Propan-1-ol

July 2007

Section A2

## **Identity of Active Substance**

	section ex Point)					Official use only
2.1	Common name (IIA2.1)	1-Propanol				
2.2	Chemical name (IIA2.2)	Propan-1-ol				
2.3	Manufacturer's development code number(s) (IIA2.3)	not applicable				
2.4	CAS No and EC numbers (IIA2.4)					
2.4.1	CAS-No	71-23-8				
2.4.2	EC-No	200-746-9 (EINECS)				
2.4.3	Other	EC No: 603-003-00-0 RTECS No: UH8225000			X	
2.5	Molecular and structural formula, molecular mass (IIA2.5)					
2.5.1	Molecular formula	$C_3H_8O$				
2.5.2	Structural formula	CH <sub>3</sub> -CH <sub>2</sub> -CH <sub>2</sub> -OH				
2.5.3	Molecular mass	60.09 g/mol				
2.6	Method of manufacture of the active substance (IIA2.1)	As reported in the EU RAR, propan-1-ol is produced almost exclusively by the reaction of ethene with synthesis gas. Reaction is performed at 25-30 MPa and 140-180 °C in the liquid phase in the presence of cobalt carbonyl hydrogen as catalyst. After the separation of the catalyst the raw mixture obtained can be hydrogenated in the gaseous phase and in the liquid phase. By means of subsequent distillation the production of pure propan-1-ol is achieved.				
2.7	Specification of the	g/kg	g/1	% w/w	% v/v	
	purity of the active substance, as			Min. 99.5		
	appropriate (IIA2.7)			(BASF: 99.93; SASOL: 99.86)		
2.8	Identity of impurities and additives, as appropriate (IIA2.8)	Impurities of marketed propan-1-ol may vary depending on the purity grade (p.a. or technical product), on manufacturers handling and on single production specifications represented by a lot number.  Impurities other than water are volatile components as well as acid contents in variable compositions.				
		H <sub>2</sub> O		Max. 0.5%	· ·	
		Other organic impu	rities	Max. 0.3%		
		Non volatile matter		Max. 50 ppm		
2.8.1	Isomeric composition	not applicable				
2.9	The origin of the natural active	not applicable				

Task Force "1-Propanol"	Propan-1-ol	July 2007
RMS: Germany		

## Section A2 Identity of Active Substance

substance or the precursor(s) of the active substance (IIA2.9)

	Evaluation by Competent Authorities		
	Use separate "evaluation boxes" to provide transparency as to the comments and views submitted		
	EVALUATION BY RAPPORTEUR MEMBER STATE		
Date	2008/07/04		
Materials and methods	The correct quotation of the following field entry is:		
	Other: Index No: 603-003-00-0 RTECS No: UH8225000		
Conclusion	¥1		
Reliability	2		
Acceptability	acceptable		
Remarks	<u>12</u>		
	COMMENTS FROM		
Date	Give date of comments submitted		
Results and discussion	Discuss additional relevant discrepancies referring to the (sub)heading numbers and to applicant's summary and conclusion.  Discuss if deviating from view of rapporteur member state		
Conclusion	Discuss if deviating from view of rapporteur member state		
Reliability	Discuss if deviating from view of rapporteur member state		
Acceptability	Discuss if deviating from view of rapporteur member state		
Remarks			

Task Force "1-Propanol" RMS: Germany Section A2.8 Annex Point IIA2.8		Propan-1-ol	July 2007
		Identity of impurities and additives (active substance) fill in one form for each impurity/additive	
Subse	ction		Official use only
2.8.1.1	Common name	Water	
2.8.1.2	Function	None, impurity	X
2.8.2	IUPAC name	Hydrogen dioxide	x
2.8.3	CAS-No	7732-18-5	
2.8.4	EC-No	231-791-2 (EINECS)	
2.8.5	Other		
2.8.6	CIPAC  Molecular formula	$H_2O$	
2.8.7	Structural formula	H <sup>O</sup> H	
2.8.8	Molecular mass	18.0153 g/mol	
2.8.9	Concentration of	g/kg g/l % w/w	% <sub>V/V</sub> X

Max. 0.5%

the impurity or

typical and range of concentrations

additive

**July 2007** 

	Evaluation by Competent Authorities			
	Use separate "evaluation boxes" to provide transparency as to the comments and views submitted			
	EVALUATION BY RAPPORTEUR MEMBER STATE			
Date	2008/03/17			
Materials and methods	The correct quotation of the following field entry is:			
	2.8.1.2 Impurity 2.8.2 Water 2.8.9 Min.0 Max. 0.5 (typical 0.1-0.2)			
Conclusion	200			
Reliability	2			
Acceptability	acceptable			
Remarks	=			
	COMMENTS FROM			
Date	Give date of comments submitted			
Results and discussion	Discuss additional relevant discrepancies referring to the (sub)heading numbers and to applicant's summary and conclusion.  Discuss if deviating from view of rapporteur member state			
Conclusion	Discuss if deviating from view of rapporteur member state			
Reliability	Discuss if deviating from view of rapporteur member state			
Acceptability	Discuss if deviating from view of rapporteur member state			
Remarks				