FOCUS (2014) makes a distinction between biodegradation endpoints used as a trigger
14 for higher tier studies (trigger endpoint) and biodegradation endpoints used in
15 quantitative environmental exposure and risk assessment (modelling endpoint). The
16 main difference in approach is that for triggering higher tier studies the best fitting
17 kinetic model is applied, for instance a biphasic kinetic model or a lag-phase model,
18 while for modelling endpoint and use of data on risk assessment the choice of the kinetic
19 model should be in agreement with the kinetics used in the environmental fate model
20 used in the risk assessment. Until now, the environmental fate models are based on
21 first-order kinetics. So in practice modelling endpoints should be derived with first-order
22 kinetics.

23 The principle for reporting biodegradation studies is that enough information should be
24 provided to allow an independent reproduction of the results and their verification with
25 alternative statistical methods/software packages. In particular, the following aspects of
26 kinetic analysis should be reported:

- 27 • Software package(s) and version/statistical test methods (with references).
28 To facilitate an independent duplication of the results it is preferred that the
29 kinetic analyses are performed with publicly available statistical software
30 packages, commonly used for such analyses.
- 31 • A listing of all original values to be used in the analysis. When data points are
32 regarded as "outliers" and discarded as part of the kinetic analyses, the
33 rationale for discarding data points should be included in the report.
- 34 • Analyses. Exact description of kinetic models used in the regressions.
35 Software options like range limits, initial values, restrictions in optimization
36 should be described.
- 37 • Visual and statistical assessment of the results. Figures of predicted and
38 observed values (i.e. concentrations) as a function of time and residual plots.
39 Other statistical endpoints that support the decision-making process should be
40 reported.

- 1 • Uncertainty (standard deviation or confidence interval) of the degradation rate
2 constant and formation rate of metabolites.
- 3 • If the degradation half-life or DT₅₀ is extrapolated beyond the experimental
4 period this should be clearly stated in the report.

5 Temperature at which to perform new simulation studies

6 According to the three OECD test guidelines (307, 308 and 309), the studies can be
7 performed at a range of temperatures, typically between 10 and 25°C. For REACH, the
8 preferred option is for new simulation degradation studies to be conducted at 12°C
9 where the principle aim of the study is to determine the half-life of the parent molecule.
10 The temperature of 12°C is the average temperature of European surface waters (see
11 Table R.16-8 in Chapter R16 "Environmental exposure estimation" of the [Guidance on](#)
12 [IR&CSA](#)). For simulation studies, the environmental media should also preferably be
13 collected from locations where conditions resemble the conditions targeted for the test.

14 If there are specific reasons for which it is not technically feasible to perform a new
15 simulation test at 12°C, a justification needs to be provided. In such cases proportionate
16 attempts should be made to bring the temperature as close to 12°C as possible.

17 If the purpose of the simulation test is principally the identification of metabolites, a
18 higher test temperature of 20°C may be appropriate to accelerate the formation of
19 degradation products and hence make their identification and characterisation easier.
20 Unless there are clear concerns for both parent and metabolites for the PBT assessment,
21 it is generally not necessary to perform one test at 12°C and one test at 20°C.

22 Temperature correction

23 Incubation temperature is one of many factors that need to be considered when
24 conducting higher tiered biodegradation studies. Others include the substance
25 concentration, test volume and geometry, airflow rate and co-metabolism.

26 Temperature is an issue within Europe due to the wide range of environmental
27 temperatures that a substance may experience in the field. Metabolic activity of a
28 microbial community, and thus degradation rates, generally increase with an increase of
29 temperature even though individual micro-organisms may show optimal activity at
30 different temperatures (Cavicchioli, 2006). Thus, rates of degradation in a test
31 conducted in the laboratory at 20-25°C are higher than those measured in the field,
32 where the average temperature is 12°C.

33 Consequently, for persistence assessments where the B and T criterion have been met,
34 and simulation data exist for degradation at 20°C, consideration should be given whether
35 temperature correction should be applied. This will be particularly important where the
36 measured half-life is close to the persistence criteria,. This correction, if applied, should
37 be based on the Arrhenius equation and extrapolate from 20°C to the temperature of the
38 environmental media at the point of sampling³⁴.

³⁴ This is in line with decisions made since the 32nd meeting of the Member State Committee where it has started to require new simulation degradation studies to be carried out at 12°C:

1 In the absence of structural substance class specific equations/models reflecting
2 temperature dependence of biodegradation, the Arrhenius equation (or a similar
3 appropriate equation designed to normalise physico-chemical degradation rates) can be
4 used as a possible means of normalisation.

5 This is:

$$6 \ln k = \ln A - (E_a/RT)$$

7

8 Where

9 k = rate constant (day^{-1})

10 A = factor equal to the rate coefficient at infinite temperature (day^{-1})

11 E_a = activation energy (kJ mol^{-1})

12 R = gas constant ($8.314 \cdot 10^{-3} \text{ kJ} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$)

13 T = temperature (K)

14

15 For first-order kinetics, the equation can be reformulated to:

16

$$17 DT50_{env} = DT50_{test} \cdot e^{\left(\frac{E_a}{R} \left[\frac{1}{T_{env}} - \frac{1}{T_{test}} \right]\right)}$$

18

19 where $DT50_{env}$ and $DT50_{test}$ are respectively the half-lives at environmental temperature
20 T_{env} (typically 285K) and test temperature T_{test} , (typically 293K).

21

22 There are potential uncertainties resulting from the use of the Arrhenius equation
23 because:

24 1) It was designed for simple chemical reactions rather than biological processes

25 2) The specific activation energy (E_a) for a substance or a chemical group is rarely
26 known

27 A generic E_a of 65.4 kJ/mol has been derived by EFSA (2007). It corresponds to the
28 median value of available pesticide E_a data. In the absence of valid substance specific
29 data, this value should be used.

30 In the absence of valid substance specific data, the Arrhenius equation with the generic
31 E_a -value should be used if temperature correction is needed.

32 No temperature correction is required for sewage treatment plants simulations (OECD
33 303).

34 Determination of degradation products

35 By monitoring parent substance, bio-transformation products/metabolites and NER, and
36 CO_2 as a function of time, it may be possible to assess the fate of the test substance in
37 the simulation test system of the specific environmental compartment. When a
38 substance is not fully degraded or mineralised, degradation products should preferably
39 be determined by chemical analysis. The methods will normally be substance specific
40 and consequently no guidance on choice of method can be given. For some substances,

1 radio-labelling and specific chemical analyses may allow reasonable fate assessment by
2 measuring subsequent metabolite formation and decay.

3 Where analytically possible, identification, stability, behaviour, molar quantity of
4 metabolites relative to the parent substance should be evaluated. Additionally, the
5 predicted degradation rate of the parent substance, $\log K_{ow}$ of the degradation
6 metabolites relative to the parent substance, and the potential toxicity and
7 bioaccumulation potential of metabolites should be considered and may need to be
8 investigated. The first step in a PBT assessment of metabolites should be the assessment
9 of their degradation half-life. If the metabolites are long-lived or persistent, they should
10 be assessed for bioaccumulation and toxicity.

11 However, it should be highlighted that the simulation studies (e.g. OECD 307, 308 and
12 309) available for high tier testing are usually designed to be environmentally relevant
13 and therefore use low concentrations of test substance. This means there are often
14 technical limitations associated with identifying transformation products. The
15 identification of the transformation/degradation products should be done according to
16 these guidelines.

17 Where the potential toxicity of significant degradation metabolites is considered, it is
18 worth noting that microbial degradation processes usually lead to more polar
19 degradation metabolites than the parent substance, but in some cases to less polar
20 degradation metabolites. This can be seen in the HPLC-RAD chromatographs routinely
21 produced during simulation tests. Reduced lipophilicity/hydrophobicity may be one
22 indication that the metabolites are less toxic and bioaccumulative than the parent
23 substance. Preliminary information on toxicity and bioaccumulation potential can be
24 obtained with the help of measured K_{ow} values as input data for for QSAR model
25 predictions on these endpoints for hypothetical or identified degradation metabolites.

26 Non-extractable residues (NER)

27 Knowledge of non-extractable residues (NERs) also needs to be considered. The
28 formation of NERs should not be confused with the degradation phenomenon. NERs
29 should be differentiated with regard to the binding, the potential remobilisation and the
30 hazard potential. NERs may potentially be re-mobilised as parent substance or
31 transformation product if they are heavily sorbed or bound by physical inclusion and thus
32 pose a potential risk. On the other hand, the NER may be bound by covalent bonds or
33 incorporated into the biomass and can in that case be considered to be irreversibly
34 bound. In the latter situation NER can be regarded as a potential removal pathway.
35 However, a standardised tiered extraction scheme to enable a differentiation and
36 qualitative assessment of NER, regarding whether the binding is reversible under certain
37 conditions which may occur in the environment or not, does not currently exist. When
38 quantifying NER, the extraction process of the sample often takes place with a suitable
39 organic solvent and it is generally repeated 3 or 4 times until no further yield is
40 achieved. Typically a range of solvents of increasing polarity is used (e.g. methanol,
41 acetone, acetonitrile and hexane etc.) under ambient conditions. If the entire residual
42 radioactivity cannot be recovered then appropriate solvent may be mixed with weak
43 acids or bases or coupled to ultrasonic extraction. This aims to provide different
44 conditions that may lead to the parent substance or metabolite being released back into
45 solution. Finally, the use of strong acids, bases or refluxing could undoubtedly extract
46 the sample more thoroughly but could alter both the substances of interest and the

1 soil/sediment matrices. Such severe extraction techniques are therefore rarely employed
2 in e.g. routine soil or sediment/water testing. The extraction methods and efficiencies as
3 well as analytical methods and detection limits should always be reported.

4 Ultimately, it is the bioavailability of the residues which is actually of environmental
5 significance, rather than the extent to which they can or cannot be extracted by the
6 above mentioned methods.

7 The amount and kind of NER is operationally defined by the extraction method
8 employed. At the moment there is no extraction scheme/ procedure available to predict
9 how much of the NERs are irreversibly bound and which part is reversibly bound and
10 may become available under changed environmental conditions.

11 These considerations should aid in determining the following environmental assessments
12 for hazard classification, PBT/vPvB assessment and exposure (risk) assessment.

13 Suspended matter in water simulation tests

14 In OECD TG 309, two test options are described: the 'pelagic test' and the 'suspended
15 sediment test'. In both cases, the coarse particles are to be removed from the water
16 sample, for example by filtration through a filter with 100 µm mesh size or with a coarse
17 paper filter, or by sedimentation. For the 'suspended sediment test', surface sediment is
18 added afterwards to obtain a suspension; the allowed concentration of suspended solids
19 for the 'suspended sediment test' is between 10 mg/L and 1 g/L. However, it is worth
20 noting that the 'pelagic test' will actually also contain suspended matter. Indeed, only
21 coarse particles will be removed from the prior filtering/sedimentation operation; some
22 undissolved matter and fine particles will remain suspended in the water sample³⁵.

23 The concentration, nature and size of the suspended particles are highly variable
24 amongst water bodies. In lowland areas (with low current velocity), the suspended
25 particles are small and usually rich in organic matter. They will pass through a 100 µm
26 mesh. On the contrary, in the headwaters (with high current velocity), the suspended
27 particles can be quite large and they are usually poor in organic matter. Most will be
28 stopped by a 100 µm mesh. However it is acknowledged that for most industrial
29 substances, direct releases occur to large rivers or to marine water. For large rivers, the
30 concentration of suspended matter (SPM) is reasonably constant and an EU default of
31 15 mg_{dw}/L has been proposed, e.g. for the implementation of the Water Framework
32 Directive (European Communities, 2011) or in EUSES. For marine waters, a default SPM
33 concentration of 3 mg_{dw}/L has been proposed for the Water Framework Directive.
34 Similarly a default SPM concentration of 5 mg_{dw}/L has been implemented in EUSES for
35 marine waters.

36 For the purpose of REACH, using natural surface water containing between 10 and 20
37 mg_{dw}/L SPM for simulation tests in freshwater and c.a. 5 mg_{dw}/L for simulation tests in

³⁵ By convention, 'dissolved' matter is operationally defined as passing through a filter mesh of 0.45 µm. By extension, 'suspended matter' (SPM) is defined as the matter which does not pass through a filter mesh of 0.45 µm. A simulation test performed with water filtered through a filter mesh of 0.45 µm would not be sensible. Virtually all degrading microorganisms would indeed be filtered out as the size of bacteria is typically in the order of 10 µm.

1 marine water is considered acceptable. The 'suspended sediment test' which implies the
2 subsequent addition of suspended matter is generally not recommended as the water
3 sample should be chosen to already contain naturally a proper SPM concentration.

4 SPM contains a significant fraction of organic carbon (a default fraction of 10% is
5 generally assumed) to which the test substance may adsorb. Hence the organic carbon
6 (OC) concentration in surface water simulation tests is typically 2 to 3 orders of
7 magnitude higher than the test substance concentration. One should therefore
8 acknowledge that the formation of NERs may be significant in surface water tests too.
9 Therefore, as for soil and sediments simulation tests, the NERs should be quantified and
10 the extraction procedure and solvent used should be explained and scientifically justified
11 for simulation tests in water too.

12 The amount of OC being much higher in sediment and soil, the potential for the
13 formation of NERs is also much higher in soil and sediment simulation tests than in the
14 water test. Therefore to minimise the formation of NERs, simulation tests in water should
15 generally be preferred over simulation tests in soil or in sediment. More guidance is
16 given in chapter R.11.4.1.1.2 of the [Guidance on IR&CSA](#).

17 Environmental hazard classification

18 When a substance is not fully mineralised, but rapidly degraded to less degradable
19 degradation products, the environmental hazard of these should be considered before a
20 final judgement of whether a substance is readily or rapidly degradable.

21 PBT and vPvB assessment

22 When a substance is not fully mineralised, but degraded into other substances, the
23 PBT/vPvB properties of these should be evaluated before a final judgement of whether
24 the parent substance fulfils the PBT/vPvB criteria. More guidance is given in chapter R.11
25 of the [Guidance on IR&CSA](#).

26 Exposure and risk assessment

27 When a substance is not fully mineralised, but degraded to more persistent degradation
28 products, the environmental exposure concentrations should be determined for these
29 products. Consequently, the safety assessment should also consider the degradation
30 products, including their potential (eco)toxicity, degradability and bioaccumulation
31 potential.

32 **R.7.9.4.2 Field data on degradation/biodegradation**

33 In higher tier studies biodegradation is not always visible as a separate process. Other
34 processes like transport, adsorption, volatilisation, uptake in plants or organisms,
35 hydrolysis also contribute to the fate of the substance simultaneously. In order to derive
36 biodegradation rate inverse modelling can be applied to quantitatively separate
37 biodegradation from other processes.

38 Measured concentrations in the mesocosm, lysimeter, or field experiments are compared
39 with simulated concentrations in an environmental model, and the biodegradation rate
40 constant is computed by a parameter estimation procedure (manually by trial and error
41 or automated by a software package for example PEST) until the modelled concentration

1 fit to the measured data. Procedures are described in FOCUS (2011), an example is
2 published by Dubus *et al.* (2004). In addition, the following guidances may be
3 considered in order to assess dissipation in the soil compartment: NAFTA (2006), EFSA
4 (2014) and OECD (2016) and Deneer *et al.*, (2015) for aquatic field studies.

5

6 **R.7.9.4.3 Exposure considerations for degradation/biodegradation**

7 The major factors that are related to exposure within the context of degradation relate
8 to:

- 9 • the use of the substance;
- 10 • the substance emission pattern (continuous or intermittent release);
- 11 • the compartment to which the substance is released (this can be more than
12 one compartment);
- 13 • the amount per time unit or rate of substance released;
- 14 • the rate of degradation; and
- 15 • the physico-chemical properties of the substance.

16 The physico-chemical properties of the substance and the compartment to which the
17 substance is released will have a large influence on where the substance will be
18 transported to and distributed to within the environment (see Chapter R.16). The
19 emission pattern (continuous or intermittent) will influence the ability of competent
20 microorganisms to establish themselves and for biodegradation to occur. The amount of
21 substance released will also influence the kinetics of biodegradation.

22 The identification of the environmental compartment(s) is of primary importance for a
23 PBT, vPvB or /and risk/exposure assessments. A simulation test will normally not be
24 required for all environmental compartments. The compartments of highest exposure
25 and risk should be tested first if testing is required for refinement of quantitative risk
26 assessment:

- 27 • If testing is triggered for PBT assessment different types of considerations
28 should be made described in more detail in chapter R.11 of the [Guidance on](#)
29 [IR&CSA](#).
- 30 • The K_p or K_{oc} values may be used as indicators of whether testing in a water-
31 sediment system or in soil may be warranted. Substances with e.g. $\log K_{oc}$
32 >4 have a high potential for adsorption to soil and sediment.
- 33 • Multi-media modelling (e.g. Mackay level 3 models) or exposure models (e.g.
34 FOCUS for agrochemicals) could also be explored in order to evaluate the
35 environmental compartment(s) of primary concern. The results of such
36 models should be interpreted with care, as the predictions are strongly
37 dependent of the default assumptions used for those models (e.g. size of the
38 environmental compartments, or the emission parameters employed in the
39 modelling).

- 1 • Nevertheless a case-by-case evaluation of the results of such models may be
2 useful and may even indicate whether or not substances may expose pristine
3 environmental compartments (e.g. open sea) to a significant extent (i.e.
4 indicate a significant potential for long range environmental transport via the
5 atmosphere).

6 One of the key aspects for consideration is the volatility of the compound. By affecting
7 the partitioning to other media and compartments from the source compartment(s) and
8 the kinetics of that transfer, volatility is a key physico-chemical properties that greatly
9 influence the overall persistence of a substance in the environment, as defined by the
10 mean time that a molecule resides in the system taking into account all intra-media and
11 transfer processes (OECD, 2002b).

12 Webster *et al.* (1998) have pointed out the inconsistencies which result when using only
13 specific degradation half-lives for determining the environmental persistence and
14 ignoring the mode/compartment of entry and the effects of partitioning to other media.

15 Usually, intra-media and transfer processes are ignored in the assessment of
16 persistence, whereas it should be considered that:

- 17 • compartment specific degradation half-lives might be overly conservative
18 when a substance does not partition significantly into that compartment;
- 19 • compartment specific degradation half-lives are not independent of each
20 other;
- 21 • the amount lost by degradation in a specific compartment is determined both
22 by the compartment specific degradation rate constant and the amount of
23 substance present in that compartment (Wania & Mackay, 2000).

24 There are several parameters that impact on the volatility of a substance and its inter-
25 compartmental partitioning, including aqueous solubility and vapour pressure (VP).
26 There are also a number of parameters that may be useful for assessing volatility and
27 inter-compartmental transport, including octanol-air partitioning constant and the
28 Henry's law constant. When assessing the persistence of a substance with high volatility,
29 it is therefore recommended not to rely only on specific-medium degradation half-lives
30 but to also consider on a case-by-case basis if these half-lives will cover the overall
31 persistence of a substance in the environment. This might be achieved by the use of
32 multimedia fate models.

33 **R.7.9.4.4 Remaining uncertainty for degradation/biodegradation**

34 Substances that fulfil the criteria for ready biodegradability are likely to undergo rapid
35 degradation in the environment under most conditions (OECD, 2006b). However, it must
36 be recognised that these tests are very stringent and most substances will not fulfil the
37 pass criteria for ready biodegradability. For substances that exhibit between 40 and 60%
38 mineralisation in ready biodegradability test, extensive primary biodegradation would
39 have occurred even though the use of non-specific endpoints such as DOC and BOD do
40 not directly measure this. Therefore there will remain a large degree of uncertainty
41 about the biodegradability of many substances and testing at higher levels or tiers will
42 be required.

1 At present the data set for biodegradation of general substances in higher tiered studies
2 such as the OECD 308 test is relatively small. These tests were originally designed for
3 plant protection products and have not been routinely applied to general substances.
4 Even though such tests constitute the highest tier testing of biodegradation there are
5 uncertainties connected with their use.

6 One example is that degradation half-lives may vary between different sites from where
7 the environmental compartments inoculum and test media are sampled. Another
8 example is, that it is uncertain what the value of conducting the strict anaerobic test part
9 of the OECD 308 test is, and how these data can be used in CSA.

10 Identifying the compartments of concern can also be problematic in the absence of
11 accurate use and emission data or data concerning the potential for environmental long-
12 range transport. Confidence can be improved if such data are comprehensive and
13 accurate.

14 **R.7.9.5 Conclusions for degradation/biodegradation**

15 **R.7.9.5.1 Concluding on suitability for Classification and Labelling³⁶**

16 Environmental hazard classification requires information on aquatic toxicity, degradation
17 and bioaccumulation. In the previous EU classification system (Council Directive
18 67/548/EEC) and in the "Globally Harmonised System of classification and labelling of
19 chemicals (GHS)" (United Nations GHS (Rev.1) 2005³⁷) / CLP, the determination of the
20 appropriate risk phrases or hazard statements are often based on an integration of this
21 information. However, this integrated approach is not considered here, as the ITS is
22 concerning degradation aspects alone.

23 Under the degradation part of the EU and GHS classification criteria two aspects need to
24 be evaluated:

25 Previous EU system (DSD):

- 26 • Whether "the substance is readily degradable or not"
- 27 • Whether "additional scientific evidence concerning degradation" is available,
28 i.e. whether there is "a proven potential to degrade rapidly in the
29 environment"

30 GHS/CLP:

- 31 • Whether there is a "lack of rapid degradability"
- 32 • Whether there is "other evidence of rapid degradation"

³⁶ For more up-to-date information please see the *Guidance on the Application of the CLP Criteria*, section 4.1.3.2.3.2 and Annex II which have been updated in April 2012.

³⁷ Please note that Please note that rev. 4 is available
(http://www.unece.org/trans/danger/publi/ghs/ghs_rev04/04files_e.html)

1 Some guidance on interpretation of information on degradation is available given in
2 Annex VI of Directive 67/548/EEC and this has been further developed in part 4 and
3 Annex 9 to the GHS criteria (United Nations GHS (Rev.1) 2005²¹)/ CLP. This latter
4 guidance, which has been internationally agreed by OECD, forms the principal basis for
5 this guidance on the suitability of degradation data on classification. For the purposes of
6 decisions on classification and testing strategies, the two terms 'not readily degradable'
7 and 'lack of rapid degradation' may be considered as synonymous.

8 The decision criteria for evaluating the suitability of available information on use in a
9 decision on environmental hazard classification should consequently be focused on these
10 aspects. At each step of the ITS, the available information will need to be evaluated
11 against the aspects described above. The definition of ready (or rapid) degradability
12 covers both biotic and abiotic degradation. Under most environmental conditions
13 hydrolysis will be the major abiotic removal process. Data on either or both biotic or
14 abiotic degradation would be sufficient to make a decision on rapid degradation.

15 Degradation can be monitored by either measuring the complete breakdown of the
16 substance to carbon dioxide and water (ultimate degradation), or simply the measuring
17 the disappearance of the parent substance, primary degradation. While ultimate
18 degradation is preferred, primary degradation can be used to define the pass levels in
19 each of the degradation tests provided certain conditions are met. Data on primary
20 biodegradability may be used for demonstrating rapid degradability only when it can be
21 satisfactorily demonstrated that the degradation products formed do not fulfil the criteria
22 for classification as hazardous to the aquatic environment.

23 In general, where experimental data are not available, and there are no additional data
24 from structurally similar substances, a substance must be considered as *not rapidly*
25 *degraded*. The following types of non-test data may be considered, however, as
26 contributing to a decision on *ready or rapid* degradation for classification purposes.

27 **QSAR Data**

28 In the absence of experimental or environmental data, the predictions from QSARs
29 models described in Section [R.7.9.3.1](#) may be considered. No formal decision has been
30 taken on how to use (Q)SAR derived information on biodegradability for classification
31 purposes in the EU. In relation to the development of the GHS, the usefulness of
32 (Q)SARs for predicting ready biodegradability is considered (United Nations GHS (Rev.1)
33 2005). It is stated that (Q)SARs for predicting ready biodegradation are normally not yet
34 sufficiently accurate to predict rapid degradation. However, it is a general rule that when
35 no useful information on degradability is available - either experimentally derived or
36 estimated - the substance should be regarded as not readily or not rapidly degradable
37 and (Q)SAR prediction can be used as supporting evidence of this.

38 The reason for this discrimination on usability of different outcomes of (Q)SAR
39 predictions is that currently conducted validations and comparisons between test data
40 and (Q)SAR predictions often seem to suggest that the probability of a correct prediction
41 of a slow biodegradation is high, while the probability of a correct prediction of a fast
42 biodegradation is significantly lower (e.g. OECD 2004). This is however according to
43 validation studies where (Q)SAR predictions have been compared with ready
44 biodegradability test data and the sensitivity and specificity of not ready biodegradability
45 predictions seem to be dependent on the particular (Q)SAR model in question (cf. OECD

1 2004:ENV/JM/TG(2004)26Rev1 and references therein). Generally however when a
2 substance is estimated to be *slowly* biodegradable, sufficient information is normally
3 considered available on biodegradability for hazard classification purposes, when no test
4 data are available. When a substance is estimated to biodegrade *fast*, further
5 information gathering is normally necessary (United Nations GHS (Rev.1) 2005³⁸).

6 **Structurally related substances**

7 When no experimental data are available, the potential for rapid degradation in the
8 aquatic environment may also be assessed by examining available data on structurally
9 related substances. There will always need to be an element of expert judgement in such
10 an evaluation, but this approach may be particularly relevant where the QSAR prediction
11 described above suggests rapid degradation. If such a prediction is supported by
12 experimental evidence from structurally similar substances, then this can be considered
13 as convincing evidence for rapid degradation for classification purposes. Equally, of
14 course, such data on similar structures may provide evidence of a lack of rapid
15 degradation. In general, expert judgement should be used in a conservative way.

16 **Degradation data suitable for use in classification**

17 Ready Biodegradation

18 Ready biodegradability is defined in the OECD Test Guidelines No. 301 (OECD 1992). All
19 organic substances that degrade to a level higher than the pass level in a standard OECD
20 ready biodegradability test or in a similar test should be considered readily
21 biodegradable and consequently also rapidly degradable. Many literature test data,
22 however, do not specify all of the conditions that should be evaluated to demonstrate
23 whether or not the test fulfils the requirements of a ready biodegradability test.
24 However, provided a test is conducted within the constraints and quality criteria defined
25 in Section [R.7.9.4](#), it may be considered as a ready biodegradability test for the purposes
26 of classification. In the context of classification, the individual test *pass* levels are
27 considered an important part of the criteria.

28 When contradictory results in ready biodegradability tests are obtained the positive
29 results could be considered valid irrespective of negative results, when the scientific
30 quality of the former is good and the positive test results are well documented, including
31 assurance of the use of non-pre-exposed (non-adapted) inoculum. (United Nations GHS
32 (Rev.1) 2005³⁹). Before a decision is made on the appropriate result to use, however,
33 the data should be carefully examined to determine whether there is a simple or clear
34 explanation for the differences in result. Not all of the various screening tests are
35 suitable for the testing of all types of substances, and results obtained by the use of a
36 test procedure which is not suitable for the specific substance should be evaluated
37 carefully before a decision on the use is taken (see Section [R.7.9.4](#)). Equally, where
38 possible, the inoculum source should be checked to ensure a positive result is not the
39 result of artificially pre-adapted inoculum.

³⁸ Please note that rev. 4 is available (http://www.unece.org/trans/danger/publi/ghs/ghs_rev04/04files_e.html)

³⁹ Please note that rev. 4 is available (http://www.unece.org/trans/danger/publi/ghs/ghs_rev04/04files_e.html)

1 Nevertheless, where a positive result has been obtained using a standard and valid
2 methodology, this will be used to indicate rapid degradation for classification,
3 irrespective of other negative results. This will hold true unless there are strong *Weight*
4 *of Evidence* or structural reasons to question this result.

5 Modified ready biodegradation tests

6 There are circumstances when it may be necessary to modify the standard guidelines in
7 order to test a particular substance. This is particularly true for poorly water soluble
8 substances, and also those that show toxicity to micro-organisms at the concentrations
9 of the test. These modifications are described in Section [R.7.9.4](#). Such tests are
10 regarded as ready biodegradation tests and can be used directly in classification.

11 BOD5/COD

12 Information on the 5-day biochemical oxygen demand (BOD5) can be used for
13 classification purposes only when no other measured degradability data are available.
14 Thus, priority is given to data from ready biodegradability tests and from simulation
15 studies regarding degradability in the aquatic environment. The BOD5 test is a
16 traditional biodegradation test that is now replaced by the ready biodegradability tests.
17 Therefore, this test should not be performed today for assessment of the ready
18 biodegradability of substances. Older test data may, however, be used when no other
19 degradability data are available. For substances where the chemical structure is known,
20 the theoretical oxygen demand (ThOD) can be calculated and this value should be used
21 instead of the chemical oxygen demand (COD).

22 Test duration less than 28 days

23 Sometimes degradation is reported for tests terminated before the 28 days period
24 specified in the standards (e.g. the MITI (1992) test data). These data are of course
25 directly applicable when degradation greater than or equal to the pass level is obtained.
26 When a lower degradation level is reached, the results need to be interpreted with
27 caution. One possibility is that the duration of the test was too short and that the
28 chemical structure would probably have been degraded in a 28-day biodegradability test.
29 If substantial degradation occurs within a short time period, the situation may be
30 compared with the criterion BOD5/COD ≥ 0.5 or with the requirements on degradation
31 within the 10-days time window (OECD 301A,C,D,E and F) or 14-days time window
32 (OECD 301B). In these cases, a substance may be considered readily degradable (and
33 hence rapidly degradable), if:

- 34 • the ultimate biodegradability exceeds 50% within 5 days and
- 35 • the ultimate degradation rate constant in the test system in this period is
36 greater than 0.1 day^{-1} corresponding to a half-life of 7 days in the test system
37 (see Section R.7.9.11).

38 Other convincing scientific evidence

39 Rapid degradation in the aquatic environment may be demonstrated by other data than
40 referred to using the standard assessment methods covered above. This may be data on
41 biotic and/or abiotic degradation. Data on primary degradation can only be used where it

1 is demonstrated that the degradation products shall not be classified as hazardous to the
2 aquatic environment, i.e. that they do not fulfil the classification criteria.

3 Scientific evidence must be provided that the substance is degraded in the aquatic
4 environment to a level of >70% within a 28-day period. If first-order kinetics is
5 assumed, which is reasonable at the low substance concentrations prevailing in most
6 aquatic environments, the degradation rate will be relatively constant for the 28-day
7 period. Thus, the degradation requirement will be fulfilled with an average degradation
8 rate constant, $k > 0.043 \text{ day}^{-1}$ which corresponds to a degradation half-life of 16 days. In
9 determining whether this half-life criterion is met, care should be taken to ensure that an
10 appropriate account has been taken of the temperature of the study.

11 The evaluation of data on fulfilment of this criterion should be conducted on a case-by-
12 case basis by expert judgement. However, guidance on the interpretation of various
13 types of data that may be used for demonstrating a rapid degradation in the aquatic
14 environment is given below. In general, only data from aquatic simulation tests are
15 considered directly applicable. However simulation test data from other environmental
16 compartments could be considered as well, but such data require in general more
17 scientific judgement before use.

18 Hydrolysis

19 Hydrolysis is not an ultimate degradation and various intermediate degradation products
20 may be formed, some of which may be only slowly degradable. Only when it can be
21 satisfactorily demonstrated that the hydrolysis products formed do not fulfil the criteria
22 for classification as hazardous for the aquatic environment, data from hydrolysis studies
23 could be considered.

24 When a substance is quickly hydrolysed (e.g. with $t_{1/2} < \text{a few days}$), this process is a
25 part of the degradation determined in biodegradation tests. Often, hydrolysis is the initial
26 transformation process in biodegradation.

27 Aquatic simulation tests

28 Aquatic simulation tests are tests conducted in laboratory, but simulating environmental
29 conditions and employing natural samples as inoculum. It should be noted that the OECD
30 303 test is not simulating conditions in the aquatic environment but in sewage treatment
31 plants and consequently, results from this test are not valid for classification. Results of
32 aquatic simulation tests (mineralisation rate, degradation half-life) may be used directly
33 for classification purposes when realistic environmental conditions in surface waters are
34 simulated. Such tests are described in Section [R.7.9.3](#).

35 Soil and sediment degradation data

36 It has been argued that for many non-sorptive (non-lipophilic) substances more or less
37 the same degradation rates are found in soil and in surface water. For adsorptive
38 substances, a lower degradation rate is generally expected in soil than in the water-
39 phase due to partly immobilization caused by sorption. Thus, when an adsorptive
40 substance has been shown to be degraded rapidly in a soil simulation study, it is most
41 likely also rapidly degradable in the aquatic environment. It is therefore proposed that
42 an experimentally determined degradation in soil is sufficient documentation for a rapid
43 degradation in surface waters. Such tests are described in Section [R.7.9.3](#).

1 Field investigations

2 Parallels to laboratory simulation tests are field investigations or mesocosm experiments.
3 In such studies, fate and/or effects of substances in environments or environmental
4 enclosures may be investigated. Fate data from such experiments might be used for
5 assessing the potential for a rapid degradation. This may, however, often be difficult, as
6 it requires that an ultimate degradation can be demonstrated. This may be documented
7 by preparing mass balances showing that no non-degradable intermediates are formed,
8 and which take the fractions into account that are removed from the aqueous system
9 due to other processes as e.g. sorption to sediment or volatilisation from the water
10 environment. In general, mesocosms and field studies are not used for classification and
11 labelling purposes.

12 Monitoring data

13 Representative monitoring data may demonstrate the removal of contaminants from the
14 aquatic environment. Such data are, however, very difficult to use for classification
15 purposes. The following aspects should be considered before use:

- 16 • is the removal a result of degradation, or is it a result of other processes as
17 e.g. dilution or distribution between compartments (sorption, volatilisation)?
- 18 • is formation of non-degradable intermediates excluded?

19 Only when it can be demonstrated that removal as a result of ultimate degradation fulfils
20 the criteria for rapid degradability, such data might be used directly for classification
21 purposes. In general, monitoring data can only be used as supporting evidence for
22 demonstration of either persistence in the aquatic environment or a rapid degradation.

23 **Degradation data not suitable for use in classification**

24 Inherent biodegradability tests

25 Substances that are degraded more than 70% in tests for inherent biodegradability have
26 the potential for ultimate biodegradation (OECD Test Guidelines). However, because of
27 the optimum conditions in these tests, the rapid biodegradability of inherently
28 biodegradable substances in the environment cannot be assumed. The optimum
29 conditions in inherent biodegradability tests stimulate adaptation of the microorganisms
30 thus increasing the biodegradation potential, compared to natural environments.
31 Therefore, positive results in these tests should not be interpreted as evidence for rapid
32 degradation in the environment.

33 STP simulation tests

34 Results from tests simulating the conditions in a sewage treatment plant (STP) (e.g. the
35 OECD 303) cannot be used for assessing the degradation in the aquatic environment.

36

37 Photochemical degradation

38 Information on photochemical degradation (cf. OECD GD(97)21) is difficult to use for
39 classification purposes. The actual degree of photochemical degradation in the aquatic
40 environment depends on local conditions (water depth, suspended solids, turbidity, etc.)

1 and the hazard of the degradation products is usually not known. Probably only seldom
2 will enough information be available for a thorough evaluation based on photochemical
3 degradation.

4 Volatilisation

5 Substances may be removed from some aquatic environments by volatilisation. In
6 general these data do not represent degradation and are not used in classification. The
7 reason is that the degree of volatilisation from the aquatic environment is highly
8 dependent on the environmental conditions of the specific water body in question, such
9 as the depth and the gas exchange coefficients (depending on wind speed and water
10 flow). In general, therefore, the Henry's Law constant cannot be used for assessment of
11 the degradation (here removal of a substance from the water phase) in relation to
12 aquatic hazard classification of substances. However, substances that are gases at
13 ambient temperature may be exempted from this general recommendation.

14 **R.7.9.5.2 Concluding on suitability for PBT/vPvB assessment**

15 Guidance on the suitability for PBT/vPvB assessment is provided in Chapter R.11 of the
16 [Guidance on IR&CSA](#).

17

18 **R.7.9.5.3 Concluding on suitability for use in chemical safety** 19 **assessment**

20 Degradation data are used in the chemical safety assessment to:

- 21 • determine the level of removal of a substance from waste water in a Sewage
22 Treatment Plant
- 23 • determine the initial soil concentration for the purposes of calculating a PEC_{soil}
24 local
- 25 • determine the steady state $PEC_{regional}$ for each environmental compartment.

26 Ready biodegradation

27 Data on ready biodegradation can be used, and is a requirement of Annex VII. The data
28 should contain information of the pass or fail status against the appropriate test
29 thresholds, including whether the 10-day window criteria has been met. For poorly
30 soluble substances, adjustments to the test protocol as described in Section [R.7.9.4](#) are
31 acceptable. Equally, test thresholds may be applied on the basis of primary degradation
32 if these data are available, but if primary degradation is considered as the principal
33 degradation route, further information on the degradation products may be required. For
34 readily biodegradable substances, regional environmental concentrations in
35 environmental media i.e. surface water, sediment and soil can be calculated by the use
36 of Mackay level 3 models. The default degradation rates for such readily biodegradable
37 substances can be used as input values (see Section R.16.4.2.3 and appendix
38 A.16.3.2.2. in Chapter R.16 of the [Guidance on IR&CSA](#)).

39 Hydrolysis

1 Data from the hydrolysis test may be used if hydrolysis is a dominant route of
2 degradation. These data may also be used to indicate:

- 3 • where problems may arise in generation and interpretation of aquatic toxicity
4 data
- 5 • where degradation can occur such that further consideration may need to be
6 given to major degradation products
- 7 • where the degradation rate constant may need adjusting in the determination
8 of the PEC_{regional}

9 Rapid hydrolysis, for example, may influence the fate of a substance entering an STP in
10 the same way as primary biodegradation and may require further investigation of
11 potential hydrolysis products. Where data are only available for the screening part of the
12 hydrolysis study, little quantitative information is available and the calculation of an
13 environmental rate constant is not possible. Nevertheless, where the estimated
14 degradation half-life is <24 hours, this will provide clear evidence of environmental
15 degradation, and consideration must be given to the identification and further evaluation
16 of any degradation products.

17 Hydrolysis data are needed over the range of environmentally relevant pHs from 4 to 9
18 (See TG 111) and should be corrected for temperature before use in the CSA (see
19 Section [R.7.9.4](#)).

20 Inherent biodegradation

21 Where information on inherent biodegradation is available, particularly from the Zahn-
22 Wellens, or the MITI (II) studies (OECD 302B & C), these data should be examined to
23 determine whether the special criteria detailed in Section [R.7.9.4](#) are met. Where these
24 criteria are met, the information may be used in the CSA to help determine the fate of
25 the substance in an STP and by use of default degradation rates for inherently
26 degradable substances in calculating the regional environmental concentrations in
27 surface water, sediment and soil by the use of Mackay level 3 models (see chapter
28 R.16).

29 A pass level (>70%) degradation in an inherent test may be used in similar manner to a
30 pass in a ready test, where a specific STP may be considered as adapted. This is
31 described further in the CSA Guidance (see Section R.16.4.2.3. and appendix A.16.3.2.2.
32 in Chapter R.16 of the [Guidance on IR&CSA](#)). In other circumstances to those described
33 above, data from inherent biodegradation testing cannot be used in the CSA.

34 Photochemical degradation

35 Information on direct photolysis is difficult to interpret in the CSA since its significance in
36 the aquatic environment depends on local conditions (water depth, suspended solids,
37 turbidity, etc.). Nevertheless, where a degradation rate constant can be derived for site
38 specific environmentally realistic conditions, these may be used in the assessment on a
39 case-by-case basis where justified by a knowledge of local conditions. Information on
40 indirect photolytic degradation half-life may be used for estimation of generic regional
41 concentrations in air by use of generic assumptions about light intensity (latitude and
42 season, length of day) and concentration of hydroxyl radicals in the air.

1 Refining a Chemical Safety Assessment

2 Where it is necessary to develop further the screening assessment, the following
3 information and testing can be considered if available, or generated as a result of testing
4 according to Annexes VI to X.

5 Sewage Treatment Plant Simulation Test

6 At screening level, models such as SIMPLETREAT are used to predict the level of
7 degradation in an STP based on simple biodegradation screening tests as described
8 above. A STP simulation test should give a direct measure of substance removal under
9 realistic operating conditions. The assessment of biodegradability and/or removal in
10 sewage treatment plants should therefore be based on results from tests simulating the
11 conditions in treatment plants such as the OECD 303A or the OECD 314 series tests. It
12 should be noted that the former test does not give a direct measurement of degradation
13 but rather removal of the test substance including both degradation and adsorption as
14 characterised by a STP. Normally inflow and outflow DOC or specific analysis is used and
15 the concentrations material may be used and a full mass balance obtained.

16 Data from non-standardised tests and/or tests not performed according to the principles
17 of GLP may be used if expert judgement has confirmed them to be equivalent to results
18 from the standardised degradation tests on which the calculation models, e.g.
19 SimpleTreat, are based. The same applies to STP monitoring data, i.e. in-situ
20 influent/effluent measurements.

21 Environmental Simulation Tests

22 The CSA will sometimes require the generation of a 'regional' or background steady state
23 concentration that might arise from a particular emission or load to an environmental
24 compartment. These are calculated using standard fugacity models that require inputs of
25 the transport characteristics between environmental compartments and the degradation
26 rates for each compartment. At screening level, these are estimated from simple
27 screening data described above. Where refinement of these degradation rates is needed,
28 data from environmental simulation testing can be used. The particular tests chosen
29 should seek to simulate the compartment(s) of concern. These tests are requirements
30 listed in Annexes IX to X. The decision on which specific test should be selected is
31 considered in Section [R.7.9.4](#) and [R.7.9.6](#).

32 In addition, the soil environment simulation test may also be used to further refine the
33 local PEC soil where an initial concentration is calculated based on an assumption of a
34 number of years of exposure, followed by an addition load from land spreading of
35 sewage sludge. Both the initial concentration, and added concentration can be refined by
36 a soil degradation rate constant measured from a simulation test.

37 Field data

38 A range of field investigation approaches such as mesocosms, lysimeters etc are
39 described in Section [R.7.9.4](#). These are not normally designed to measure just
40 degradation processes and thus cannot be considered to yield a degradation half-life that
41 can be read directly against the criteria.

1 **R.7.9.5.4 Information not adequate**

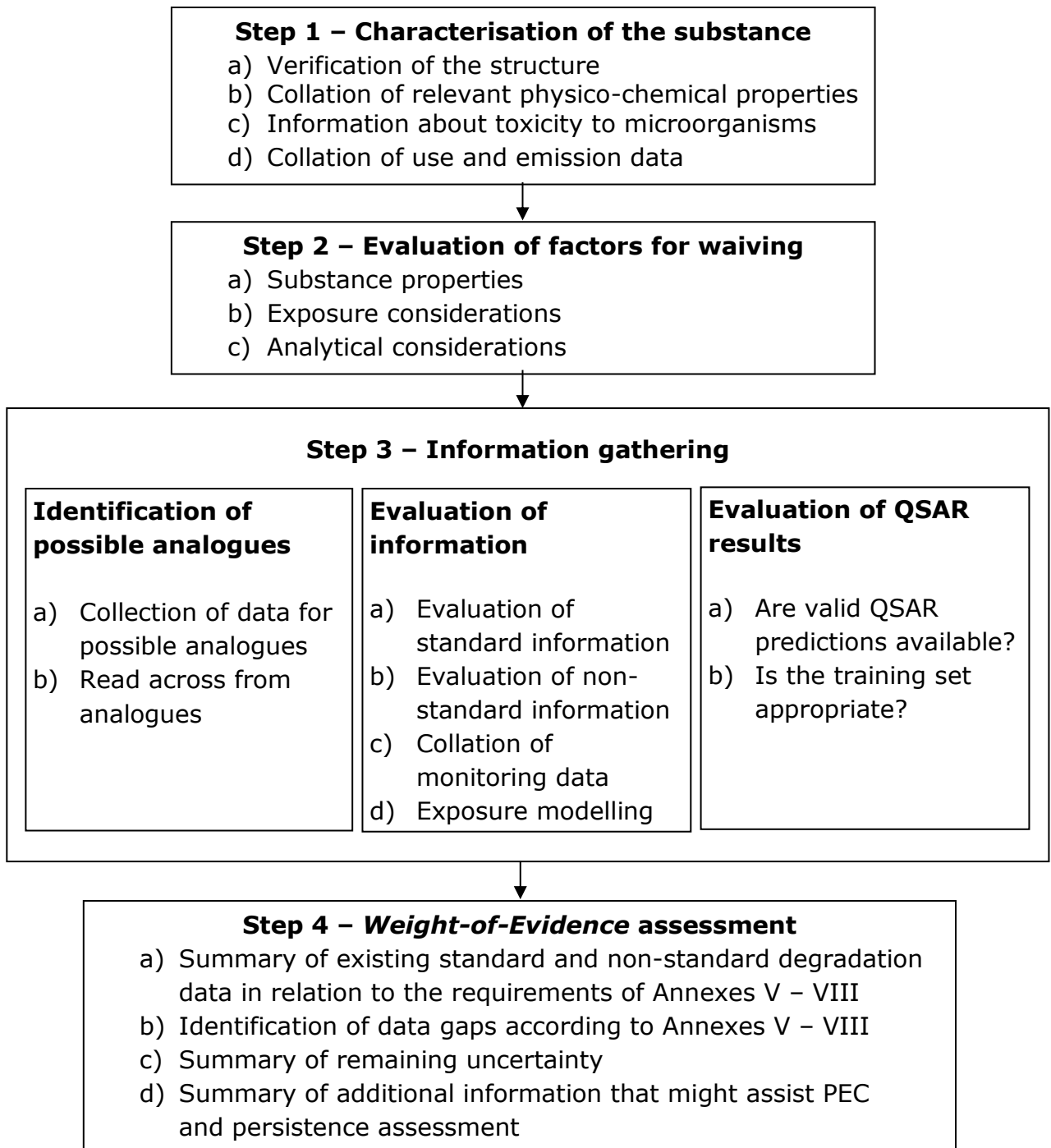
2 The prerequisite for use of other information than those types specified by the
3 information requirements of REACH is that such information alone or in combination with
4 other information is:

- 5 • equivalent to the results that would be obtained by standard testing, and
- 6 • adequate for the three regulatory endpoints: Classification and Labelling, PBT
7 assessment and chemical safety assessment. The equivalence and adequacy
8 will have to be substantiated by a *Weight of Evidence* approach using expert
9 judgement and making best use of all existing information.

10 *Weight of Evidence* is closely linked to “integrated testing strategies (ITS)”, in that the
11 available evidence can help to determine the subsequent testing steps. Results from
12 these subsequent tests affect the *Weight of Evidence*, which leads to a new decision on
13 whether there is any need of further testing, and so on. The ITS’s are designed to be
14 flexible and applied on a case-by-case basis.

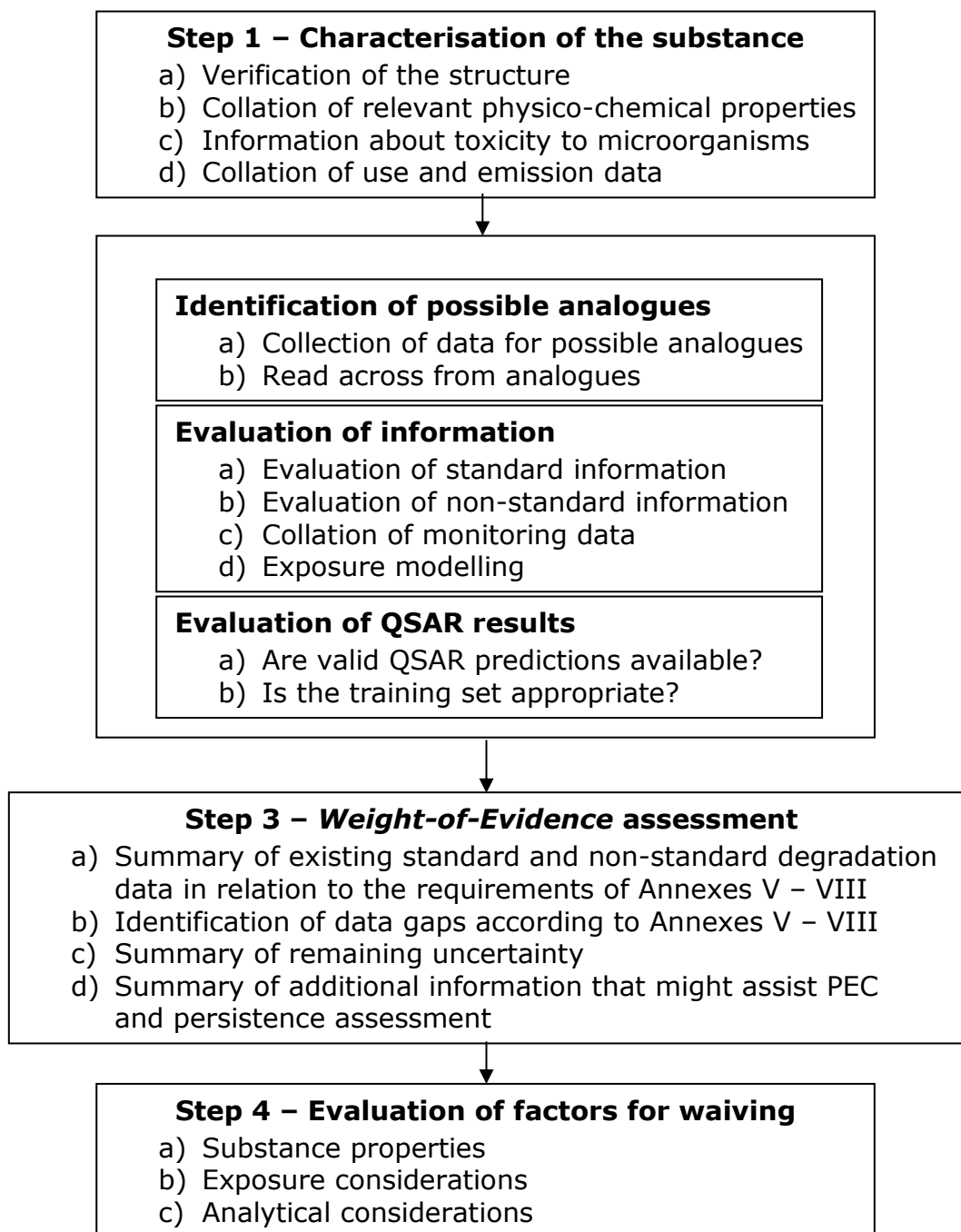
15 The following scheme outlines a systematic approach how to use all available
16 degradation data on a *Weight-of-Evidence* decision ([Figure R.7.9–2](#)). It provides a step-
17 wise procedure for the assessment of different types of information, which might be
18 helpful to come to an overall conclusion that may include the requirement for additional
19 data. The scheme proposes a flexible sequence of steps, the order of which depends on
20 the quality and quantity of data. Step 1, which is a collection of information on physico-
21 chemical properties rather than an assessment of available information, is a prerequisite
22 for the further assessment of other information. All steps are associated with three
23 distinct activities: (i) the gathering of information, (ii) the evaluation of the quality of a
24 distinct piece of information, and finally (iii) the overall assessment of all available
25 information.

26 *Weight of Evidence* is a decision-making activity aiming at concluding on degradation of
27 a substance based on integration of information from different sources and various
28 aspects of uncertainty. It will often require expert judgement. To make this expert
29 judgement transparent and comprehensible it is essential that all information used, all
30 steps carried out in the evaluation process and all conclusions drawn are fully
31 documented and justified.

1 **Figure R.7.9—2 A Weight-of-Evidence Approach for Assessing Degradation**

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1 **Step 1 – Characterisation of the Substance**

2 Initially it is important gather as much data about the substance. This includes its CAS
3 number, chemical formulae, chemical structure, purity and whether there are any known
4 isomers.

5 Information on the following physico-chemical properties determined using the relevant
6 OECD technical guidelines identified is also desirable: vapour pressure, water solubility,
7 absorption - desorption using a batch equilibrium method, partition coefficient (n-
8 octanol/water), dissociation constants in water, partition coefficient (n-octanol/water) -
9 HPLC method, and Estimation of the Adsorption Coefficient (K_{oc}) on Soil and on Sewage
10 Sludge using High Performance Liquid Chromatography (HPLC).

11 Prior to assessing existing biodegradability data or requiring new biodegradation data it
12 is important to assess information about the substances toxicity to microorganisms. Data
13 from tests such as the activated sludge respiration inhibition test (OECD 209) are
14 appropriate.

15 Finally, any information that can be gathered about the use and emission of the
16 substance will help determine the potential relevance of existing data, and it will also
17 assist in prioritising additional degradation data requirements in Steps 2 and 3.

18 **Step 2 – Evaluation of factors for Waiving**

19 There are a number of factors for waiving testing based on substance and exposure
20 properties. These include:

- 21 • Biodegradability studies are not required for inorganic substances as they
22 cannot be tested for biodegradability.
- 23 • Hydrolysis tests are not required for readily biodegradable substances, as the
24 test will provide little additional information since rapid mineralisation of the
25 substance in the environment is assumed. In addition, if the substance does
26 hydrolyse this will occur in the ready biodegradation test and if it is
27 accompanied with mineralisation >60% then it is unlikely that any terminal
28 degradation products will exist. Hydrolysis tests are also difficult to conduct
29 with substances that are highly insoluble in water and their relevance is likely
30 to be low as such substances are unlikely to be associated water in the
31 environment.
- 32 • Simulation studies in surface water, soil and sediment are not required for
33 readily biodegradable substances as it is assumed that they will undergo rapid
34 degradation in the environment. Specific simulation studies are also not
35 required if direct or indirect exposure is unlikely. When it is not necessary for
36 PBT-assessment (e.g. the substance not either vB or not B or T) it may not be
37 required for risk assessment purposes either if the exposure is so low that no
38 refinement of the PEC_{regional} is indicated.
- 39 • Identification of degradation products are not required for readily
40 biodegradable substances as the 60% pass criteria assumes that the
41 remaining 40% has been assimilated into new microbial biomass and any
42 transient metabolites have been degraded.

1 Step 3 – Information gathering

2 For substances where known analogues exist, relevant physico-chemical and degradation
3 data need to be collated. In the case of biodegradation, where the biochemistry of
4 biodegradation is known, analogues can include substances that are known to be
5 degraded through identical mechanisms e.g. β -oxidation of certain hydrocarbons. It is
6 also known that different pathways for biodegradation can exist for closely related
7 analogues. Particular care will need to be taken with respect to differences in physico-
8 chemical properties as simple structural changes to a chemical molecule can alter the
9 behaviour of the substance in the environment.

10 In the substance dossier mixed types of information is usually available. The information
11 could be arranged according to information type each with its characteristics according
12 to accuracy, interpretability and relevance for the particular regulatory type of decision:

- 13 • monitoring studies and field studies,
- 14 • simulation test data,
- 15 • inherent biodegradability data,
- 16 • ready and modified ready biodegradability studies
- 17 • enhanced screening studies indicating lack of persistence
- 18 • non-standard test data (including pure microbial culture data)
- 19 • poorly described test data
- 20 • marine biodegradability data
- 21 • abiotic degradation data
- 22 • sewage treatment plant removal data
- 23 • QSAR data

24 It should always be considered that a combination of information sources should give the
25 most comprehensive assessment. When no reason can be found for lack of agreement
26 between relevant and reliable testing and non-testing data then the non-testing data
27 should normally not be decisive.

28 For substances where a range of degradation data is available, a *Weight of Evidence*
29 approach should be employed. When more than one simulation test result is available, a
30 suitable degradation half-life in the higher end of the observed range should be selected
31 taking into account the realism, relevance, quality and documentation of the studies in
32 relation to environmental conditions (e.g. test substance concentration and
33 temperature). When more than one screening test result is available, positive test
34 results should be considered valid, irrespective of negative results, when the scientific
35 quality is good and the test conditions are well documented, i.e. guideline criteria are
36 fulfilled, including the use of non-adapted inoculum (cf. OECD, 2001c). It should also be
37 noted that the results of screening tests may be negative due to toxic effects of the test
38 substance, whereas simulation tests employing a low concentration of the test substance
39 may give a more realistic estimate of the degradation in the environment.

1 When judging poorly reported or non-standard data (e.g. biochemical studies using
2 mixed or pure culture) then the following information should be extracted in order to
3 maximise the potential use of the data:

- 4 • The source and density of the inoculum should be defined; ideally this should
5 not be taken from an industrial site and the density should be equivalent to
6 that of a ready biodegradation test.
- 7 • Any pre-treatment of inoculum including pre-exposure to the test substance.
- 8 • The test substance, its purity and the concentration that is used in the test.
- 9 • The motivation for the study (e.g. isolation of competent microorganism or
10 determination of the pathway for biodegradation)
- 11 • The analyte being measure (e.g. parent compound, DOC, BOD or CO₂
12 evolution)
- 13 • Either a removal percentage over a define time period or a degradation rate.

14 An example review of published literature has been provided for Toluene in the case
15 studies provided with this guidance.

16 For substances that have been identified as readily biodegradable, any known
17 metabolites of these compounds can also be considered as readily biodegradable. The
18 public domain literature and the Minnesota Biodegradation Database might assist in
19 identifying such metabolites (<http://umbbd.ethz.ch/>).

20 For substances where monitoring data exist it is important to gather these data together
21 with appropriate metadata (e.g. sample points, dates, times, frequency, relevant
22 hydrogeological and meteorological data etc.) associated with the monitoring
23 programme.

24 Using the information gathered up to this point, it may be possible to model the
25 exposure of the substance at this stage to 1) identify environmental compartments of
26 concern to determine the relevance of the available information and 2) to determine
27 whether any available monitoring data supports the exposure model predictions.

28 The reliability of the prediction of a QSAR model should be taken into account based on
29 an evaluation of the validation status for the models (sensitivity and specificity etc.) and
30 based on an evaluation of whether the prediction falls within the applicability domain of
31 the model. Similar considerations apply when judging the robustness of chemical
32 categories relating to degradability. Often use of predictions from more QSAR models – if
33 feasible supported by read-across or chemical categorisation - may enhance the overall
34 possibility to make a robust overall prediction of ready biodegradability (see also Section
35 [R.7.9.4.1](#)).

36 By using all available degradability test data, it may be possible to establish a
37 comprehensive evaluation of the degradability of the substance. For example in
38 particular ready biodegradation test data that demonstrated significant mineralisation
39 (>40%) but fails to reach the pass criterion for ready biodegradability may exist. In
40 certain cases where such data are available together with other evidence of
41 biodegradation such as through the use of a valid QSAR and/or other test data that

1 indicating rapid degradation without the presence of any significant metabolites, then
2 this could together be used as evidence for non-persistence.

3 **Step 4 – Weight of Evidence Assessment**

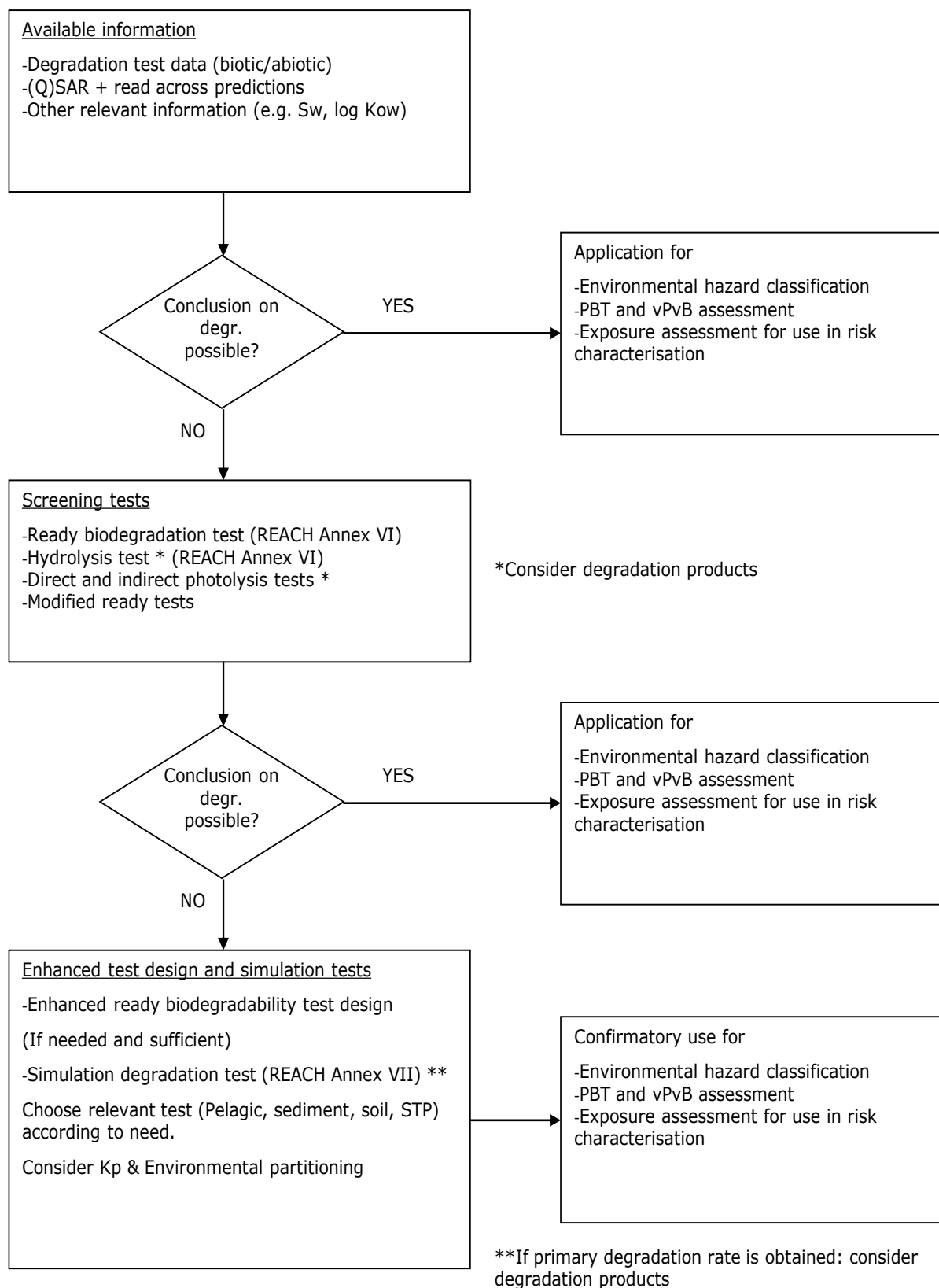
4 Once all the relevant information has been gathered in relation to the requirements of
5 REACH, it needs to be determined whether sufficient information exists to draw
6 conclusions for each of the three regulatory endpoints: hazard assessment (e.g. for
7 classification and labelling), exposure assessment (for determination of the PEC) and
8 persistence assessments (for PBT/vPvB assessment).

9 If insufficient information exists then the data gaps for each regulatory endpoint need to
10 be identified together with a summary of any remaining uncertainty. For substances at
11 tonnages that require simulation data, the most appropriate environmental
12 compartments to support both P/vP assessment and exposure assessment should be
13 identified.

14 **R.7.9.6 Integrated Testing Strategy (ITS) for** 15 **degradation/biodegradation**

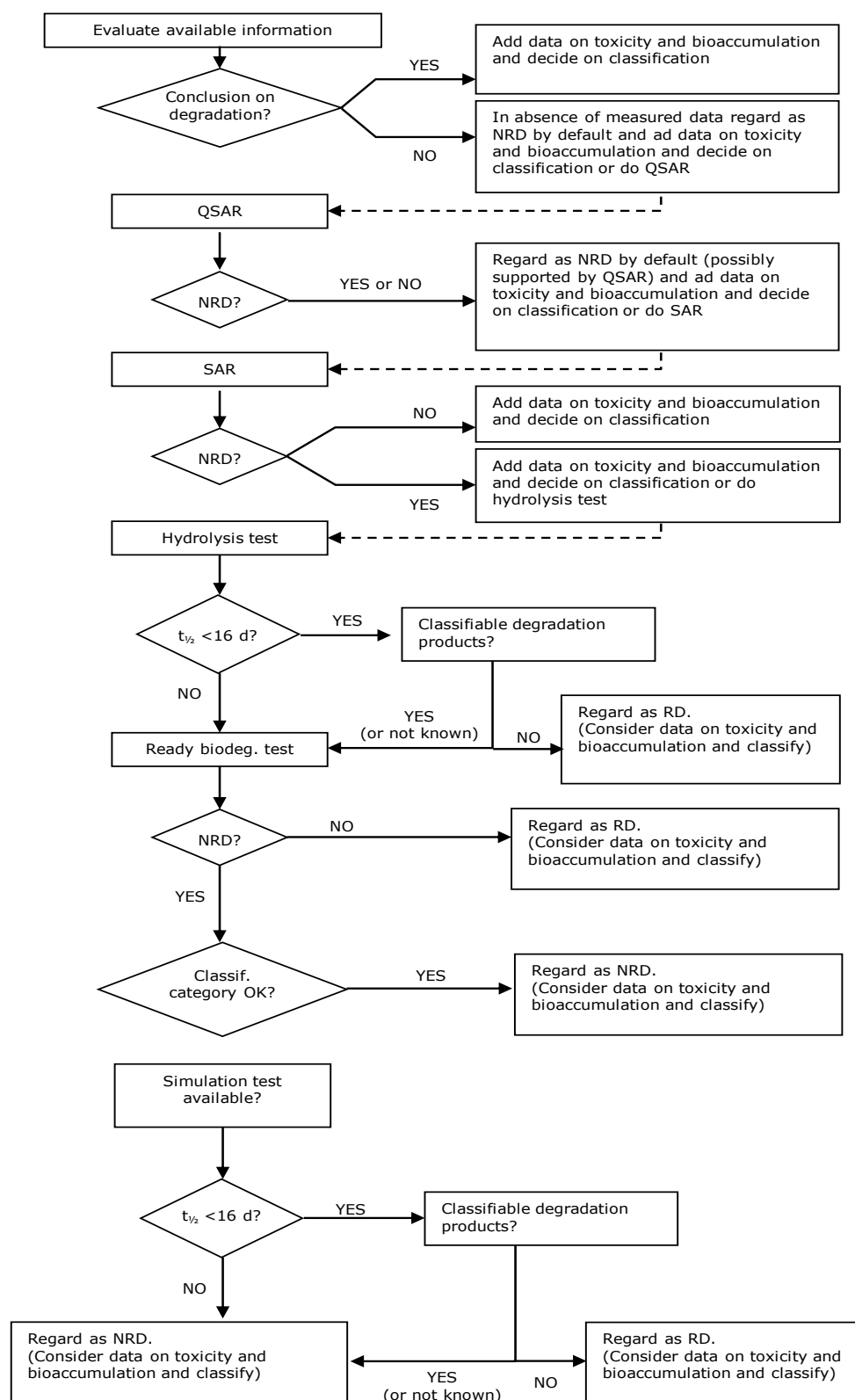
16 The ITS presented in [Figure R.7.9–3](#) attempts to summarise the approach required to
17 maximise the use of degradation data against all three regulatory endpoints. The
18 scheme starts with collating all available information before requiring tests at the
19 screening and simulation test levels.

1 **Figure R.7.9—3 Overview decision scheme on degradation for the three**
 2 **regulatory needs Environmental hazard classification, PBT/vPvB assessment**
 3 **and Exposure assessment for use in risk characterisation**



1 **R.7.9.6.1 Classification and Labelling**

2 An ITS to determine the suitability of degradation data on classification and labelling is
 3 provided in [Figure R.7.9–4](#).

4 **Figure R.7.9–4 An ITS for the use of degradation data in C&L.**

1 Hazard classification should be considered regardless of the tonnage level and based on
 2 available information (GHS, Annex 9 [1]). Information on ready biodegradability is
 3 required already at a tonnage level of 1 t per year for the purpose of environmental
 4 hazard classification of a substance (OECD Test Guidelines 301 A-F, or OECD TG 310, or
 5 QSAR predictions). The choice between the six OECD 301 test guidelines, or the OECD
 6 TG 310 *head space variant* of OECD TG 301B, depends on the characteristics of the
 7 substance (see OECD introduction 'Degradation of Organic Chemicals' [2] and
 8 information in the individual test guidelines).

9 **R.7.9.6.2 Chemical safety assessment**

10 A chemical safety assessment (CSA) under REACH, including environmental hazard
 11 assessment and PBT/vPvB assessment, only has to be carried out for substances with an
 12 annual tonnage exceeding 10 tonnes per registrant. An exposure assessment (PEC
 13 characterisation) as well as a risk characterisation (PEC/PNEC ratios) has to be carried
 14 out if the substance meets the criteria for any of the Article 14(4) hazard classes,
 15 categories or properties.

16 [Table R.7.9–3](#) shows the relevant information on the ITS on degradation and which at a
 17 minimum should be available for each annual tonnage level above 10 tonnes per
 18 registrant.

19 **Table R.7.9–3 Required test data of interest for the ITS on degradation**

Tonnage band (t/y/registrant)	Required degradation data	Other relevant information
10-100	Ready biodegradability Hydrolysis	Log K _{ow} Vapour pressure Water solubility Adsorption/desorption
100-1000	Ready biodegradability Hydrolysis Simulation of biodegradability in water ¹ Simulation of biodegradability in sediment ² Simulation of biodegradability in soil ³	Log K _{ow} Vapour pressure Water solubility Adsorption/desorption Dissociation constant Degradation products BCF ⁴

<p>>1000</p>	<p>Ready biodegradability</p> <p>Hydrolysis</p> <p>Simulation of biodegradability in water¹</p> <p>Simulation of biodegradability in sediment²</p> <p>Simulation of biodegradability in soil³</p> <p>Further testing shall be proposed if the CSA indicates a need for additional data on the degradation of the substance</p>	<p>Log K_{ow}</p> <p>Vapour pressure</p> <p>Water solubility</p> <p>Adsorption/desorption</p> <p>Dissociation constant</p> <p>Degradation products</p> <p>BCF⁴</p>
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- 1 1. Not needed if the substance is highly insoluble in water and/or is readily
2 biodegradable (see Section [R.7.9.2](#))
- 3 2. Not needed if the substance is readily biodegradable and/or direct and indirect
4 exposure of sediment is unlikely (see Section [R.7.9.2](#))
- 5 3. Not needed if the substance is readily biodegradable and/or direct and indirect
6 exposure of soil is unlikely (see Section [R.7.9.2](#))
- 7 4. Not needed if the substance has a low potential for bioaccumulation (for instance
8 a log K_{ow} <3) and/or a low potential to cross biological membranes and/or direct
9 and indirect exposure of the aquatic compartment is unlikely.

10 An exposure assessment can be carried out on the basis of information on ready
11 biodegradability. If an environmental risk assessment of a substance leads to the
12 conclusion *no risk*, using only information on ready biodegradability, then there is no
13 need for further testing of the biodegradability.

14 However, further testing of the biodegradability (and/or ecotoxicity) of the substance
15 may be required, if the risk assessment indicates a potential risk to one or more
16 environmental compartments.

17 In the exposure assessment, rates for the biodegradation in the various compartments
18 are used for the derivation of the associated PEC-values. These compartments include:

- 19 • Sewage treatment plant
- 20 • Freshwater
- 21 • Freshwater sediment
- 22 • Marine water
- 23 • Marine water sediment
- 24 • Soil

25 Additional consideration will be needed to whether or not inherent biodegradation test
26 data (OECD 302) or sewage treatment simulation test data are required to refine the
27 PEC_{local} and PEC_{regional}. These tests are not currently required under the REACH

1 Annexes but can be used to refine the PEC and may help to determine whether either
2 simulation tests are required or which simulation test may be the most relevant.

3 [Table R.7.9–4](#) shows an approach for selection of additional biodegradability tests,
4 which may either simulate realistic conditions in the external environment (freshwater,
5 marine or soil) or simulate the biodegradation and removal of the substance in the
6 sewage treatment plant (estimates of effluent concentration, e.g. based on CAS test).

7 **Table R.7.9–4 Selection of appropriate biodegradation studies for PEC**
8 **assessments**

Relevant environmental compartment ¹	Recommended biodegradation studies
Freshwater	Freshwater simulation test (e.g. OECD 309) and/or CAS test (OECD 303)
Freshwater sediment	Freshwater water/sediment simulation test (e.g. OECD 308) and/or CAS test (OECD 303)
Marine water	Marine water simulation test (e.g. OECD TG 309) and/or CAS test (OECD 303)
Marine water sediment	Marine water sediment simulation test (e.g. OECD 308) and/or CAS test (OECD 303)
Soil	Soil simulation test (e.g. OECD 307)

9 ¹: The relevant environmental compartment(s) may be identified on the basis of an
10 analysis of the intrinsic properties of the substance, modelling of transport and fate.

11 **R.7.9.6.3 PBT/vPvB assessment**

12 The information gathered through the steps outlined in the previous sections enables an
13 assessment to be carried out for PBT/vPvB. Guidance for this is given in Chapter R.11 of
14 the *Guidance on IR&CSA*.

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Appendices to Section R.7.9

- Appendix R.7.9—1 International Guidelines for Assessing Biodegradability**
- Appendix R.7.9—2 Reporting Requirements**
- Appendix R.7.9—3 Testing the Biodegradability of Poorly Water Soluble Substances**
- Appendix R.7.9—4 Guidance for Testing of multi-constituent substances (e.g. UVCB Petroleum Substances) for biodegradation**

1 **Appendix R.7.9—1 International Guidelines for Assessing Biodegradability**

Method	Test duration	Inoculum	Test conditions	Measurements	Limitations
Ready Biodegradability Tests					
OECD 301A DOC die away (ISO 7827)	Up to 28 days	Micro-organisms ($\sim 10^7 - 10^8$ cells/ml) in surface waters, unchlorinated sewage treatment works effluents or activated sludge. Not pre-adapted inoculum	Agitation in the dark or diffuse light under aerobic conditions at 20-24°C	DOC removal	Test substance has to be soluble, non-volatile, not sorbed to vessel or sludge and non-toxic at test conc.
OECD 301B CO ₂ evolution test (ISO 9439, OPPTS 835.3120)	Up to 28 days	Micro-organisms ($\sim 10^7 - 10^8$ cells/ml) in surface waters, unchlorinated sewage treatment works effluents or activated sludge. Not pre-adapted inoculum	Agitation in the dark or diffuse light under aerobic conditions at 20-24°C	CO ₂ production	Test substance must be non-toxic at test concentration.
OECD 301C Modified MITI Test	Up to 28 days	Micro-organisms ($\sim 10^7 - 10^8$ cells/ml) in surface waters, unchlorinated sewage treatment works or industrial effluents or activated sludge. Not pre-adapted inoculum	Agitation in the dark under aerobic conditions at 24-26°C	O ₂ uptake	Test substance has to be non-toxic at test concentration, subject to interference from nitrification.
OECD 301D Closed bottle test (ISO 10707)	Up to 28 days	Micro-organisms ($\sim 10^5$ cells/ml) in surface waters or unchlorinated sewage treatment works effluents Not pre-adapted inoculum	Agitation in the dark under aerobic conditions at 20-24°C	O ₂ uptake	Test substance has to be non-toxic at test concentration, subject to interference from nitrification.

Method	Test duration	Inoculum	Test conditions	Measurements	Limitations
OECD 301E Modified OECD screening test (ISO 7827)	Up to 28 days	Micro-organisms ($\sim 10^7 - 10^8$ cells/ml) in unchlorinated sewage treatment works effluents Not pre-adapted inoculum	Agitation in the dark or diffuse light under aerobic conditions at 20-24°C	DOC removal	Test substance has to be soluble, non-volatile, not sorbed to vessel or sludge and non-toxic at test conc.
OECD 301F Manometric respirometry test (ISO 9408)	Up to 28 days	Micro-organisms ($\sim 10^7 - 10^8$ cells/ml) in surface waters, unchlorinated sewage treatment works effluents or activated sludge Not pre-adapted inoculum	Agitation in the dark or diffuse light under aerobic conditions at 20-24°C	O ₂ uptake	Test substance has to be non-toxic at test concentration, subject to interference from nitrification.
OECD 310 (Headspace test) ISO 14593	Up to 28 days	Inoculum of aerobic mixed micro-organisms (approx $10^7 - 10^8$ cells/l). Not pre-adapted inoculum	Batch culture, aerated aquatic test using the test substance as the sole carbon source at 20-25°C. Assesses ultimate biodegradation.	CO ₂ production in sealed vessels giving % degradation	Test substance must be non-toxic at test concentration.

Simulation Tests for Freshwater and Sediment Systems

Method	Test duration	Inoculum	Test conditions	Measurements	Limitations
OECD 308 Aerobic and anaerobic transformation in aquatic sediment systems	Less than 100 days	Microorganisms in sediment (not pre-adapted)	Static test with natural water and sediment, with non-volatile ¹⁴ C labelled compounds at natural levels.	Chemical analysis of transformation products or ¹⁴ CO ₂ analysis where labelling used.	Simulates suspended sediment only. Test substance has to be non-toxic, non-volatile and soluble. Site specific with respect to sediment. Sorption to sediment may be misleading if ¹⁴ C not used.
OECD 309 Aerobic mineralisation in surface water	Up to 90 days for the batch test	Microorganisms in surface water (not preadapted) May include suspended sediment and/ or semi-continuous operation			
ISO 14592-1 (OPPTS 835.3170)	No fixed duration	Micro-organisms in surface water samples filtered through 100 um filter for a 'pelagic test' which may be amended with an aerobic sediment slurry from the study site for a 'suspended sediment test'.	Agitation in the dark or diffuse light under aerobic conditions at field temperature or 20-25°C	Specific chemical or radio-chemical analysis (and DOC or TOC if possible) giving 1 st order rate const.	Test substance has to be non-toxic, non-volatile and soluble. Site specific with respect to sediment. Sorption to sediment may be misleading if ¹⁴ C not used.

Method	Test duration	Inoculum	Test conditions	Measurements	Limitations
ISO 14592-2	No fixed duration but <60 days	Micro-organisms in surface water	Natural diffuse daylight or constant illumination of artificial white light (400-700 nm) with an energy of 50 uE/m ² /s at the water surface	Specific chemical or radio-chemical analysis giving 1 st order rate const.	Test substance has to be non-toxic, non-volatile and soluble. Site specific with respect to sediment if used – glass beads may not be representative of sediment. Sorption to sediment may be misleading if ¹⁴ C not used.
OPPTS 835.3180 Sediment/ water microcosm	Less than 60 days	Natural microbial assemblage.	Sediment microcosms using intact cores with (semi) continuous water replacement. ¹⁴ C labelling at environmentally realistic levels recommended.	Chemical analysis of transformation products or ¹⁴ CO ₂ analysis where labelling used.	Test substance has to be non-toxic, non-volatile and soluble. Site specific with respect to sediment. Sorption to sediment may be misleading if ¹⁴ C not used.
Sewage Treatment Simulation Tests					
OECD 303A Aerobic sewage treatment: coupled unit test (ISO 11733)	Up to 12 weeks	Aerobic sewage	Elimination of test substances (20 mg.l ⁻¹ DOC) from continuously fed laboratory scale coupled sewage treatment units.	DOC or COD giving % degradation	Test substance must be water soluble and non-volatile.

Method	Test duration	Inoculum	Test conditions	Measurements	Limitations
<p>OECD 314 Simulation tests to assess the biodegradability of chemicals discharged in wastewater</p> <p>A: Biodegradation in a Sewer System Test</p> <p>B: Biodegradation in Activated Sludge Test</p> <p>C: Biodegradation in Anaerobic Digester Sludge Test</p> <p>D: Biodegradation in Treated Effluent-Surface water Mixing Zone Test</p> <p>E: Biodegradation in Untreated Wastewater-Surface water Mixing Zone Test</p>	<p>314 A: typically < 96 hrs but can be extended</p> <p>314 B: typically 28 days but can be extended or shortened</p> <p>314 C: typically 60 days but can be extended or shortened</p> <p>314 D: typically 28 days but can be extended or shortened</p> <p>314 E: typically 28 days but can be extended or shortened</p>	<p>314 A: raw wastewater</p> <p>314 B: activated sludge</p> <p>314 C: anaerobic sludge</p> <p>314 D: surface water amended with treated effluent</p> <p>314 E: surface water amended with untreated effluent</p>	<p>Open batch system or sealed, flow-through batch system.</p> <p>For volatile test materials, appropriate modification must be made to quantify losses due to volatilisation.</p>	<p>Specific chemical or radio-chemical analysis.</p>	<p>This guideline describes methods for determining the extent and kinetics of primary and ultimate biodegradation of organic chemicals during key phases of wastewater transit as well as treatment and environmental release. It is relevant for organic chemicals whose route of entry into the environment begins with their discharge to wastewater. It should not be used as a replacement for simulation tests for degradation in environmental compartments such as surface water, sediment or soil.</p>

Method	Test duration	Inoculum	Test conditions	Measurements	Limitations
Primary Biodegradability Tests					
OPPTS 835.3220 Porous Pot Method,	At least 21 days	Activated sludge mixed liquor from a domestic plant.	Test and control pots filled with inoculum and 10-20 mgC/l test substance.	Primary biodegradation determined by test substance removal, DOC analysis provides measure of ultimate biodegradation.	Test substance has to be soluble, non-volatile, not sorbed to vessel or sludge and non-toxic at test conc.
Simulation Tests for Marine Waters					
OECD 306 (ISO 7827 and 10707, OPPTS 835.3160)	Up to 60 days	Micro-organisms ² in test seawater Not pre-adapted inoculum	Agitation in the dark or diffuse light under aerobic conditions at 15-20°C. Concentrations 5-40 mg DOC.l ⁻¹	DOC	Test substance must be non-toxic at test concentrations, soluble and not sorbed by vessel. Closed bottle test subject to interference from nitrification. High nutrient concentrations with respect to seawater
Simulation Tests for Soil					

Method	Test duration	Inoculum	Test conditions	Measurements	Limitations
OECD 307 Aerobic and anaerobic transformation on soil	Up to 120 days, longer under some circumstances				
Inherent Biodegradation Tests – Water					
OECD 302A Modified SCAS test (OPPTS 835.3210)	Months (often up to 120 days).	Settled domestic sewage and activated sludge. Inoculum to be sourced from a domestic treatment plant	Test substance (20 mg DOC.l ⁻¹) aerated with settled domestic sewage and activated sludge (ca. 2500 mg.l ⁻¹ TSS) for 23h at 20-25°C. Aeration stopped, sludge settled and supernatant removed. Fresh sewage and test substance are added and the cycle repeated. ¹⁴ C-radiolabelled substances can be used for increased sensitivity.	DOC CO ₂ production in sealed vessels giving % degradation. Potential to measure ¹⁴ CO ₂	Test substance must be non-volatile, not lost by foaming and non-toxic at test conc. Sorption potential needs to be determined.
OPPTS 835.5045 Modified SCAS for insoluble and volatile substances	Months (often up to 120 days).	Settled domestic sewage and activated sludge.		CO ₂ production in sealed vessels giving % degradation Potential to measure ¹⁴ CO ₂	

Method	Test duration	Inoculum	Test conditions	Measurements	Limitations
OECD 302B Zahn Wellens (ISO CD9888) (OPPTS 835.3200)	28 days	Inoculum of 200 - 1000 mg.l ⁻¹ (TSS) of activated sludge. Unadapted or pre-adapted inoculum	Aerated batch culture, using the test substance as the sole carbon source (50 – 100 mg.l ⁻¹ DOC) and with the inoculum at 20-25°C. Assesses ultimate biodegradation.	DOC or COD or Specific analysis for primary transformations	Test substance must be non-volatile, not lost by foaming and non-toxic at test conc. Sorption potential needs to be determined.
OECD 302C MITI (II)	14-28 days	Aerobic mixed, specially grown, unadapted micro-organisms at 100 mg.l ⁻¹ (TSS, or approx. 3×10^7 - 3×10^8).	Agitated batch culture, using the test substance as the sole carbon source (30 mg ThOD.l ⁻¹) with inoculum. Assesses ultimate biodegradation.	O ₂ demand and possibly specific chemical analysis	Test substance must be non-volatile, not lost by foaming and non-toxic at test concentration.
OPPTS 835.3100 Aerobic aquatic biodeg	28 days after pre-adaptation	Pre-adapted inoculum	Agitated aerated aquatic test using test substance (10 mg.l ⁻¹ DOC) pre-adapted inoculum from a medium concentration of aerobic mixed micro-organisms at 20-25°C. ¹⁴ C labelled compounds may be used	DOC removal and CO ₂ evolution ¹⁴ C provides mass balance phase distribution data	Test substances must be soluble and non-volatile.

Method	Test duration	Inoculum	Test conditions	Measurements	Limitations
OPPTS 835.5045 Modified SCAS test for insoluble and volatile substances	40 to 120 days	Settled domestic sewage and activated sludge Unadapted or pre-adapted inoculum	Same principle as for OECD 302A but with a volatiles trap on the aeration unit and additional analytical requirements for trapped volatiles and sludge solids. 20 mg.l ⁻¹ DOC test concentration at 20-25°C. ¹⁴ C labelled compounds may be used.	DOC. Specific analysis can provide primary transformation data. Kinetic data and half-life determination available. >20% removal of DOC =inherent biodegradation, >70% =ultimate biodegradation.	Additional analytical requirements.
Inherent Biodegradation – Soil					
OECD 304A (ISO 14239 – biometer system) OPPTS 835.3300	Up to 64 days	Disturbed soil – alfisol, spodosol, entisol. In special cases can use soil with high silt fraction content or soil with high clay content (30%).		CO ₂ evolution giving % degradation	
Anaerobic Degradation Test Methods					
OECD 311 ISO 11734	Up to 60 days	Washed digester sludge at 1-3 /l in nutrient amended anaerobic medium, containing a redox indicator in sealed vessels.	Batch culture with test concentration of 20-100 mg.l ⁻¹ as OC, at 35°C. Assesses ultimate biodegradation	Total gas production (CH ₄ +CO ₂) using a pressure transducer and DIC	Test substance must be non-toxic at test concentration.

Method	Test duration	Inoculum	Test conditions	Measurements	Limitations
OPPTS 835.3400 Anaerobic biodegradability of organic substances	Up to 56 days.	Sludge from an anaerobic sludge digester. Recommendations are for a well-mixed primary sludge from a digester with a retention time of 15 to 25 days.	Test sample concentrations at around 50 mg.l ⁻¹ with tests carried out at 35°C.	CO ₂ and CH ₄ production.	Not applicable to toxic substances, reproducibility not yet fully defined. Uses high concentrations of test substances.

1 **Appendix R.7.9—2 Reporting Requirements**

2

3 **Hydrolysis Test Requirements (OECD 111)**

4 The test report should include the following information:

5 • Test substance:

6 - common name, chemical name, CAS number, structural formula
7 (indicating position of label when radiolabelled material is used) and
8 relevant physico-chemical properties;

9 - purity (impurities) of test substance;

10 - label purity of labelled substance and molar activity (where
11 appropriate).

12 • Buffer solutions:- buffers and waters used;- molarity and pH of buffer
13 solutions.

14 Test conditions:

15 - amount of test substance applied;

16 - solvents (type and amount) used for application of the test substance;

17 - volume of buffered test substance solutions incubated;

18 - description of the incubation system used;

19 - pH and temperature during the study;

20 - sampling times;

21 - method(s) of extraction;

22 - methods for quantification and identification of the test substance and
23 its hydrolysis products in the buffer solutions;

24 - number of replicates.

25 • Results:

26 - repeatability and sensitivity of the analytical methods used;

27 - recoveries;

28 - replicate data and means in a tabular forms;

29 - mass balance during and at the end of the studies (when labelled test
30 substance is used);

31 - results of preliminary test;

32 - discussion and interpretation of results;

1 - all original data and figures.

2 The following information is only required when the hydrolysis rate is determined:

- 3 • plots of concentrations versus time for the test substances and, where
4 appropriate, for the hydrolysis products at each pH value and temperature;
- 5 • tables of results of Arrhenius equation for the temperature 20 °C/25 °C, with
6 pH, rate constant [h^{-1} or day^{-1}], degradation half-life or DT50, temperatures
7 [°C] including confidence limits and the coefficients of correlation (r^2) or
8 comparable information;
- 9 • proposed pathway of hydrolysis.

10 **Ready biodegradability test requirements (OECD 301 series and OECD 310)**

- 11 • Test substance:
 - 12 - physical nature and, where relevant, physico-chemical properties;
- 13 • Test conditions:
 - 14 - inoculum: nature and sampling site(s), concentration and any pre-
15 conditioning treatment;
 - 16 - proportion and nature of industrial waste water in sewage, if known;
 - 17 - test duration and incubation temperature;
 - 18 - in the case of poorly soluble test substances, methods of preparation
19 of test solutions/suspensions;
 - 20 - test method applied; scientific reasons and explanation for any change
21 of procedure;
 - 22 - details of controls.
- 23 • Results:
 - 24 - data in tabular form;
 - 25 - any observed inhibition or toxicity;
 - 26 - any observed abiotic degradation;
 - 27 - specific chemical analytical data, if available;
 - 28 - analytical data on intermediates, if available;
 - 29 - the graph of percentage degradation against time for the test and
30 reference substances to include the lag phase, degradation phase, the
31 10-d window and slope (see Annex I for definitions);
 - 32 - percentage removal at plateau, at end of test, and/or after 10-d
33 window.
- 34 • Discussion of results

35

1 Marine Biodegradability Test Requirements (OECD 306)

- 2 • Test substance:
 - 3 - physical nature and, where relevant, physico-chemical properties;
- 4 • Test conditions:
 - 5 - location and description of the sampling site; pollution and nutrient
 - 6 status (colony count, nitrate, ammonium, phosphate if appropriate);
 - 7 - characteristics of the sample (date of sampling, depth, appearance,
 - 8 temperature, salinity, DOC (optional), delay between collection and
 - 9 use in the test;
 - 10 - method used (if any) for ageing of the seawater;
 - 11 - method used for pre-treatment (filtration/sedimentation) of the
 - 12 seawater;
 - 13 - method used for DOC determination;
 - 14 - method used for specific analysis (optional);
 - 15 - method used for determining the number of heterotrophs in the
 - 16 seawater (plate count method or alternative procedure) (optional);
 - 17 - other methods (optional) used to characterise the seawater.
- 18 • Results:
 - 19 - the course of the degradation test is represented graphically in a
 - 20 diagram showing the lag phase (tL), slope, and time (starting from the
 - 21 end of the lag phase) to reach 50 per cent removal (t50). The lag
 - 22 phase may be estimated graphically as shown in the figure in the
 - 23 "Validity and interpretation of results" section or conveniently taken as
 - 24 the time needed for 10 per cent degradation;
 - 25 - percentage degradation measured after 60 days, or at end of test.
- 26 • Discussion of results.

27 Inherent Biodegradability Test Requirements (OECD 302 Series)

28 The test report should include the following information:

- 29 • Test substance:
 - 30 - physical nature and, where relevant, physico-chemical properties;
- 31 • Inoculum:
 - 32 - source, concentration, pre-treatment and status of adaptation.
- 33 • Test conditions:
 - 34 - analytical methods used;

- 1 - procedure control and compound used in the control.
- 2 • Results:
- 3 - biodegradation curve;
- 4 - toxicity evaluations;
- 5 - the degree of biodegradation attained at the end of the test after 28d,
6 or earlier if complete degradation is attained in less than 28d, as
7 "inherent biodegradability in the static test after x days";
- 8 - any significant difference between the DOC (or COD) in the first
9 sample at 3h after starting the test and the value calculated from the
10 amount of test compound added as "adsorbed by the activated sludge"
11 (OECD 302B);
- 12 - the adaptation phase (days), the biodegradation phase (days) and the
13 endpoint of biodegradation reached after x days as identified from the
14 biodegradation curve.
- 15 • Discussion of the results.
- 16
- 17

1 **Appendix R.7.9—3 Testing the Biodegradability of Poorly Water** 2 **Soluble Substances**

3

4 This appendix discusses the technical issues associated with conducting biodegradability
5 assays with poorly water-soluble substances and the data-reporting requirements that
6 would improve confidence in the data generated for such substances. The OECD and ISO
7 Guidance 10634 (1995) for testing poorly water-soluble substances will form the basis of
8 discussion. Whilst the focus of this document will be towards methods for assessing the
9 ready biodegradability of poorly water-soluble substances (OECD 301 series and the
10 OECD 310 test) the issues equally apply to other biodegradability assays.

11 *OECD Evaluation of the Biodegradability of Poorly Soluble Substances*

12 OECD requires that when assessing biodegradability of poorly soluble compounds OECD
13 the following aspects should receive special attention (OECD, 1992: Annex III):

- 14 • While homogeneous liquids will seldom present sampling problems, it is
15 recommended that solid materials be homogenised by appropriate means to
16 avoid errors due to non-homogeneity. Special care must be taken when
17 representative samples of a few milligrams are required from multi-
18 constituent substances or substances with large amounts of impurities.
- 19 • Various forms of agitation during the test may be used. Care should be taken
20 to use only sufficient agitation to keep the substance dispersed, and to avoid
21 overheating, excessive foaming and excessive shear forces.
- 22 • An emulsifier which gives a stable dispersion of the substance may be used. It
23 should not be toxic to bacteria and must not be biodegradable or cause
24 foaming under the test conditions.
- 25 • The same criteria apply to solvents as to the emulsifiers.
- 26 • It is not recommended that solid carriers be used for solid test substances but
27 they may be suitable for oily substances.
- 28 • When auxiliary substances such as emulsifiers, solvents and carriers are used,
29 a blank run containing the auxiliary substance should be performed.
- 30 • Any of the four respirometric tests (301 B, 301 C, 301 D, 301 F) can be used
31 to study the biodegradability of poorly soluble compounds.

32 Whilst OECD raise a series of valid issues that require careful considerations in testing
33 the biodegradability of poorly soluble substances they do not constitute explicit
34 guidance. The only critical guidance provided is the applicability of a restricted range of
35 the 301 test series (point 7) and the requirement of additional control vessels where
36 emulsifiers, solvents and carriers are used (point 6). Tests conducted with draft OECD
37 310 test "Ready Biodegradability – CO₂ in sealed vessels (Headspace Test)" are also
38 suitable for assessing the biodegradability of poorly soluble substances.

39 Whilst advocating the use of emulsifiers, solvents and carriers, none are specifically
40 identified and no guidance is provided regarding the acceptable level of each that can be
41 introduced into the test system. Consequently, numerous approaches of introducing the

1 test substance can be applied and this will make it difficult to identify a set of core
2 acceptable or workable solutions.

3 *ISO Guidance for the preparation and treatment of poorly water-soluble organic*
4 *compounds for the subsequent evaluation of their biodegradability in aqueous medium*

5 In 1995 the International Standards Organization (ISO) concluded that the development
6 of a single method for evaluating the biodegradability of poorly water-soluble organic
7 substances might not be realized in the immediate future. Consequently, ISO proposed
8 a series of methods where the final selection was based on a judgment of the physico-
9 chemical properties of the test substance (ISO, 1995).

10 The ISO standard (1995) addressed four techniques for preparing poorly water-soluble
11 substances and introducing them into the test apparatus. It must be noted that water-
12 soluble test substances are usually introduced into the test medium via a concentrated
13 stock solution. The methods proposed by ISO for poorly soluble substances were 1)
14 direct addition, 2) ultrasonic dispersion, 3) adsorption on an inert support, and 4)
15 creating a dispersion or emulsion. All of these techniques proposed by ISO are suitable
16 for including within the OECD 301 and 310 test guidelines. ISO does not provide any
17 advice about the use of suitable poorly soluble reference standards. Each of the ISO
18 methods will be described below with a brief commentary or assessment.

19 Direct addition

20 ISO proposed introducing the test compound by either 1) weighing the substance
21 directly into the test vessel, 2) weighing the test compound on to an inert support
22 (typically a glass cover slip or piece of foil) and introducing this into the test vessel, or 3)
23 preparing a solution of the test substance in a volatile solvent and removing the solvent
24 prior to testing.

25 Direct addition is applicable for a variety of substances e.g. crystalline solids and non-
26 viscous liquids. These are introduced using either high precision micro-pipettes or direct
27 weighing. In the case of direct weighing some replicate-to-replicate variability can be
28 expected for crystalline compounds as they are usually being introduced at the very low
29 mg weight range. Whilst direct pipetting using viscous liquids can be problematic, the
30 use of a cover slip or foil can overcome this. However care should be taken to ensure
31 that the cover slip remains face up, if this becomes inverted then the microbiota will not
32 be able to access the test substance.

33 It must be noted that control flasks will be needed where carrier solvents have been
34 used to ensure that all the solvent has been eliminated. In this case the same volume of
35 the solvent needs to be introduced into the test system as in the test flask, but without
36 the test substance. Even low levels of respiration associated with the solvent will need
37 to be accounted for when interpreting data from the test flasks. Whilst controls should
38 be used for cover slips etc. it is unlikely that any background respiration will be
39 observed.

40 Direct addition, particularly via direct weighing (or pipetting) or using a support, should
41 act as a 'bench mark' and be applied in the assessment of all poorly water-soluble
42 substances i.e. they should be used in parallel to any of the other guidance methods
43 recommended by ISO. Direct addition is likely to give the most conservative estimate of
44 biodegradation.

1 Ultrasonic dispersion

2 ISO (1995) recommend that a dispersion of the compound can be prepared using an
3 ultrasonic probe prior to introducing it into the test vessel. Specific guidance are
4 provided with respect to the frequency of the ultrasonication required to make a 20
5 times concentrated stock solution, however total carbon analysis is required to confirm
6 the concentration achieved.

7 It must be noted that this approach is not suitable for substances that undergo thermal
8 decomposition and that a stable emulsion is rarely formed. Consequently, this may not
9 be the most appropriate approach recommended within the ISO guidance. This is
10 particular true when stable emulsions cannot be formed and large numbers of sacrificial
11 test flasks are being prepared as the possibility exists for introducing reduced
12 concentrations to each flask with time i.e. a concentration gradient. If this technique is
13 to be applied to tests using sacrificial analysis (e.g. OECD 310) the test flasks need to be
14 sacrificed randomly for analysis at each time point.

15 Adsorption on to an inert support

16 ISO (1995) recommend the use of silica gel, glass filter or any other non-biodegradable
17 inert supports that do not release organic carbon into the test media. Supporting
18 evidence is required to demonstrate that the support is inert and carbon free and the
19 amount of support used should be minimal. Silica-based gels that are used for
20 chromatography represent an inert support that has been used successfully.

21 The test compound is usually introduced into the inert support at the required
22 concentration via a carrier solvent (e.g. acetone or dichloromethane). Rotary
23 evaporation and oven drying are then used to remove the solvent. A parallel procedure
24 is required using the inert support and carrier solvent without the test substance for use
25 in the control test flasks. Inert supports can also be used with insoluble solids.

26 Prior to testing the carbon level of the inert support containing the test substance or the
27 specific substance contained in the inert support needs to be quantitatively determined
28 and compared to nominal. The required amount of the inert support can then be directly
29 weighed into the test vessel. Any biodegradation of the solvent should be taken into
30 account through the use of parallel control vessels.

31 This procedure is applicable for compounds that will not be lost during the rotary
32 evaporation and oven drying procedures. It does enable the amount of material to be
33 directly weighed into the test flask to be increased thus increasing accuracy between
34 replicate test flasks.

35 Dispersion with an emulsifying agent.

36 ISO (1995) recommend using emulsifying agents to enhance the availability of the
37 poorly soluble test substance that are non-biodegradable and non-toxic under the
38 conditions of the biodegradation test. Synperonic PE/P94, Synperonic PE/P103 or Tween
39 85 have been identified as commercial substances that could be used as emulsifying
40 agents. Carrier solvents that are also non-toxic and non-biodegradable are also required
41 to form these emulsions.

1 ISO recommends that three emulsions be prepared prior to selecting the most
2 homogeneous emulsion for use in the biodegradation test. Very clear guidance is also
3 provided that states that the degradation observed in the control vessel (solvent and
4 emulsifier with no test compound) must not exceed 10% of the degradation observed in
5 the test flasks for the test to be consider valid.

6 Supporting evidence should be provided to demonstrate that neither the solvent or the
7 emulsifying agents are toxic to microbes or are biodegradable.

8 **Minimum Test and Data Requirements for Poorly Water Soluble Substances**

9 The following information should be reported:

- 10 • Information on the substance's water solubility, vapour pressure and
11 adsorption characteristics are essential.
- 12 • The solubility of the substance in other solvents should be stated (especially
13 those being used to disperse the substance in emulsifications and on to inert
14 supports).
- 15 • The chemical structure or formula should be identified in order to calculate
16 theoretical values and/or check measured values of parameters, e.g. ThOD,
17 ThCO₂, DOC, TOC, and COD. Information on the purity or the relative
18 proportions of major constituents of the test material is required in order to
19 interpret the results obtained, especially when the result lies close to the pass
20 level.
- 21 • Information on the toxicity of the test substance, or any emulsifiers or carrier
22 solvents, to bacteria may be very useful for selecting appropriate test
23 concentrations and preparation strategies.
- 24 • Any pre-treatment of the compound before the test.
- 25 • The method of test substance introduction should be described in detail with
26 supporting evidence especially regarding the use of solvents, emulsifiers and
27 inert supports.
- 28 • Nominal versus measured carbon concentrations where inert supports and
29 emulsions are used to generate concentrated stock preparations of the test
30 substance prior to use. This should include the degree of recovery.
- 31 • Duration of any pre-treatment.
- 32 • Rate of degradation observed in the control flasks (treatment minus test
33 substance).
- 34 • Suitable positive reference poorly soluble data (see below).

35

36 **Conclusions & Recommendations on biodegradability testing of poorly water-** 37 **soluble substances**

1 There is no single method for assessing the biodegradability of poorly water-soluble
2 substances. The state of the science has not changed since ISO published its guidance in
3 1995. A combination of approaches should be used and these should at the very
4 minimum be compared to biodegradation observed by direct addition. Direct addition will
5 usually provide the most conservative estimate of biodegradation.

6 Normal positive reference substances such as sodium acetate, sodium benzoate, aniline
7 or glucose offer little support in the assessment of poorly soluble substances other than
8 demonstrate that the inoculum is active. In order to 'bench mark' methods to assess
9 poorly soluble substances common poorly soluble reference substances should be used.
10 Two examples are provided in the Annexes of the ISO guidance. These are
11 biodegradation curves for diisooctylphthalate (where adsorption on inert support and
12 dispersion with an emulsifying agent enhances degradation compared to direct addition)
13 and anthraquinone (where adsorption on inert support and dispersion with an
14 emulsifying agent enhances degradation compared to direct addition). In both cases the
15 use of ultrasonication did not provide any significant benefit.

16 Greater confidence in the methods for increasing the availability of poorly soluble
17 substances will be gained by using either diisooctylphthalate or anthraquinone as a
18 positive control. The reference control should be introduced to the test system by direct
19 addition and the choice of preparation. Therefore for any given biodegradation
20 assessment there will need to be the following series of flasks:

- 21 • Blank Control (inoculum & media with no test compound);
- 22 • Positive reference for biodegradation (sodium acetate, sodium benzoate,
23 aniline or glucose);
- 24 • Poorly soluble positive control (either diisooctylphthalate or anthraquinone
25 introduced by direct addition);
- 26 • Test substance (introduced by direct addition for conservative assessment);
- 27 • Direct addition control;
- 28 • Test substance with choice of introduction (e.g. adsorption on an inert
29 support);
- 30 • Poorly soluble positive control using the same choice of introduction as the
31 test substance; and
- 32 • Choice of introduction control (e.g. inert support and solvent without the test
33 substance).

34 The above set of flasks appears onerous but they do not constitute a great deal of extra
35 effort or expense. The long-term value of providing the additional information will be one
36 of greater confidence in assessing poorly-soluble material against agreed bench mark
37 standards.

38

1 **Appendix R.7.9—4 Guidance for Testing of multi-constituent** 2 **substances (e.g. UVCB Petroleum Substances) for biodegradation**

3
4 For the guidance on PBT/vPvB assessment of UVCB and well-defined multi-constituent
5 substances, please, see Section R.11.4.2.2 in Chapter R.11 of the [Guidance on IR&CSA](#).

6 Due to derivation from natural crude oils and subsequent production from use of various
7 refining processes, petroleum substances are complex substances containing multiple
8 hydrocarbon constituents, and are often of variable composition. Many petroleum
9 substances are produced in very high tonnages to a range of technical specifications,
10 with the precise chemical composition of unique structures, rarely if ever characterised.
11 Since these materials are typically separated on the basis of distillation, the technical
12 specifications usually include a boiling point range. These ranges correlate with
13 approximate carbon number ranges, while the nature of the original crude oil and
14 subsequent refinery processing influence the types of hydrocarbon structures present.
15 The CAS definitions established for the various petroleum substance streams generally
16 reflect this detail, including final refinery process; boiling range; carbon number range
17 and predominant hydrocarbon types present.

18 For most petroleum substances, the complexity of the chemical composition is such that
19 that it is beyond the capability of routine analytical methodology to obtain complete
20 characterisation. There are techniques like GC-MS and GCxGC (CONCAWE, 2012) that
21 are useful, however, these are not routine. Typical substances may consist of
22 predominantly straight and branched chain alkanes, single and multiple naphthenic ring
23 structures (often with alkyl side chains), single and multiple aromatic ring structures
24 (often with alkyl side chains). As the molecular weights of the constituent hydrocarbons
25 increase, the number and complexity of possible structures (isomeric forms) increases
26 exponentially.

27 Environmental testing strategies for petroleum substances must necessarily reflect the
28 complexity of their composition. Reflecting the properties of the constituent
29 hydrocarbons, petroleum substances are typically hydrophobic and exhibit low solubility
30 in water. However, individual constituent hydrocarbons will exhibit a wide range of water
31 solubilities. When adding incremental amounts of a complex petroleum substance to
32 water, a point will be reached where the solubility limit of the least soluble constituent is
33 exceeded and the remaining constituents will partition between the water and the
34 undissolved hydrocarbon phases. Consequently, the composition of the total dissolved
35 hydrocarbons in water will be different from the composition of the parent substance.
36 The complex composition and generally low water solubility impacts the choice and
37 conduct of biodegradation studies. A further complication is the volatility of constituent
38 hydrocarbons, which shows a wide variation across the range of carbon numbers and
39 hydrocarbon structures present in petroleum substances. It has been the practise to
40 assess the inherent hazards of petroleum substances by conducting testing in closed
41 systems (going to great lengths to ensure that volatile losses are minimised), even
42 though under almost all circumstances of release into the environment, there would be
43 extensive volatilisation of many of the constituent hydrocarbons.

1 Biodegradation Testing Methods

2 Lower molecular weight hydrocarbons tend to be readily biodegradable in standard OECD
3 tests, and although biodegradability decreases as molecular weight increases
4 (corresponding to decreasing water-solubility and thus reduced bioavailability)
5 hydrocarbons are generally regarded as being inherently biodegradable. The initial
6 metabolites of hydrocarbons will be carboxylic acids and hence of less concern than the
7 parent structures.

8 Typically, laboratory studies of the aquatic biodegradability of petroleum substances
9 have evaluated the biodegradation potential of the whole substance, not just the portion
10 which is soluble in water. To achieve adequate sensitivity, most biodegradation tests
11 utilise higher concentrations of substances than would commonly be found in the
12 environment. For a petroleum substance, this means that there will be a large proportion
13 of the substance in the undissolved phase and hence, not fully available to the degrading
14 organisms. This will result in an underestimate of its true potential to biodegrade in the
15 environment. It is also likely that the rate of biodegradation will be affected; firstly, the
16 rate of biodegradation is likely to be limited by the rate of dissolution and solubility of
17 individual hydrocarbon constituents. Secondly, the fact that petroleum substances
18 contain a complex composition of constituents results in a stepwise, sequential
19 adaptation of the microorganisms to utilise individual hydrocarbons, again resulting in
20 deviation from 'typical' kinetics. For these reasons, typical logarithmic growth phase
21 (Monod) biodegradation kinetics (which are assumed to occur in RB tests) may not be
22 observed with petroleum substances, so that even if individual constituents are readily
23 biodegraded, the petroleum substance may not achieve the '10-day window' defined by
24 OECD (Deneer *et al.*, 1988).

25 Some modifications of test methods to enhance dissolution rates may improve this
26 situation. Guidance on approaches to the testing of poorly soluble substances has been
27 published (Whitehouse and Mallet, 1994). Experimental methods include ultrasonic
28 dispersion, addition of an inert dispersant or emulsifier to assist in dispersion, or addition
29 of the test substance on an inert support (to increase the surface area and hence aid
30 access of the microorganisms). See Section [R.7.9.4.1](#).

31 Several accepted methods for determining biodegradation potential are unsuitable for
32 poorly soluble substances (because they are based on measurement of total dissolved
33 organic carbon) or are unsuitable for volatile substances (because volatile constituents
34 are lost by evaporation, rather than biodegradation).

35 Three basic types of biodegradation test are used to estimate the relative
36 biodegradability of substances, viz. ready, inherent and primary biodegradation
37 methods. The use of these procedures in testing petroleum substances is dealt with in
38 the following paragraphs. Usually only ready biodegradation data are used for
39 classification, although, for example under the GHS scheme, other types of information
40 may be used e.g. simulation test data or primary degradation data and consideration of
41 degradation products.

42 The rationale for using standard laboratory tests to assess biodegradation potential of
43 mixtures has been discussed in an EU workshop (European Chemicals Bureau, 1996); it
44 was agreed that the available methods were suitable for evaluating the biodegradation
45 potential of mixtures comprising homologous series of hydrocarbons (like the petroleum

1 substances), although such methods were not judged generally applicable for mixtures
2 (e.g. preparations).

3 **Ready Biodegradability tests**

4 These are the most stringent of the commonly used laboratory tests, measuring
5 complete mineralisation or Ultimate Biodegradation of the test substance (oxidation to
6 carbon dioxide and water) using an unadapted inoculum⁴⁰ over a 28-day period. Ready
7 Biodegradability is defined in terms of the pass/fail criteria agreed for each of the six test
8 methods published by OECD (and subsequently adopted by the EU) (EU, 1967; OECD,
9 2000); in particular, the required level of biodegradation must be obtained within 10
10 days of 10% biodegradation being achieved. In all the 28-day biodegradation tests, the
11 mineral salts concentration, temperature and pH are tightly controlled, and the microbial
12 inoculum is not allowed to be pre-exposed to the test substance. In addition to the OECD
13 methods, there is a surrogate procedure whereby if the BOD5:COD ratio is 0.5 or higher,
14 the substance is regarded as being readily biodegradable. Because of the stringency of
15 these test methods, it is presumed that any substance demonstrating Ready
16 Biodegradability will be rapidly biodegraded if released into the aquatic environment.

17 The Modified Sturm test (OECD 301B) for non-volatile substances and the Respirometric
18 Method (OECD 301F) are the most commonly used methods for petroleum substances.
19 More recently a test guideline that addresses the biodegradation of volatile substances
20 has also been published, OECD 310.

21 **Inherent Biodegradability Tests**

22 These laboratory methods are less stringent than the Ready Biodegradability tests, and
23 hence, increase the likelihood of observing biodegradation within a specific test system.
24 The extent of complete oxidation of the test substance to carbon dioxide and water is
25 still measured.

26 Inherent Biodegradability is again defined in terms of the percentage biodegradation
27 recorded in the test; it can be presumed that substances demonstrating Inherent
28 Biodegradability will not persist if released into the aquatic environment.

29 Unfortunately, the currently available Inherent Biodegradation test methods defined by
30 OECD (OECD, 2000) are not suitable for petroleum substances (CONCAWE, 1992).
31 However, following development and validation of a new Inherent Biodegradation test
32 within ISO (Battersby, 1997; ISO, 1996), CONCAWE has recently validated a version of
33 this Headspace Method, adapted to make it more suitable for petroleum substances; the
34 results of this trial have recently been published (Battersby *et al.*, 1999). This method is
35 still under discussion as regards its suitability.

36 **Primary Biodegradation Tests**

37 Originally developed for evaluating the biodegradability of two-stroke outboard engine
38 lubricants, the CEC L-33-A-93 biodegradation method (CEC, 1995) has been extensively

⁴⁰ The ready biodegradation testing implies use of inoculum from municipal STPs – and thus the adaptation that occurs in domestic STPs is implicitly taken into account

1 used in the oil industry for assessing the biodegradation potential of a wide range of oil
2 products. The test estimates biodegradation on the basis of a specific change in chemical
3 composition, viz. loss of the parent substance rather than mineralisation. Similar tests
4 can also be conducted using specific GC and CG-MS analytical methods, although as the
5 substance becomes more complex. Results obtained using these procedures are
6 generally of limited value for classification purposes, but may in specific cases provide
7 useful information on comparing the relative biodegradability between substances as well
8 as providing data to support persistence and risk assessment. In such cases the
9 degradation products should also be assessed to the extent necessary for the purposes
10 of the assessment.

11 **Abiotic Degradation**

12 Hydrolysis is not an important fate process for petroleum substances since hydrocarbons
13 do not undergo reaction with water. However, degradation of unsaturated hydrocarbons,
14 notably aromatic hydrocarbons by reaction with sunlight in the presence of oxygen can
15 be a significant removal process where such substances are present in, or near the
16 surface of water. Whilst current criteria for environmental hazard classification do not
17 address photodegradation, this is a significant fate process for a number of aromatic
18 hydrocarbons present in certain petroleum streams. The significance of the issue for risk
19 assessment has been reviewed (CONCAWE, 2013). The rate of direct photolysis of
20 substances in water is highly dependent on the latitude, season and the shadowing effect
21 of the water column plus suspended material in the water column.

22

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